

# Applications – MD / MC

Andrew Torda, May 2010, strukt &sim

## Basic tools

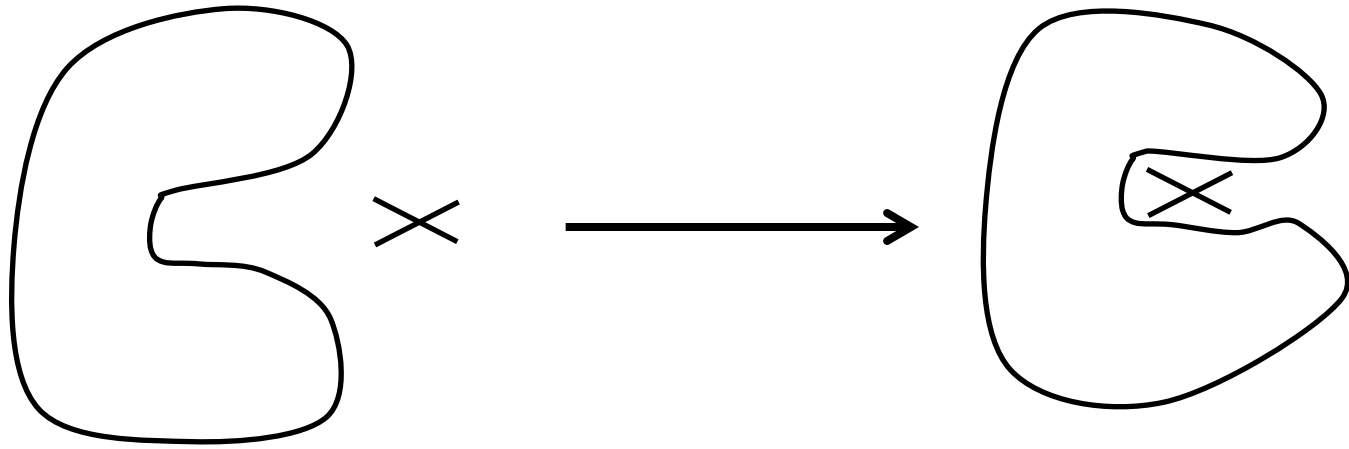
- Force field
- MD / MC

## Some application areas

- timescales
- free energy calculations
- simulated annealing
- structure refinement

# Simulating dynamics (optimistic / naïve)

- Claim
  - protein has a hinge which must open to bind ligand



- Can one see rates ?
  - rates for different ligands ?

# Timescales

- most common quantity  $\tau$ 
  - time to rotate by 1 rad
  - time for decay in  $A(t) = A(0) e^{-t/\tau}$ 
    - relaxation time
    - characteristic time
- times in proteins...

# Some typical times in proteins

	Amplitude (Å)	$\log_{10} \tau(\text{s})$
bond vibration	0.01 – 0.1	–14 to –13
rotation of surface sidechain	5 – 10	–11 to –10
protein hinge bending	1 – 5	–11 to –7
rotation of sidechain in middle of a protein	5	–4 to 0
local loss of protein structure	5 – 10	–5 to +1

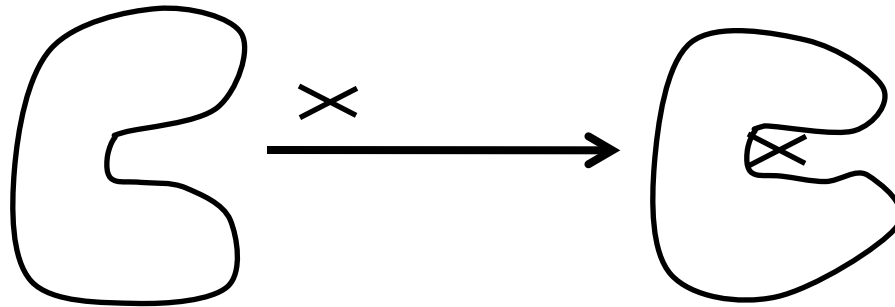
# Timescales

- Typical big simulation  $\approx 1\text{ns} = 10^{-9}\text{s}$
- Imagine event with characteristic time  $10^{-9}\text{s}$ 
  - may or may not be seen
- consider time  $10^{-10}\text{s}$ 
  - may be seen a few times
- What you would like
  - 100's or 1000's of observations
- Limits of timescales
  - fast events  $\tau \ll t_{simulation}$  OK
  - events  $\tau < t_{simulation}$  poor statistics
  - $\tau \approx t_{simulation}$  no statistics
- Previous example (drug binding)
  - it is not enough to observe an event once (or few times)

# Free Energy Calculations

- Free energy is most important
- Predicting therapeutic efficacy

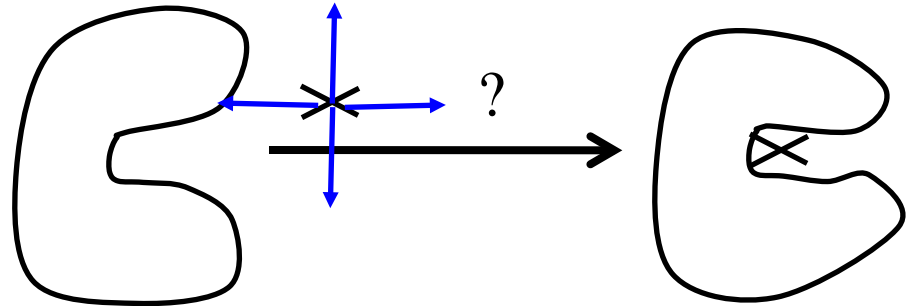
$$k_d = \frac{[\text{drug}][\text{protein}]}{[\text{drug - protein}]}$$
$$= e^{\frac{-\Delta G}{RT}}$$



- could we just look at energies ? What are contributing terms ?
  - ligand-water  $\rightarrow$  ligand + water (many interactions,  $\Delta S$ )
  - ligand+protein
  - ligand loss of entropy / water entropy change
- simulate ?

# Free simulation for binding

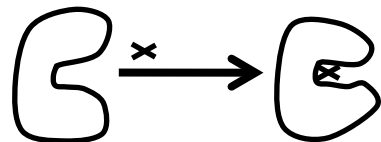
- if we simulate, where will the ligand go ?



- may take years for ligand to find protein
- short cut ?
  - force ligand to protein
    - artificial force + corrections
    - very difficult – still requires rearranging water
    - entropy estimation very difficult

# Estimating free energy differences

- $G = U - TS$ 
  - but  $S = -k \sum_{i=1}^{N_{state}} p_i \ln p_i$ 
    - so we cannot really get  $S$
    - some books write in terms of partition function
    - similar problem – especially visiting high energy regions
- forget absolute free energies
  - concentrate on  $\Delta G$
  - no problem – usually interesting property

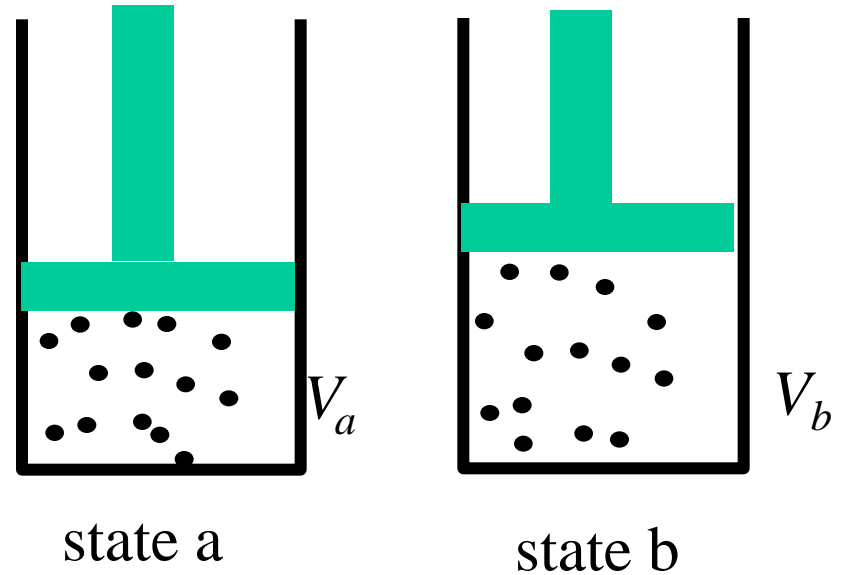




# Work and free energy changes

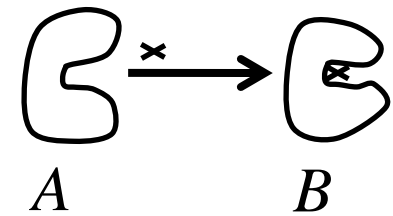
work done A to B

- free energy change
  - look at either state
    - real world automatically includes entropy

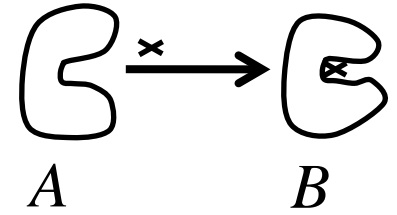


work going from unbound  $\rightarrow$  bound

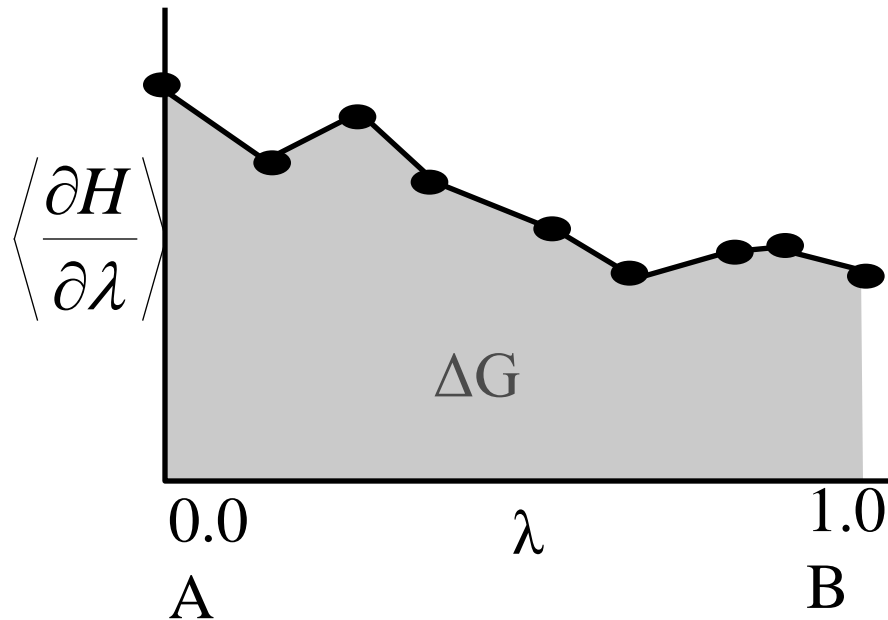
- $\Delta G_{AB}$
- what is B ? what is A ?
  - more later
- measuring work ?



# Work and free energy



- measure the work needed to move from A to B



- where  $H$  is again Hamiltonian ( $E_{kin} + E_{pot}$ )

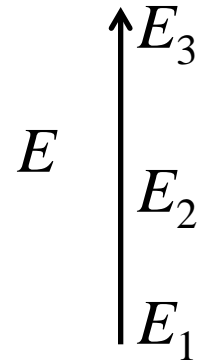
$$\Delta G = \int_A^B \left\langle \frac{\partial H(\mathbf{p}, \mathbf{r})}{\partial \lambda} \right\rangle_{\lambda} d\lambda \quad \text{or} \quad \Delta G = \sum_{i=0}^{N_{step}} (H_{i+1} - H_i)$$

# Binding energy - feasibility

- Would this approach work ?
  - $\left\langle \frac{\partial H}{\partial \lambda} \right\rangle$  must be a good average (lots of fluctuations)
    - must change  $\lambda$  slowly
  - chemistry problems: your simulation would
    - get averages with all water molecules
    - gradually remove water molecules (high energy ?)
    - find the correct binding
    - get good averaging there
  - states A and B are very different they must be well sampled
  - intermediate (higher energy states) must also be sampled
  - does not work well in practice

# Paths / Energy differences (detour)

- Problem – the path is too difficult – changes too big
- Energy differences depend on end states – not paths
- Look at  $\Delta E_{1,2} = E_1 - E_2$ 
  - would it matter if we go  $E_1 \rightarrow E_3 \rightarrow E_2$  ?
- Can we take even stranger paths ?
  - go through non existent  $E_4$  ?
    - no problem
- Same reasoning applies to free energies



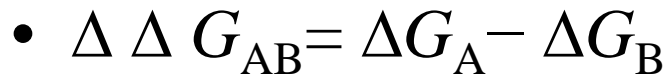
# Applying different paths

- Originally wanted ( ligand A or B, protein P)



- what if I know  $B + P \leftrightarrow BP$  ?  $\Delta G_B$

- maybe  $\Delta \Delta G_{AB}$  would be easier



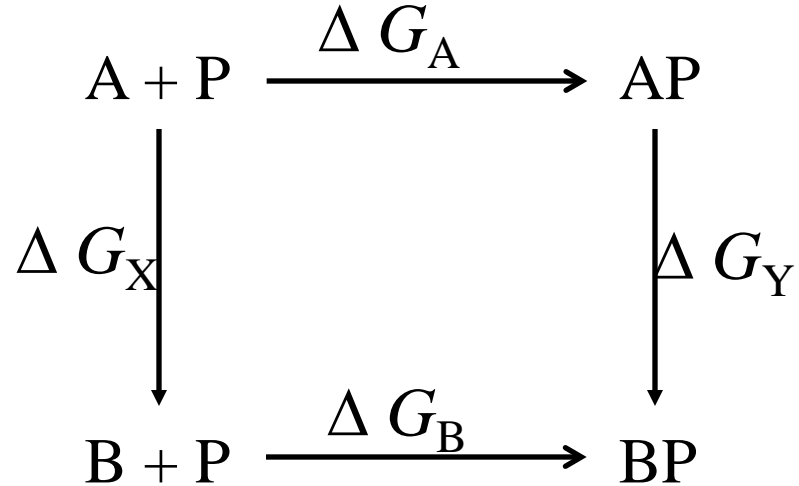
- what would  $\Delta \Delta G_{AB}$  mean ?
  - what is relative binding strength ?



# Alternative routes

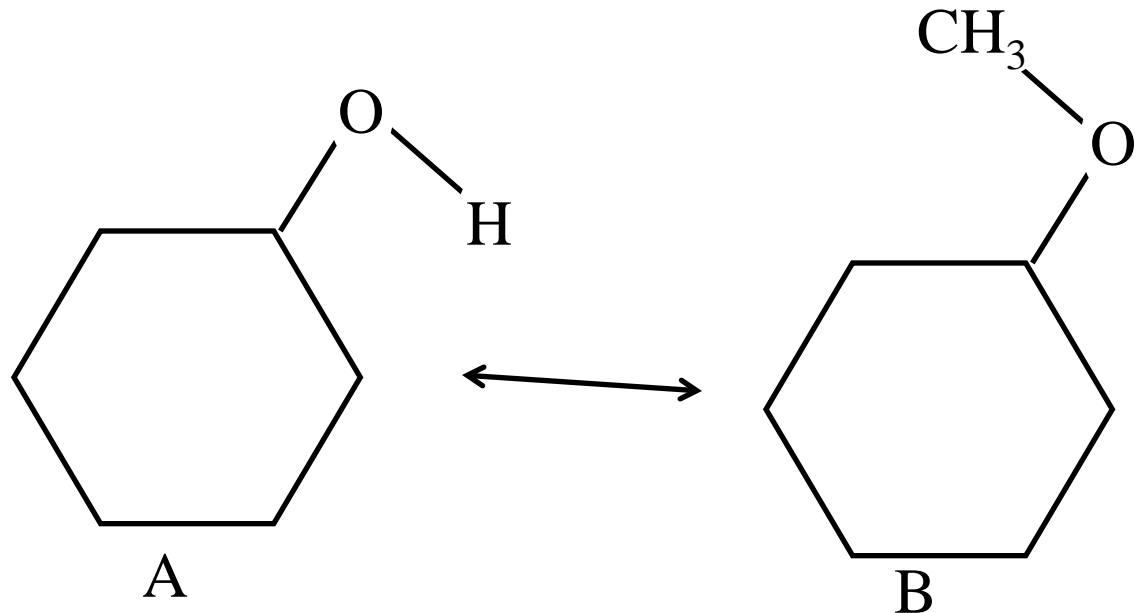
- $\Delta G_A$  and  $\Delta G_B$  too hard
- we would be happy with  $\Delta \Delta G_{AB}$
- $\Delta G_A + \Delta G_Y = \Delta G_B + \Delta G_X$
- $\Delta G_A - \Delta G_B = \Delta G_X - \Delta G_Y$  remember  $\Delta \Delta G_{AB} = \Delta G_A - \Delta G_B$

- so  $\Delta \Delta G_{AB} = \Delta \Delta G_{XY}$
- why  $\Delta G_X$  easier ?
- why  $\Delta G_Y$  easier ?



# Easier free energy changes

- if A/B are rather similar
  - $AP \leftrightarrow BP$  or
  - $B + P \leftrightarrow A + P$  (free  $A \leftrightarrow B$ )
- are small changes – smaller than
  - removing water order, removing water energy, finding protein...
- example
  - small change

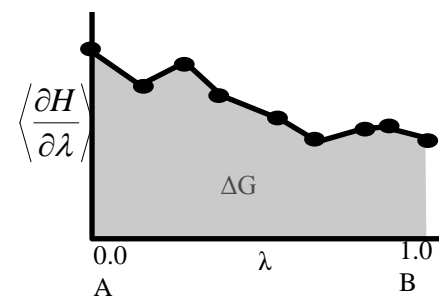
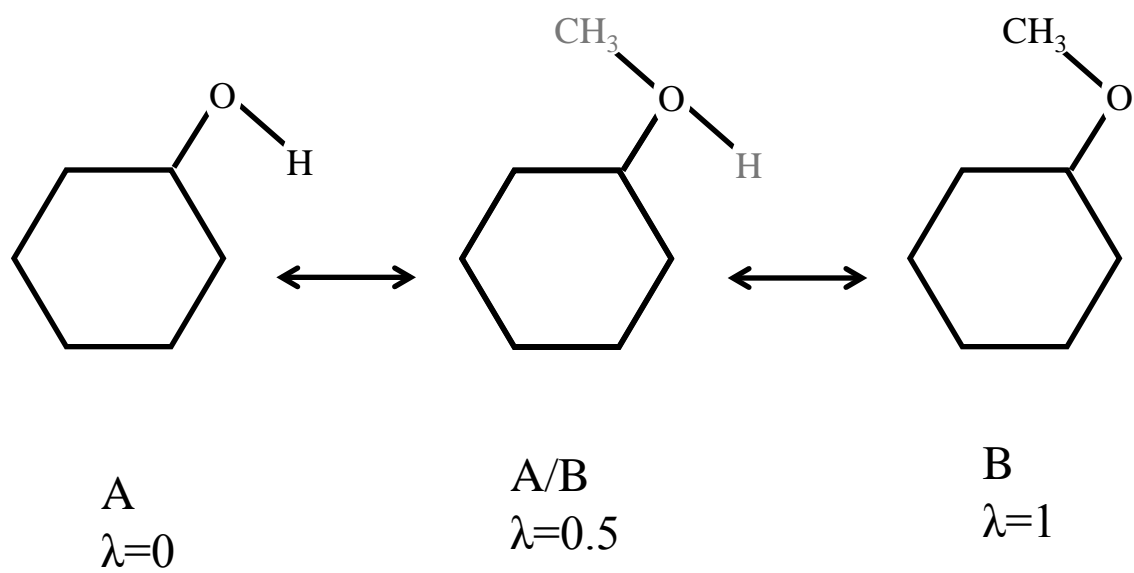


# Fictitious states

- remember formulae
- we need to make chemistry a function of  $\lambda$

$$\Delta G = \int_A^B \left\langle \frac{\partial H(\mathbf{p}, \mathbf{r})}{\partial \lambda} \right\rangle_{\lambda} d\lambda$$

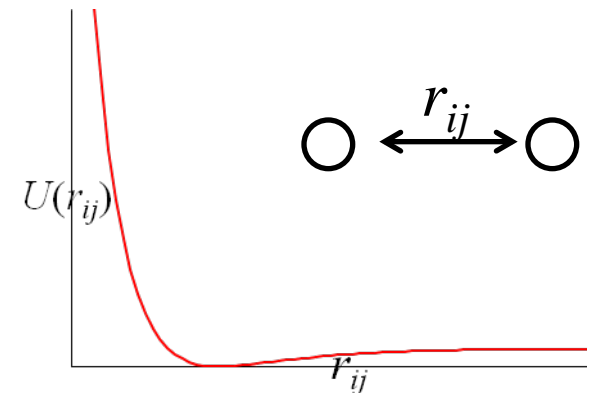
$$\Delta G = \sum_{i=0}^{N_{step}} (H_{i+1} - H_i)$$



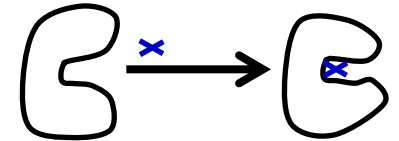
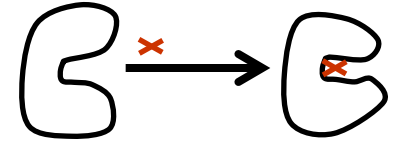


# $\lambda$ dependence

- $\lambda = 0$  an OH group
- $\lambda = 1$  an OCH<sub>3</sub> group
- $\lambda = 0.5$ 
  - charge of H – half of original charge
  - radius / size ( $\sigma$ ,  $\epsilon$ ) half of real value and so on
- atoms gradually
  - appear in one direction
  - disappear in other
- description of system is now function of  $\lambda$



# $\lambda$ dependent simulations



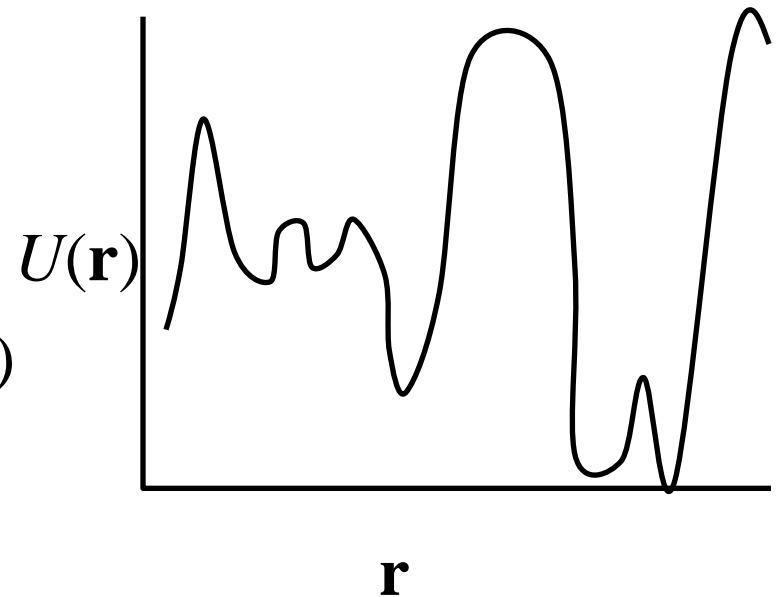
- two simulations necessary
  - $\lambda$  from 0.0  $\leftrightarrow$  1.0 in protein
  - $\lambda$  from 0.0  $\leftrightarrow$  1.0 in water
  - both from **red**  $\leftrightarrow$  **blue**
- As  $\lambda$  slowly moves from 0.0
  - water gradually feels more/less influence of some atoms
  - system should not have to rearrange itself too much
- when does method work best ?
  - when changes are small
    - comparison of similar ligands in a protein

# Summary of free energy calculations

- from first principles: "free energy differences, equilibria"
  - easy to calculate
  - in practice impossible (sampling not possible)
- forget absolute free energies
  - $\Delta G$  determine most phenomena in the world
- processes like binding still too difficult to simulate (slow, too many conformations / states to visit)
- most calculations these days use  $\Delta\Delta G$ 
  - aim to get relative binding strengths

# Simulated Annealing

- Classic reference – separate handout / not on web (naughty)
- Basic tools
  - MC or MD with control of temperature
- Use : difficult optimisation problem
  - chip layout
  - travelling salesman problem
  - protein structure
- Optimisation problem
  - several dimensional (2 to 2 000)
  - many local minima



# Procedure

while ( $T > T_{\text{end}}$ )

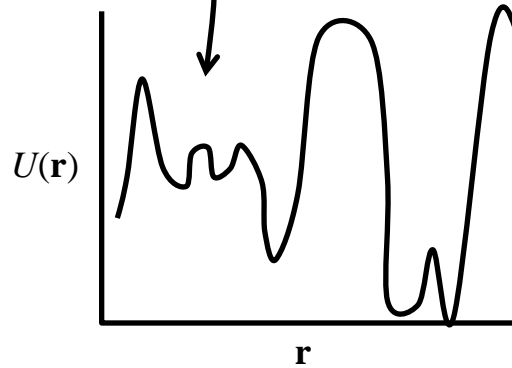
$$T(t) = T_0 e^{-ct}$$

move system (Monte Carlo)

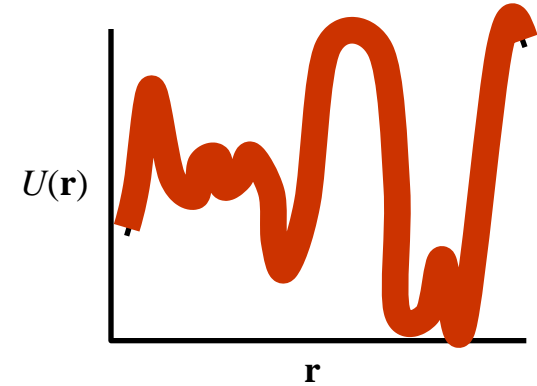
- $T_0$  initial temperature is hot
- $c$  is decay rate (rate of decrease)
- cost function is
  - $E_{\text{pot}}$  in chemistry
  - path length in travelling salesman
  - board cost in chip layout problem ...
- why may this work ?

# Simulated Annealing concept

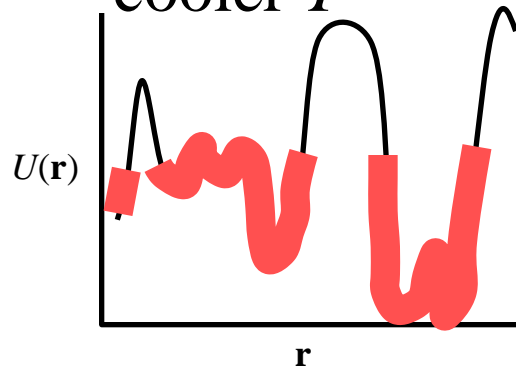
initial (poor)  
guess



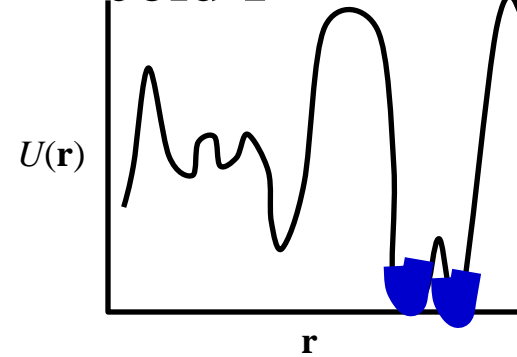
initial high  $T$   
distribution



cooler  $T$



cold  $T$



# Properties, practical issues

Admit that there may not be a best solution

- not worth spending effort between many very good solutions

Some problems have "phase transitions"

How hot should  $T_0$  be ?

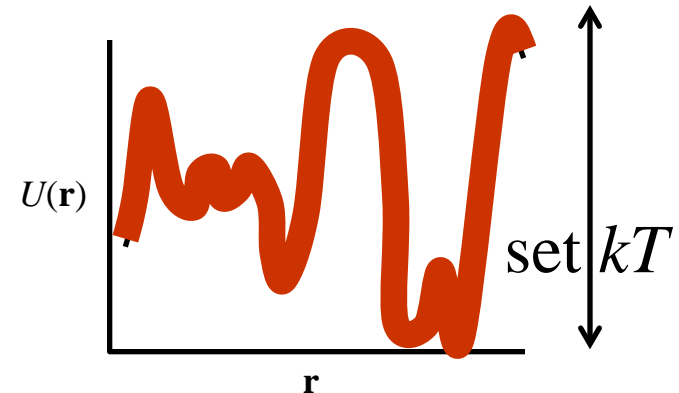
- infinite ? No : look at barriers

How slow should cooling be ( $c$ ) ?

- system should be at equilibrium
- very slow

Cool exponentially ?

- best first guess
- should certainly cool more slowly at transition points



# Anneal with MC or MD ?

Historic use of Monte Carlo

- easiest to apply to many problems

Use MD ?

- provides expected advantages (efficiency)
- uses available gradient / derivative information

Implementation

- Couple to temperature bath, make  $T$  time dependent

Use in practice ?

- simulated annealing in
  - most MD codes, refinement packages, ...



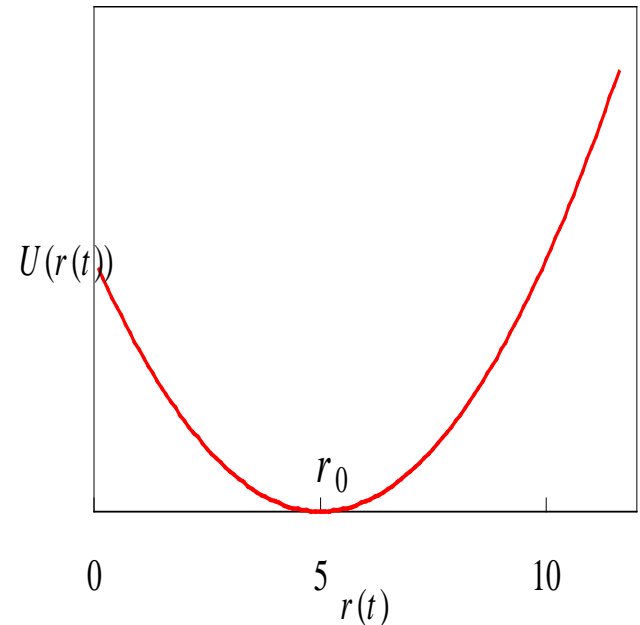
# Refinement of Structures (NMR / X-ray)

- Story from first semester
- Problem : generate protein coordinates from NMR information (or X-ray)
  - distance geometry gives an initial guess, but
    - distance geometry methods spread error across all distances
    - errors are spread across bonds, measured distances
    - chirality may be broken (causes distance problems)
- Belief
  - coordinates are not bad, but could be improved

# Pseudo – energy terms

- For some distance measurement  $i$  between some pair of atoms
  - $r_0$  measured distance
  - $r(t)$  distance between particles at time ( $t$ )
  - say  $U_i(r) = c_i (r(t) - r_0)^2$
  - add this to normal force field

$$U_{tot}(\mathbf{r}) = U_{phys}(\mathbf{r}) + \sum_{i=1}^{N_{restraints}} U_i(\mathbf{r})$$



- $U_{phys}(\mathbf{r})$  normal force field - atomistic (bonds, electrostatics...)

# result ?

- System moves to low energy + low fake energy
  - gradually moves to agree with experimental data

- Practical issues

$$U_{tot}(\mathbf{r}) = U_{physical}(\mathbf{r}) + \sum_{i=1}^{N_{restraints}} U_i(\mathbf{r})$$

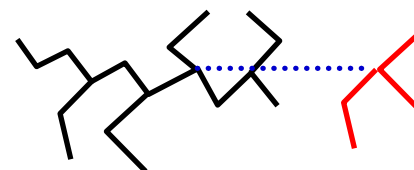
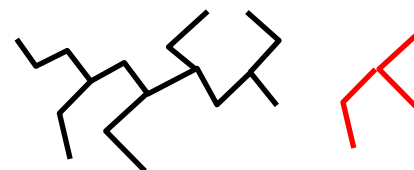
- $U_i(r) = c_i (r(t) - r_0)^2$

- big  $c$  very artificial
- small  $c$  system will be slightly biased to agree with experimental data

# Fake Energies

Fake energies for many purposes

- Refinement of
  - X-ray structures (common)
  - NMR (often)
  - others: microwave spectroscopy, ...
- Modelling problems
- you want to put a bond in a model
  - putting it in directly
    - high energy bond
    - system stuck in minimum
  - introduce a distance restraint
    - gradually increase associated constant  $c$



# Summary

- What one can do with related methods
  - look at timescales of motions (very superficial)
  - free energy calculations – important for problems such as binding of ligands
  - simulated annealing – methods used as minimizers, not necessarily to get an ensemble
  - pseudo-(potential) energies (X-ray, NMR, ...)