

# NMR (Nuclear Magnetic Resonance Spectroscopy)

- literature / background (already in Stine)
  - Thomas James chapter  
<http://www.biophysics.org/img/James.T.pdf>
  - Ferentz, A.E. and Wagner, G., Q. Rev. Biophys, 33, 29-65 (2000)

## current standing

- $\approx 13\%$  of all current structures solved by NMR
- about 1/3 of smaller structures

# Next 3 Weeks

- Background to NMR – chemistry
- Calculating structures
  - distance geometry
  - problems with structures

# History

- younger field than X-ray
  - one Nobel prize in early 90's (Ernst – technical)
  - ½ Nobel prize 2002 (Wüthrich)
- first real protein structure about 1985 or 1986

## NMR from our viewpoint

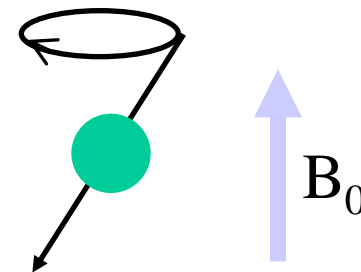
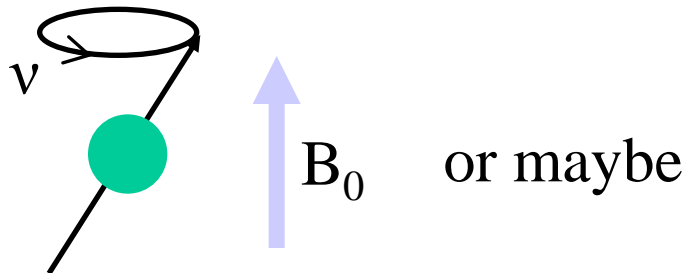
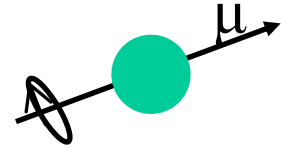
- a way to get structures
- can provide information on
  - dynamics, stability
  - interactions (other proteins, small molecules)
- we concentrate on structural aspects

# Overview – how we get coordinates

- protein in solution
- record spectra
- assign peaks to  $^1\text{H}$ ,  $^{13}\text{C}$ ,  $^{15}\text{N}$  nuclei
- record some more spectra
  - distance information (mostly)
  - some internal angles
- reconstruct structure

# Nuclei have spin

- have a charge and act like magnets
- put them in a field and they will align with it
- now apply a magnetic field
  - they "precess" around the field
  - two possible states

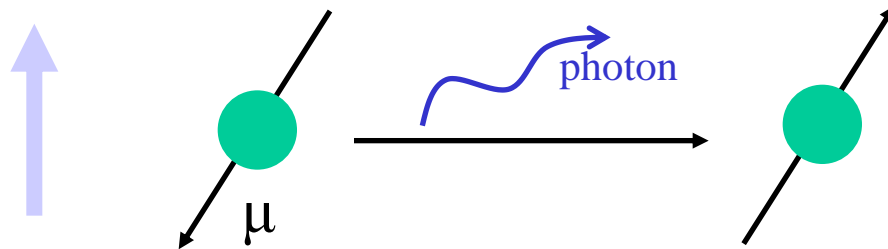


$B_0$   
 $\nu$

is applied field  
speed of rotation (many MHz /  $10^6$  Hz)

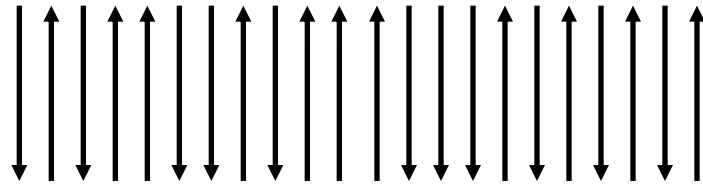
# Do nuclei like fighting the field ?

- is a nucleus really happy facing the wrong way ?
- what if we push it the wrong way ?
  - wants to get to low energy state – emits a photon



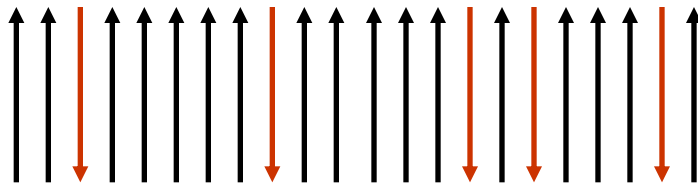
- energy difference very small

# What NMR records



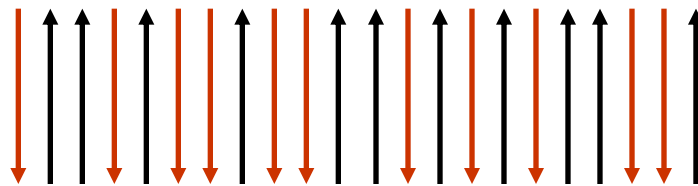
some nuclei not doing much

turn on a field



$B_0$  in an applied field, some align

put in energy

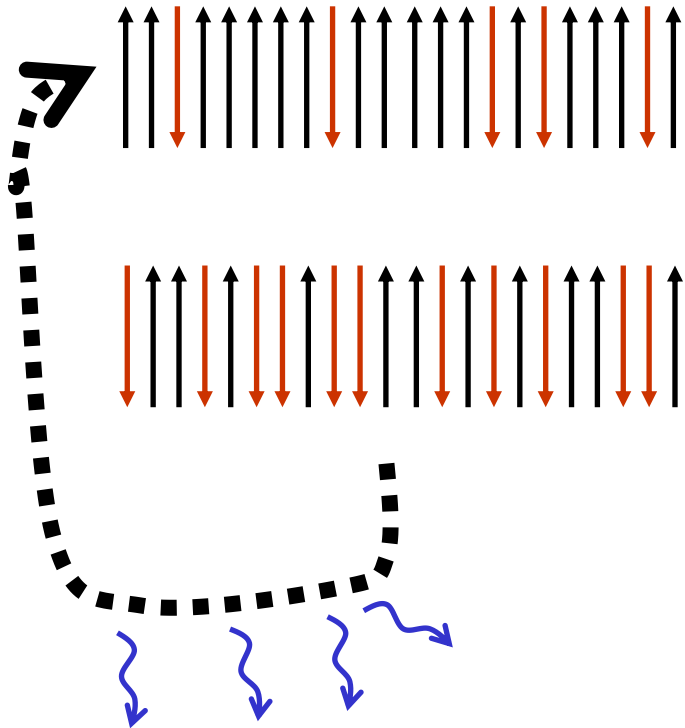


$B_0$

let them relax

Still not really interesting spectroscopy

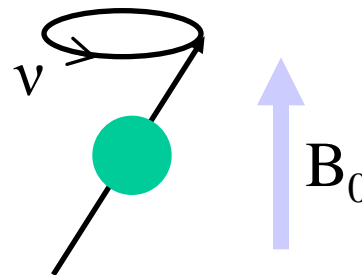
# Is this useful ?



- record some photons (radio freq)  
no information (yet)

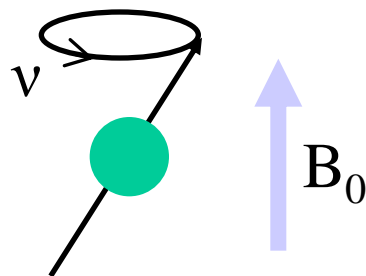
- what if the nuclei emit slightly  
different frequency energy ?

- what determines the frequency ?
  - energy difference
    - field strength



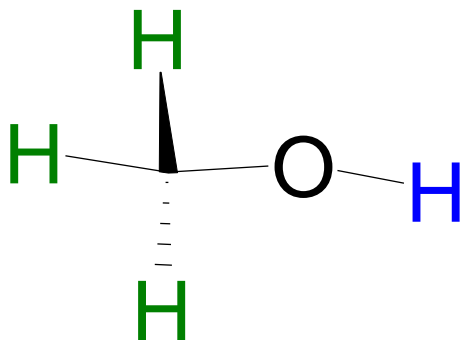


- $B_0$  applied field
- $\nu$  Larmor frequency
- $\gamma$  magic number for nucleus (gyromagnetic ratio) purely empirical



$$\nu = \gamma B_0 / 2\pi$$

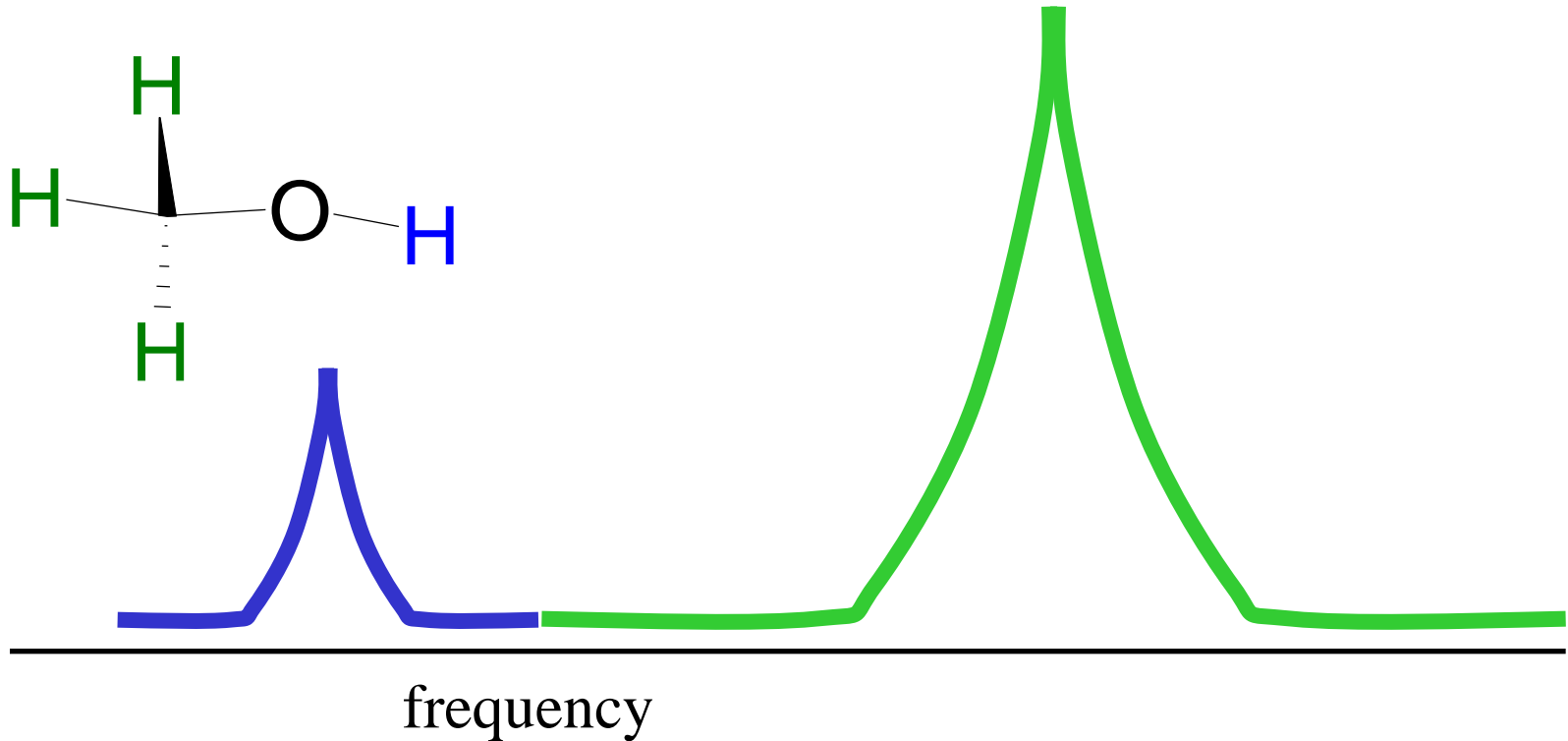
- What is the real field that a nucleus sees ?
- mixture of outside field and local environment



blue H is different to green H  
so frequency should change

# A possible toy spectrum

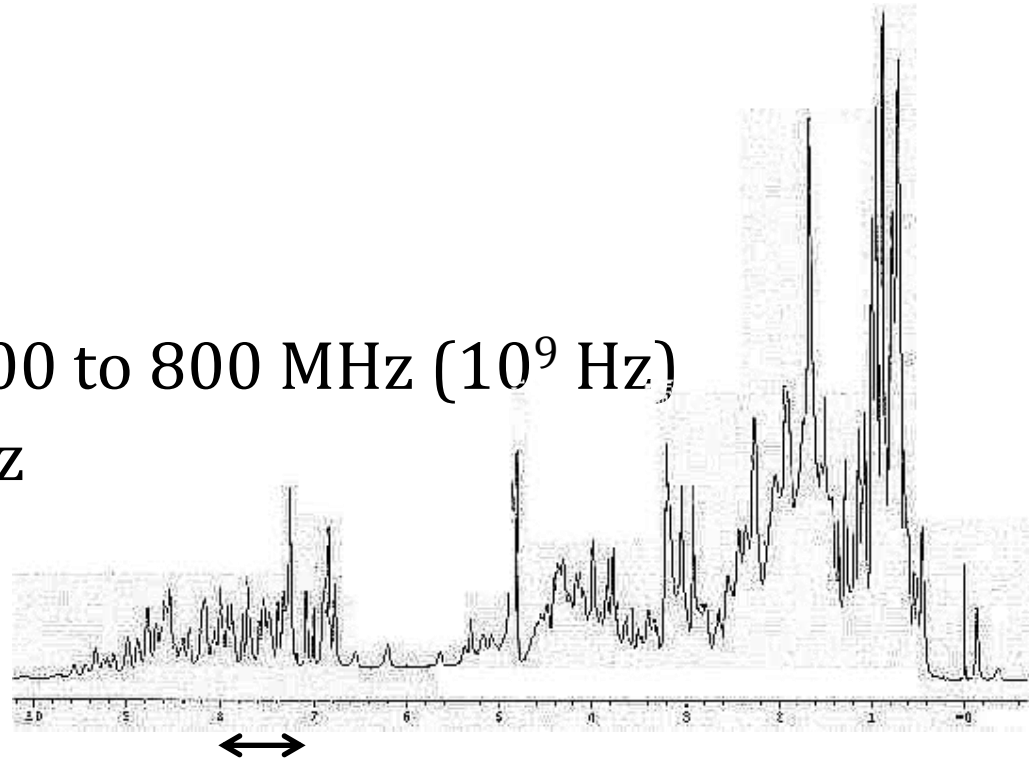
- different atoms / nuclei give different peaks



- a more interesting spectrum ...

# chemical shift / real spectrum

- some protein
  - 100's  $^1\text{H}$
- Scales ?
  - all peaks resonating 100 to 800 MHz ( $10^9$  Hz)
  - whole spectrum  $10^4$  Hz



100 Hz

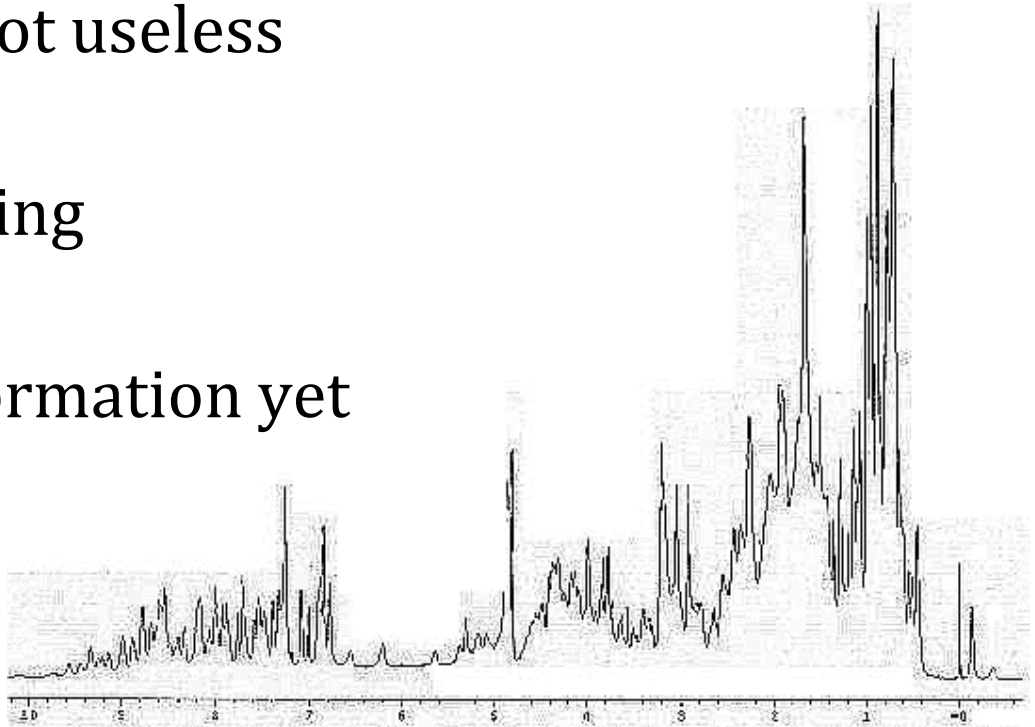
# Important nuclei (spin $\frac{1}{2}$ )

| nucleus         | sensitivity          | notes   |
|-----------------|----------------------|---|
| $^1\text{H}$    | 1                    | cheap and natural                             |
| $^{13}\text{C}$ | $1.6 \times 10^{-2}$ | expensive, but only 1% of natural abundance   |
| $^{15}\text{N}$ | $10^{-3}$            | bit less expensive, 0.4 % natural abundance   |
| $^{31}\text{P}$ | $7 \times 10^{-2}$   | fun for DNA and other $\text{PO}_4$ chemistry |

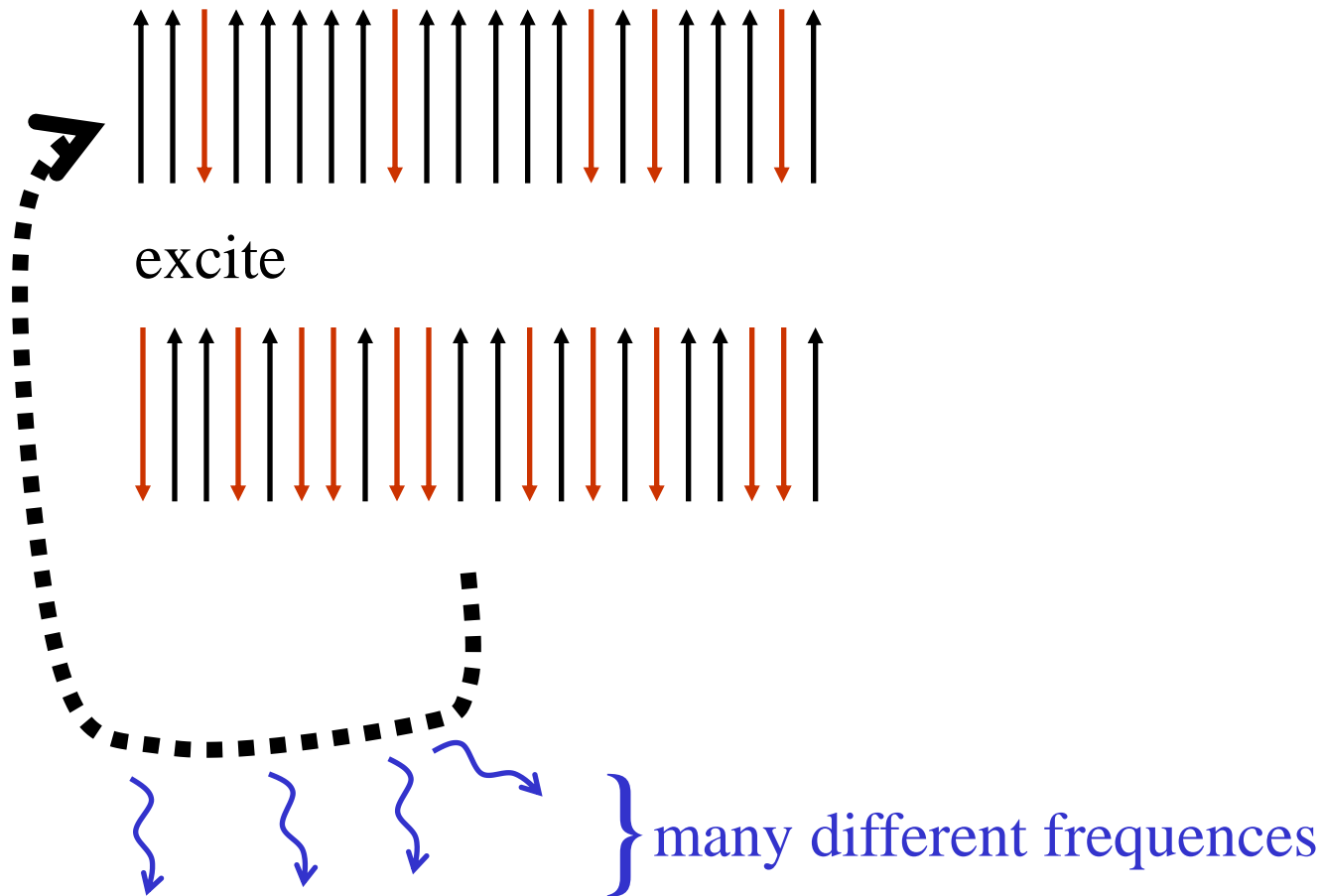
- 
- but the natural isotopes are  $^{12}\text{C}$  and  $^{14}\text{N}$ 
    - (usually) these isotopes require labelling
  - other nuclei ? ...

# A simple spectrum

- an example protein (ubiquitin)
  - lots of peaks, but not useless
- could already
  - look at ligand binding
  - $pK_a$  of residues
- no real structural information yet



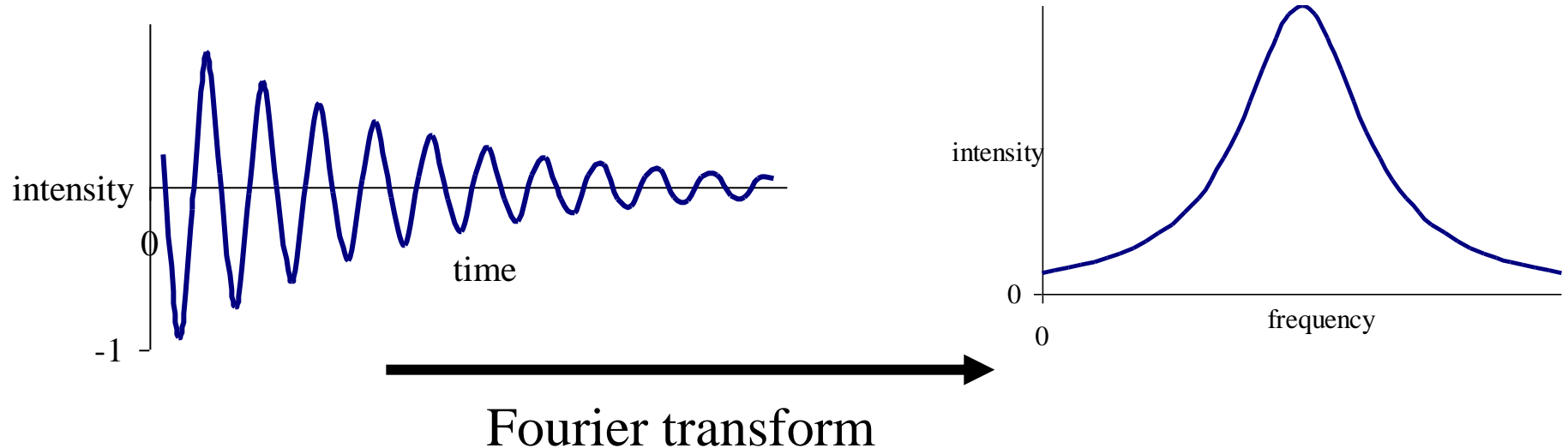
# Recording a spectrum



sort out frequencies with Fourier transform

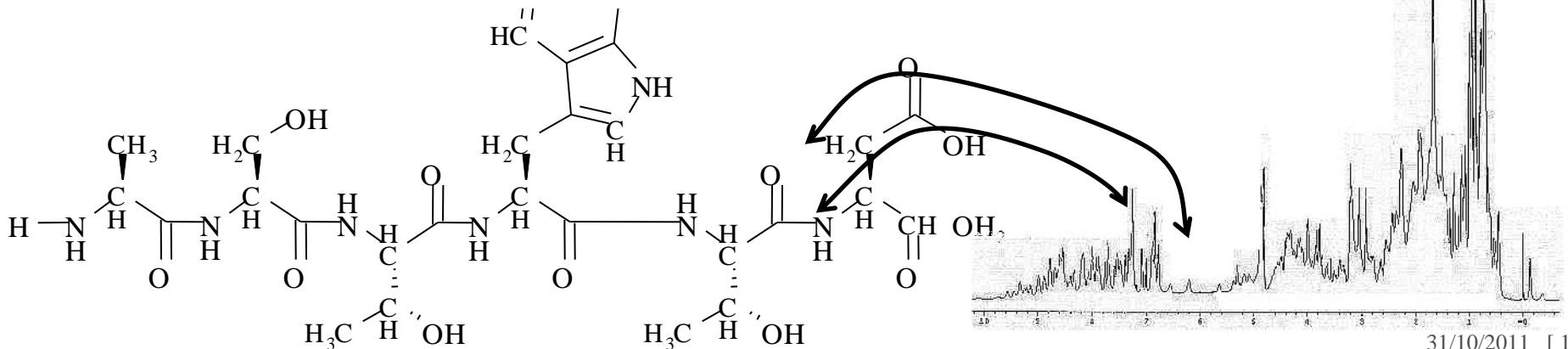
# Raw data and Fourier transforms

- raw data will be simple periodic functions + decay



# Assignment

- Peaks correspond to  $^1\text{H}$  /  $^{13}\text{C}$  from specific atoms
- structural information given by peaks
  - more later
- Assignment – which peaks correspond to which atoms
- Assumption
  - we know exactly which atoms are present
    - sequence of protein known
    - know the structure of each amino acid



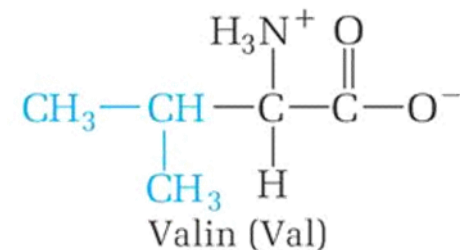


# Assignment

- Which peaks correspond to which atoms ?
  - location / chemical shift
  - with which atoms are you bonded ?
  - which atoms must you be near in space

- Example

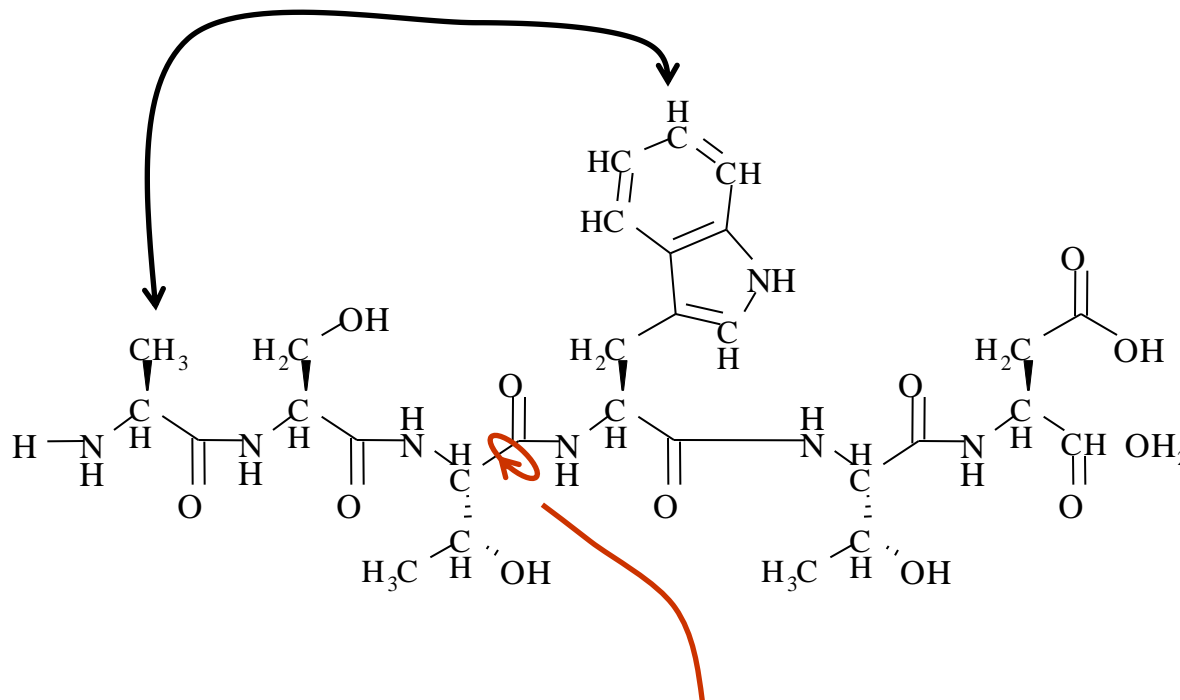
- find a methyl peak
  - connected to a methyne
    - connected to a methyl & another methyne
    - ...



- mostly automated

# To calculate structures ?

## 1. distance information



## 2. dihedral / torsion angle information

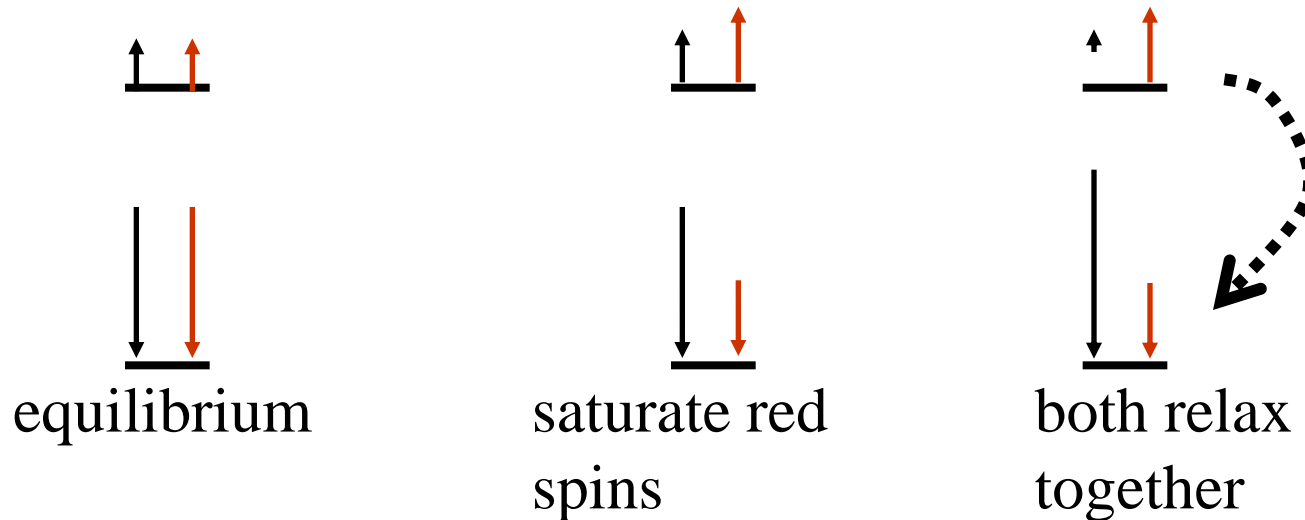
# Distance information / the NOE

- most important
  - an effect which depends on how close in space nuclei are
  - $\text{NOE} \propto r^{-6}$
  - usually only up to about 5 - 6 Å
- story
  - two spins' dipoles interact
  - saturating one spin affects populations of other spin
- who wants an explanation ?
  - cross relaxation phenomenon



- red relaxing (jumping to lower energy) affects black
- can one create this situation ?

# Cross relaxation and the NOE

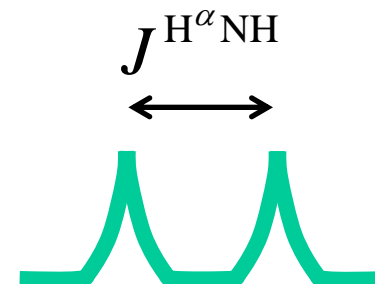
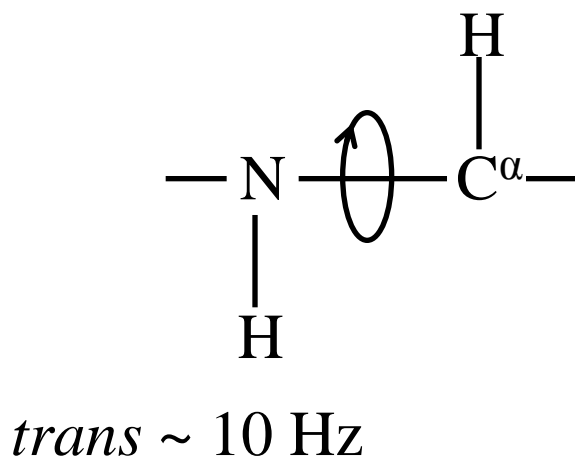
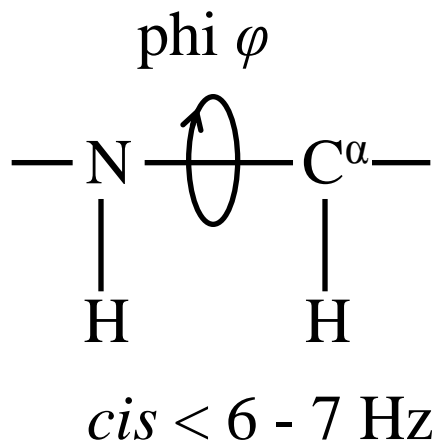


- now, the population difference is bigger than normal
  - bigger signal
- record a normal spectrum
  - red is not there
  - black is "enhanced"
- via another mechanism
  - population difference can become smaller
- only happens if nuclei are very close in space

# Other structural information

- NOE – information about short (  $< 5$  or  $6 \text{ \AA}$  ) distances
- there is more – angles
  - mainly  $J$  coupling

## Amide NH to $H^\alpha$ coupling

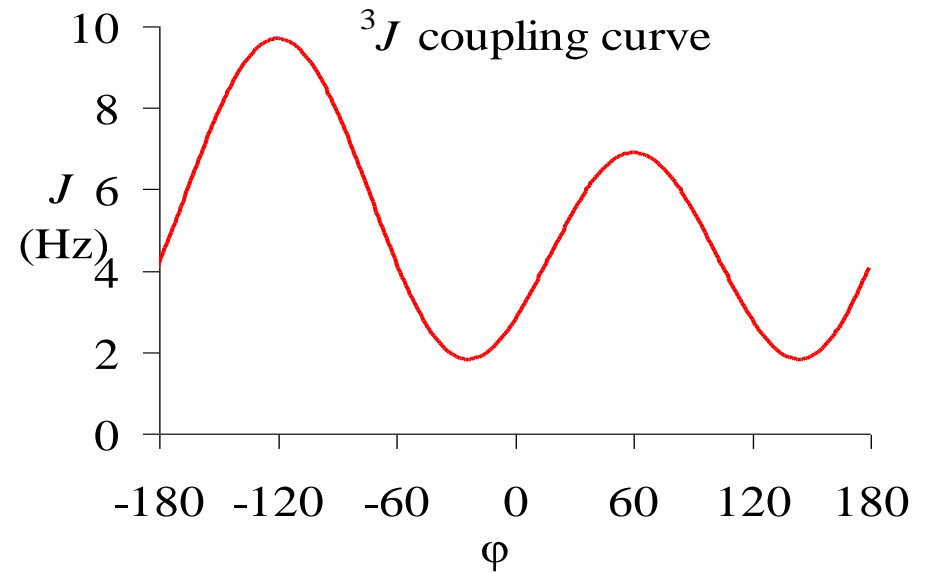


# $^3J_{\text{HN}\alpha}$ coupling

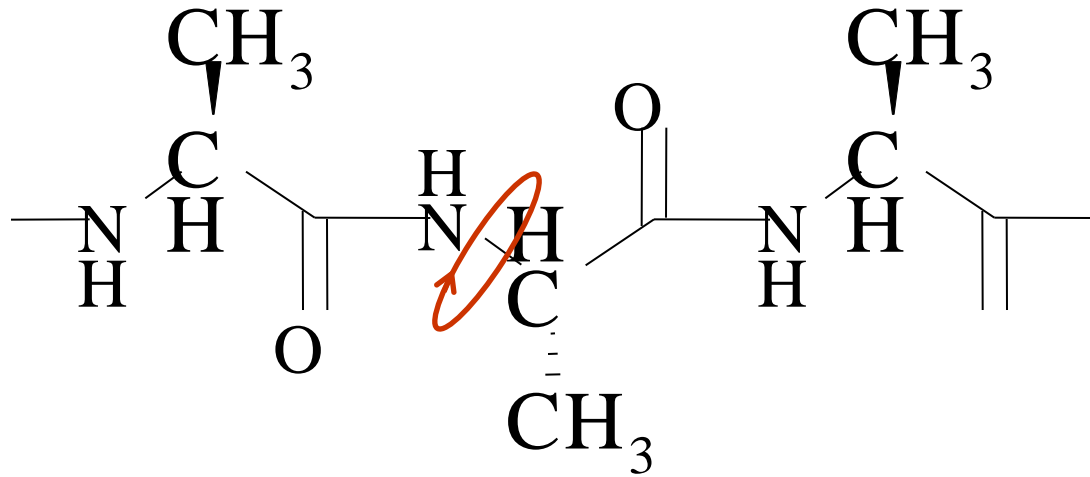
- formalised as

$$^3J_{\text{HN}^\alpha} = 6.4 \cos^2 \theta - 1.4 \cos \theta + 1.9$$

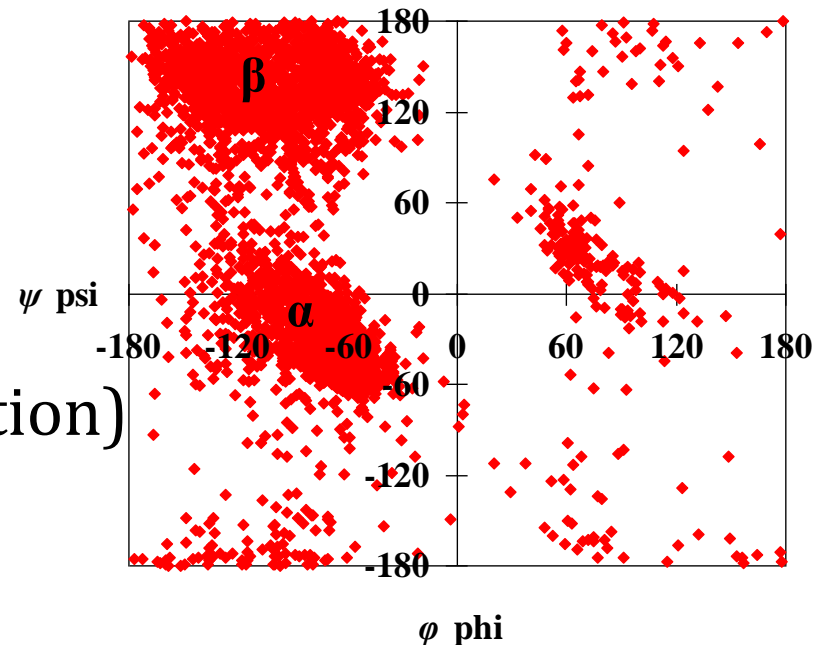
Problems later



# Amide NH to H $^{\alpha}$ coupling



- can help distinguish  $\alpha$  from  $\beta$
- not always seen (exchange / motion)
- NH not always present
- other angles ?
  - other vicinal protons
  - C $^{\alpha}$  to C $^{\beta}$



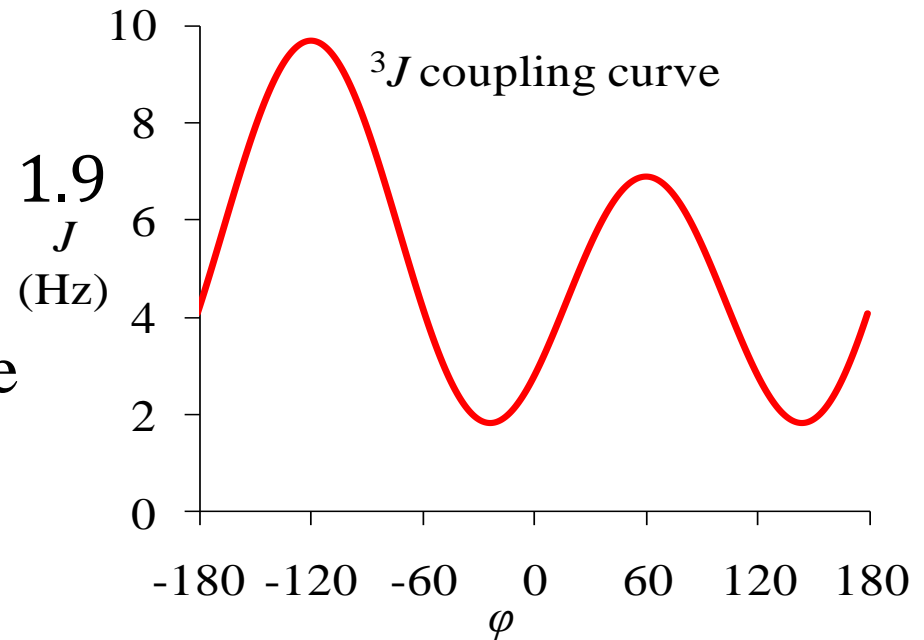
# Problems with $J$ -coupling

1. we have a formula

$${}^3J_{HN^\alpha} = 6.4 \cos^2 \theta - 1.4 \cos \theta + 1.9$$

$J$   
(Hz)

- most of the time, there is more than one solution
- only use very big  $J$  values



2. dynamics

more serious than they appear ! look around  $-90^\circ$

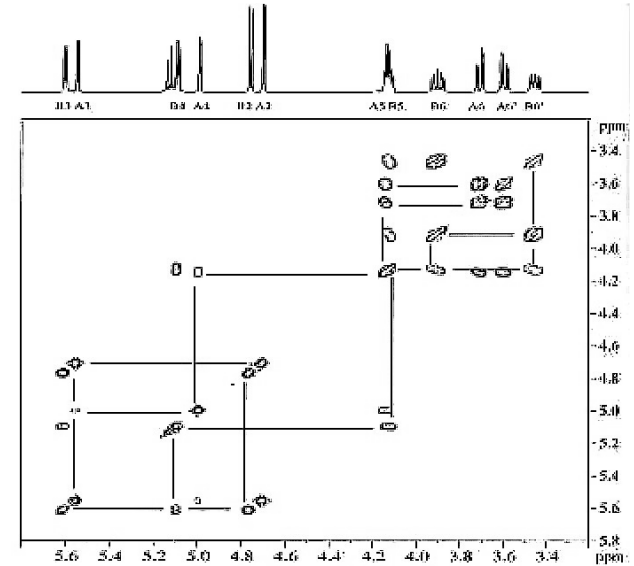


# Practical NMR

- We have some basic methods
- Real NMR
  - more techniques
    - identifying specific kinds of atom
    - spreading peaks out
- Briefly mention the most important...
  - 2D NMR

# 2D NMR

- two reasons
  - 1. spread spectrum out
    - resolve peaks / remove overlap
  - 2. add information



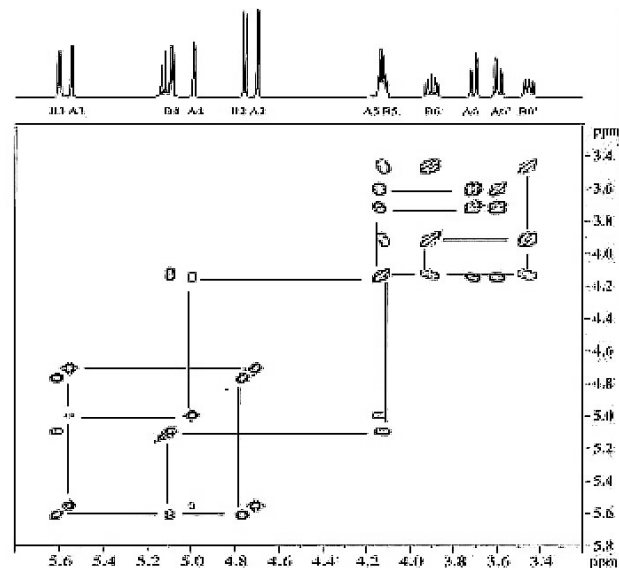
# 2D spectra information

What do the off-diagonal peaks mean ?

- depends on spectrum

Example 1

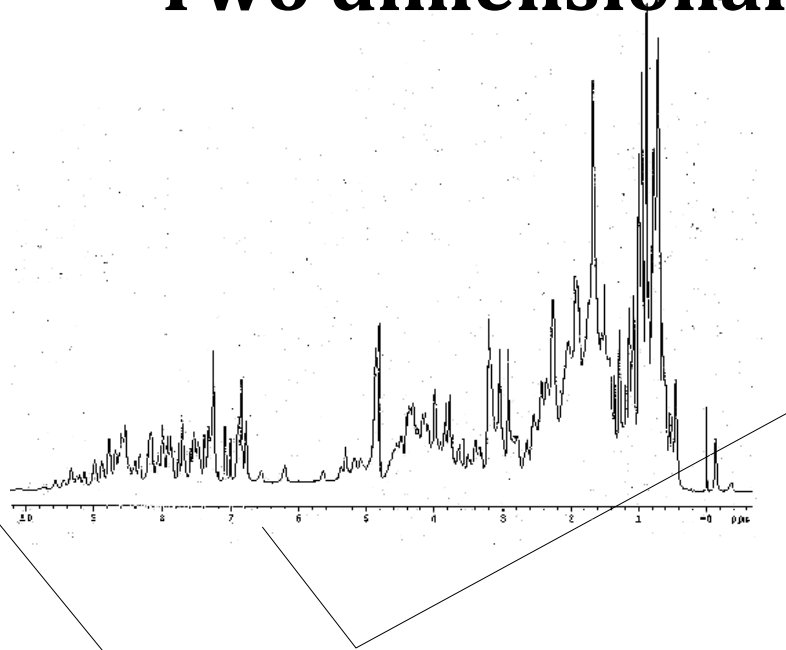
- COSY (correlated spectroscopy)
- peaks indicate J-coupling
- look at spectrum and quickly see which peaks are connected



Example 2

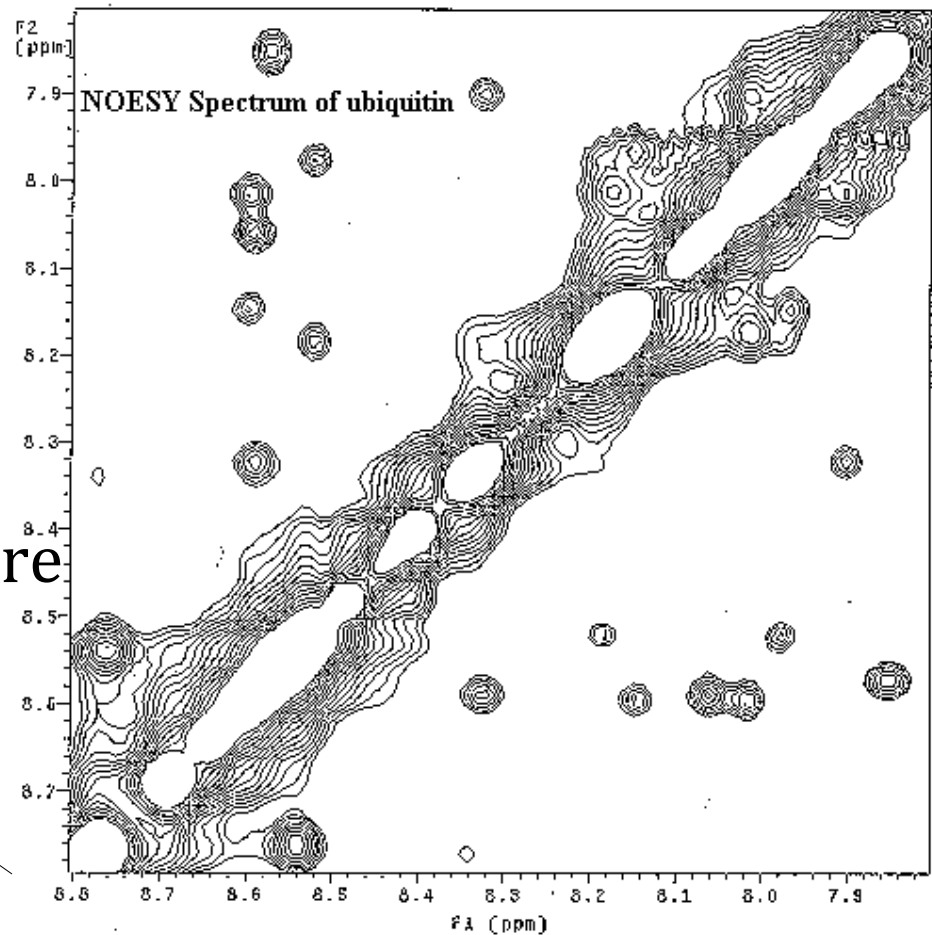
- NOESY (NOE ...)
- peaks indicate NOE
- corresponding nuclei close in space

# Two dimensional NOE spectra example



## 2 D NOE spectrum

- NOESY
- what determines if peaks are present ?



# Information summary

| phenomenon                 | assignments | structure        |
|----------------------------|-------------|------------------|
| chemical shift             | important   | not used         |
| spin-spin ( $J$ ) coupling | important   | torsion angles   |
| NOE / distances            | important   | main information |

- more spectroscopy
  - filtering according to chemistry, atom types
  - $n$ -dimensional methods
- structural information
  - labels for broadening / shifting peaks
  - orientation of bonds to reference ..

# Structures from NMR data

To come

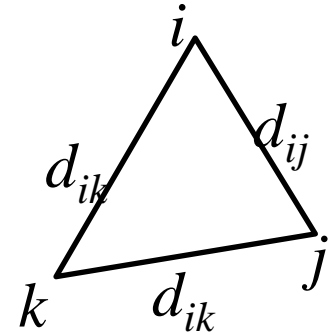
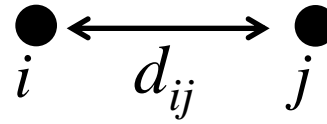
- Distances in 2 and 3 D
- Distance geometry
  - 2 approaches
- Restrained molecular dynamics (MD)

Available information

- distances
  - short (5 to 6 Å)
  - incomplete
- some dihedral / torsion angles
- does this define a structure ?
  - strictly no
  - with chemical information ?
    - still not

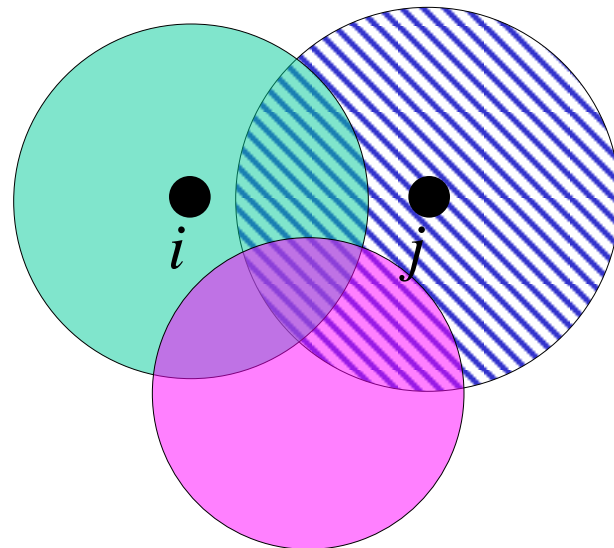
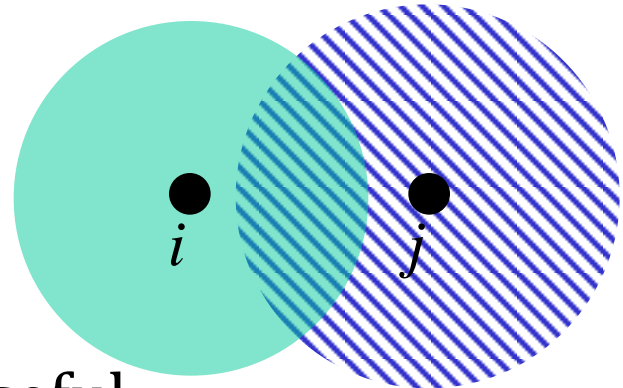
# Determining distances (ideal)

- 2 points 1 distance
- 3 points 3 distances...
  - think of  $3N_{atom}$  distances
  - remember  $N_{atom} \approx 10$  or  $20 N_{res}$



# Underdetermined distances

- think in terms of triangles ...
  - $d_{ik} < 6 \text{ \AA}, d_{jk} < 6 \text{ \AA}$
  - where is  $k$  ?
- a few more distances...
  - more and more distances are useful

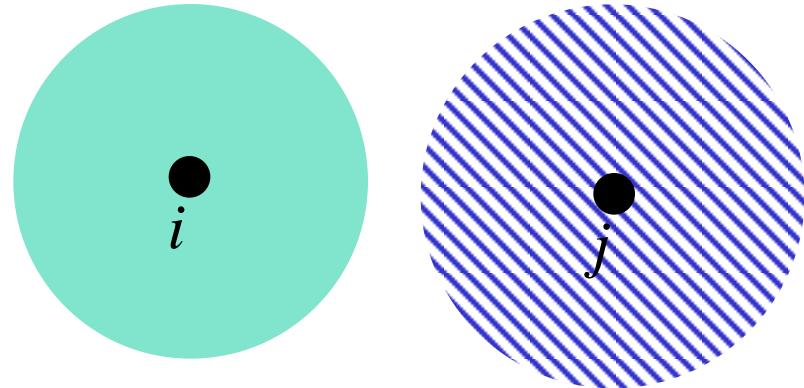




# Impossible distances

No overlap ?

- experimental error
- nowhere for  $k$  to go



## Real data

For  $N$  residue protein, maybe  $5 N_{res}$  or  $10 N_{res}$

- want more like  $3N_{atom}$  ( $30 - 60 N_{res}$ ) distances if perfect
  - needs much more data...
  - lots of chemical data

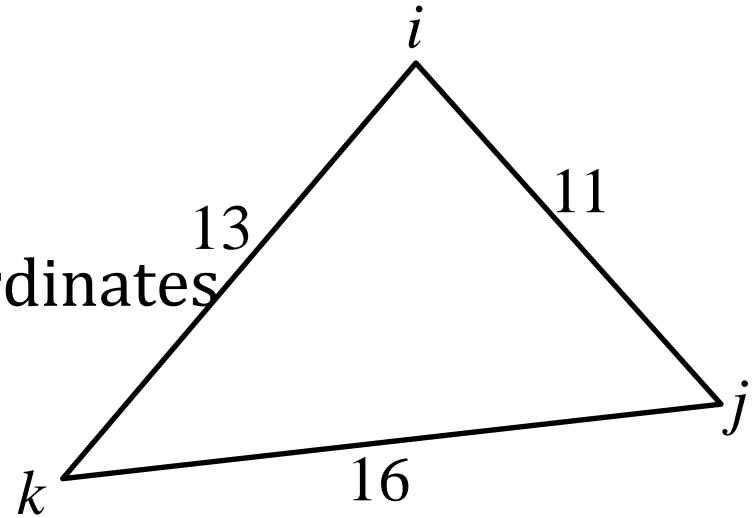
# Mission

- gather all experimental data
- mix in chemical data
- make all distance information as tight as possible
- put an upper bound on the distance between every pair of points
- put a lower bound on every distance (less important)
- somehow generate coordinates
- start with toys and triangles

# Structures from distance information

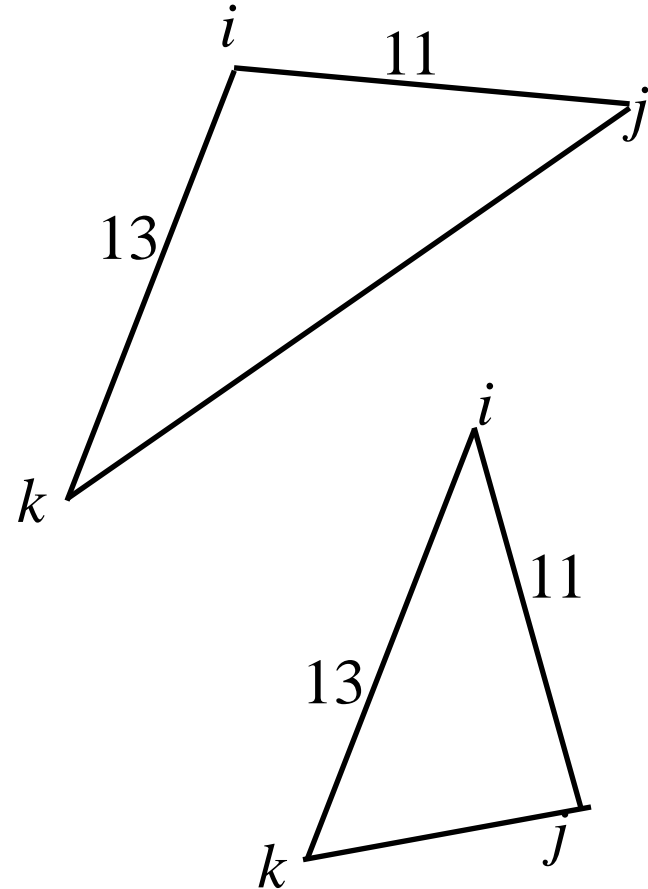
Start in two dimensions..

- ein freundliches Dreieck
  - $d_{ij}=11$     $d_{ik}=13$     $d_{jk}=16$
- fix  $i$ , put  $j$  on x-axis and make coordinates
- solve analytically



# Underdetermined data

- $d_{ij}=11$     $d_{ik}=13$     $d_{jk}=12 - 20$ 
  - more like NMR data
- unique solution ?
  - no



# Impossible data

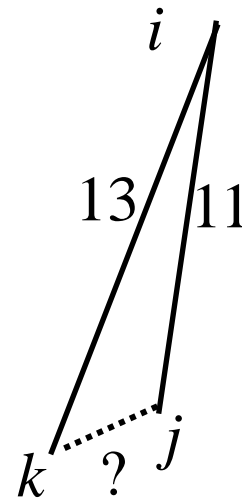
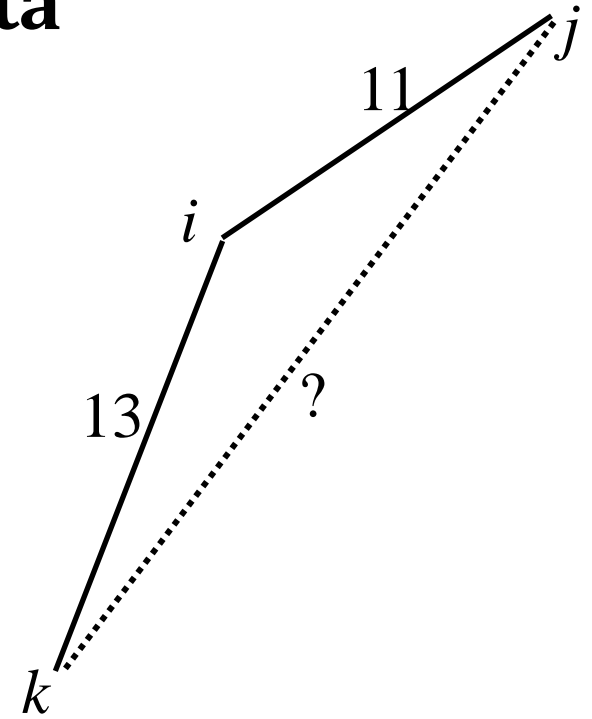
- distance too big

$$d_{ij}=11 \quad d_{ik}=13 \quad d_{jk}=25$$

- distance too small

- $d_{ij}=11 \quad d_{ik}=13 \quad d_{jk}=1$

- no 3D structure



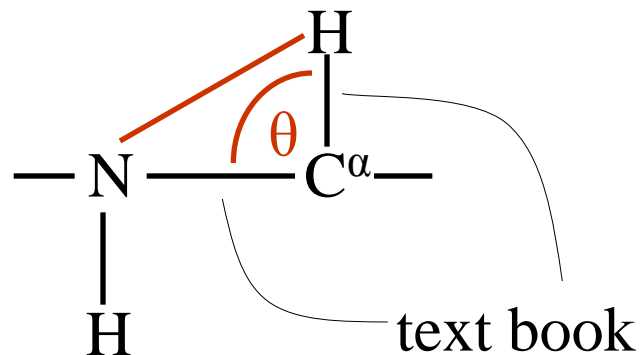
# Gathering data

- add in chemistry
- use to get more
  - mix chemistry + measurements
- what comes easily from chemistry ?

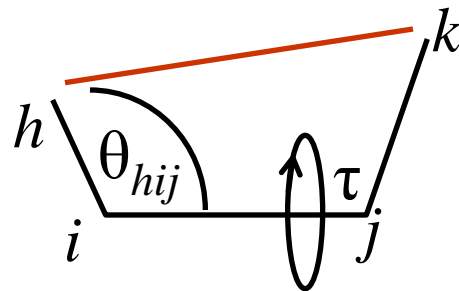
# Gather as much data as possible

Simple, geometric information

- bonds – standard
- angles – standard
- simple distances from bond angles
- dihedral / torsion angles



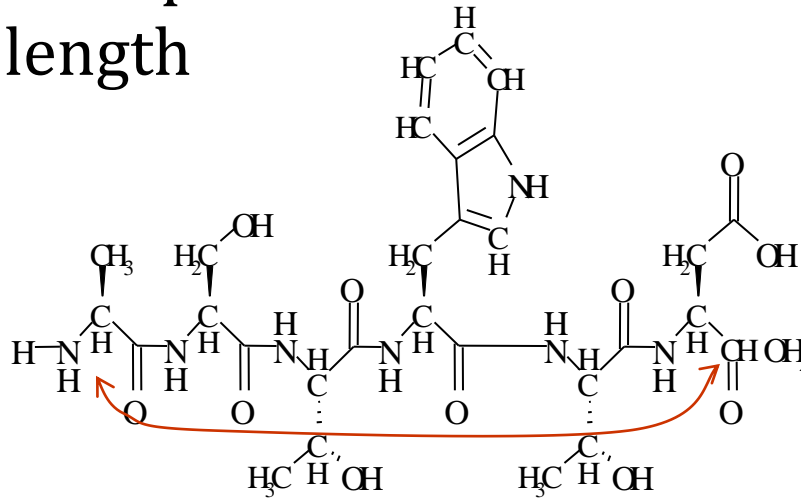
$$\begin{aligned} d_{hk}^2 = & \left( d_{ij} - d_{hi} \cos \theta_{hij} - d_{jk} \cos \theta_{ijk} \right)^2 \\ & + \left( d_{hi} \sin \theta_{hij} - d_{jk} \sin \theta_{ijk} \cos \tau_{hijk} \right)^2 \\ & + \left( d_{jk} \sin \tau_{hijk} \right)^2 \end{aligned}$$



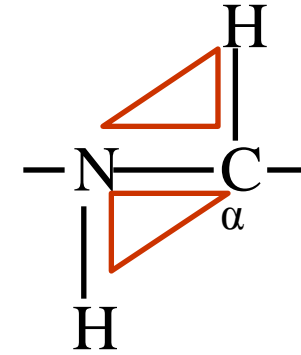
- set  $\tau = 0$ 
  - minimum
- $\tau = \pi$ 
  - maximum

# How to get more distance information

- impose some distance limits generally
- intuitively
  - stretch out a protein and there is a limit to length



??



- can we formalise this ?



# More general / triangle inequality

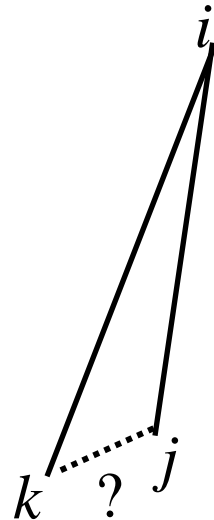
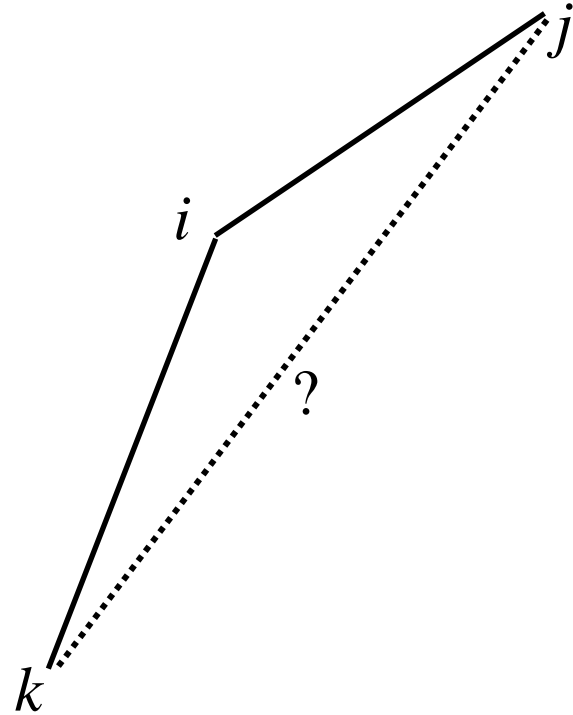
What limits can be worked out ?

- upper bound

$$d_{jk} \leq d_{ij} + d_{ik}$$

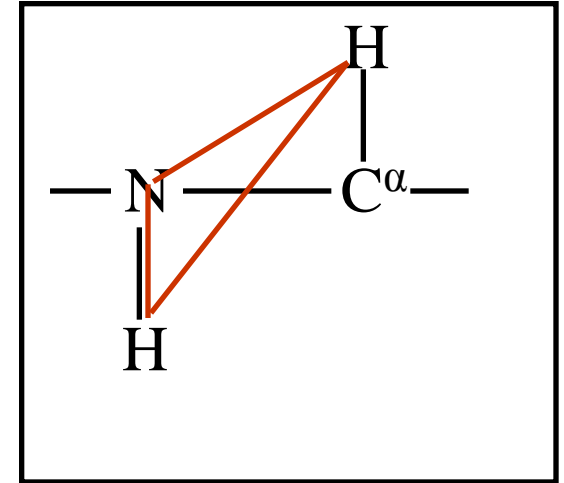
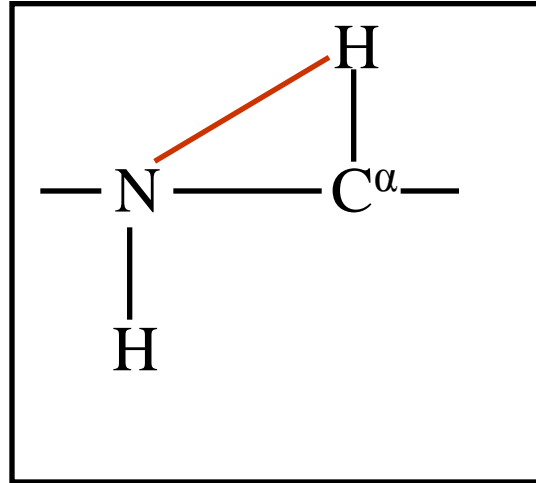
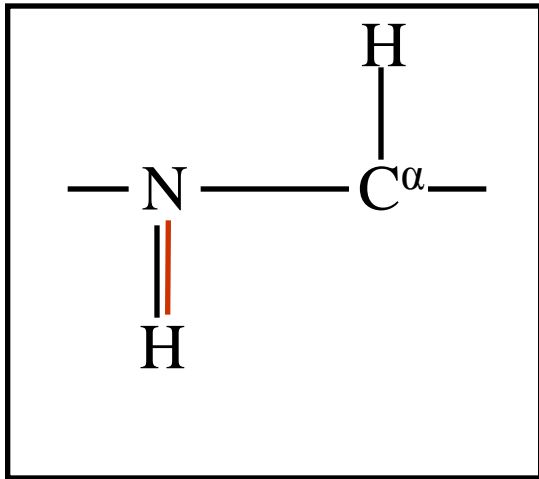
- lower bound

- $d_{jk} \geq |d_{ij} - d_{ik}|$



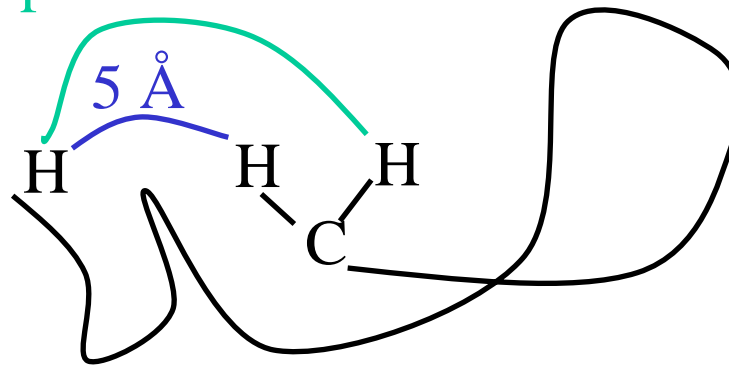
# Where to use triangle inequality

- we could avoid some ugly trigonometry



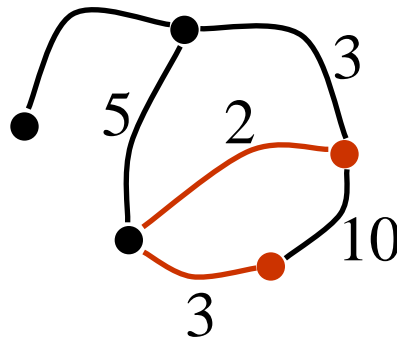
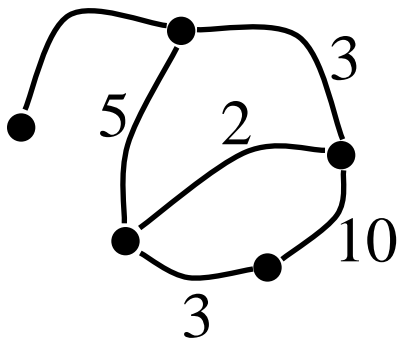
- more general

implied 6 or 7 Å

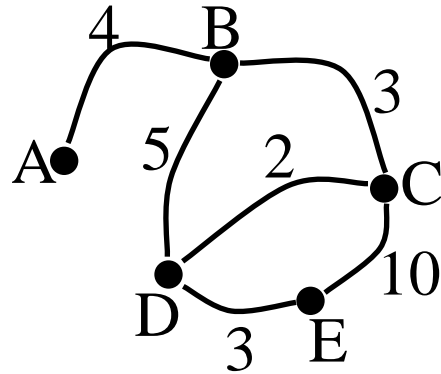


# Most general triangle bound inequality

- triangle bound should be satisfied by any three points
- chemists
  - triangle bound smoothing
- informatik
  - all points shortest path problem



# All points shortest path (Floyd)

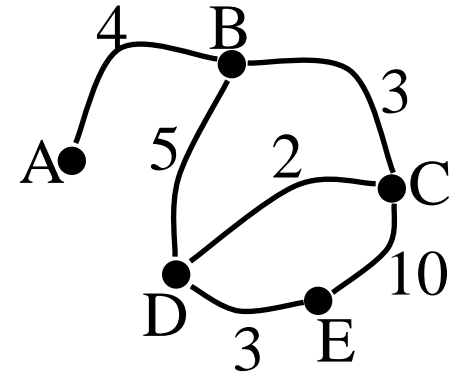


|   | A | B | C | D | E  |
|---|---|---|---|---|----|
| A |   | 4 |   |   |    |
| B |   |   | 3 | 5 |    |
| C |   |   |   | 2 | 10 |
| D |   |   |   |   | 3  |
| E |   |   |   |   |    |

|   | A | B | C   | D   | E   |
|---|---|---|-----|-----|-----|
| A |   | 4 | max | max | max |
| B |   |   | 3   | 5   | max |
| C |   |   |     | 2   | 10  |
| D |   |   |     |     | 3   |
| E |   |   |     |     |     |

# Bound smoothing / Floyd

|   | A | B | C   | D   | E   |
|---|---|---|-----|-----|-----|
| A |   | 4 | max | max | max |
| B |   |   | 3   | 5   | max |
| C |   |   |     | 2   | 10  |
| D |   |   |     |     | 3   |
| E |   |   |     |     |     |



```

for k = 0; k < n_last; k++)
    for (i = 0; i < n_last; i++)
        for (j = 0; j < n_last; j++)
            if ij > ik + jk
                ij := ik + jk
  
```

Running time

- $O(n^3)$

|   | A | B | C | D | E  |
|---|---|---|---|---|----|
| A |   | 4 | 7 | 9 | 12 |
| B |   |   | 3 | 5 | 8  |
| C |   |   |   | 2 | 5  |
| D |   |   |   |   | 3  |
| E |   |   |   |   |    |

# Distance matrix so far

- we can build a distance matrix of upper limits
  - consistent with all bonds and angles and other information
- can do the same for lower bounds
  - every pair of atoms
    - invent some lower bound (atomic radii)

## Does this define a structure ?

- almost certainly not
  - still no way to get to a 3D model

# From distances to coordinates

How would you build coordinates from distances

- stepwise ?
  - error prone, errors add
- history
  - early 80's
  - methods which are tolerant of errors
    - metric matrix method

# Metric matrix method

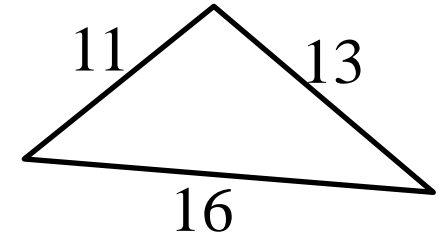
- get best upper bounds
- get best lower bounds
  - guess distances between
    - trial distance matrix
  - convert to centre of mass matrix (metric matrix)
  - magic conversion to coordinates
    - if metric matrix has three positive eigenvalues
      - error free coordinates
- real coordinates
  - lots of errors
  - initial coordinates not healthy
  - refine



# Chirality

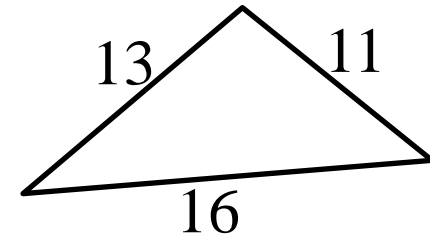
## 2D version

- can \*not\* be rotated on to each other
- can not be distinguished by distances



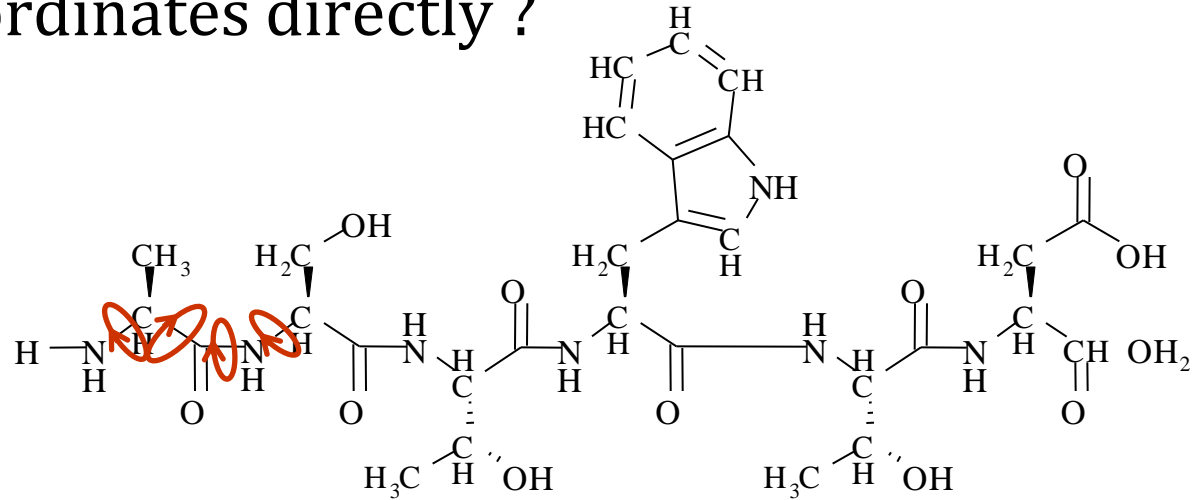
## 3D

- chirality is random
- problem ? no
  - flip all coordinates and check
- local chirality
  - mixture of good and bad
  - difficult to fix



# Other distance geometry

Can we adjust coordinates directly ?

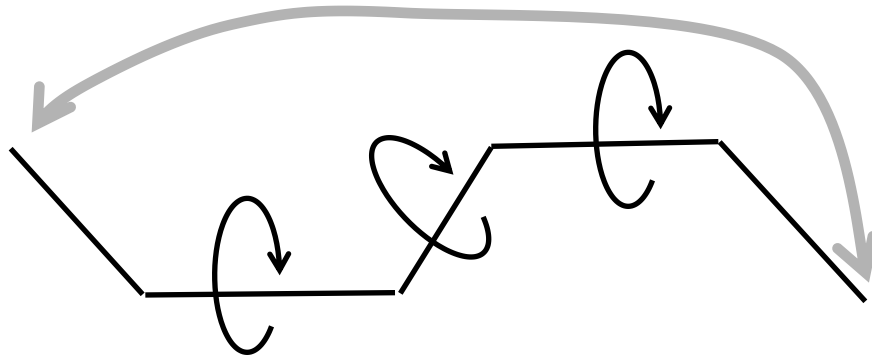


Can we work with angles ?

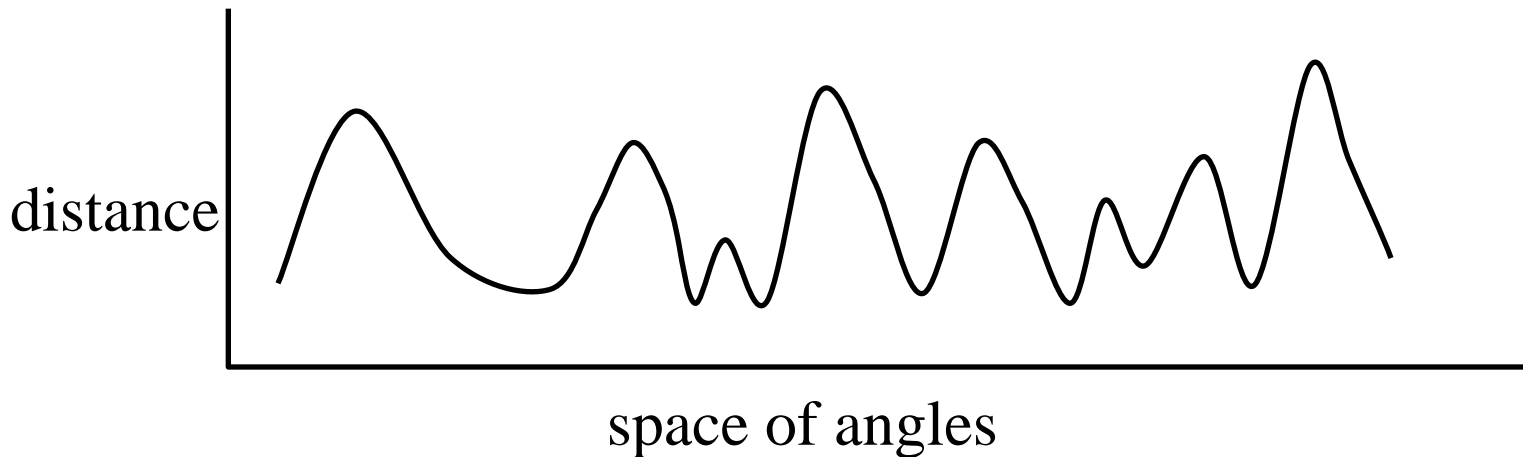
- many fewer angles than atoms

# Distances and angles

- each distance may depend on many included angles

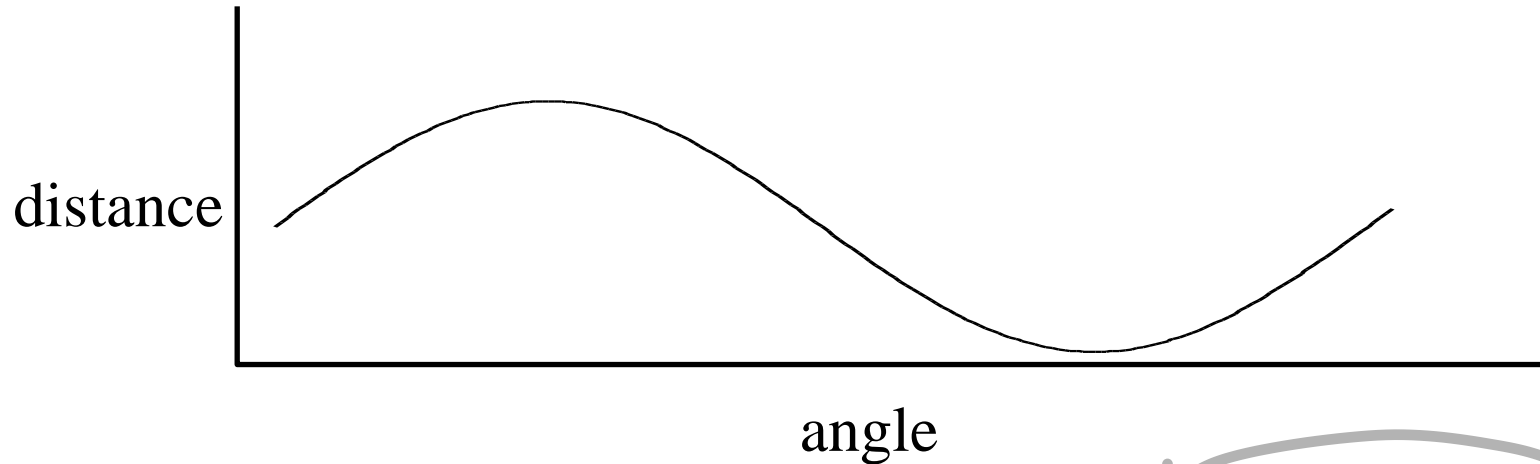


- high dimensional space of angles...

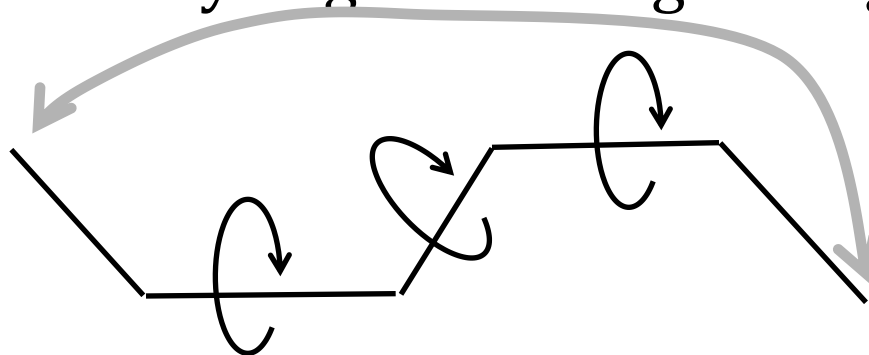
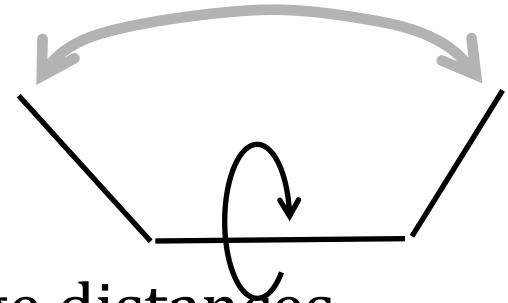


# Distances and angles

- moving one angle affects one distance simply



- one angle is very important for a
  - short range distance
- just one of many angles for longer range distances



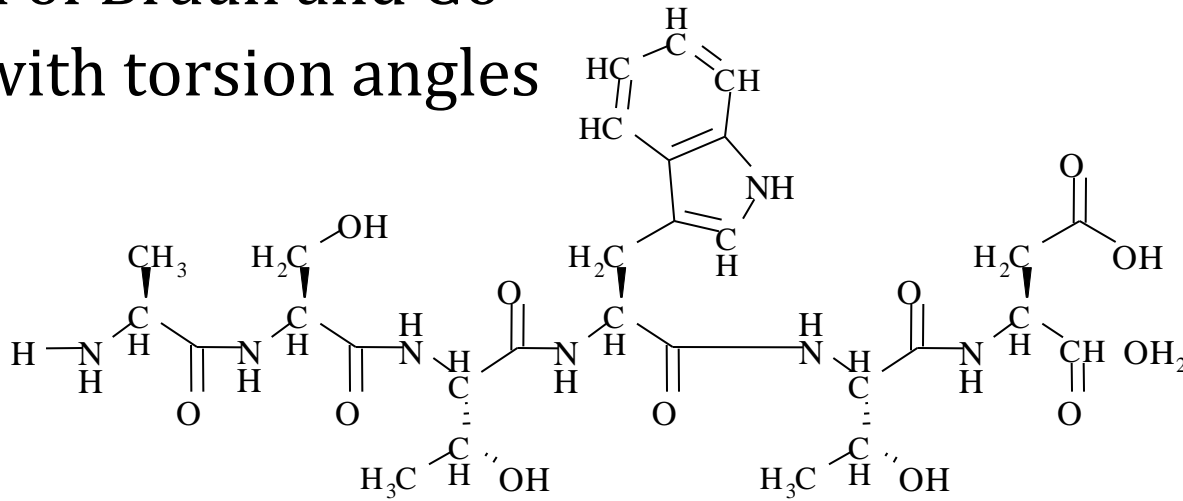
# Optimisation Strategy

- start
  - concentrate on distances with few angles in between
  - shorter distances become correct
- add in more distances
  - re-optimize
- add in more distances
  - ...

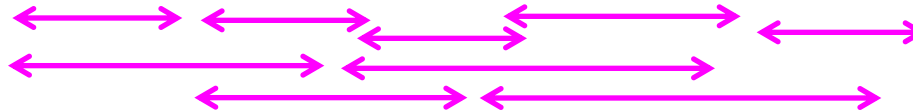
# Variable target function

approach of Braun and Gō

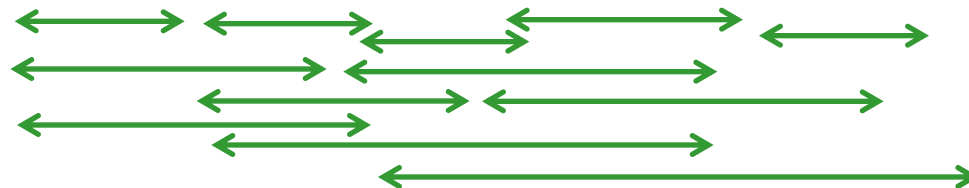
- work with torsion angles



1<sup>st</sup> step



2<sup>nd</sup> step



3<sup>rd</sup> step

# Stepwise variable target function method

- Collect experimental data

| distance<br>in<br>sequence | residue<br>1 | atom<br>1      | residue<br>2 | atom<br>2      | distance<br>in space<br>(Å) |
|----------------------------|--------------|----------------|--------------|----------------|-----------------------------|
| 1                          | 5            | H <sup>α</sup> | 6            | H <sup>N</sup> | 4.0                         |
| 0                          | 8            | H <sup>α</sup> | 8            | H <sup>γ</sup> | 4.4                         |
| 80                         | 2            | H <sup>α</sup> | 82           | H <sup>N</sup> | 4.5                         |
| 2                          | 3            | H <sup>α</sup> | 5            | H <sup>γ</sup> | 5.0                         |
| 1                          | 7            | H <sup>β</sup> | 8            | H <sup>γ</sup> | 3.8                         |
| 0                          | 3            | H <sup>α</sup> | 3            | H <sup>N</sup> | 5.0                         |

- Sort according to distance in sequence

# Stepwise variable target function method

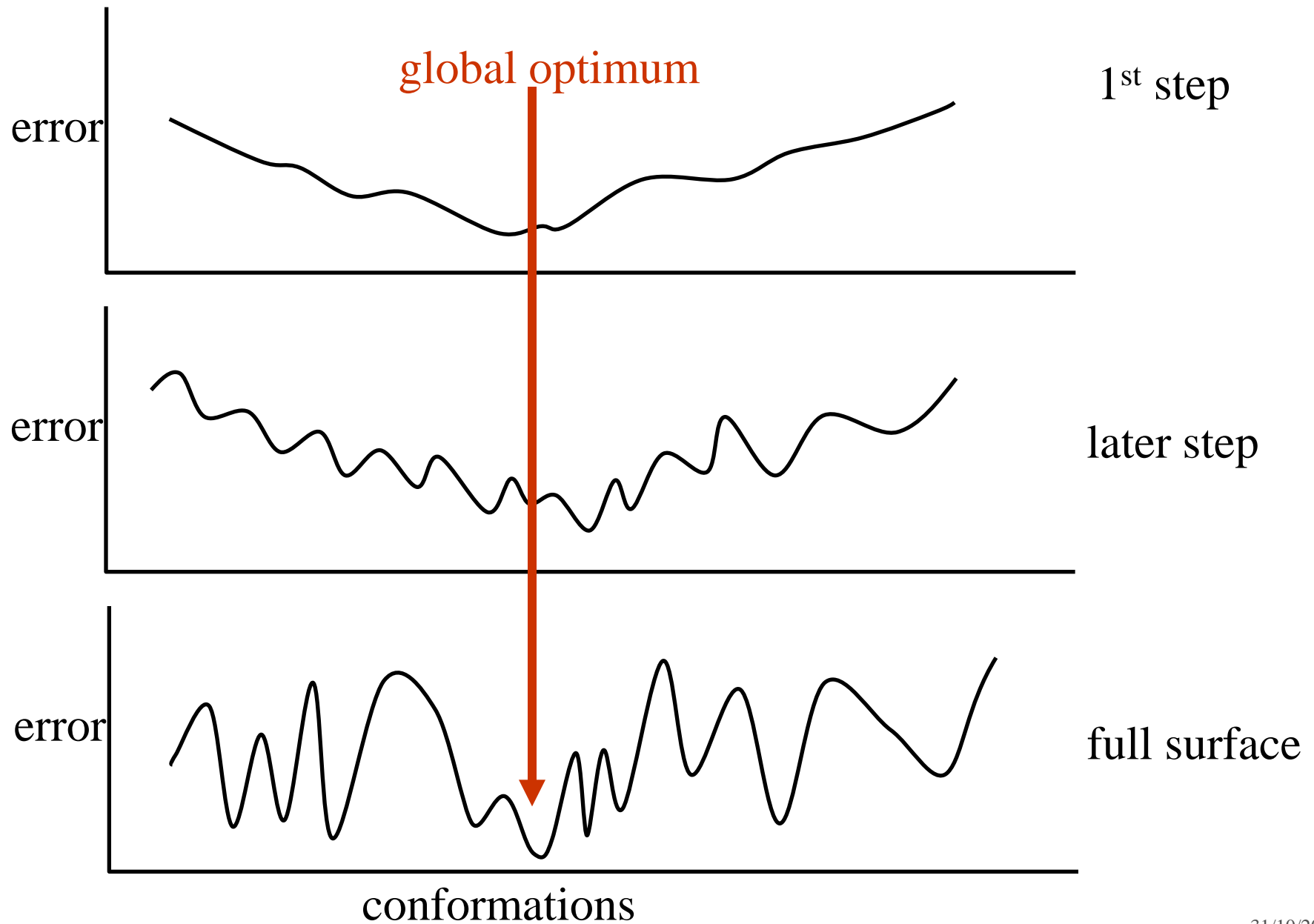
| distance<br>in<br>sequence | residue<br>1 | atom<br>1     | residue<br>2 | atom<br>2     | distance<br>in space<br>(Å) |
|----------------------------|--------------|---------------|--------------|---------------|-----------------------------|
| 0                          | 8            | H $^{\alpha}$ | 8            | H $^{\gamma}$ | 4.4                         |
| 0                          | 3            | H $^{\alpha}$ | 3            | H $^N$        | 5.0                         |
| 1                          | 5            | H $^{\alpha}$ | 6            | H $^N$        | 4.0                         |
| 1                          | 7            | H $^{\beta}$  | 8            | H $^{\gamma}$ | 3.8                         |
| 2                          | 3            | H $^{\alpha}$ | 5            | H $^{\gamma}$ | 5.0                         |
| ...                        |              |               |              |               |                             |
| 80                         | 2            | H $^{\alpha}$ | 82           | H $^N$        | 4.5                         |
| ...                        | ...          |               |              |               |                             |



# Stepwise variable target function method

| distance<br>in<br>sequence | residue<br>1 | atom<br>1      | residue<br>2 | atom<br>2      | distance<br>in space<br>(Å) | 1 <sup>st</sup> | 2 <sup>nd</sup> | 3 <sup>rd</sup> | ... | later |
|----------------------------|--------------|----------------|--------------|----------------|-----------------------------|-----------------|-----------------|-----------------|-----|-------|
| 0                          | 8            | H <sup>α</sup> | 8            | H <sup>γ</sup> | 4.4                         | ↓               | ↓               | ↓               |     | ↓     |
| 0                          | 3            | H <sup>α</sup> | 3            | H <sup>N</sup> | 5.0                         |                 |                 |                 |     |       |
| 1                          | 5            | H <sup>α</sup> | 6            | H <sup>N</sup> | 4.0                         |                 |                 |                 |     |       |
| 1                          | 7            | H <sup>β</sup> | 8            | H <sup>γ</sup> | 3.8                         |                 | ↓               | ↓               |     |       |
| 2                          | 3            | H <sup>α</sup> | 5            | H <sup>γ</sup> | 5.0                         |                 |                 |                 |     |       |
| ...                        |              |                |              |                |                             |                 |                 |                 |     |       |
| 80                         | 2            | H <sup>α</sup> | 82           | H <sup>N</sup> | 4.5                         |                 |                 |                 |     |       |
| ...                        | ...          |                |              |                |                             |                 |                 |                 |     |       |

# Hope..



# Variable target function vs metric matrix

- metric matrix vs variable target function
  - proponents of both
- variable target function probably more popular
  - no problems with chirality

# Real implementations of distance geometry

- not small programs
- what kind of input would they like ?
  - list of protein sequence
  - set of distances
- most of code
  - libraries of standard amino acids
  - code to do geometry and work with standard geometries
- other information
  - angle restraints
    - convert to distances for metric matrix
    - natural for variable target function

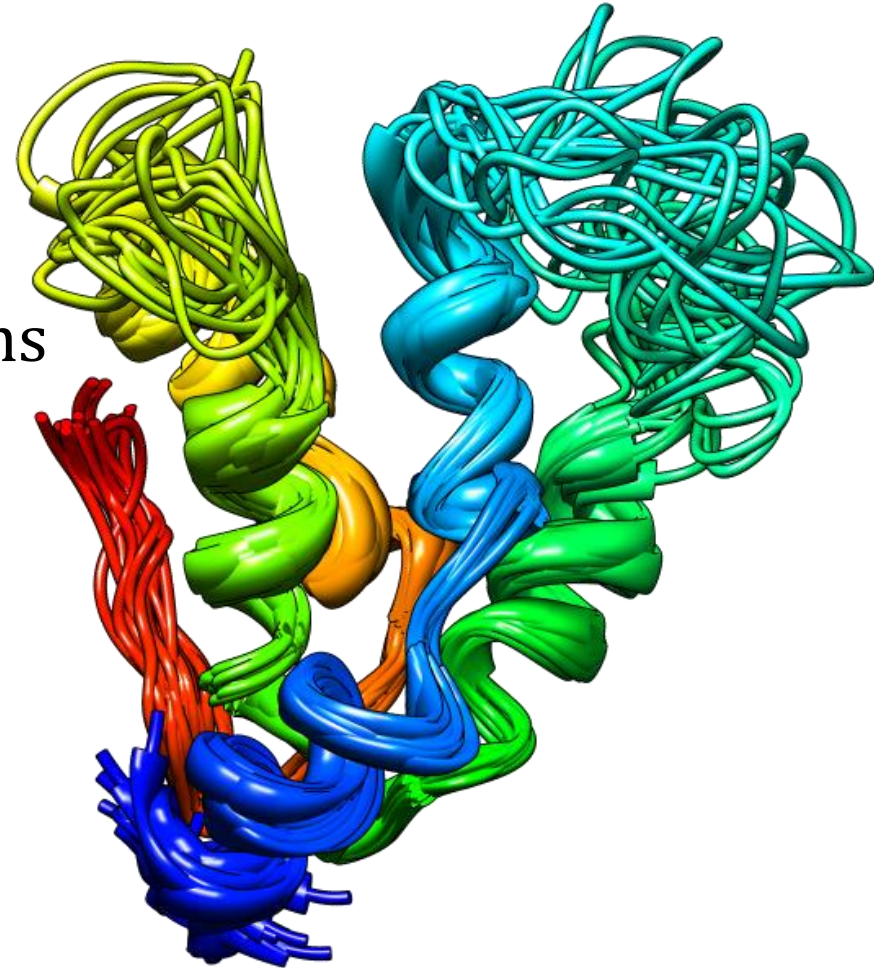
# Output from programs

Structure impossible ?

- program dies or
- best possible solution

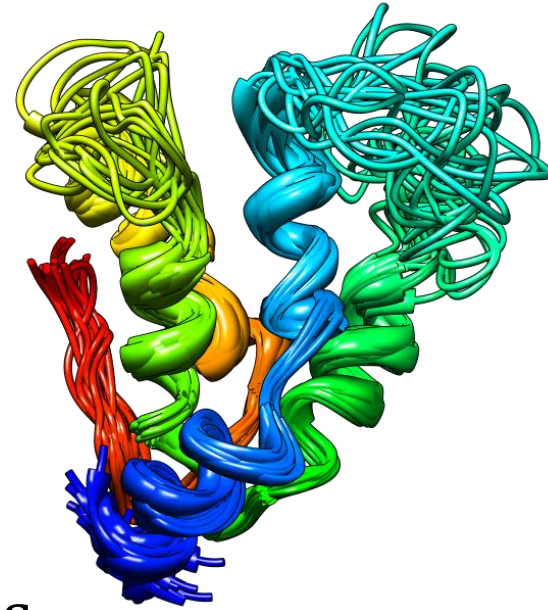
Structure not determined ?

- set of possible conformations (10 to 100 )
- example 1sm7.pdb



# Lots of models in a PDB file

- big difference compared to most x-ray coordinates
- typical
  - ends (C- and N-termini) badly defined
  - loops poorly defined
- spectroscopists say this reflects mobility
- problems with many models
  - difficult to work with
  - arbitrary which to select for calculations
  - averaging usually not a good idea
- Is this the absolute truth ? No.
  - number of models arbitrary
  - different methods (programs /details) give different results



# Are we finished with making coordinates ?

- structures may not be well defined
  - can they be improved ? probably
    - restrained molecular dynamics (more next semester)
  - normal MD  $E_{phys}(\vec{r}) = \text{bonds} + \text{angles} + \text{electrostatics} \dots$
  - restrained MD  $E_{total}(\vec{r}) = E_{phys}(\vec{r}) + E_{restr}(\vec{r})$
  - and...  $E_{restr} = \sum_i k_i (r_i^{struct} - r_i^{measured})^2$
  - where  $i$  refers to the distance restraint
- Mission - to minimise  $E_{total}$
- result ?
  - structures
    - agree with restraints + low energy

# What else can one do with NMR ?

- NMR sensitive to dynamics

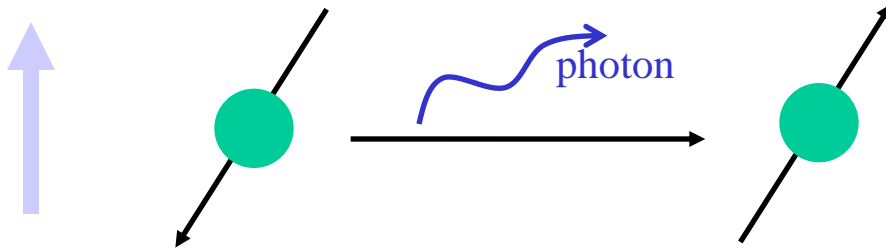
## Timescales

- for phenomena where peaks are separated by Hz
  - timescale are Hz
- fast chemistry (small molecules drift in and out)
  - completely averaged
- very special to NMR
  - relaxation and dynamics...

What makes a nucleus relax ?



# What makes a nucleus relax ?

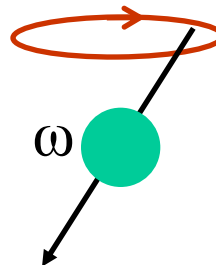


Is this really spontaneous ?

- no (think of metastable state)

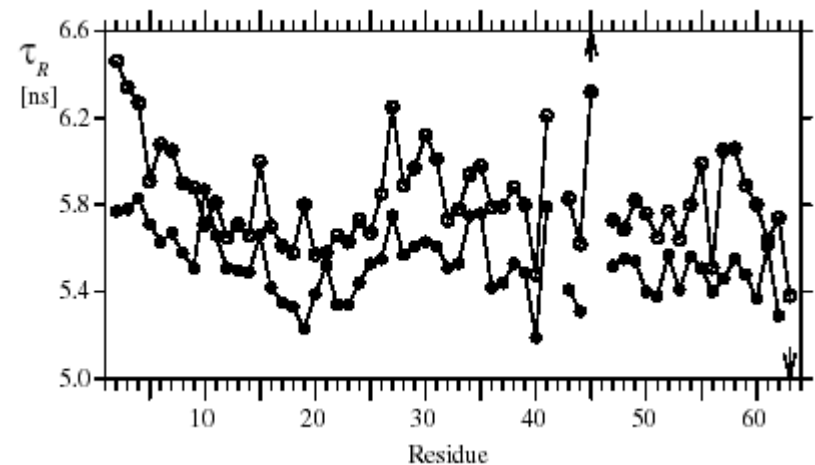
What will make it relax ?

- movement
  - overall
  - local
- Certain frequencies most important
- low frequency
  - $\omega$
  - $2\omega$



# NMR Relaxation

- different phenomena
  - NOE,  $T_1$ ,  $T_2$ , ...
  - different sensitivities to low frequency,  $\omega$ ,  $2\omega$ 
    - plus, we have  $\omega$   $^{13}\text{C}$ ,  $^{15}\text{N}$ ,  $^1\text{H}$
- different sites in molecule have different motions
- define  $\tau_c$  characteristic time of motion
- example, 64 residue protein
  - overall  $\tau_c$  5 ns
  - individual residues..
- do we see this in PDB files ?
  - no



# NMR last words (almost)

- NMR good for
  - dynamics
  - deuterium exchange
  - screening / binding / ligands
- timescales

|           |   |  |
|-----------|---|--|
| very slow | separate peaks                                | very different conformations                           |
| <hr/>     |   |  |
|           | NOE disappears<br>poorly determined structure |  |
| <hr/>     |   |  |
|           | broad peaks                                   | solvent exchange<br>sidechains turning                 |
| <hr/>     |   |  |
| fast      | averaged sharp peaks                          | fast side chain rotation (methyls)<br>ligands on / off |

# Generating Structures Summary

- Information from NMR
  - is not complete
  - may be conflicting - methods must handle these problems
- Metric Matrix method
  - use distance information directly
  - convert  $^3J$  (angle information) to distances
  - add chemical information (bonds, angles)
- Variable target function
  - angles and bonds are fixed - will generate good chemical geometry
  - attempts to solve an optimisation problem with a smoothing procedure (remove local minima and gradually add them)