

Protein Struktur

- Biologen und Chemiker dürfen mit Handys spielen (leise)
- go home, go to sleep
- wake up at slide 39

Proteins - who cares ?

Most important molecules in life ? Ask the DNA / RNA people

- structural (keratin / hair)
- enzymes (catalysts)
- messengers (hormones)
- regulation (bind to other proteins, DNA, ..)
- industrial – biosensors to washing powder
- receptors
- transporters (O_2 , sugars, fats)
- anti-freeze ...

Proteins are easy

- data (protein data bank, www.rcsb.org)
 - $\approx 10^5$ files
- literature on function, interactions, structure
- software
 - viewers, molecular dynamics simulators, docking, ..
- nomenclature and rules

Proteins are not friendly

- one cannot take a sequence and predict structure/function
- data formats are full of surprises
- data contains error and mistakes

Protein Rules, Physics, Folklore

Physics / Chemistry

- protein + water = set of interacting atoms
 - can be calculated (not really)

Rules (not quantified)

- proteins unfold if you heat them (exceptions ?)
- many charged amino acids.. they are soluble
- if they are more than 300 residues, they have more than one domain,
- proteins fold to a unique structure (could you prove this ?)
 - lowest free energy structure

Protein chemistry

Chemists / biochemists

- sleep, go home
- one tiny surprise at the end of the lectures

Short version

- proteins are sets of building blocks (amino acids, residues, Reste)
- 20 types of residue
- chains of length few to 10^3 (100 or 200 typical)
- small ones ($< \approx 50$ residues) are peptides
- they fold up to nice stable structures – why ?

Longer version..

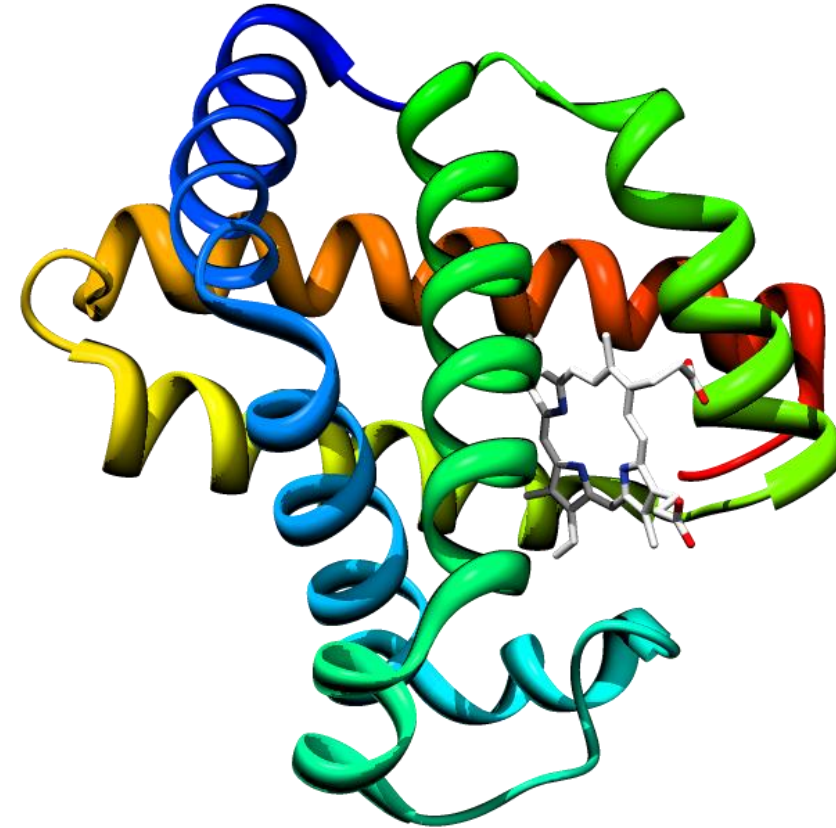
The Plan

- polymers
- different kinds of sidechain
- structure due to backbone (secondary structure)
- properties of sidechains
- representation

Sizes

$1 \text{ \AA} = 10^{-10} \text{ m}$ or 0.1 nm

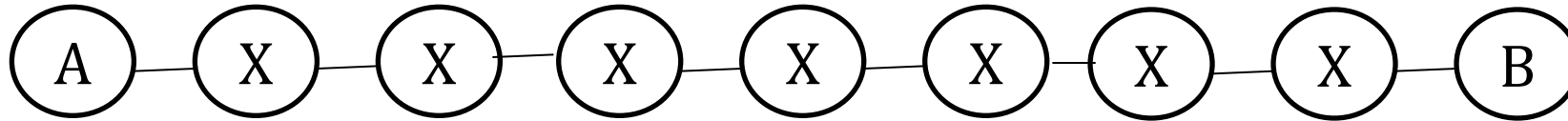
structure		size
bond	CH	1 \AA
	CC	1.5 \AA
protein radius		$10 - 10^2 \text{ \AA}$
α -helix spacing		$5 \frac{1}{2} \text{ \AA}$
C^α_i to C^α_{i+1}		3.8 \AA



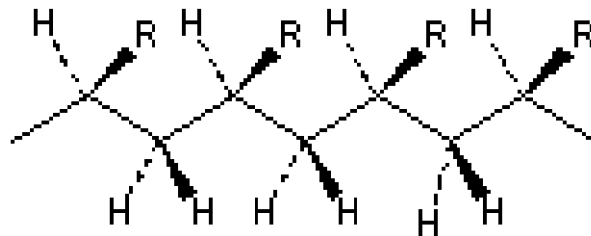
Proteins are polymers

- simple polymers 

many times gives



example

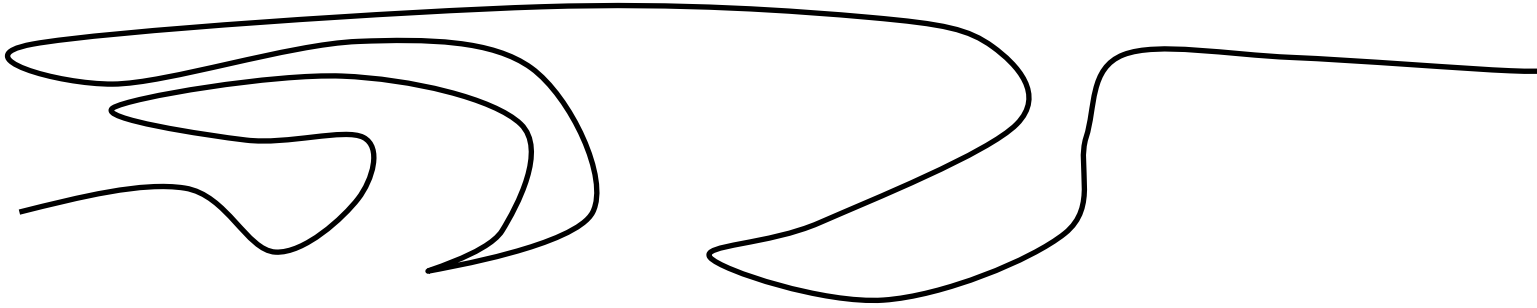


what kind of polymer would this give ?

Do you know what R is ?

Why are proteins interesting polymers ?

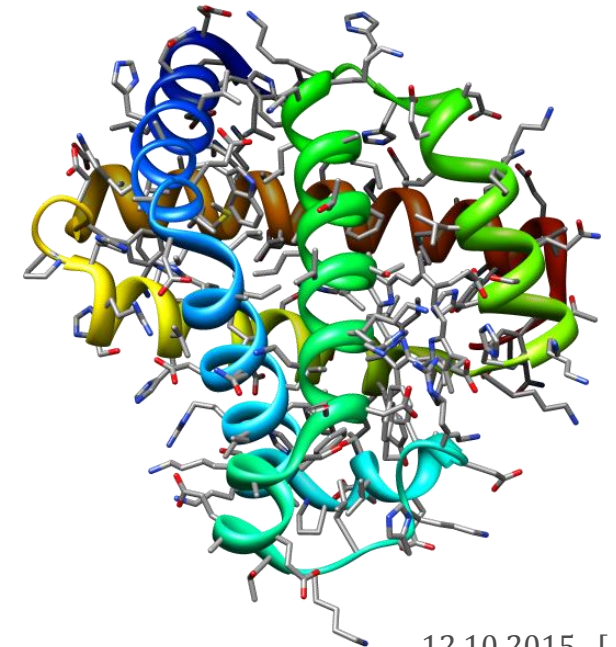
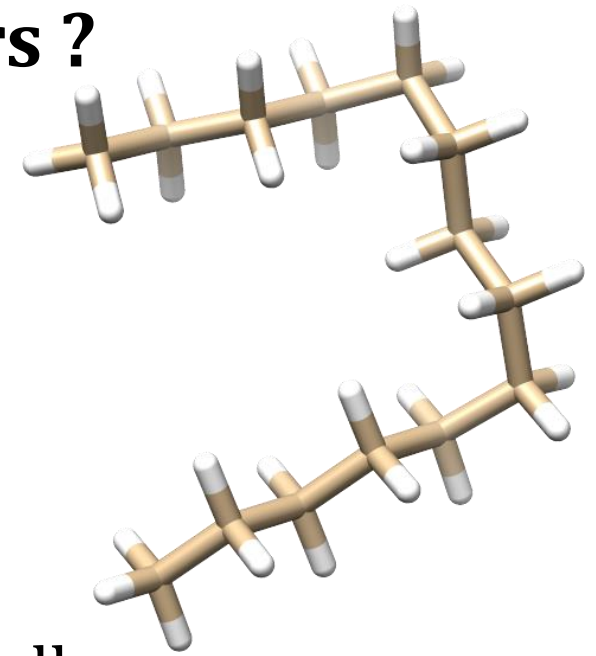
Boring polymer gives irregular structures



Each part of polymer wants to interact with all other parts equally

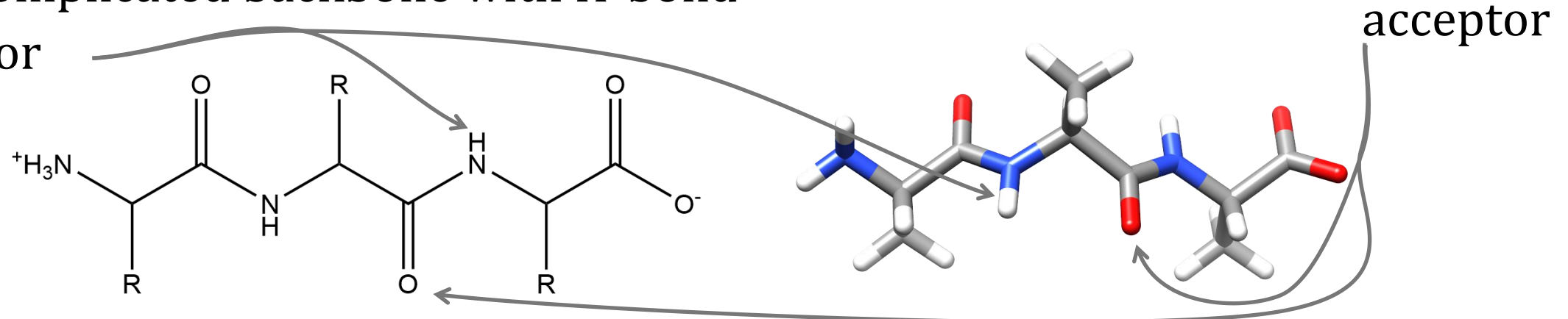
- no structural preferences
- plastic bags, Haushaltsfolie
- no regular structures

Properties that make proteins different from plastics ..



Giving proteins character 1

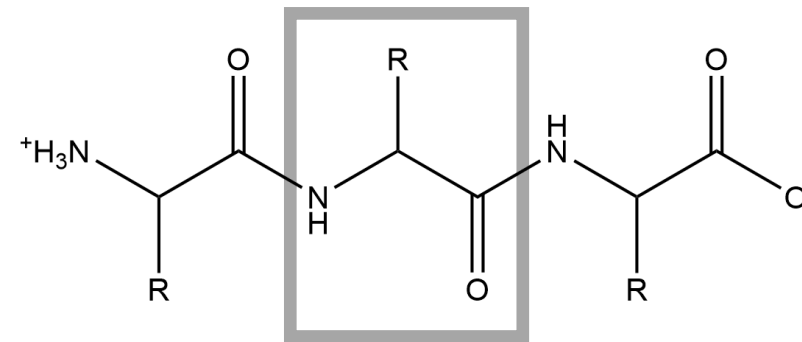
More complicated backbone with H-bond



- basis of standard regular structures in proteins (secondary structure)
- repeating polymer unit:

If this was all there was

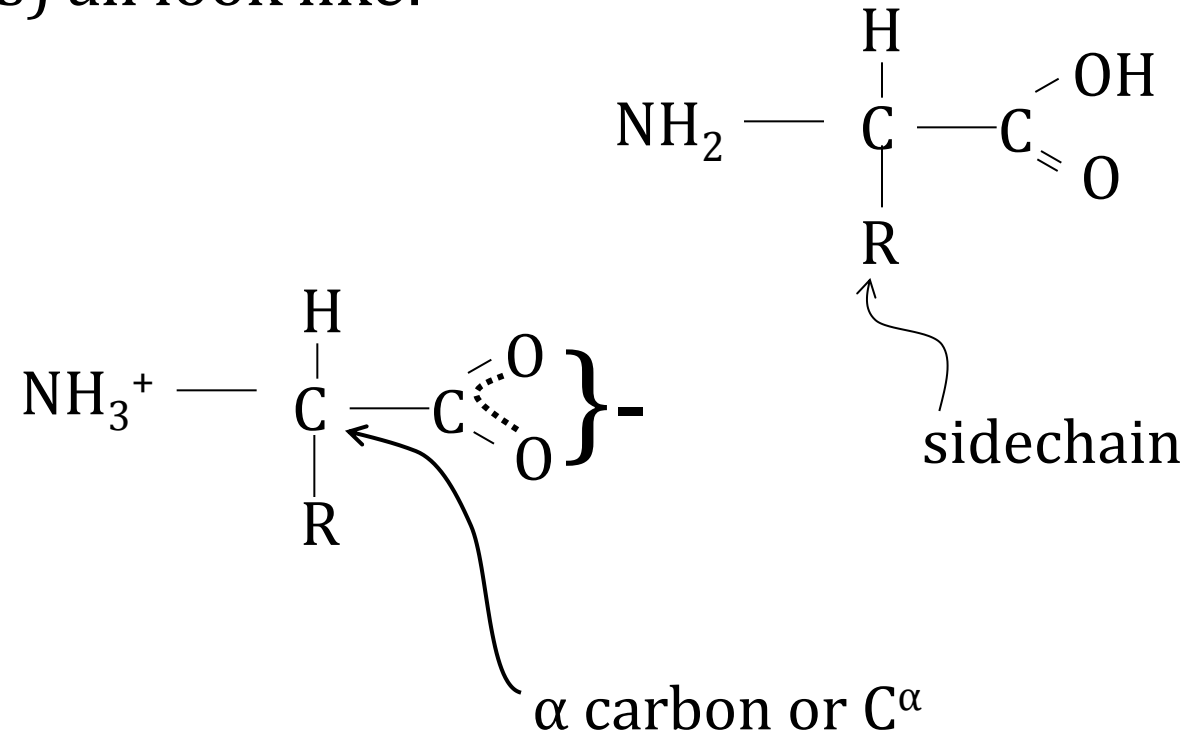
- all proteins would be the same



protein chemistry

amino acids (monomers) all look like:

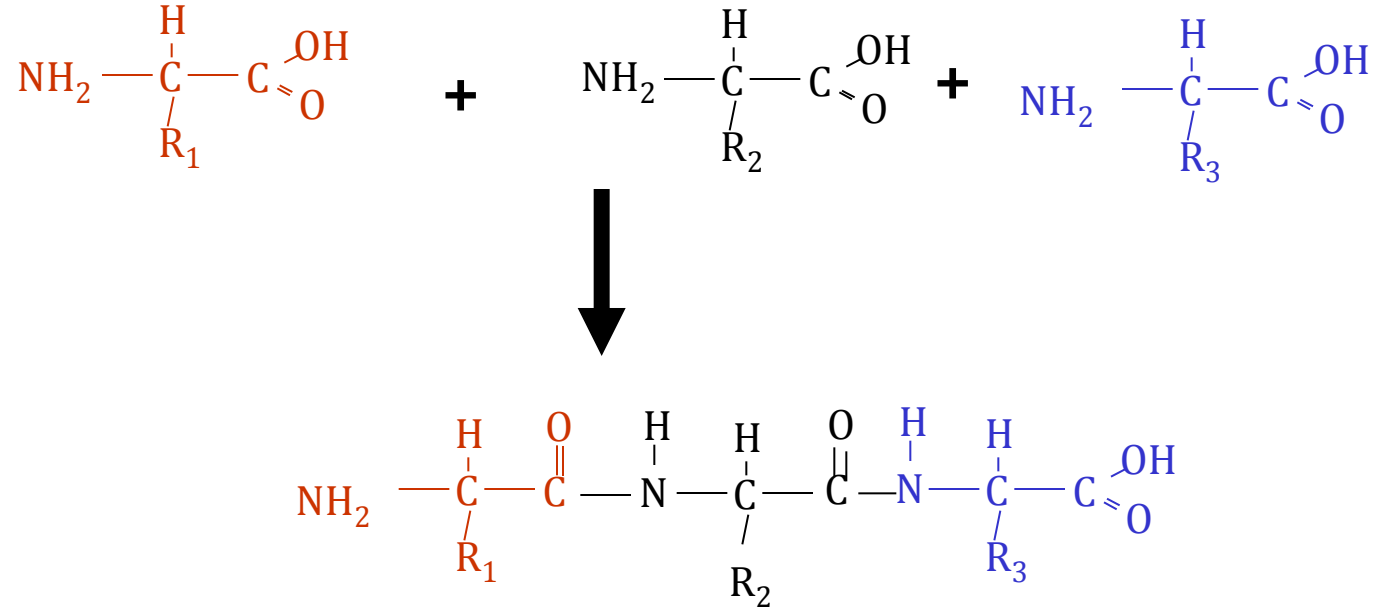
maybe



How can we construct specific structures ?

- different kinds of "R" groups

Putting monomers together



- protein synthesis story (biochemistry lectures)
- peptides and proteins
 - < 30 or 40 residues = peptide
 - > 30 or 40 residues = protein

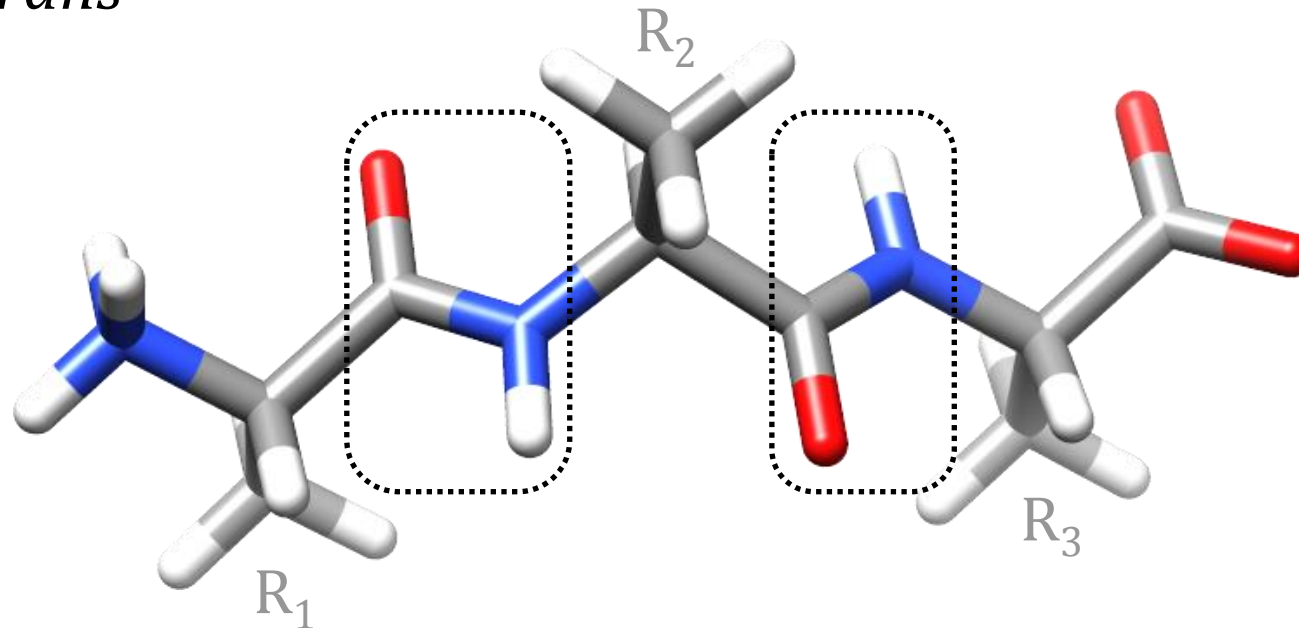
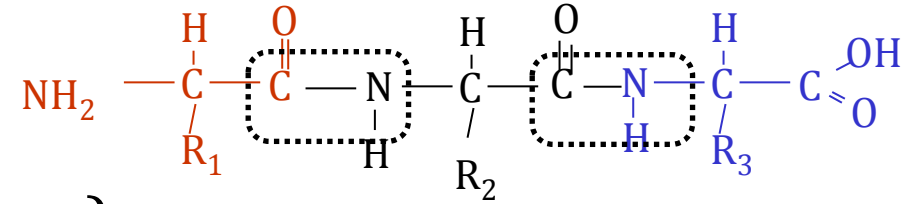
Backbone peptide bonds

How many backbone angles ?

- 3 (ϕ , ψ , ω)

Peptide bond ω is planar

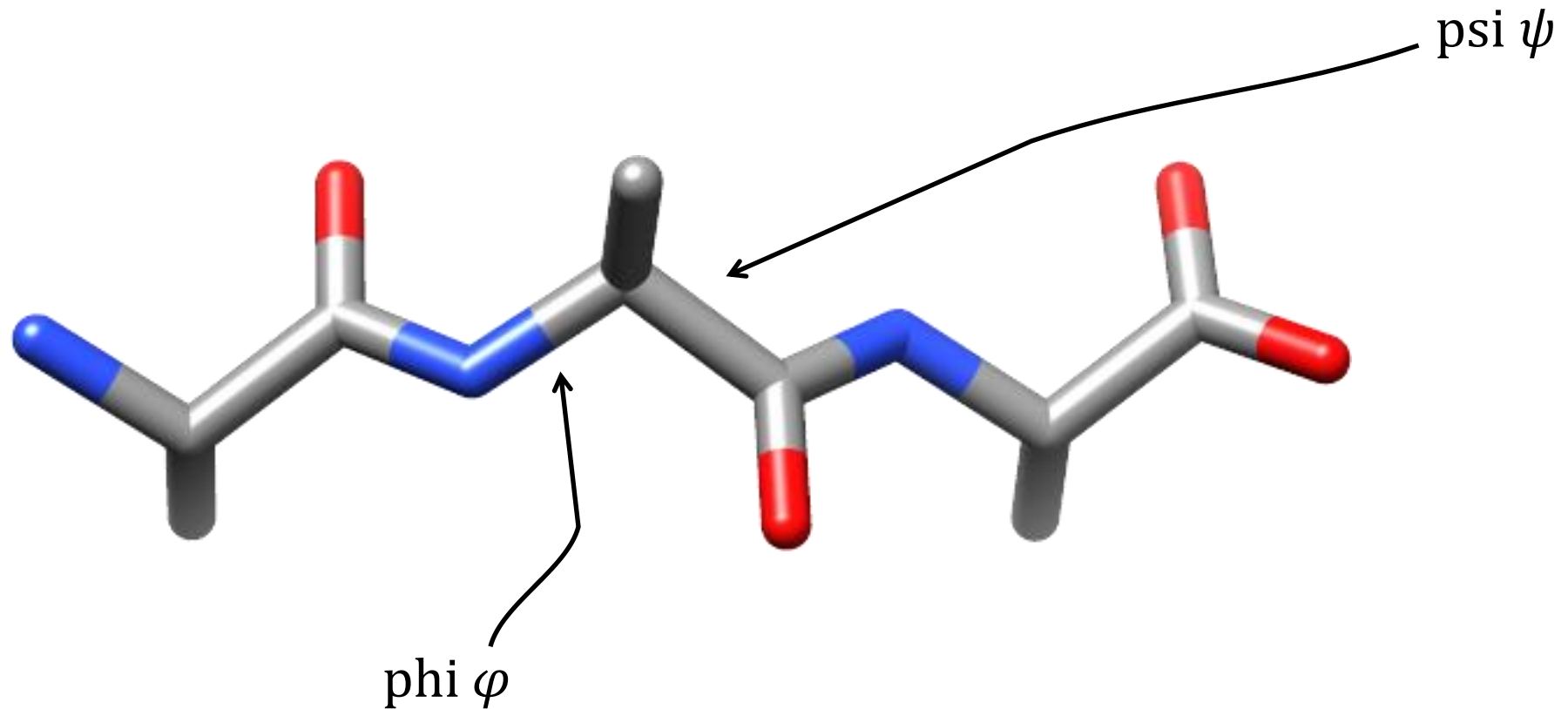
- partial double bond character (resonance forms)
- shorter than other C-N
- nearly always *trans*



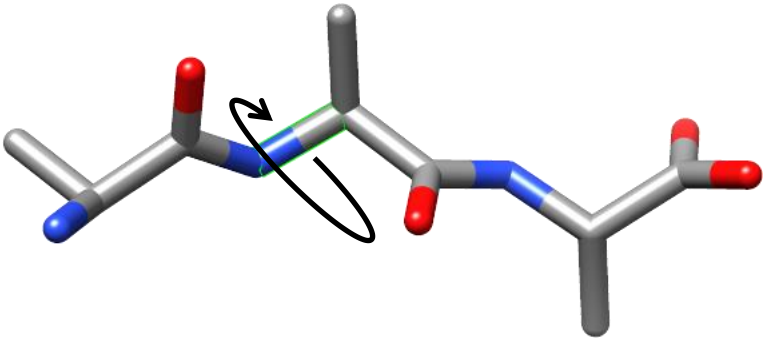
Note:
usually we
do not draw
H atoms

Backbone rotatable angles

Two rotatable angles ϕ , ψ

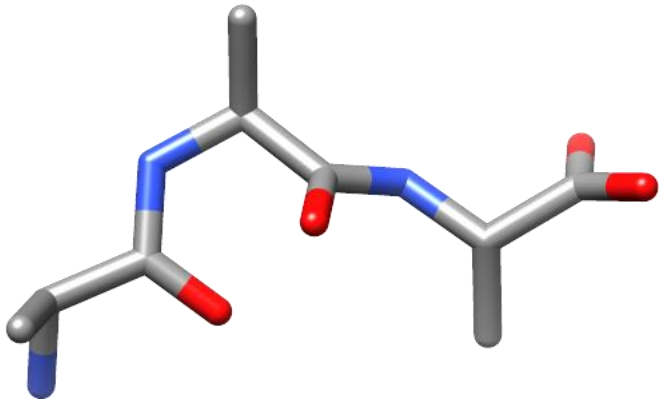
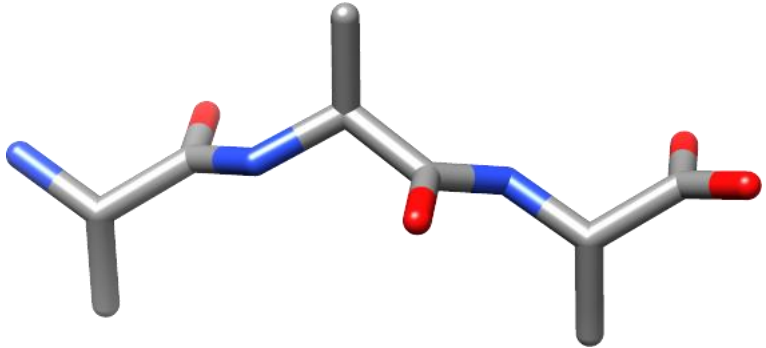


some ϕ rotations



can we rotate freely ?

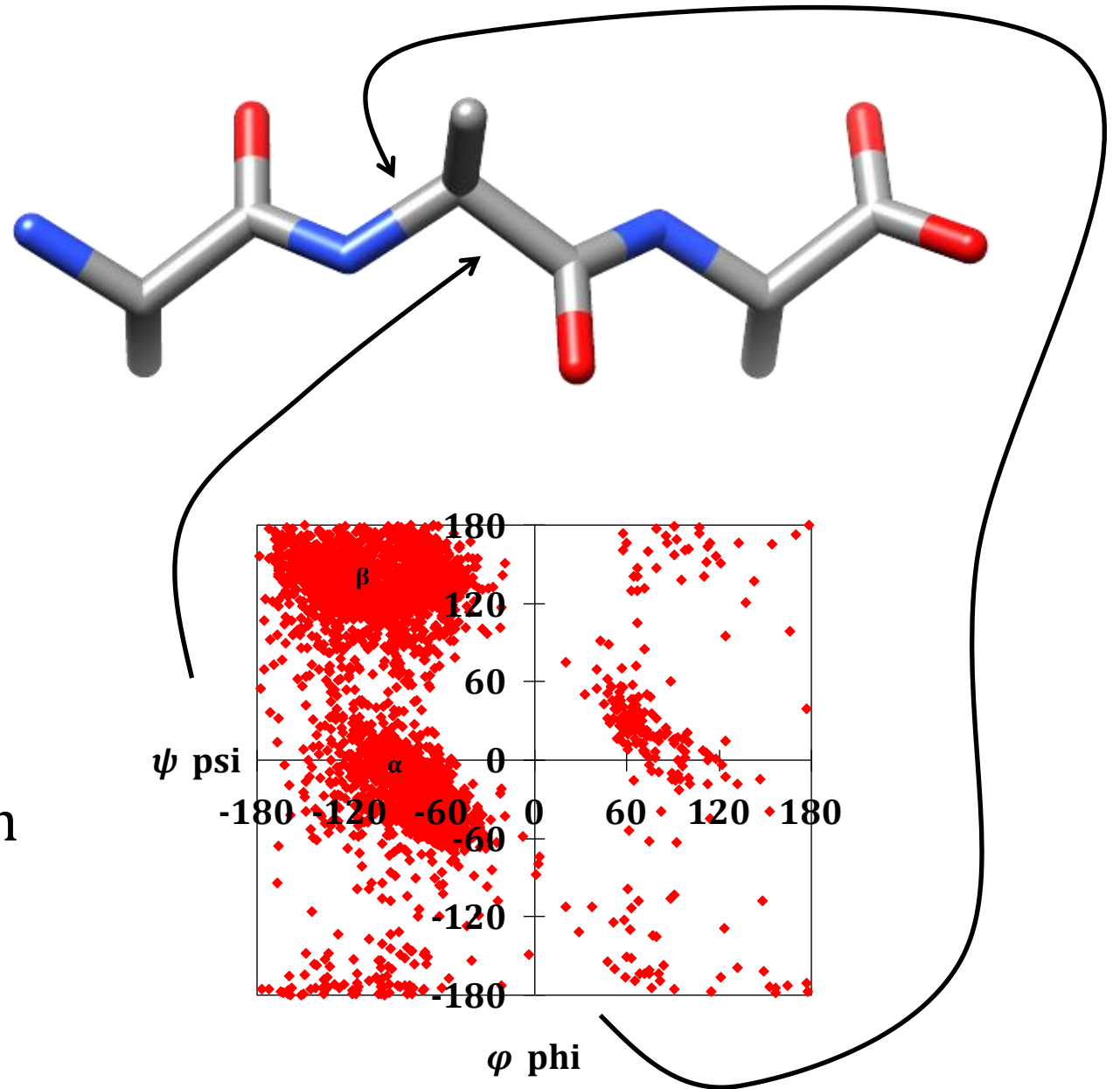
- no... steric hindrance
- look at bottom – two unhappy O atoms



ramachandran plot

can we rotate freely ?

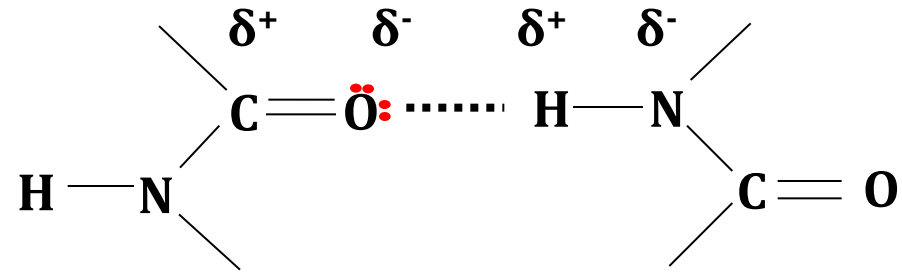
- no... steric hindrance



Ramachandran plot
will reappear very often

Backbone H bonds

- oxygen is slightly negative
- NH bond is polar



H-bonds

- can be near or far in sequence
- fairly stable at room temperature

Secondary structure

Regular structures using information so far

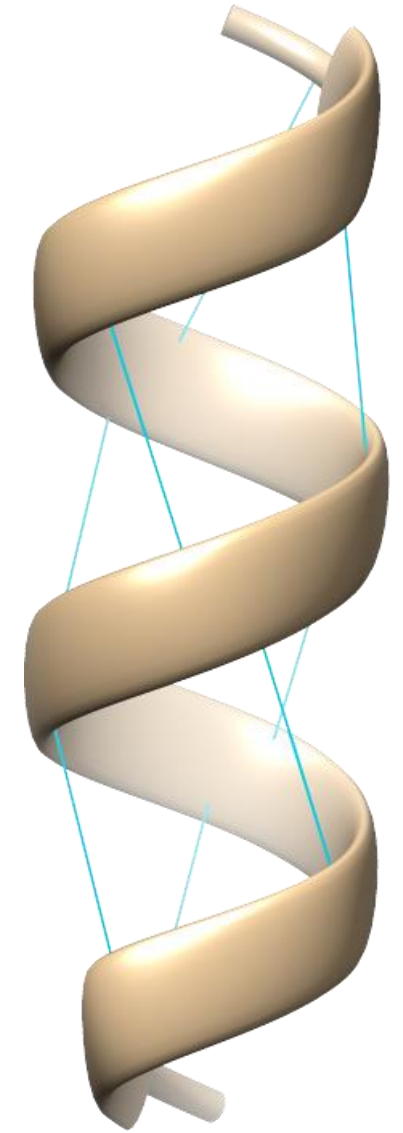
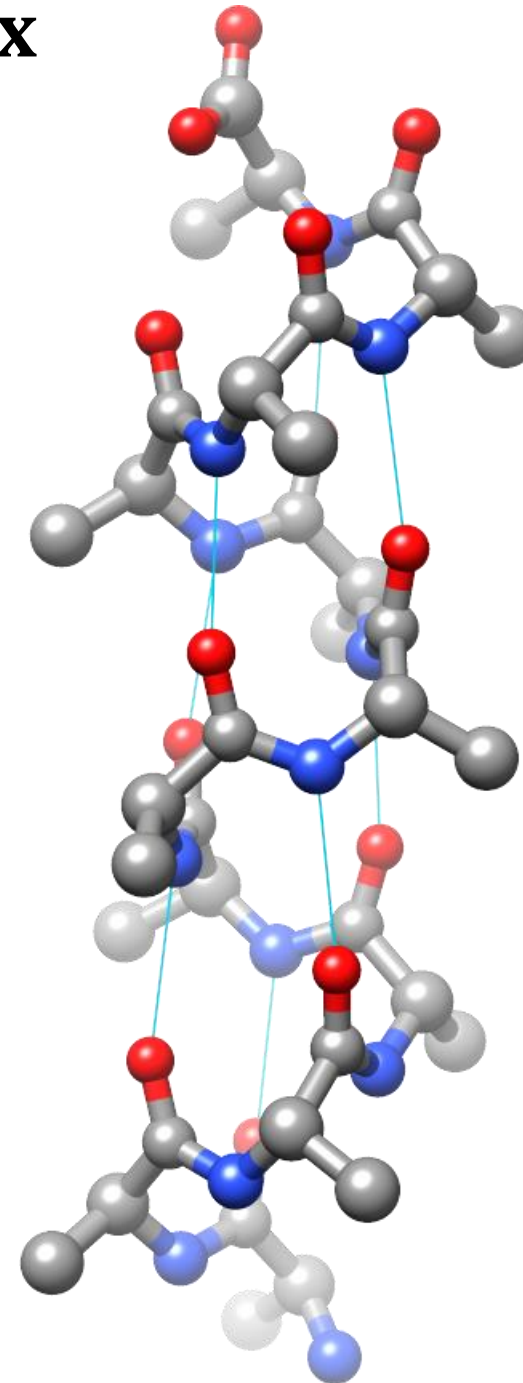
- rotate phi (φ), psi (ψ) angles so as to
 - form H-bonds where possible
 - do not force side chains to hit each other (steric clash)

Two common structures

- α -helix
- β -strand / sheet

α helix

- each CO of residue i H-bonded to N of $i+4$
- 3.6 residues per turn
- 2 H-bonds per residue
- side chains well separated



β -sheet

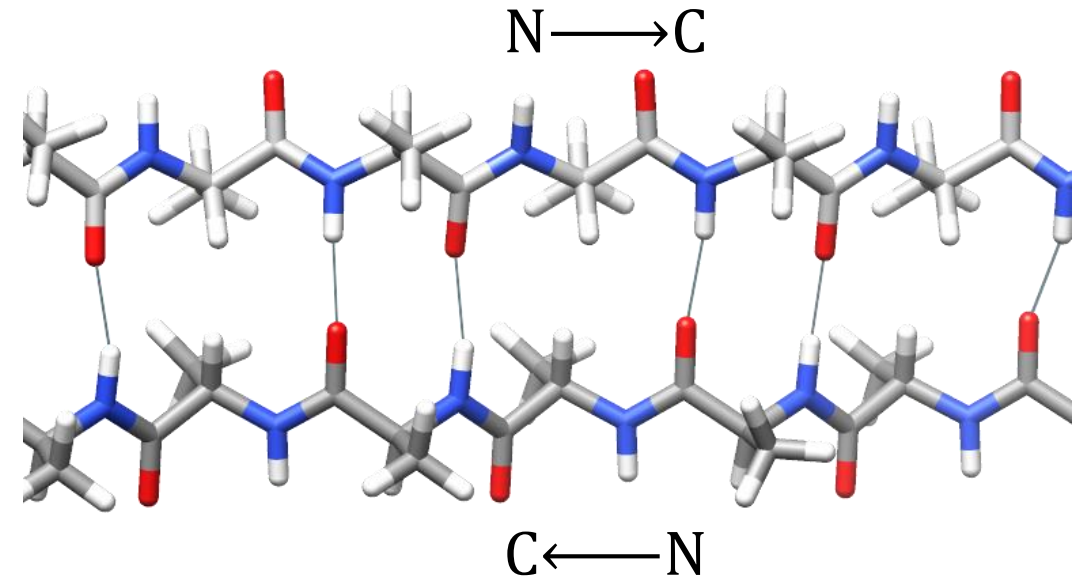
β -strand

- stretch out backbone and make NH and CO groups point out

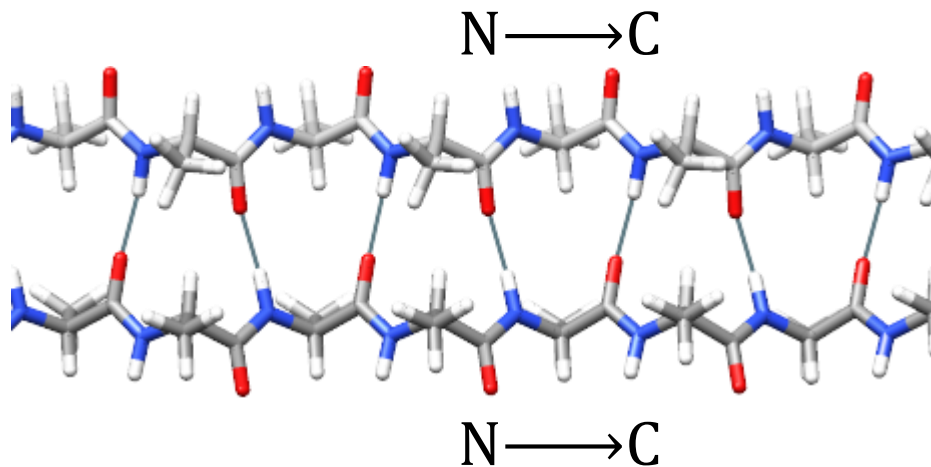
β -sheet

- join these strands together with H-bonds (2 H-bonds/residue)

anti-parallel



or parallel



After α -helix and β -sheet

Do helices and sheets explain everything ? No

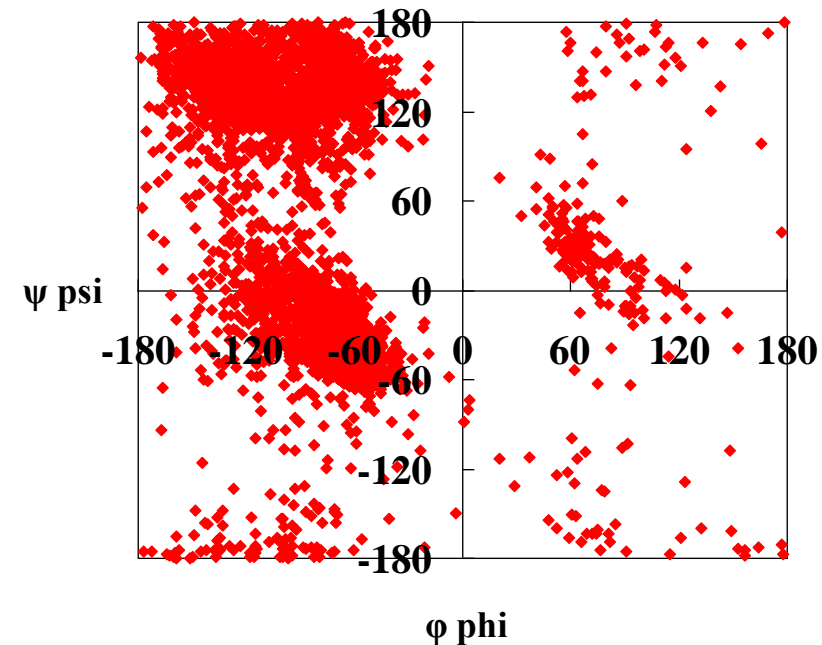
- there is flexibility in the angles (look at plot)
 - geometry is not perfectly defined
- there are local deviations and exceptions

Other common structures

- tighter helices
- some turns

Other structure

- coil, random, not named



What determines secondary structure ?

So far

- secondary structure pattern of H-bonding

Almost all residues have H-bond acceptor and donor

- almost all could form α -helix or β -sheet

Difference ?

- sequence of side-chains – overall folding

Why else are sidechains important

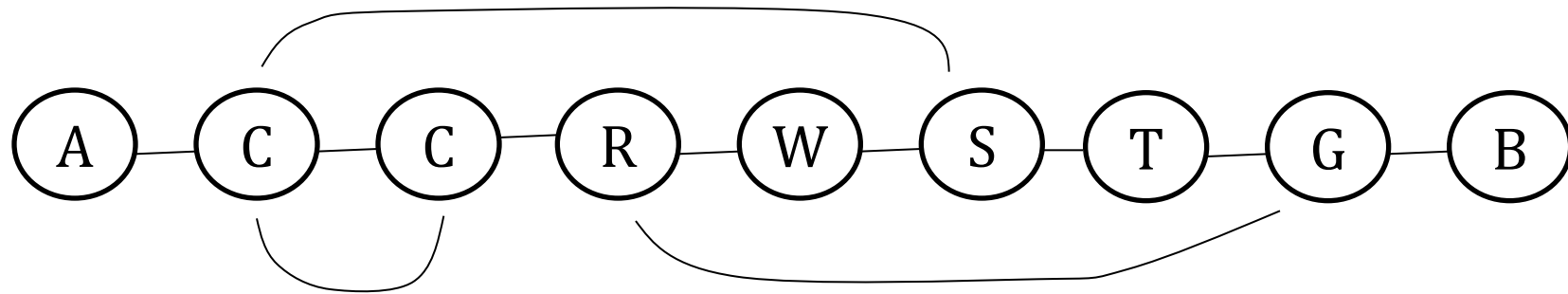
- chemistry of proteins (interactions, catalysis)

Fundamental dogma

- the sequence of sidechains determines the protein shape

side chain possibilities

- big / small
- charged +, charged -, polar
- hydrophobic (not water soluble), polar
- interactions between sites...



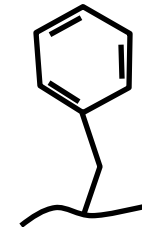
Side chain properties

properties

- big / small
- neutral / polar / charged
- special (...)

example

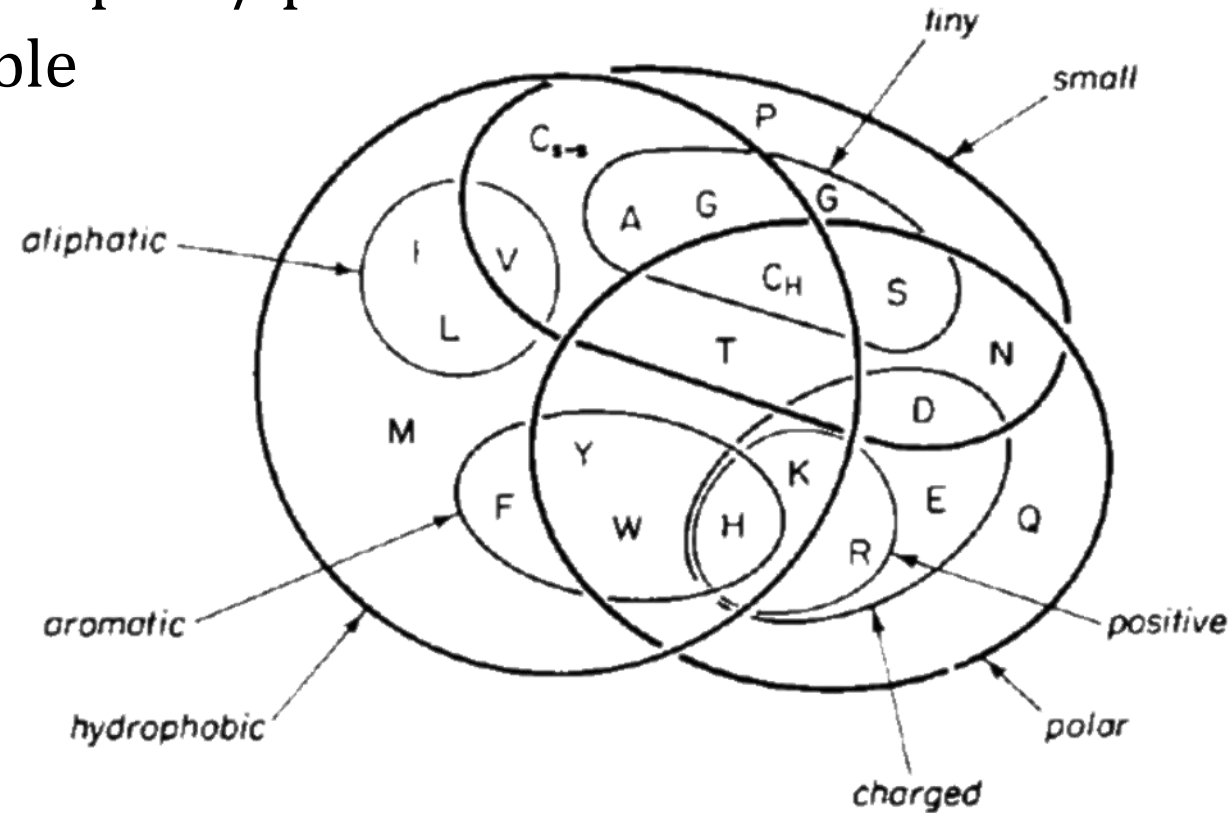
- phenylalanine side chain looks like benzene (benzin)
 - very insoluble
 - benzene would rather interact with benzene than water
 - what if you have phe-phe-phe... poly-phe ?
 - does not happen in nature (can be made)
 - would be insoluble
 - not like a real peptide
 - phe is a constituent of real proteins – has a role



Properties are not clear cut

You can be big / small, hydrophobic / polar

- combinations are possible

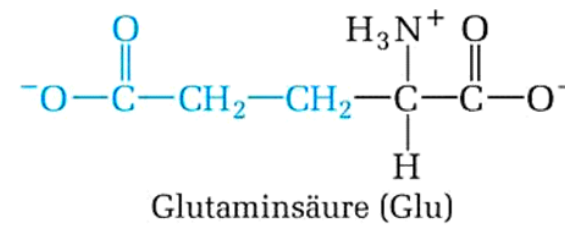
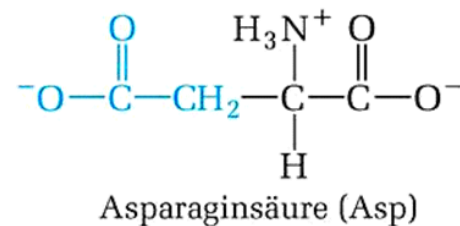
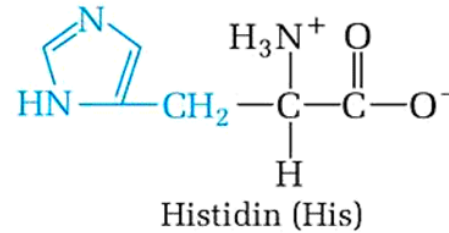
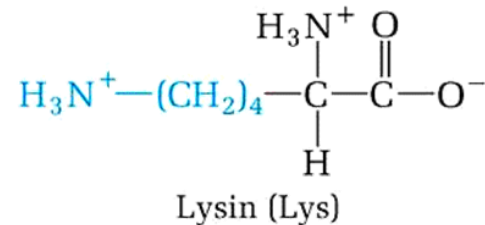
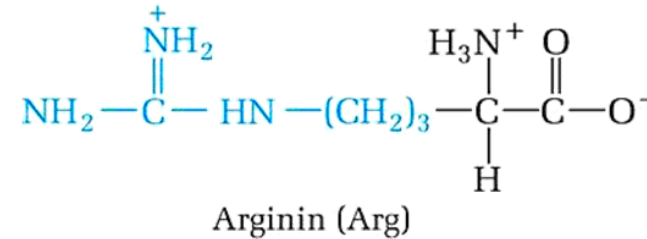
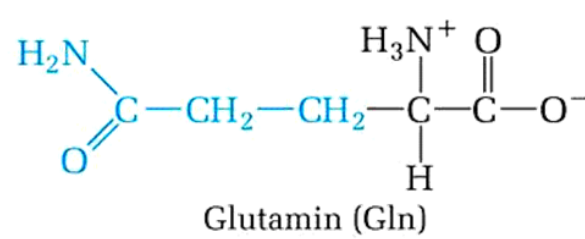
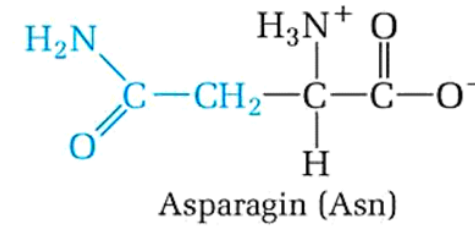
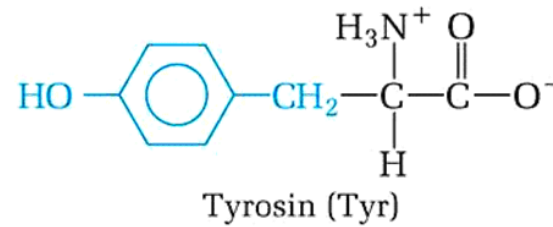
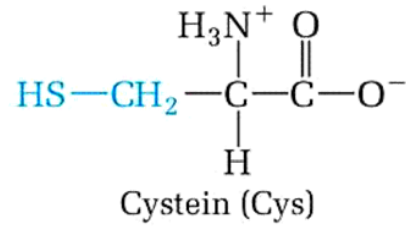
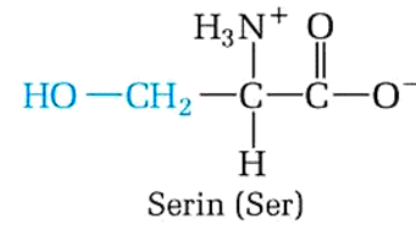
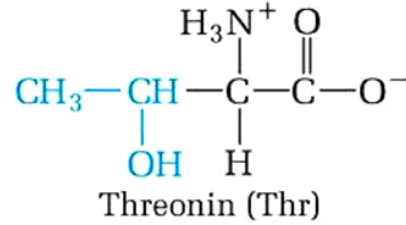
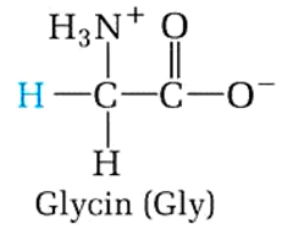


Do not memorise this figure

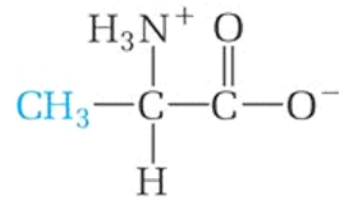
Sidechain interactions

- ionic (if the sidechains have charge)
- hydrophobic (insoluble sidechains)
- H-bonds (some donors and acceptors)
- repulsive

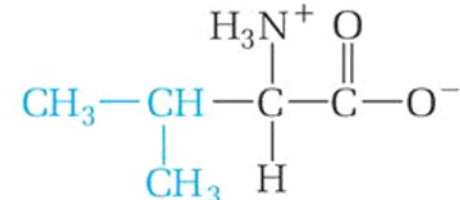
Summary of amino acids (first dozen)



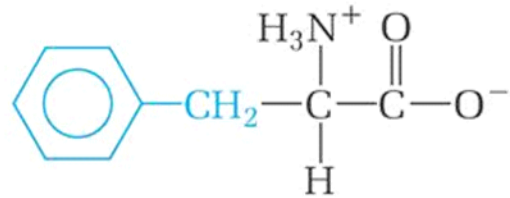
summary of amino acids (part 2)



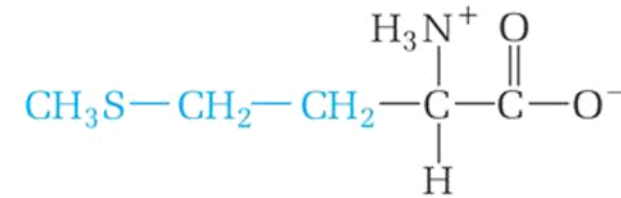
Alanin (Ala)



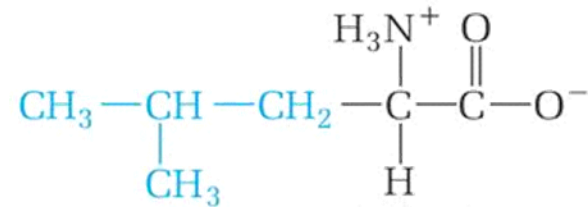
Valin (Val)



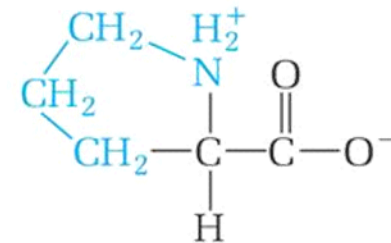
Phenylalanin (Phe)



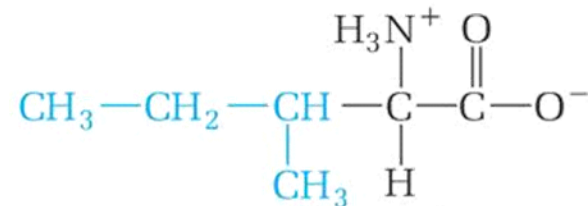
Methionin (Met)



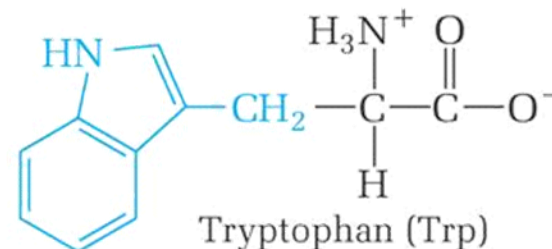
Leucin (Leu)



Prolin (Pro)



Isoleucin (Ile)

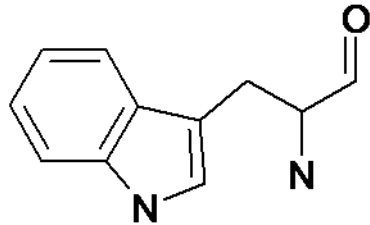


Tryptophan (Trp)

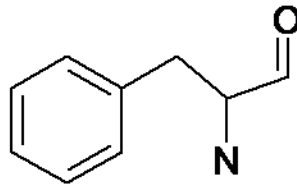
Amino Acids by property

aromatic

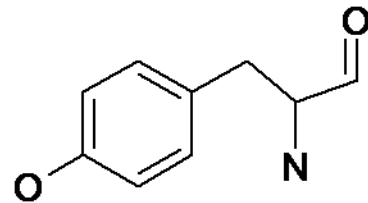
tryptophan



phenylalanine

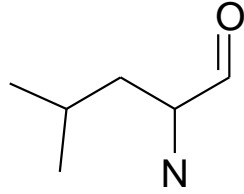


tyrosine

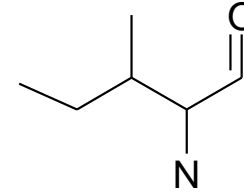


rather hydrophobic

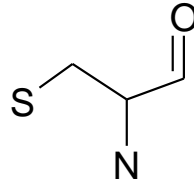
leucine



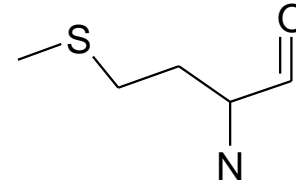
isoleucine



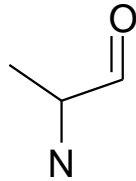
cysteine



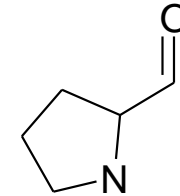
methionine



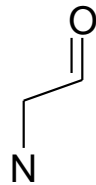
alanine



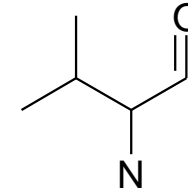
proline



glycine

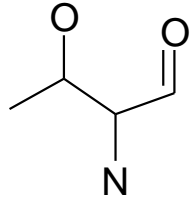


valine

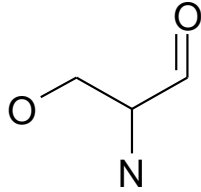


Polar

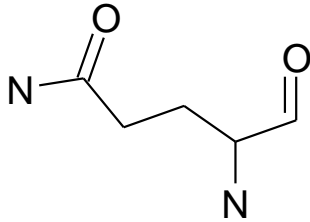
threonine



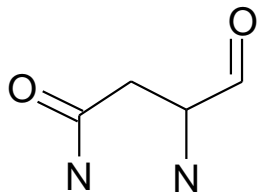
serine



glutamine

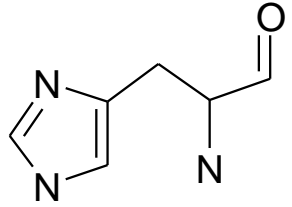


asparagine

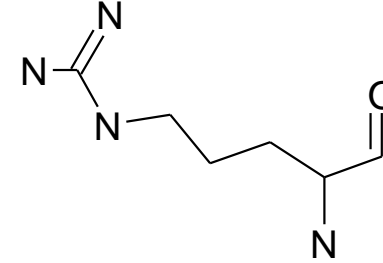


charged

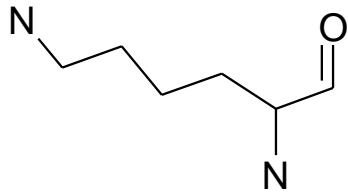
histidine



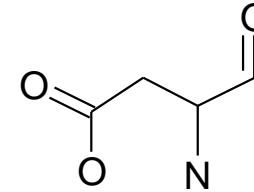
arginine



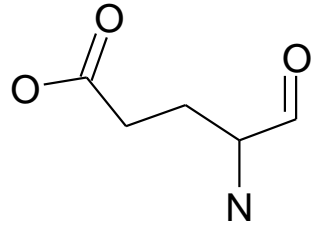
lysine



aspartate



glutamate



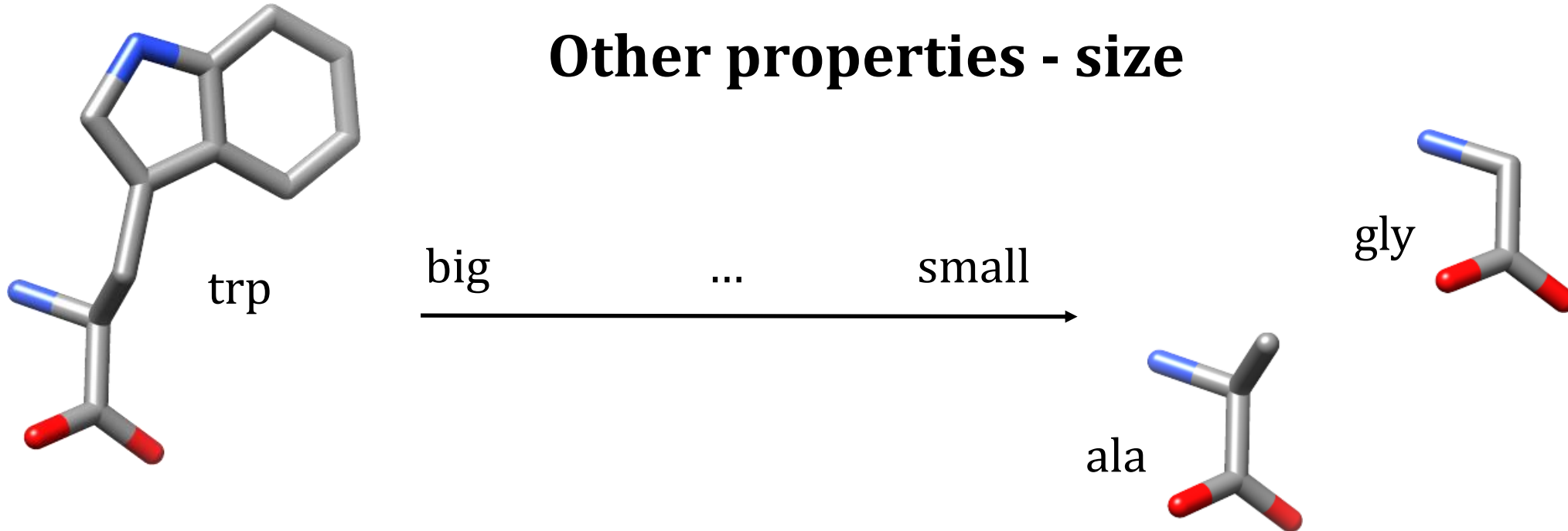
- Muss ich alle Strukturen für die Klausur wissen ?

Hydrophobicity – how serious ?

Very serious, but simplified

- the lists above are
 - pH dependent
 - difficult to measure experimentally (some aspects)
- Is there a single definition for hydrophobicity ?

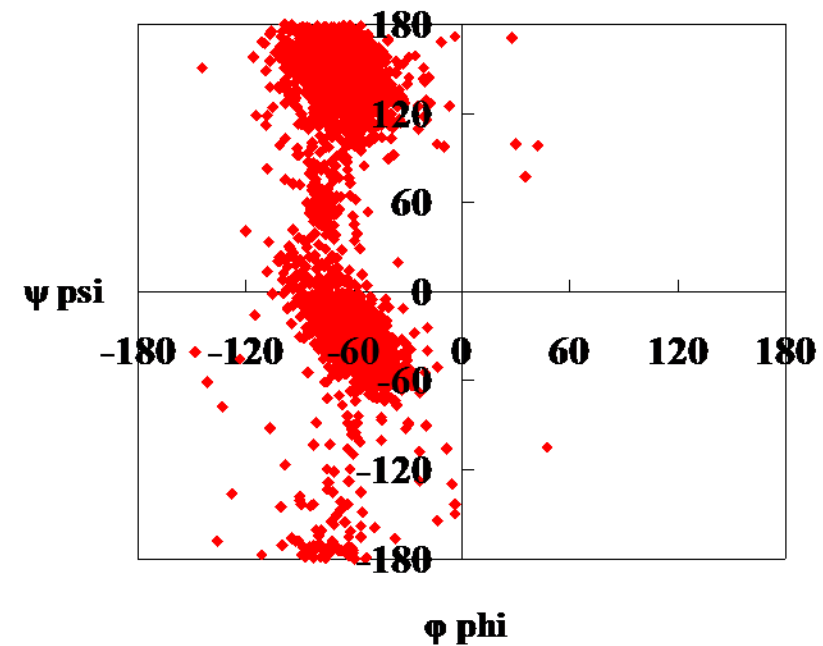
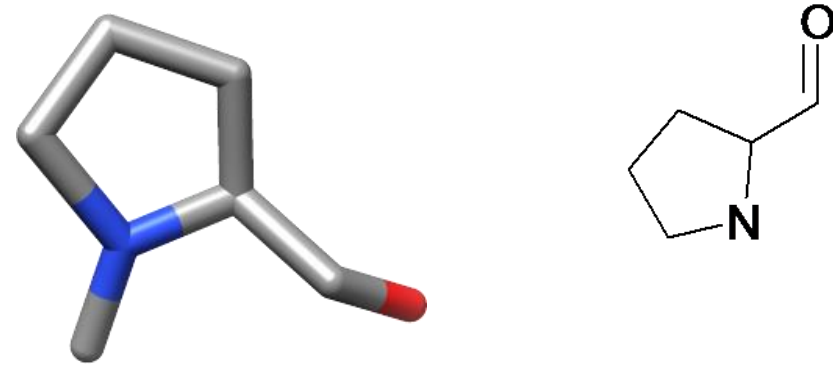
Other properties - size



Other properties – chemistry / geometry

Proline

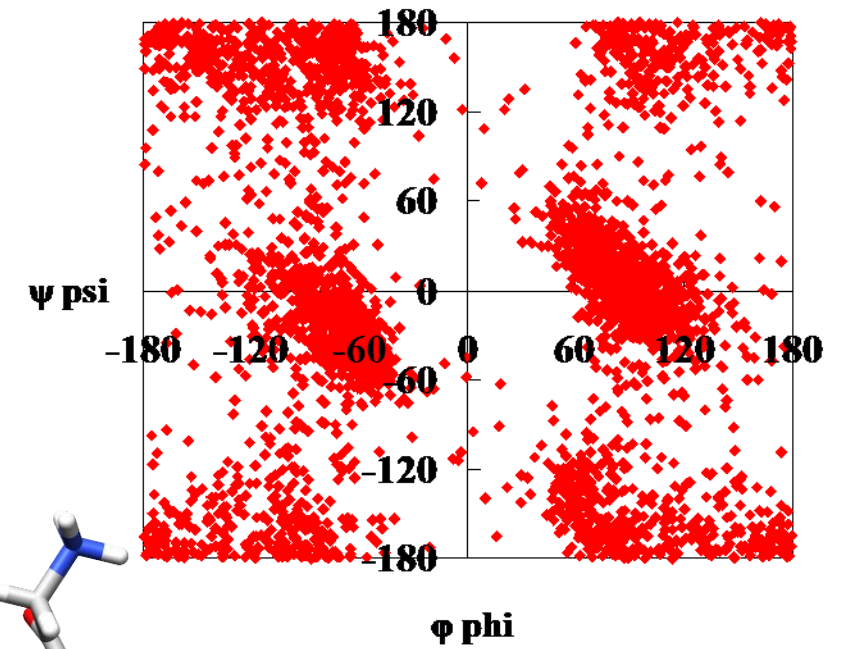
- only one rotatable angle !
- peptide bond sometimes *cis*
- pro ramachandran plot



gly and cys

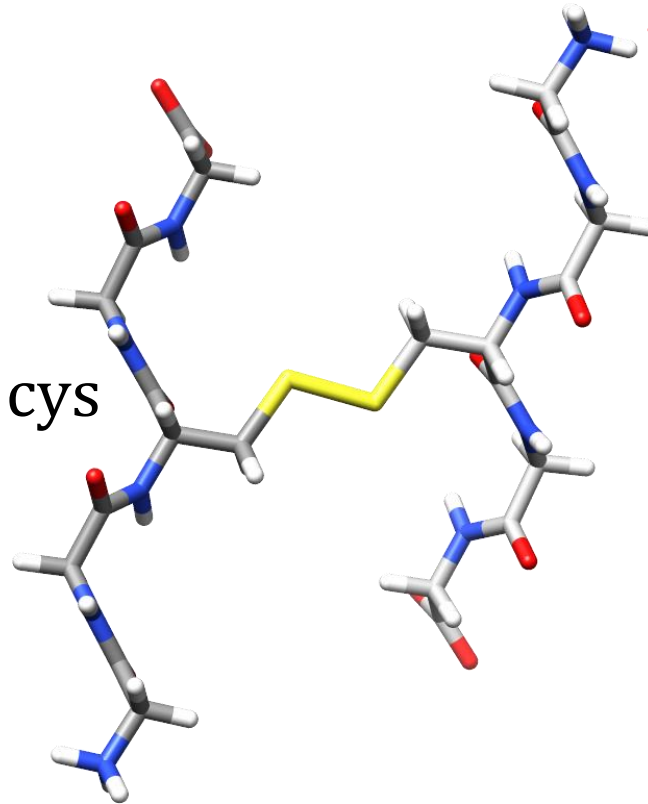
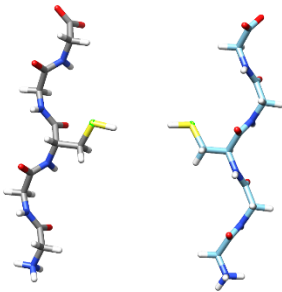
glycine

- no side chain
- can visit forbidden parts of phi-psi map



cysteine

- forms covalent links with other cys



Summary so far

- proteins are heteropolymers
- backbone forms α -helices and β -strands (and more)
 - not sequence specific
- side-chains determine the
 - pattern of secondary structure
 - overall protein shape
- special amino acids
 - cys (forms disulfide bridges)
 - gly (can visit "forbidden" regions of ramachandran plot)
 - pro (no H-bond donor)
- how many sequences can one have ? 20^n

Nomenclature

Some rules are unavoidable

Alanine	Ala	A
Cysteine	Cys	C
Aspartic acid	Asp	D
Glutamic acid	Glu	E
Phenylalanine	Phe	F
Glycine	Gly	G
Histidine	His	H
Isoleucine	Ile	I
Lysine	Lys	K
Leucine	Leu	L
Methionine	Met	M
Asparagine	Asn	N
Proline	Pro	P
Glutamine	Gln	Q
Arginine	Arg	R
Serine	Ser	S
Threonine	Thr	T
Valine	Val	V
Tryptophan	Trp	W
Tyrosine	Tyr	Y

Always write from N to C terminal
(convention)

Definitions, primary, secondary ...

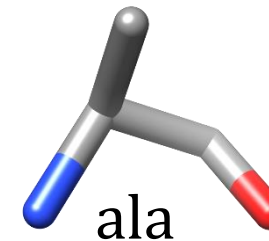
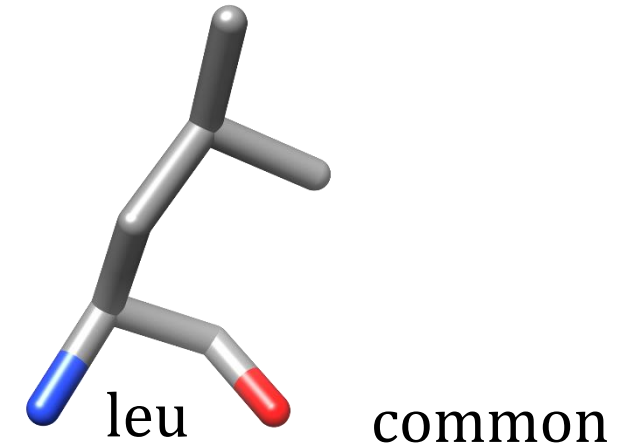
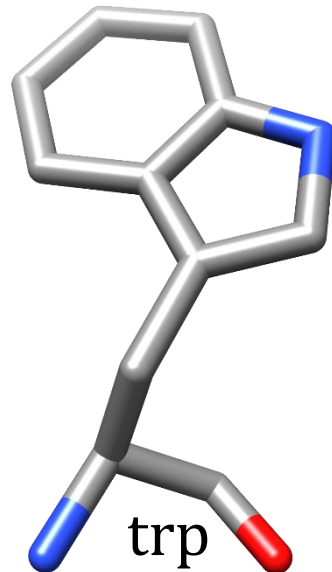
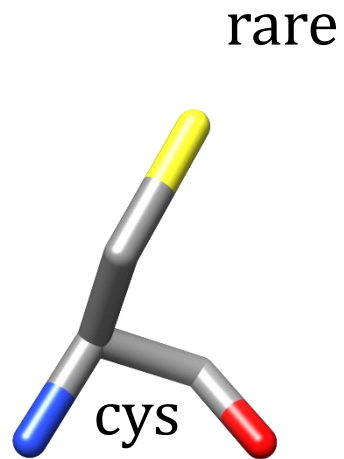
More definitions

- primary structure
 - sequence of amino acids
 - ACDF (ala cys asp phe...)
- secondary structure
 - α -helix, β -sheet (+ few more)
 - structure defined by local backbone
- tertiary structure
 - how these units fold together
 - coordinates of a protein

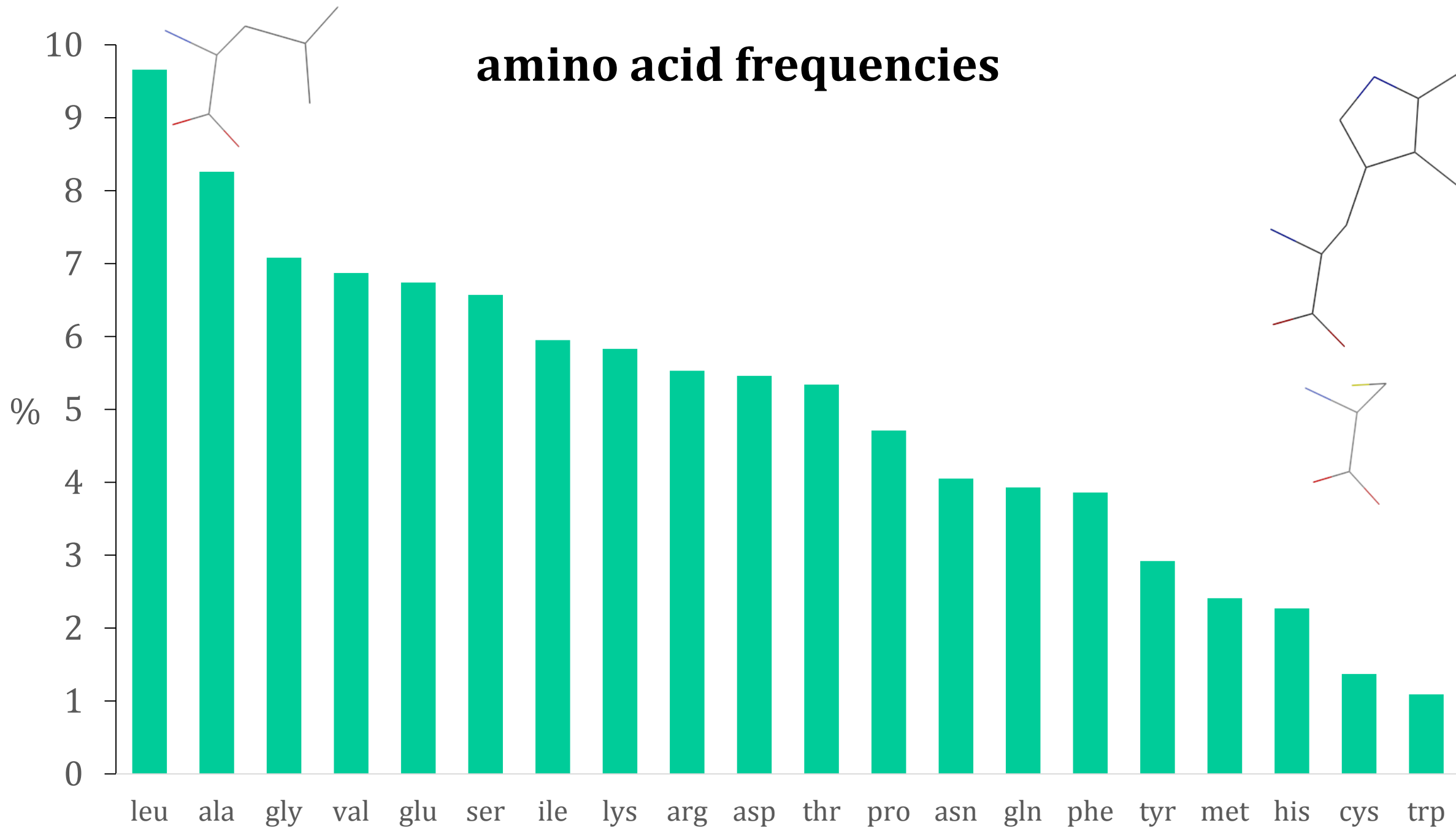
distributions of residue types

Surprise coming

- 20 amino acid types – are they all equally common ?
- Are you made of $1/20 = 5\%$ of ala, leu, cys, ... ?



amino acid frequencies



What would Darwin say ?

Why ?

- so much ala, leu
- so little trp, cys, his, met

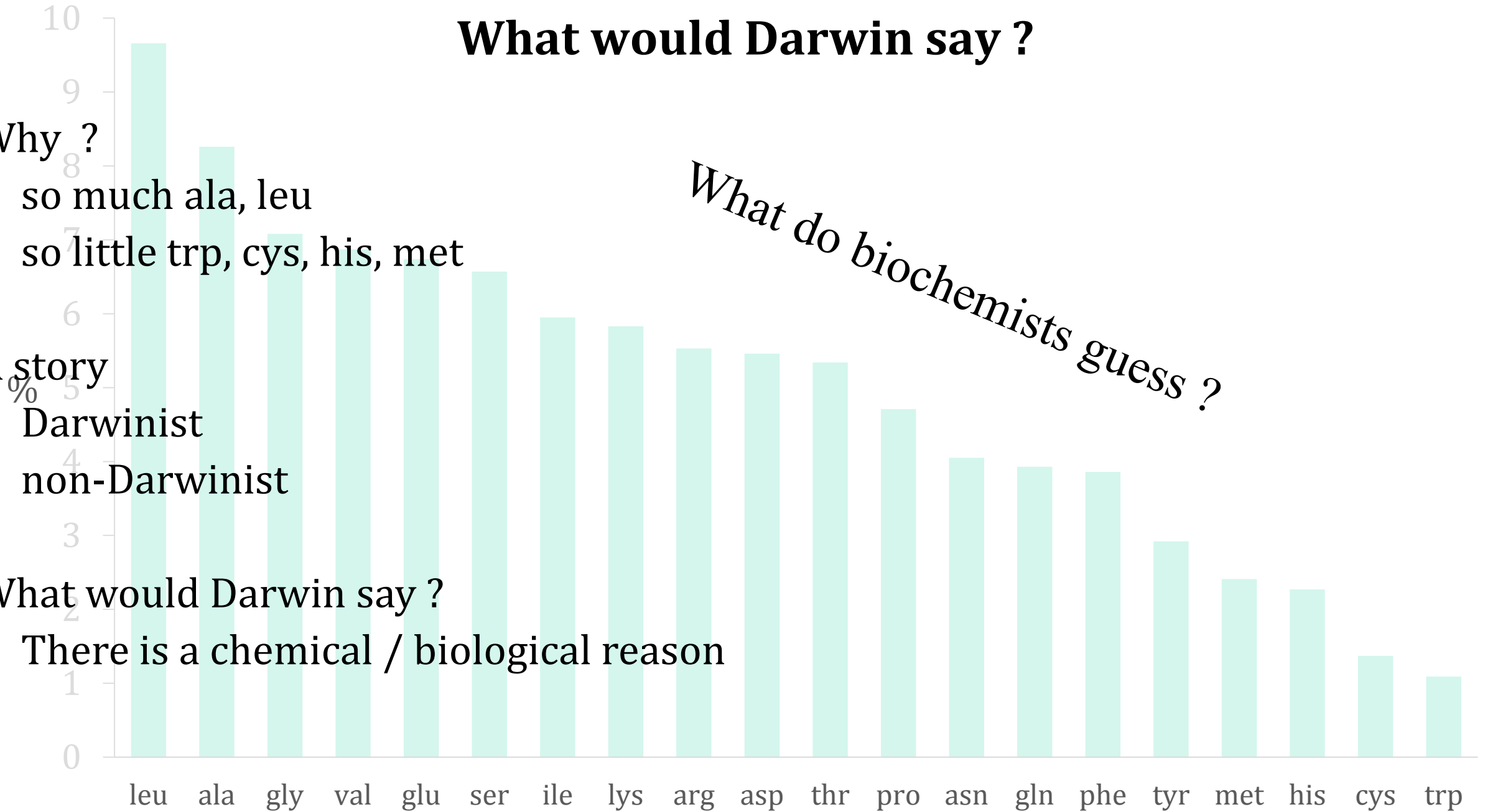
A story

- Darwinist
- non-Darwinist

What would Darwin say ?

- There is a chemical / biological reason

What do biochemists guess ?



Think Darwinist

Empirical fact

- trp, cys, met are rare in proteins

Consequence

- too much trp is bad for you / expensive / dangerous

Possibilities

- metabolic cost issues
 - does it cost energy / nutrients to make trp ? cys with its sulfur ?
- protein structure – lots of chemical differences between amino acid types
 - if you put lots of trp / cys / met in a protein
 - does it not fold ? Does it become unstable ?
- if free trp toxic ?

Common amino acids

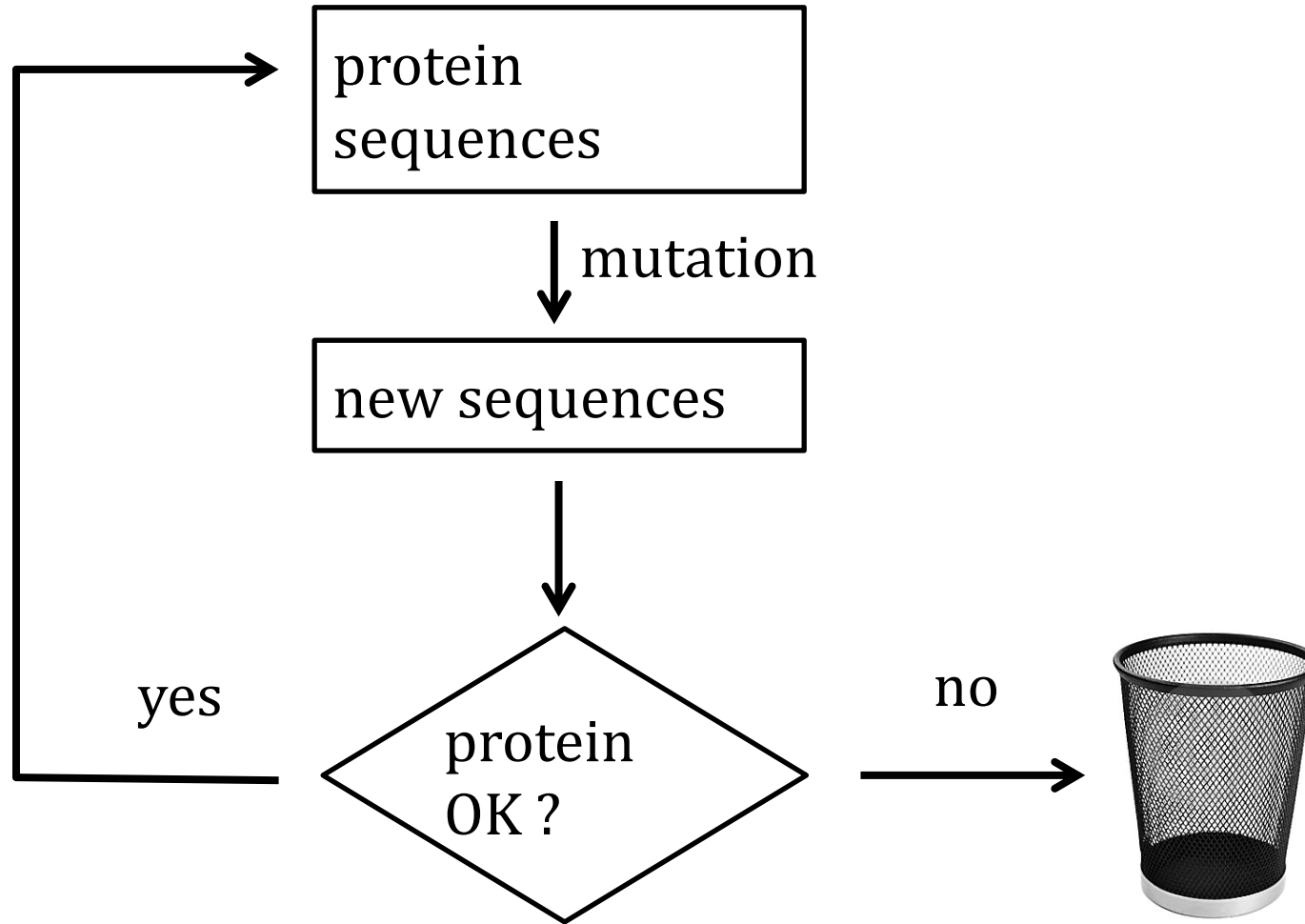
Leu and ala

- cheap to synthesise ?
- do you get them as by-products from other biochemistry ?
- what is their advantage in protein structure ?
 - stability ? rigidity ? flexibility ?

Forget Darwin – think neutral evolution

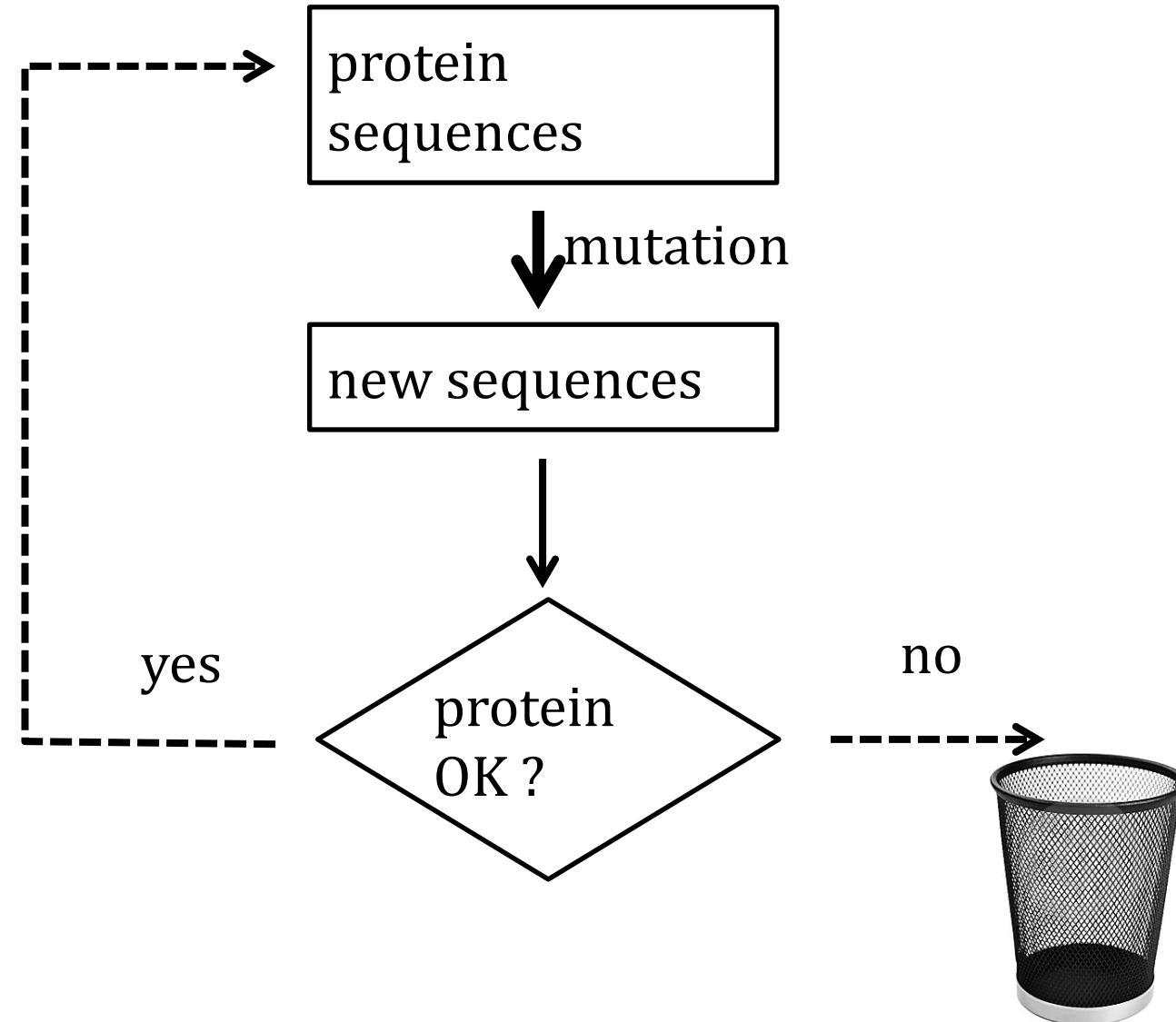
- what do we mean by Darwinism ?

Very Darwinist



Think neutralist

- OK/not OK step (selection) less important
- What determines the sequences you see ?
 - "mutation" step
- mutation step looks very simple
 - not really
- consider the meaning and biases



Codon bias

- look at the most rare amino acids...

ser	UCU, UCA, UCC, UCG, AGU, AGC
leu	CUU, CUA, CUC, CUG, UUA, UUG
...	
his	CAU, CAC
met	AUG
trp	UGG

- number of codons not quite everything

- some bases are more common than others

$$p(\text{his}) = 0.22 \cdot 0.3 \cdot 0.22 + 0.22 \cdot 0.30 \cdot 0.22 \approx 0.03$$

- does this predict the probability of all amino acids ?
- if yes, there is no selection for amino acids
 - Darwinism at the amino acid selection level

U	22 %
A	30 %
C	22 %
G	26 %

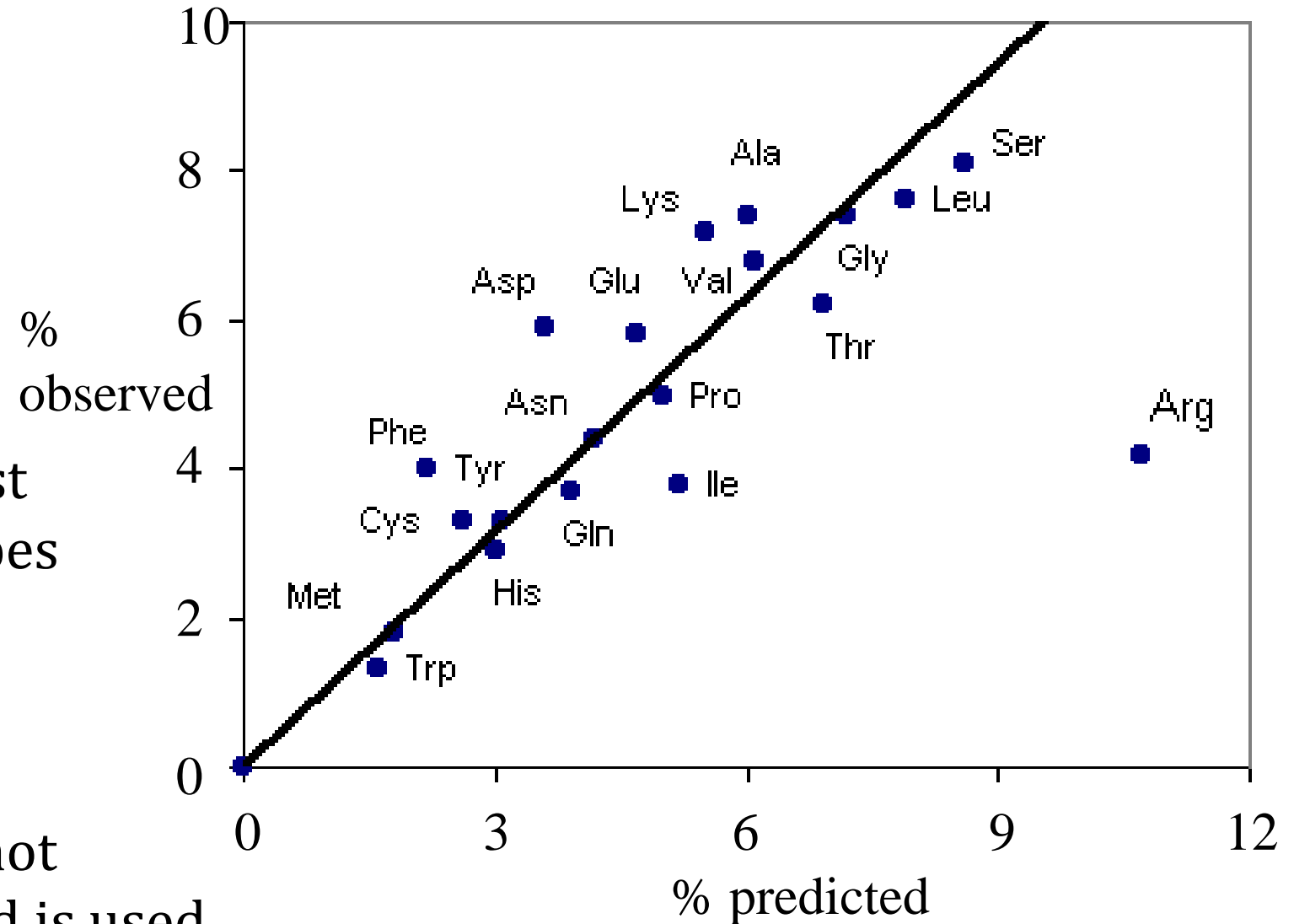
How relevant is Darwinism ?

One outlier (arg)

Little evidence of Darwinist selection in amino acid types

Logical consequence

- there are many sites in proteins where it does not matter which amino acid is used



Forget Darwinism and selection of amino acids ?

No

- arg example
- lots of mutation data
 - for an enzyme
 - most mutations are a bit bad, some do not matter
- Do not be a pure Darwinist
- do not interpret everything you see in terms of fitness

Representation

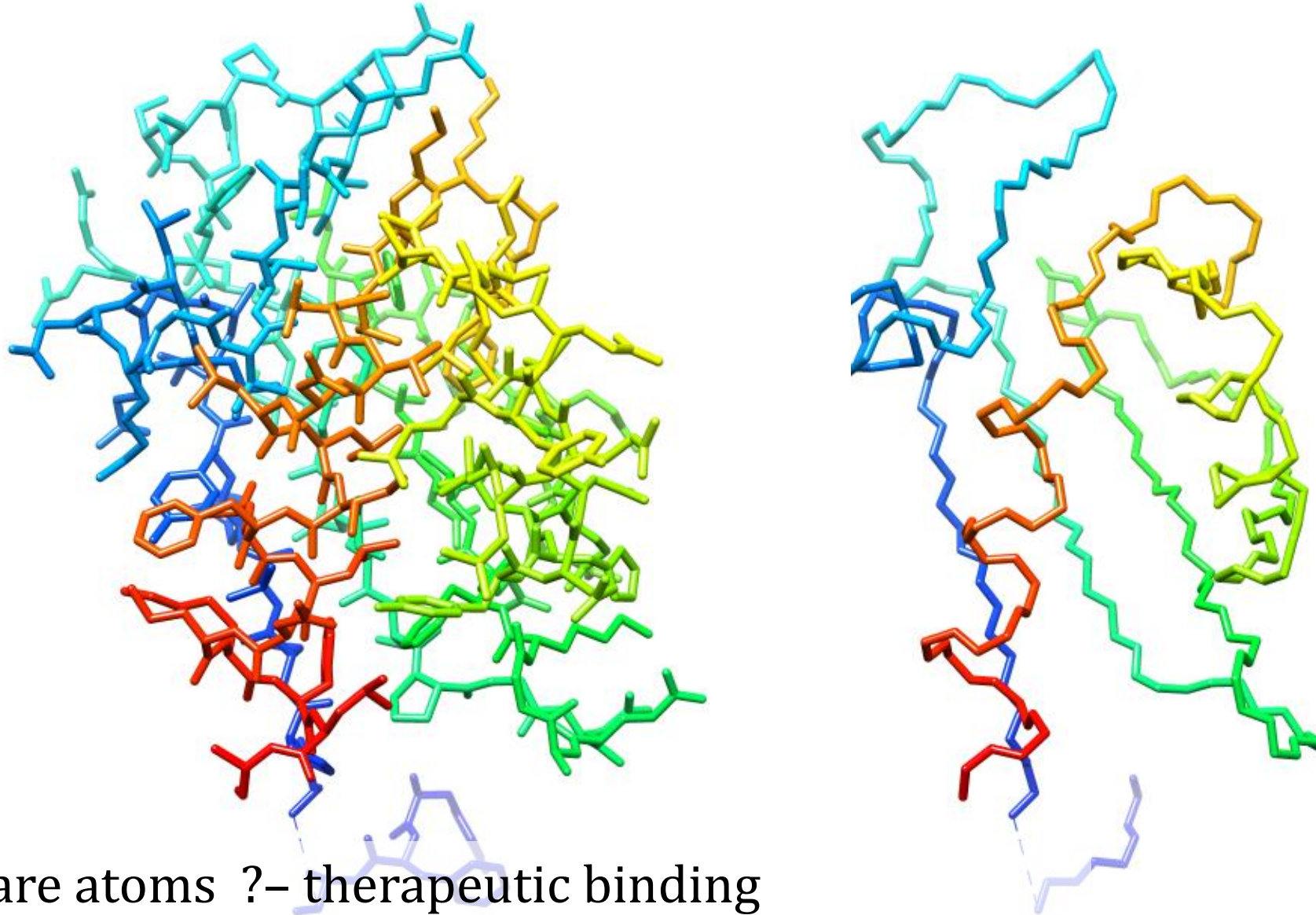
Ultimately, our representation of a structure...

ATOM	1	N	ARG	1	31.758	13.358	-13.673	1.00	18.79	1BPI	137
ATOM	2	CA	ARG	1	31.718	13.292	-12.188	1.00	14.26	1BPI	138
ATOM	3	C	ARG	1	33.154	13.224	-11.664	1.00	18.25	1BPI	139
ATOM	4	O	ARG	1	33.996	12.441	-12.225	1.00	20.10	1BPI	140
ATOM	5	CB	ARG	1	30.886	12.103	-11.724	1.00	16.74	1BPI	141
ATOM	6	CG	ARG	1	29.594	11.968	-12.534	1.00	15.96	1BPI	142
ATOM	7	CD	ARG	1	28.700	13.182	-12.299	1.00	15.45	1BPI	143
ATOM	8	NE	ARG	1	27.267	12.895	-12.546	1.00	12.82	1BPI	144
ATOM	9	CZ	ARG	1	26.661	13.087	-13.727	1.00	17.38	1BPI	145
ATOM	10	NH1	ARG	1	27.370	13.558	-14.735	1.00	18.38	1BPI	146
ATOM	11	NH2	ARG	1	25.367	12.797	-13.838	1.00	25.73	1BPI	147
ATOM	12	N	PRO	2	33.800	13.936	-10.586	1.00	17.07	1BPI	148
ATOM	13	CA	PRO	2	34.976	13.367	-9.840	1.00	14.99	1BPI	149
ATOM	14	C	PRO	2	34.960	11.922	-9.660	1.00	13.11	1BPI	150
ATOM	15	O	PRO	2	33.962	11.306	-9.391	1.00	10.57	1BPI	151
ATOM	16	CB	PRO	2	34.922	14.145	-8.523	1.00	15.81	1BPI	152
ATOM	17	CG	PRO	2	34.058	15.391	-8.737	1.00	18.91	1BPI	153
ATOM	18	CD	PRO	2	33.371	15.273	-10.096	1.00	19.41	1BPI	154
ATOM	19	N	ASP	3	30.190	8.73	-9.707	1.00	8.73	1BPI	155

x, y, z
coordinates

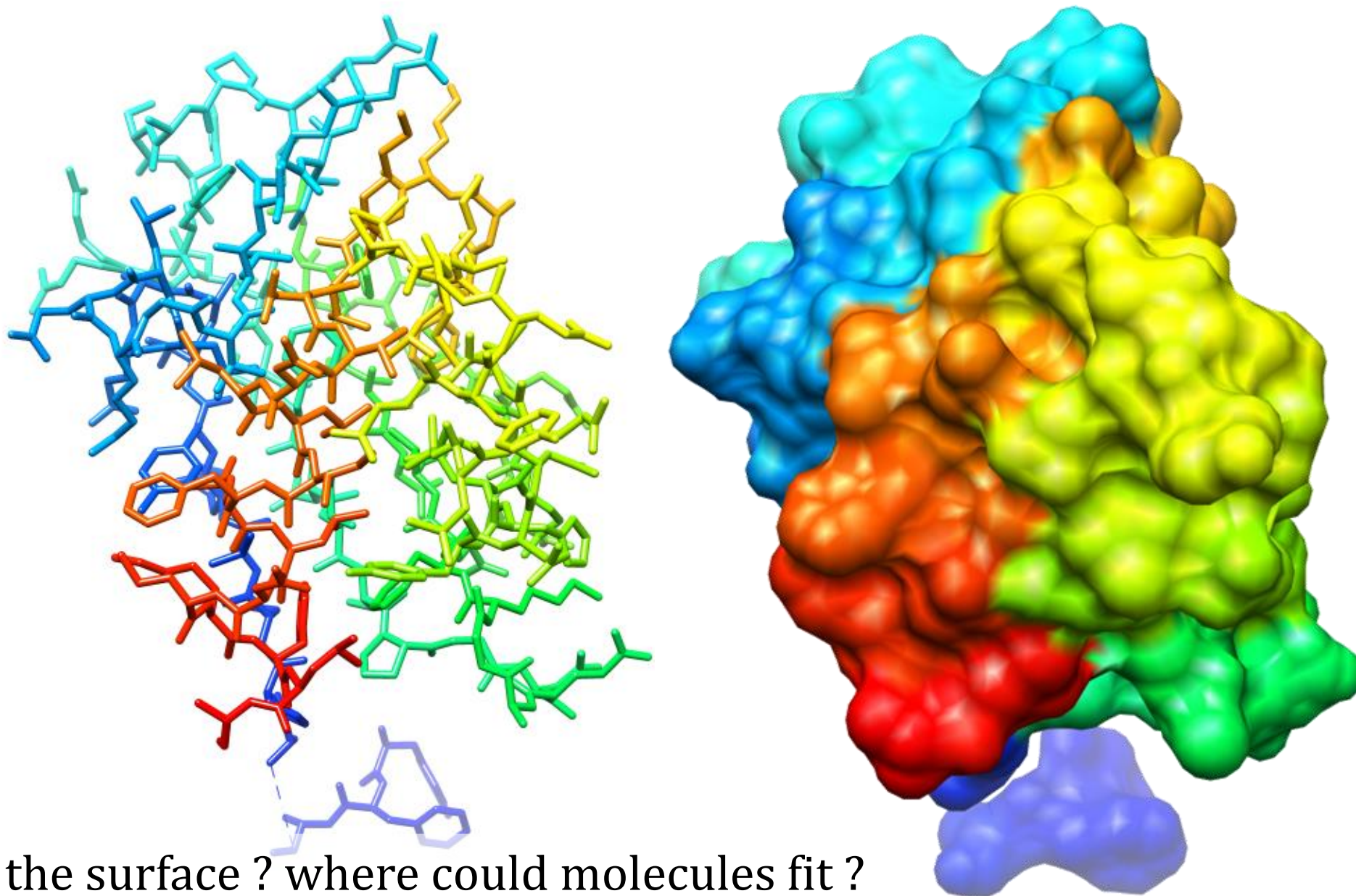
Drawing the structure ?

Representations



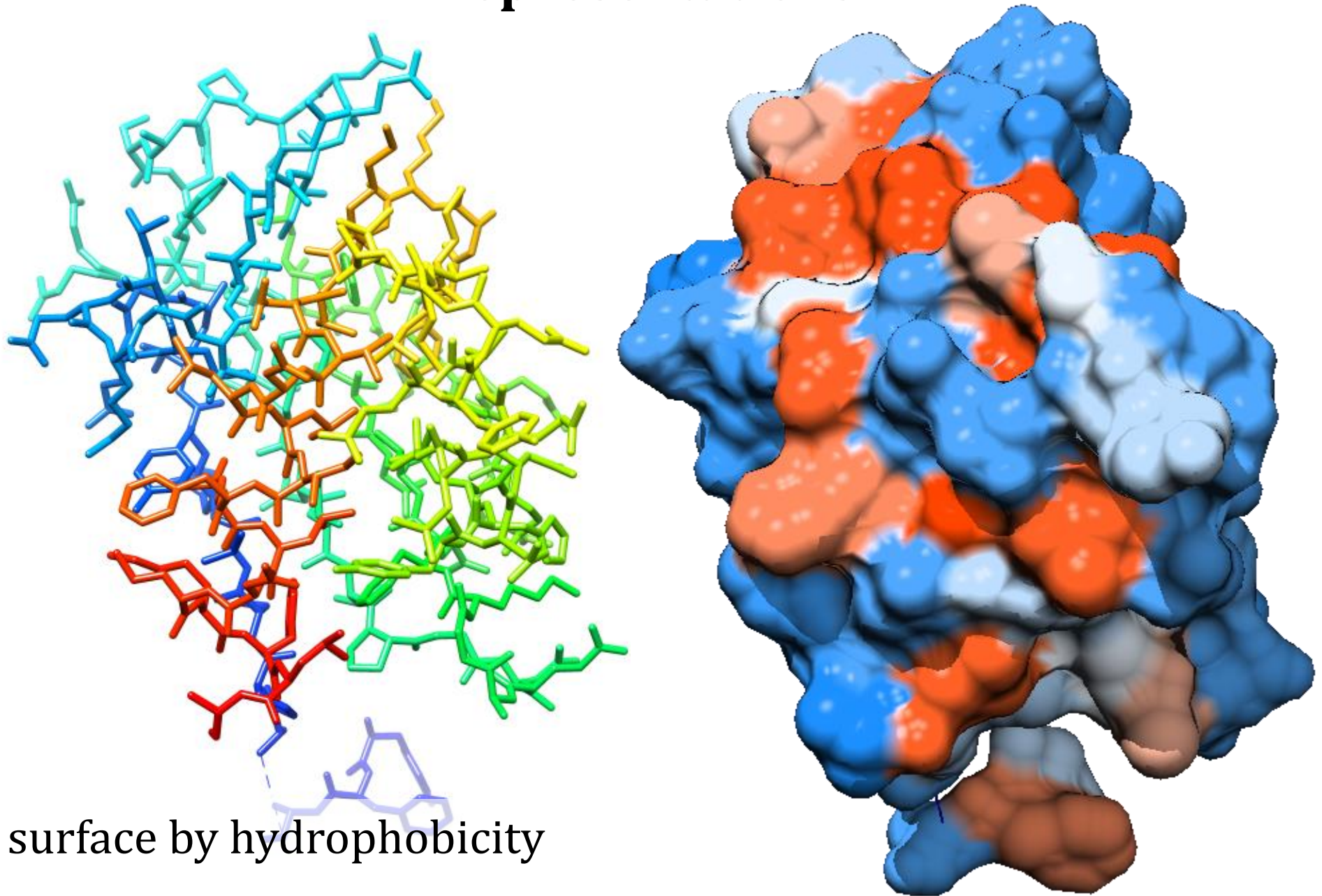
- where are atoms ?– therapeutic binding
- which residues could be involved in interactions ?

Representations



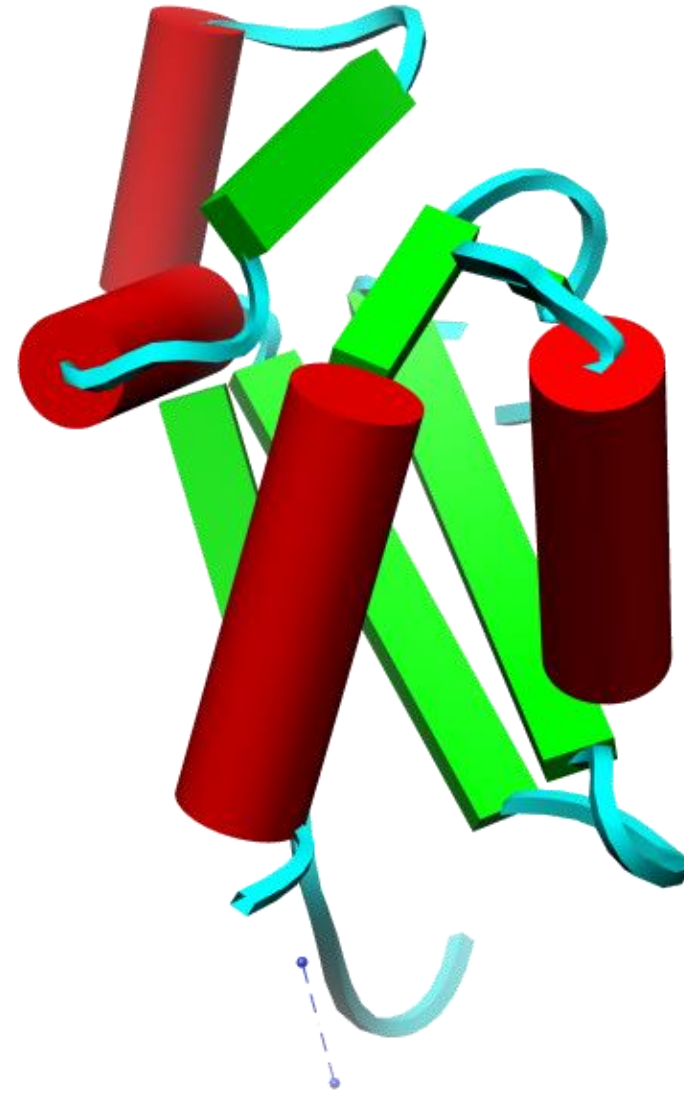
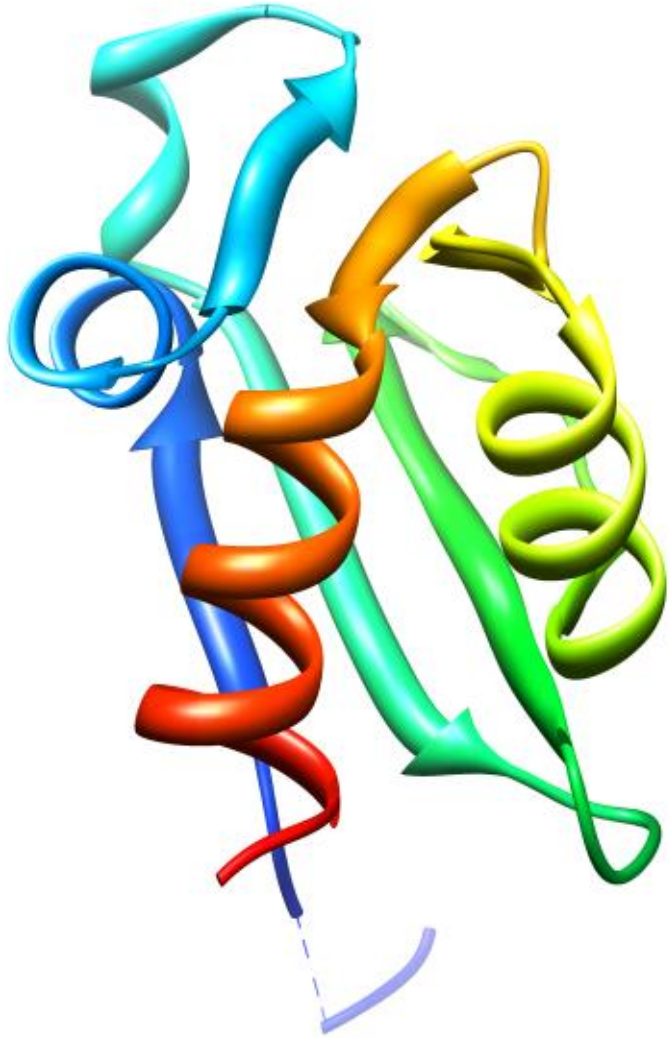
What is the surface ? where could molecules fit ?

Representations



Colour surface by hydrophobicity

Representations



Highlight / emphasise regular structures

Why does structure matter ?

- what residues can I change and preserve function ?
- what is the reaction mechanism of an enzyme ?
- what small molecules would bind and block the enzyme ?
- is this protein the same shape as some other of known function ?

Where do structures come from ?

- X-ray crystallography
- NMR
- + a bit of small angle X-ray scattering, electron diffraction, neutron diffraction...

resolution, precision, accuracy

Coordinates 27.370 13.558 -14.735

- what do they mean ?

Random errors

- non-systematic / noise / uncertainty
- should be scattered around correct point

X-ray crystallography has model for data

- uncertainty (probability)
- resolution (experimental)
 - $< 1 \text{ \AA}$ (unusually good)
 - $> 5 \text{ \AA}$ (bad, but examples..)

3LJ5 Full Length Bacteriophage P22 Portal Protein

3M0C X-ray Crystal Structure of PCSK9 in Complex with the LDL receptor

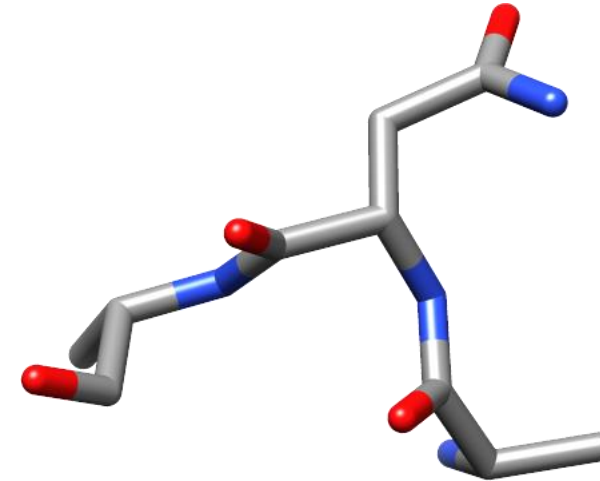
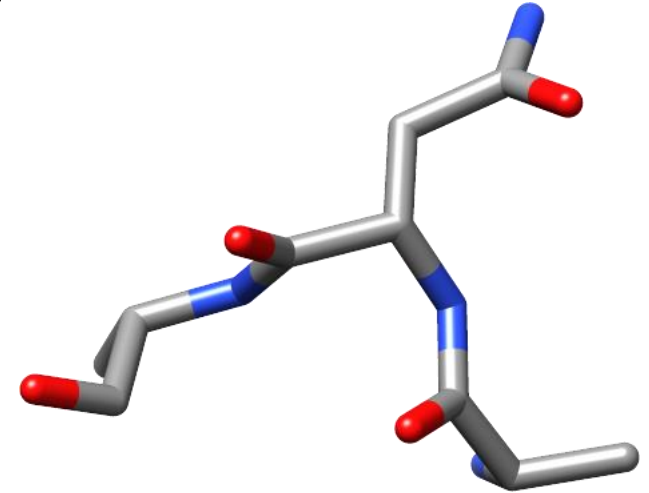
X-ray crystallography

Non-systematic errors

- small problems: (O and N look the same)
- few huge problems
- newer structures are better

Proteins are not static

- overall motion
- local motion



NMR structures

Different philosophy to X-ray

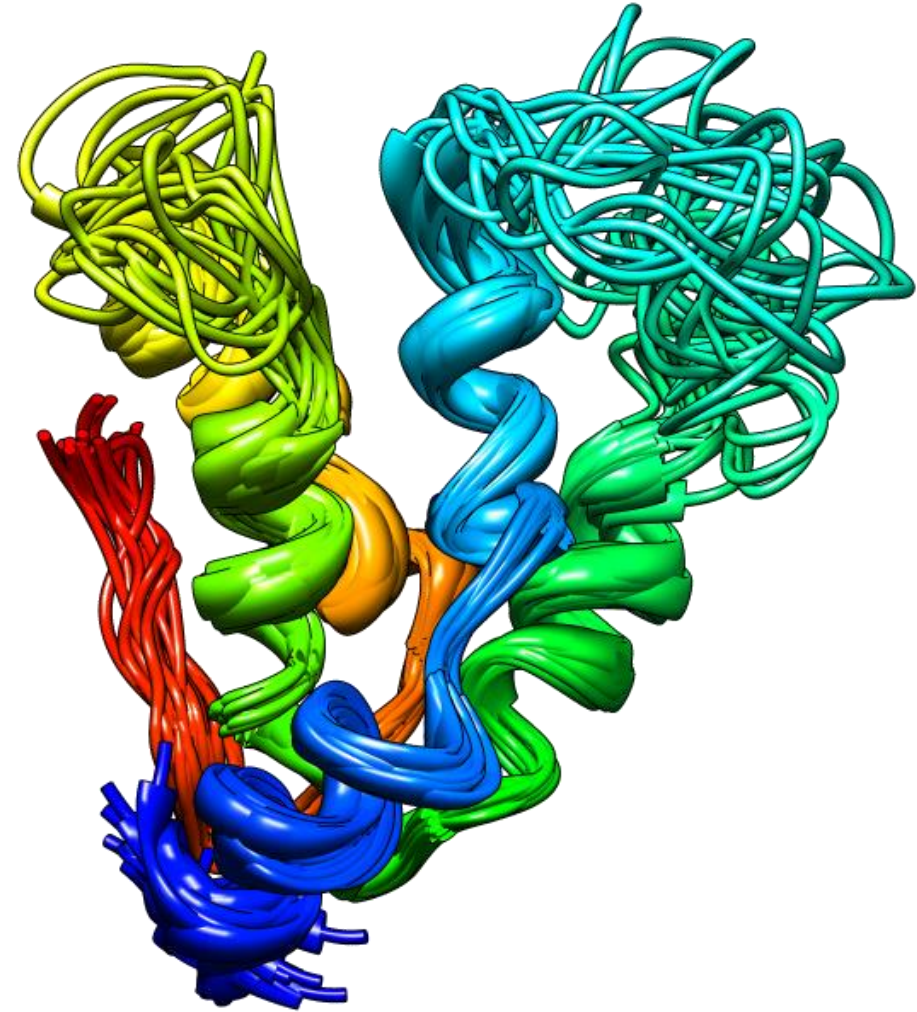
- lots of little internal distances
- do not quite define structure

Generate 50 or 10^2 solutions

- look at scatter of solutions

As with X-ray

- some parts are well defined
- some not



Summarise and stop

- roles of proteins
- heteropolymers – 20 types of amino acid / residue
- geometry – avoiding atomic clashes, forming H bonds
 - leads to regular secondary structure
- chemistry of amino acids very different to another
- unique structure for a sequence reflects these differences
- representations of structures
- structures in PDB are experimental – have errors

some questions

- (Asp)₁₀₀
 - is it soluble ? Is it acidic / basic ?
 - would it form a compact regular structure ?
- How big is sequence space ? How much has been tried by evolution ?
- if you have a protein of poly-trp, would it form a specific structure ? How would it behave in solution ?
- for length n , do all / many / few of the n^{20} sequences form specific structures ?
- how would a Darwinist explain the uneven distribution of amino acid usage ?
- why would you want to represent a protein by its surface ?
- why might you draw it as a series of helices and strands ?
- what is the biggest chain in the protein data bank ? Examples
 - fatty acid synthase $> 2 \times 10^3$ residues/chain
 - dynein heavy chain motor domain $> 4 \times 10^3$ residues/chain