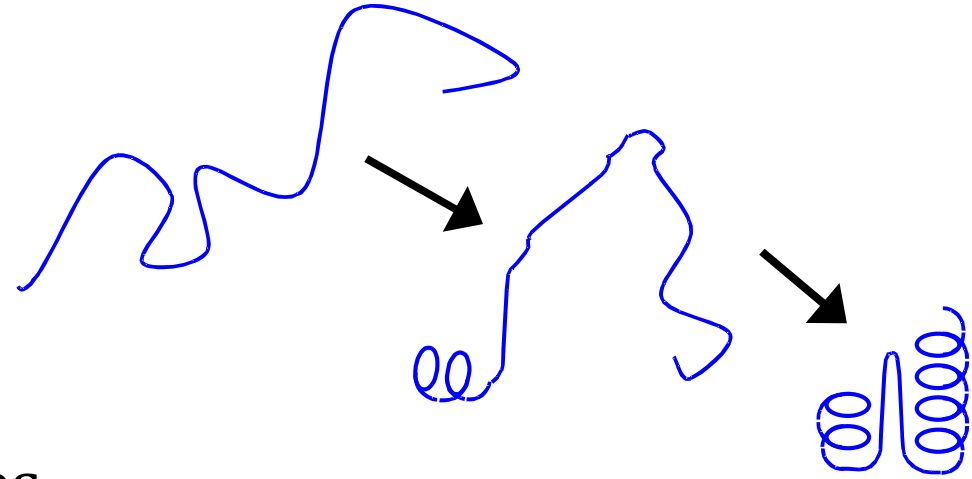


# Protein folding

How does a protein do this ?



Ideas

- kinetic vs thermodynamic structures
- experimental methods for following protein folding
- pathways for folding
- entropic barriers

# Background / stories

From biochemistry Übungen (protein folding – easy)

- take lysozyme / ribonuclease...
- put in 8M urea (unfolds)
- remove urea (refolds)

Conclusion ?

- the protein sequence is all you need to fold a protein
- is this true ? Not always

Alternative (logical reasoning)

- protein folding should be impossible...

# Protein folding should be difficult

## Levinthal's paradox

- each amino acid has 2 or 3 or  $n$  conformations
  - for a protein of  $m$  residues, it should visit  $n^m$
- if it spends  $10^{-15}$ s at each conformation ?
- time to find one conformation for  $n=3$  and 100 residues
  - $3^{100} \times 10^{-15} \text{ s} \approx 1.6 \times 10^{25} \text{ years}$

## Consequence

- proteins cannot be exploring space randomly
- historic idea of "folding pathway"

# Who cares about protein folding ? Faith..

If we could understand folding we could

- predict structure
- design proteins that fold better (more stable)
- identify essential residues for folding (not suitable for mutagenesis)

# Issues / Questions

- Kinetic versus thermodynamic
- What order do events happen in ? (collapse vs secondary struct)
- Is unfolding the same as folding ?
- Is folding in a test tube the same as nature ?

# Are proteins in energy minima ?

Anfinsen story..

- proteins can be unfolded and refolded alone
- all the information is in the sequence
- native conformations are the (free) energy minimum
- thermodynamic belief

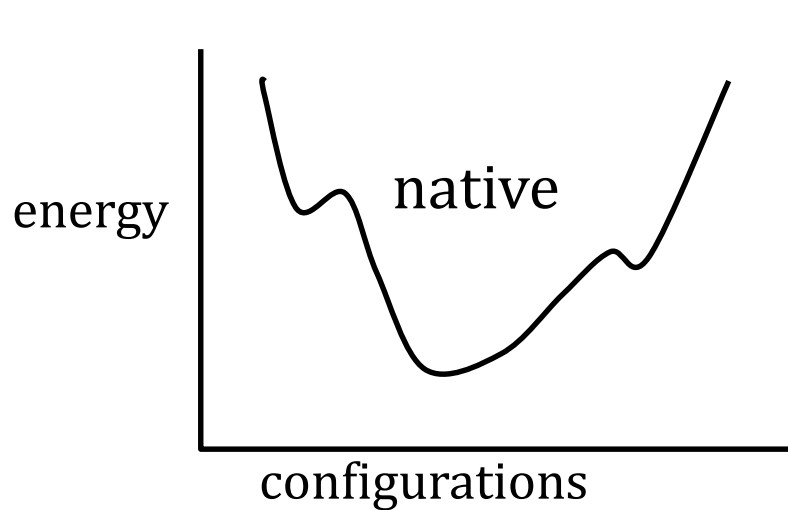
More modern

- many many proteins cannot be refolded in the lab
  - consequence .. maybe they need something else
  - maybe they are not always in free energy minimum

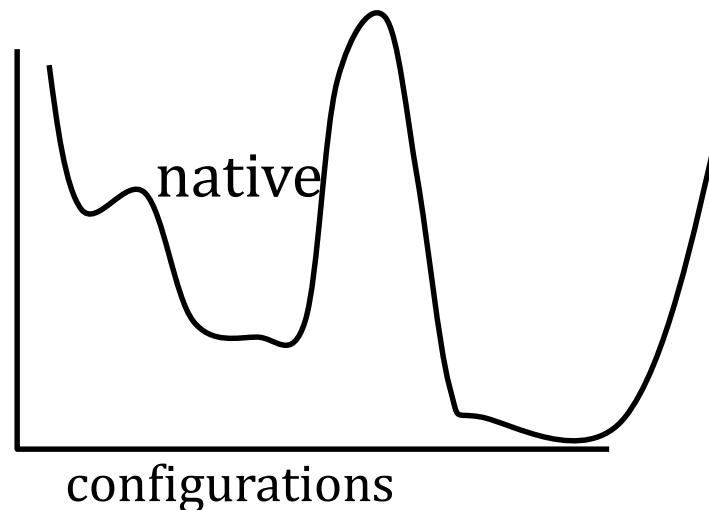
Kinetically trapped proteins..

# Kinetic versus thermodynamic

- If proteins fold spontaneously and remain folded, they are thermodynamically determined
- If you leave a protein long enough and it unfolds, it was not in an energy minimum



thermodynamic



kinetically  
trapped

# Consequences

## Thermodynamic

- protein structure prediction
  - just a matter of modelling the real world

## Kinetically trapped

- we cannot predict structure from sequence just by energies



# Consequences

Can we see which is the case ?

- leave a protein for 10 minutes
  - see if it finds another state
- leave it for 10 years ?
- depends on barriers

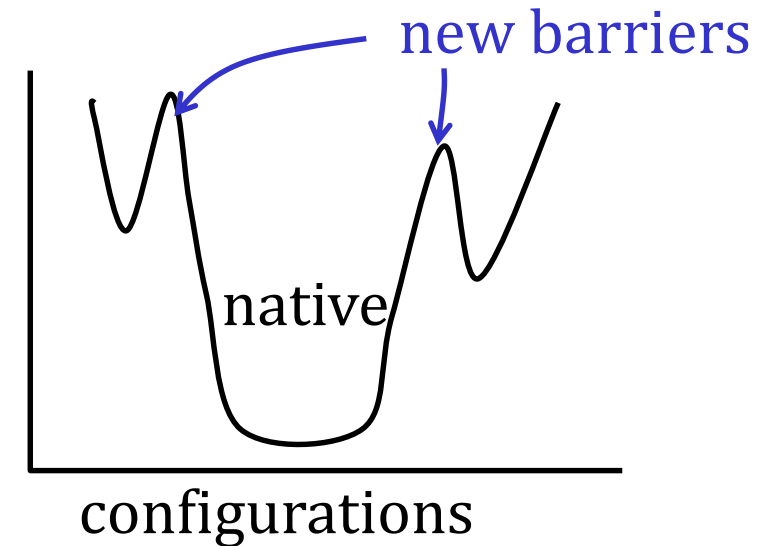
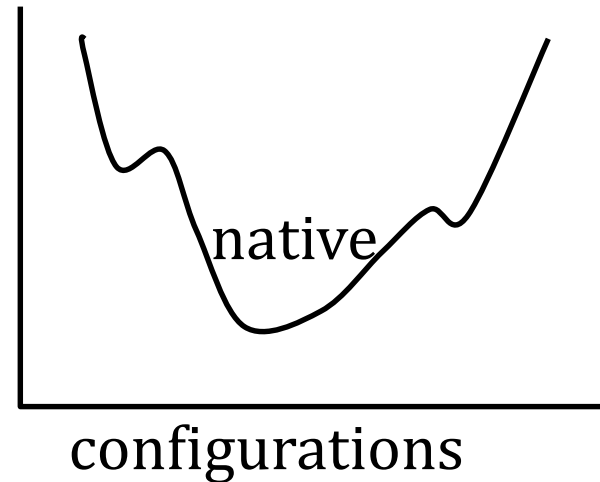
Empirically

- some evidence of kinetic trapping
- some proteins do have other states
  - $\beta$ -fibrils, Alzheimers, mad cow disease

# Evolution / design consequences

Imagine I can predict structure and stability

- I design a better / more stable protein



- my new protein may be more stable
- it may never fold
- evolutionary implications
  - protein sequences may evolve for folding (+ structure and function)

# Change of direction

- enough background on folding
- brief overview of experiments
- simulation

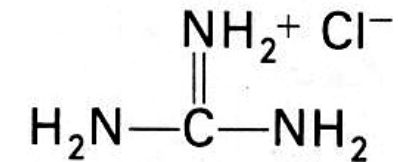
# Experiments

## Timescales

- maybe  $10^{-6}$  s for folding
- maybe orders of magnitude slower or faster

## Experimental approaches

- force protein to unfold
  - chemistry (guanidinium HCL, 8M urea)
  - temperature (heat, cold)
  - change conditions and watch
- try to measure very fast
- try to change timescale
- try to measure unfolding



# Experiments - problems

Very difficult to measure on the  $\mu\text{s}$  /  $\text{ms}$  time scale

- temperature jump
- stop - flow
- fluorescence
- NMR
- circular dichroism (CD)

# Are experiments relevant ?

- Technical difficulties (obvious)
- Tradeoff
  - fast methods – less information
  - more information – too slow
- How real is it ?
- Imaginary technique :
  - I can take any protein in denaturant
  - suddenly bring back native conditions
  - follow every detail
  - is this what happens in nature ?...

# How real is experiment ?

Our bodies – about 150 mM salt, regulated pH, temperature, ...

Denature a protein with high salt

- is the partially folded state natural ?
  - it comes from disrupting a very special set of ionic interactions

# How real is experiment ?

Heat the protein

- guaranteed to visit high energy states which are not natural

Hope ..

- the strongest interactions are formed first – last to be broken

Do proteins fold like this in nature ?

- proteins made from N to C terminus
- N terminus gets a chance to find structure, before rest of protein is there
- would permit very specific paths / kinetic trapping

Next ... simulation and theory...



# Monster Simulations

Months of cpu time

- very hard to fold
  - 1 copy
  - 1 protein

2011 record holder Shaw Research

- some ms simulations, small proteins

more feasible

- simulate unfolding

# Simulating unfolding

Atomistic simulation of real protein too slow

- take native structure at 300 K
- gradually heat up
- watch it fall apart
- what breaks first ?
  - secondary structure ?
  - overall fold ?
  - everything ?

Reasons to believe

- the last interactions to form (folding) may be first to break (unfolding)

# Problems simulating unfolding

## Problems

- the system is visiting high energy states which may not really exist
- force fields are parameterised for 300 K
- property of unfolded state(s)
  - statistics may be dominated by huge number of partially folded states (more later)
  - cannot visit these states in realistic time

# Forgetting atomistic detail

- What are questions we can ask ?
- What can we guess without any calculations ?
- Questions
  - is there such a thing as a folding pathway ?
  - how should we look at folding ?
    - secondary structure forms first and is rearranged
    - hydrophobic residues come together and then secondary structure forms ?
    - a few important contacts are formed, then structure forms

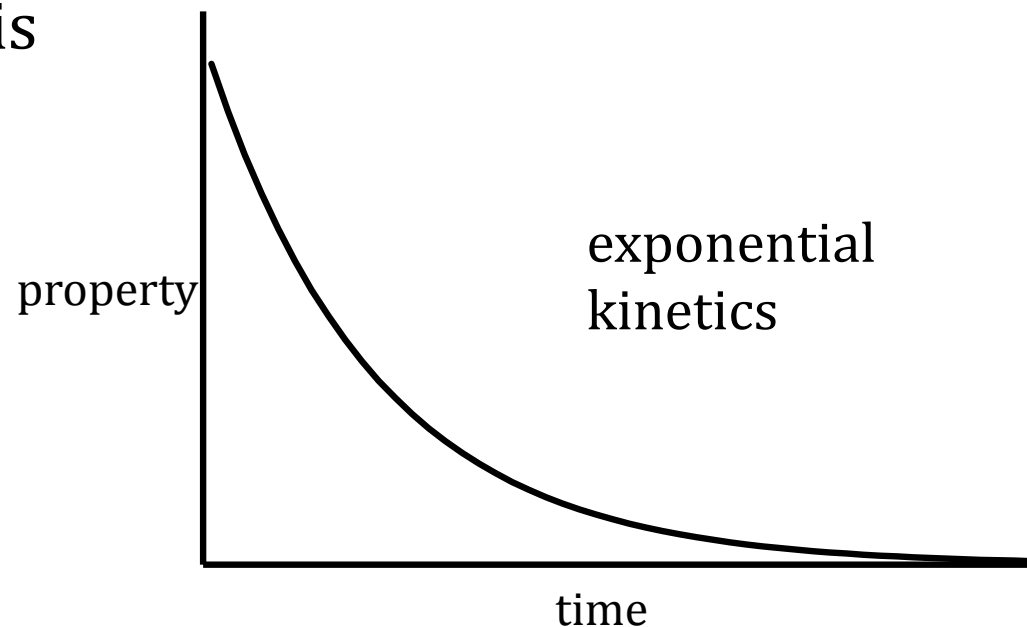
# General kinetics

What have we seen so far ?

- most properties have something like exponential decay
- property =  $a e^{-\alpha t}$
- rate of change proportional to quantity present

A model should explain at least this

- later by ensemble view..  
where are barriers ?



# What do we know

## Possibilities

- Proteins form secondary structure first
- helices and sheets then arrange themselves

OR

## Hydrophobic collapse

- hydrophobic residues find each other
- backbone rearranges and secondary structure is fixed

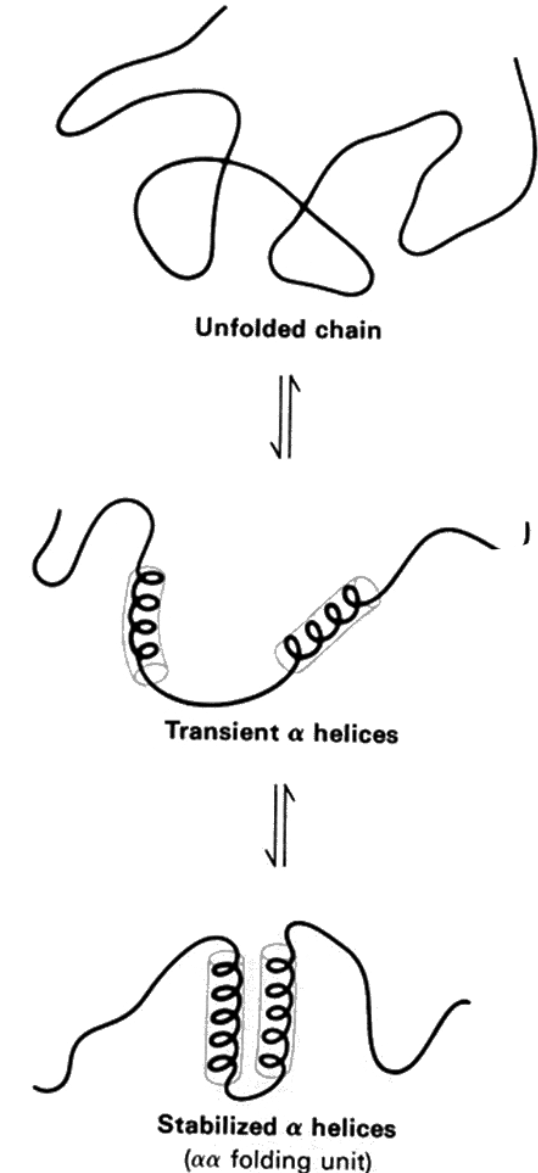
# Side chain vs backbone driven

Old textbook

- local secondary structure forms, then reorganises
- secondary structure depends largely on backbone

Alternative

- sidechains are very important



# Sidechains might be important

- backbone view does not predict collapse of protein
- $\alpha$ -helix and  $\beta$ -strand propensity is weak
  - isolated peptides are not stable
  - $\beta$ -strands often depend on long-range H bonds
- helix / strand formation depends on environment / solvent and is not known in open structure
- fold is largely predictable / characterised by pattern of HP (sidechains)
- proteins are most sensitive to mutations in core  
(they are important for stability)



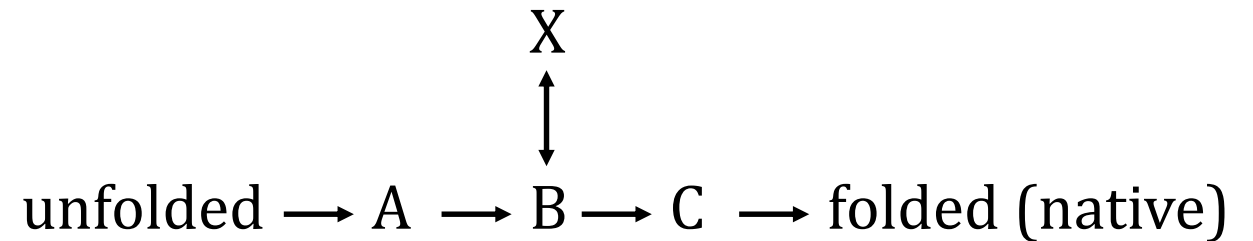
# Types of pathway

From classical chemistry we would like a path

unfolded  $\rightarrow$  A  $\rightarrow$  B  $\rightarrow$  C  $\rightarrow$  folded (native)

Slightly more complicated – detours and sideproducts (X) allowed

Basic idea



- molecules may get sidetracked, but
  - every molecule sees A, B, C...
- where does it come from ?
  - Levinthal's answer to paradox
    - there must be a preferred pathway
- old view / microstate pathway approach

# Consequence of simple pathway

Two state kinetics

- $A \rightarrow B$  or  $B \rightarrow C$  might be part of transition barrier
- pathway with detours explains multi-state kinetics (if necessary)

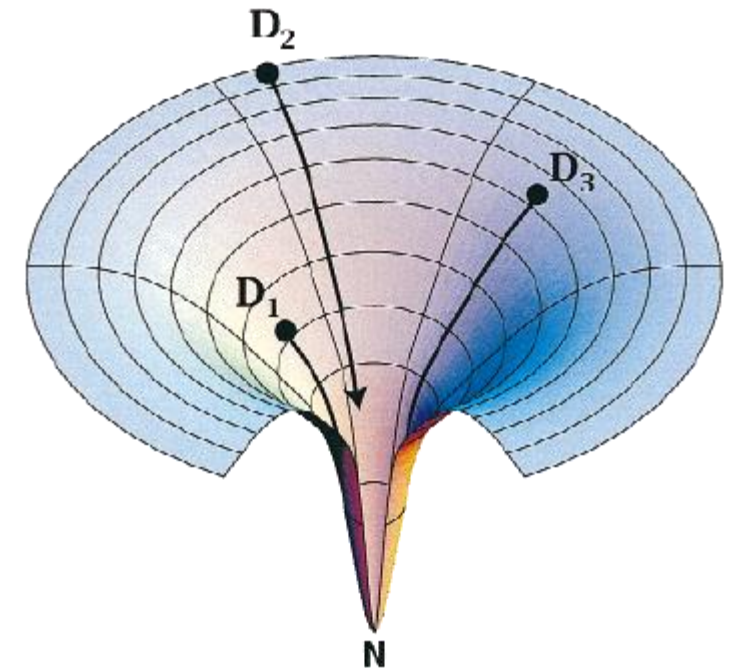
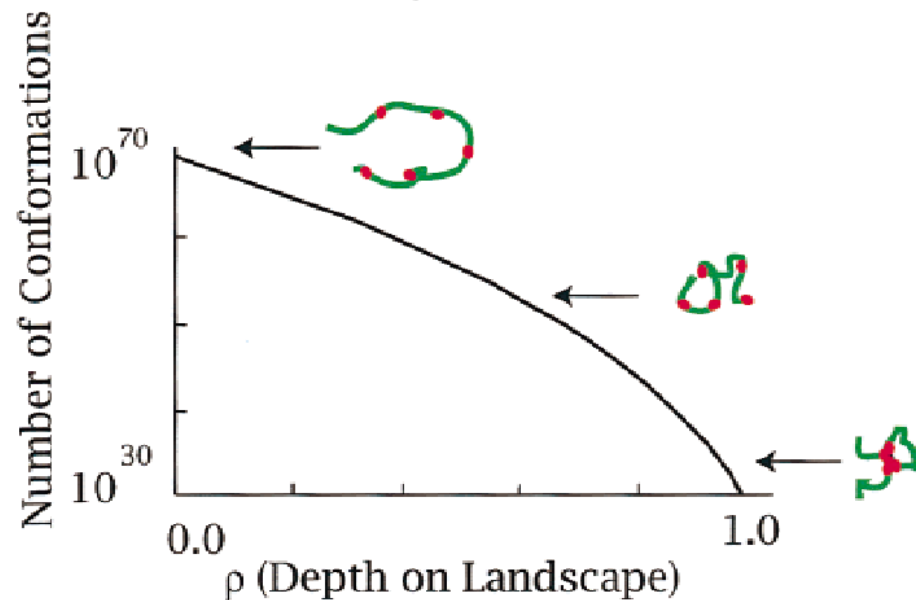
Does it sound intuitively reasonable ?

- what if a mutation perturbs A or B or C ?
  - whole pathway might break
  - maybe OK (this is why some mutants do not fold)

Do you need conventional pathway to explain barriers ?

# Ensemble view

- conformation space is huge
- will a protein be able to find a neat path through it ?
- should we even look for paths ?
- consider a multitude of paths...
- is this merely a cute picture ?
- first implication...
  - entropy vs energy



Dill, K.A., Protein Sci., 8, 1166-1180, 1999  
Polymer principles and protein folding

# Consequence of ensemble view

As a protein folds

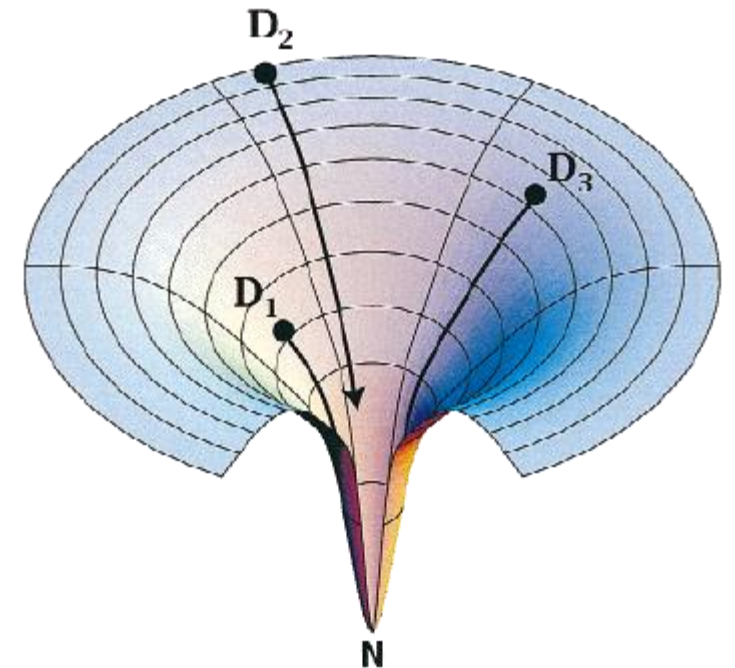
- potential energy goes down (happy)
- number of possible states goes down
- entropy goes down (unhappy)

Interpretation in chemical terms

- do we have a off-pathway intermediate ?

Do we have a reaction coordinate ?

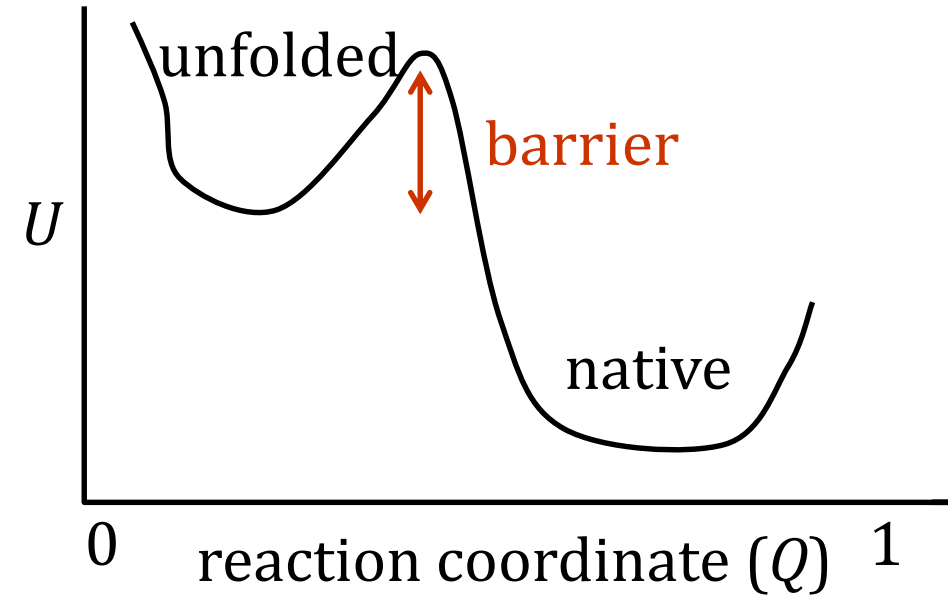
- not a classic one
- can we invent one ? yes
  - if two atoms are in contact in the final structure
    - native contact
  - $Q$  = fraction of contacts which are correct



# Reaction path

Is this like a chemical reaction ?

- no
- many molecules have same  $Q$ , but different conformations



We want at least two state kinetics

- where does barrier come from ?

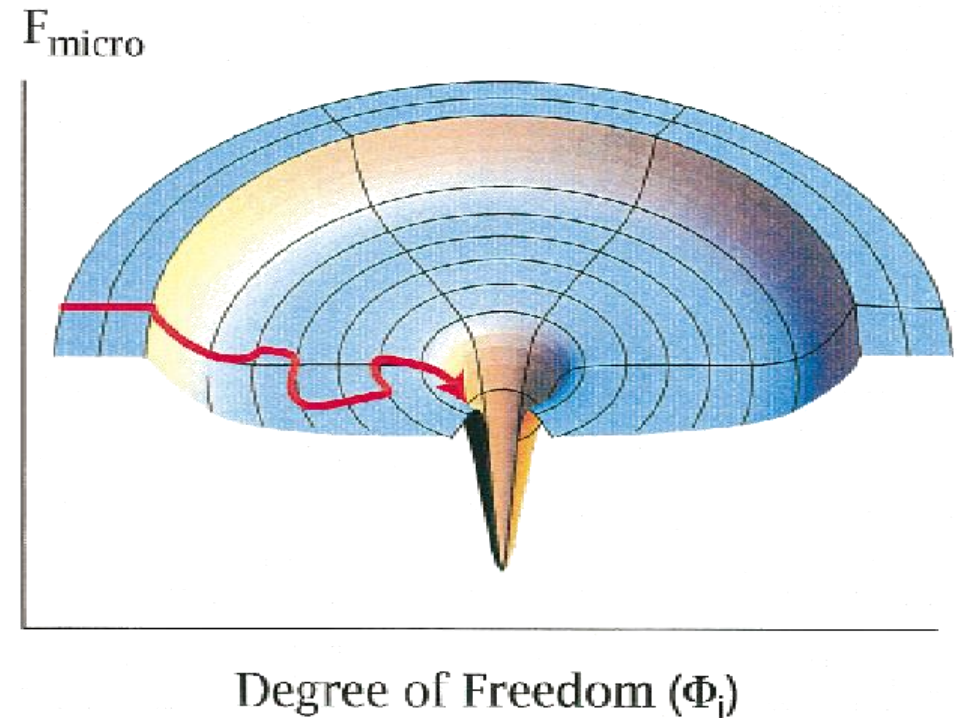
# Entropic barriers

Nature cares about free energies

$$\Delta G = U - T\Delta S$$

If a molecule walks around

- it takes a long time
- looks the same as an energy barrier
- Are these pictures useful ?
- Do they agree with calculation ?

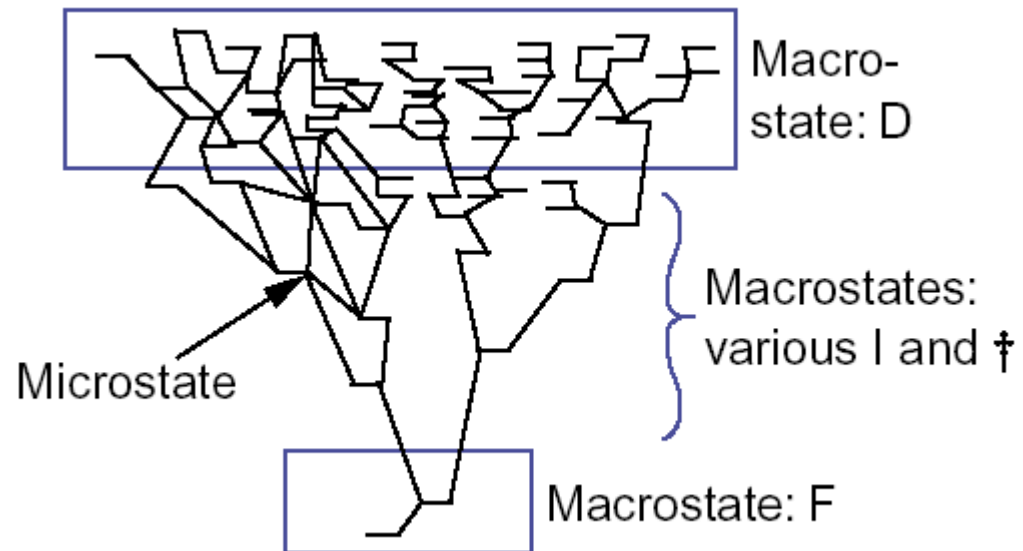


# Interpretation of landscape

Does this disagree with conventional pathway ?

$A \rightarrow B \rightarrow C$  ?

- mostly at early stages
- there is a multiplicity of "A"
- when near native, there are relatively few conformations, so there may be something more like a pathway



# Agreement with other ideas

Agreement with experiment ?

- experiment says most about average properties
  - these are the same in landscape picture
  - should we expect to find well defined, early intermediates ?

Agreement with MD simulation ?

- peptide results – argue that they correspond to near native view

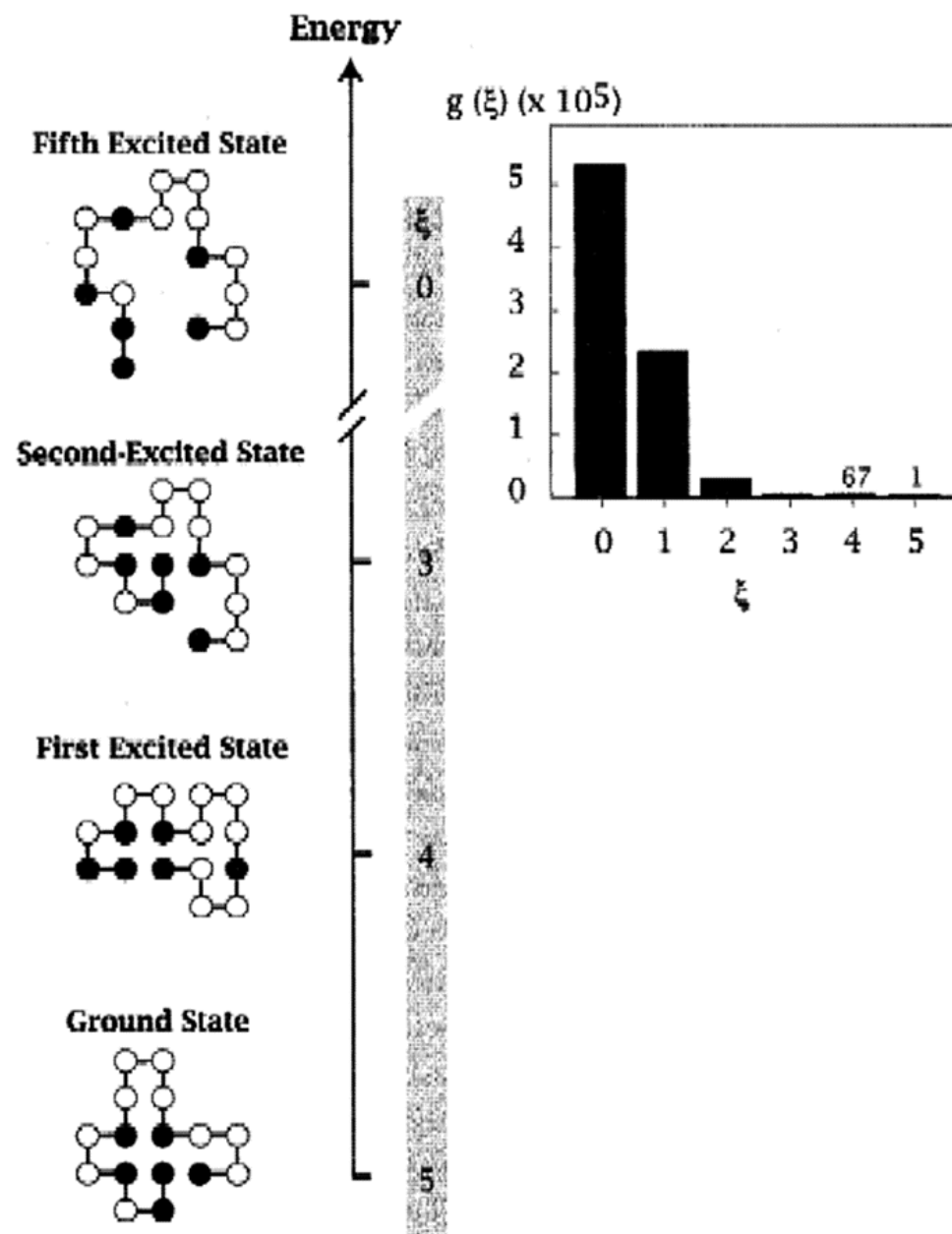


# Simple models

Return of HP models

- $g(\xi)$  is density of states
- how many conformations have  $\xi$  correct contacts

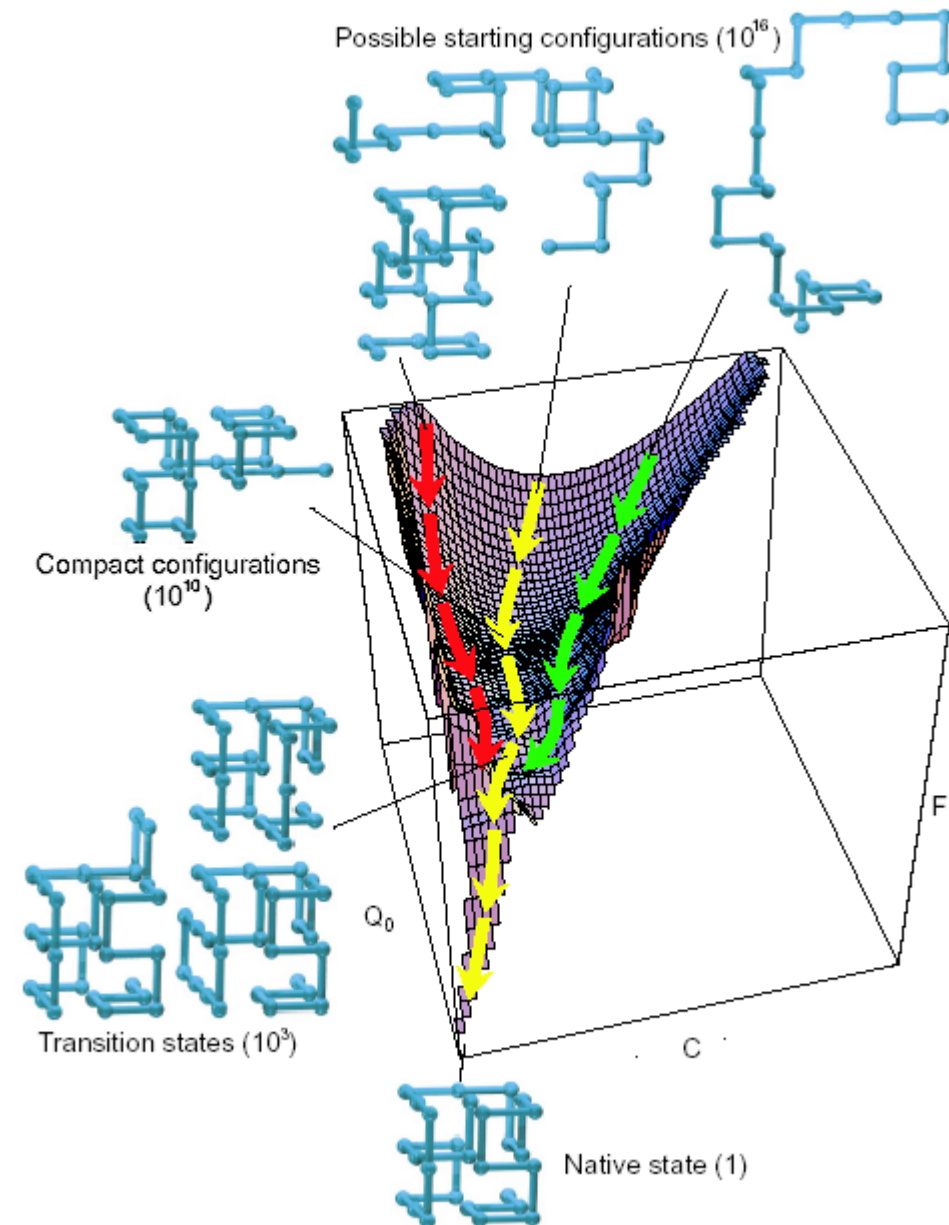
A bigger calculation



# A larger calculation

27 residue

- simple lattice model
- estimations by sampling
  - not exhaustive
  - $Q_0$  correct contacts
  - $C$  total contacts
  - $F$  free energy



# Summary

- Experiment vs. theory
  - experiment usually gives us averages
  - most calculations look at details
- Very different views on folding may be hard to distinguish – are predictions different ?
- Folding may be guided by sidechains (not hierarchical)
- Early folding may be best modelled by very crude models (so space can be sampled)
- Any useful model should predict exponential kinetics (or more complicated)