

# NMR (Nuclear Magnetic Resonance Spectroscopy)

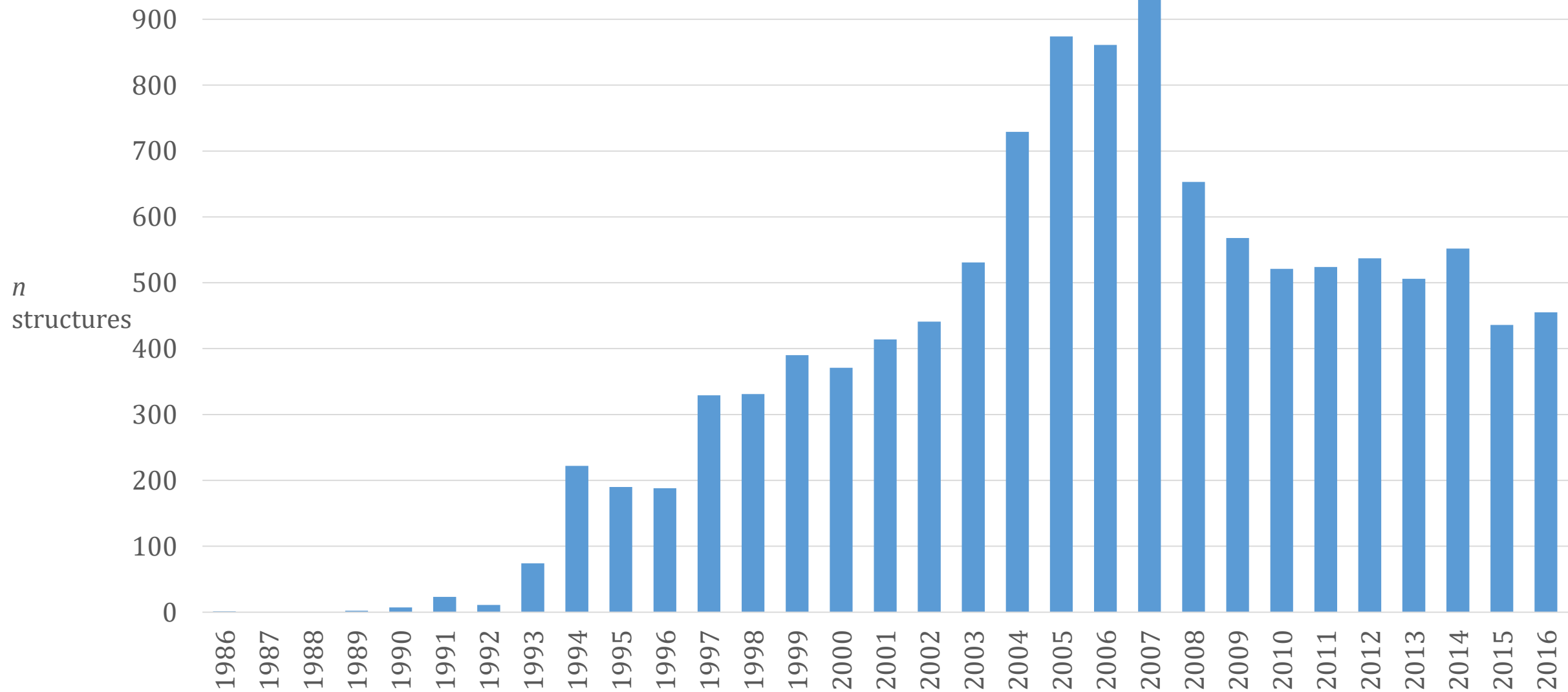
Literature / background (already in Stine)

- Ferentz, A.E. and Wagner, G., Q. Rev. Biophys, 33, 29-65 (2000) – in Stine

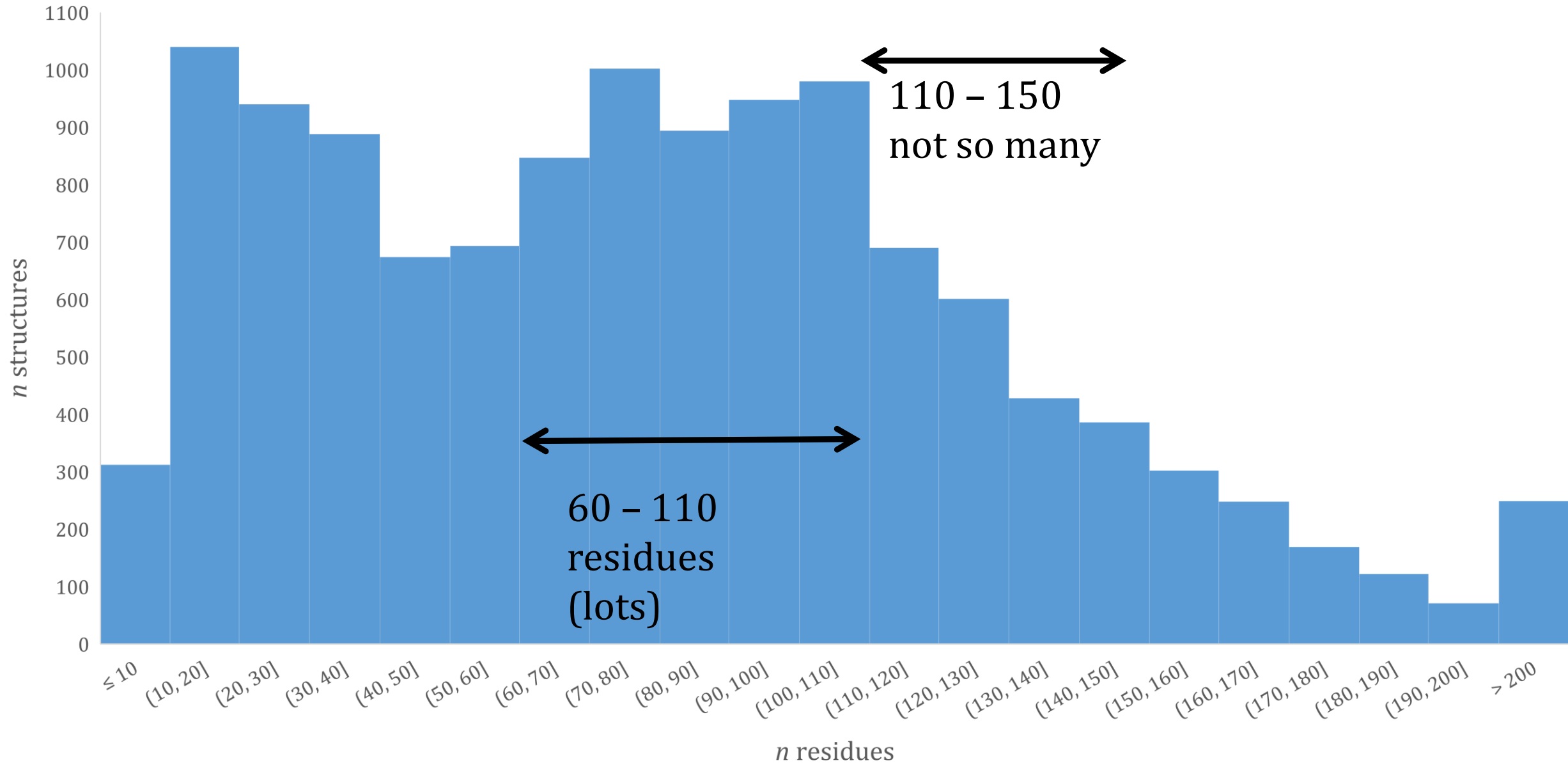
Current standing

- $\approx 11$  % of current structures solved by NMR (11 950 structures, 9287 proteins)
- about 1/4 of smaller structures (<100 residues)

# How many structures by NMR ?



# sizes of NMR structures in protein data bank



# What is coming

Background to NMR – chemistry

Calculating structures

- distance geometry
- problems with structures

For chemists: no

- chemical shifts
- 2D and higher
- residual dipole coupling, spin labels
- ...

# History

Younger field than X-ray

- 1 ½ Nobel prizes (Ernst, Wüthrich)

First real protein structure about 1985 or 1986

## **NMR from our viewpoint**

A way to get structures - our focus

Can provide information on

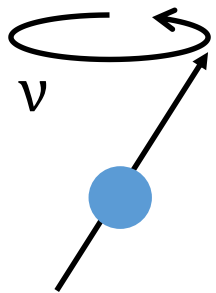
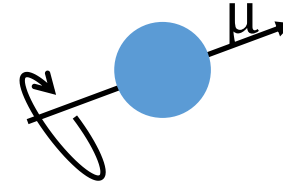
- dynamics, stability
- interactions (other proteins, small molecules)

# Overview – how we get coordinates

- protein in solution
- record spectra
- assign peaks to  $^1\text{H}$ ,  $^{13}\text{C}$ ,  $^{15}\text{N}$  nuclei
- record some more spectra
  - distance information (mostly)
  - some internal angles
- reconstruct structure

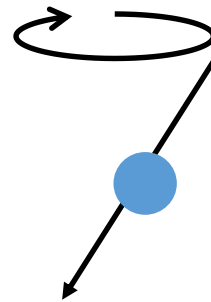
# Nuclei have spin

- have a charge and act like magnets
- put them in a field and they will align with it
- now apply a magnetic field
  - they "precess" around the field
  - two possible states



$B_0$

or maybe



$B_0$

$B_0$

$\nu$

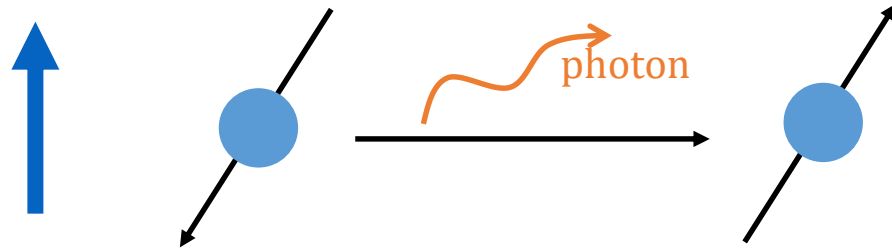
is applied field

speed of rotation (many MHz /  $10^6$  Hz)

# Do nuclei like fighting the field ?

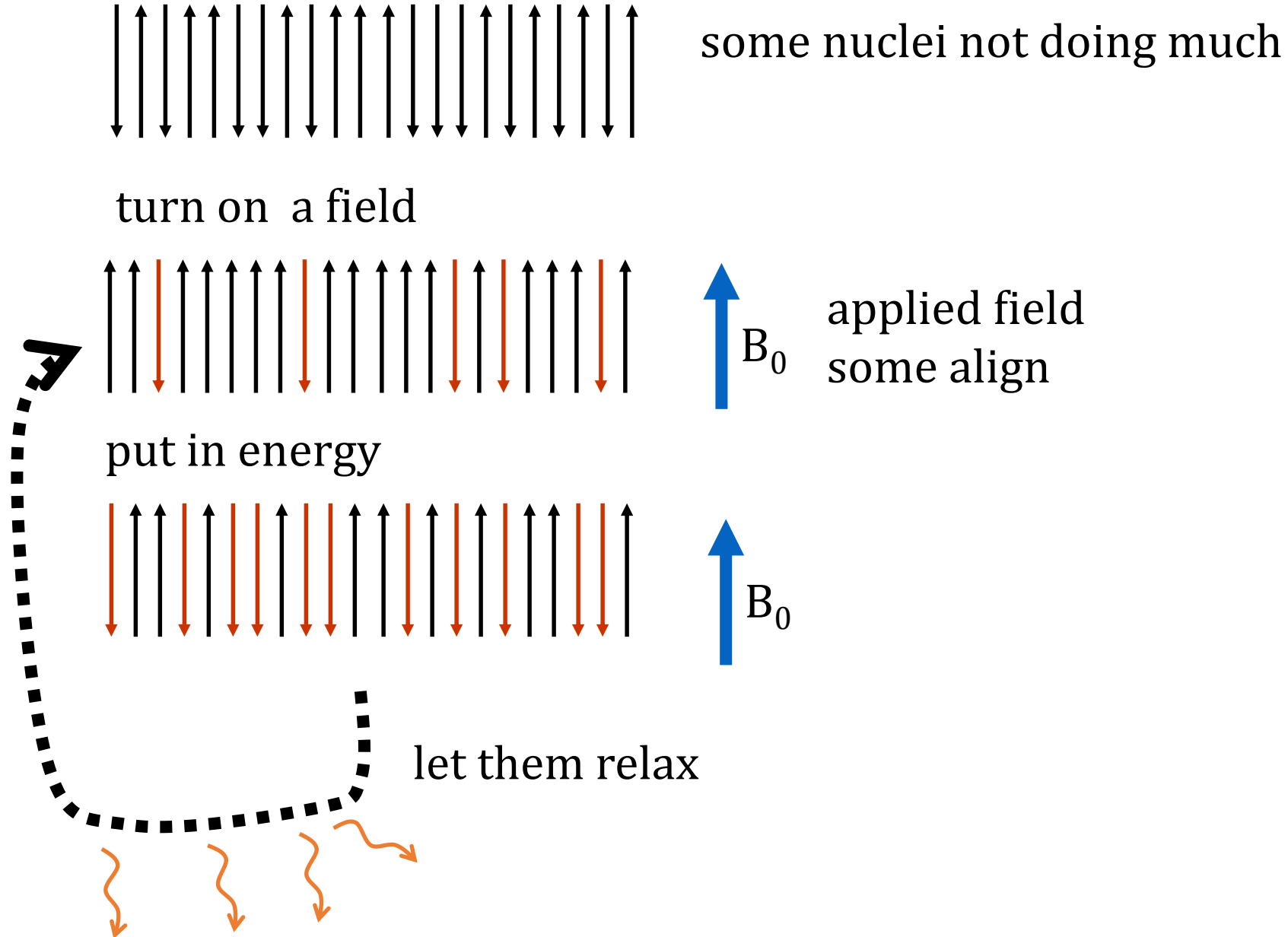
Is a nucleus really happy facing the wrong way ?

- what if we push it the wrong way ?
  - wants to get to low energy state – emits a photon





# What NMR records



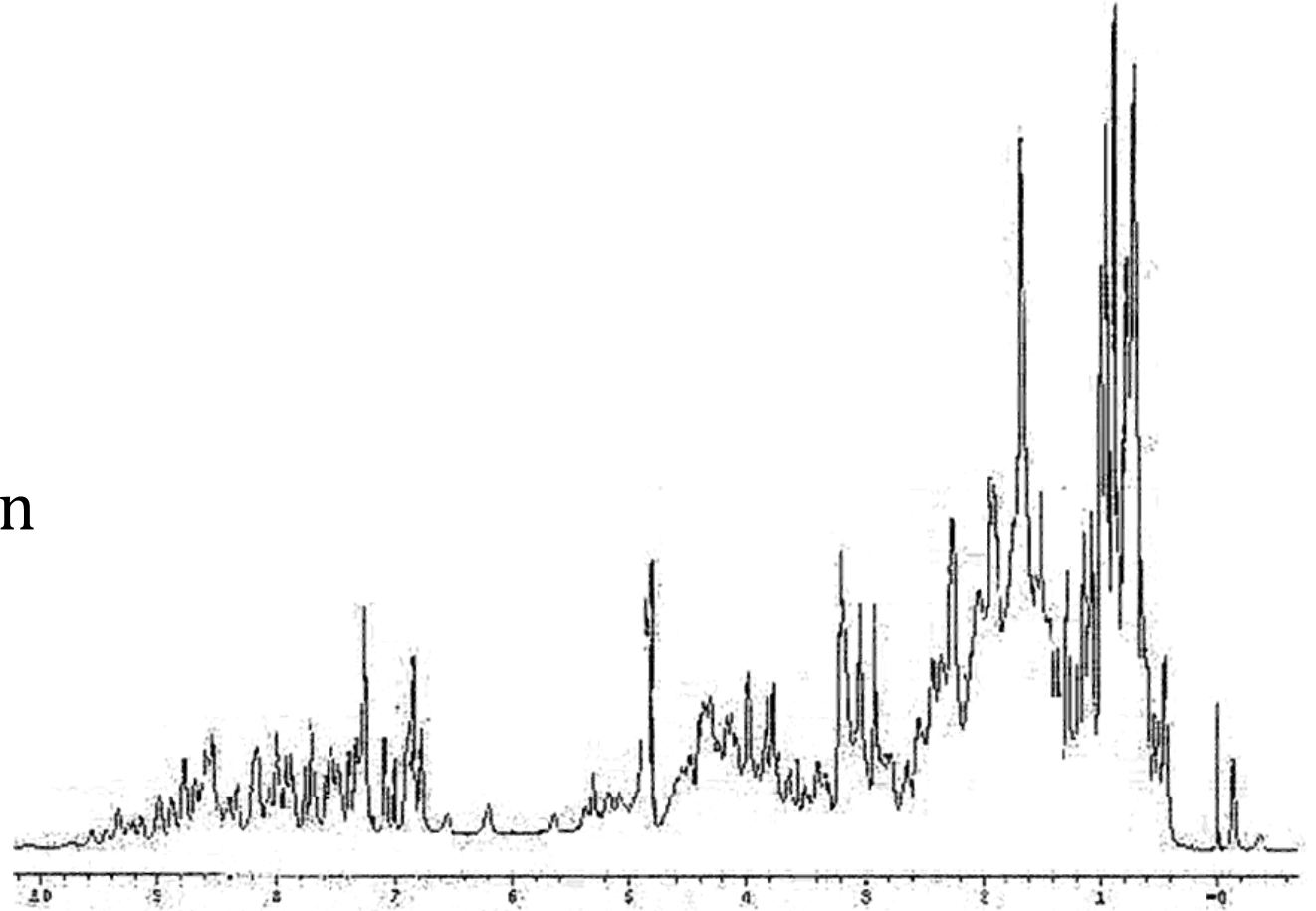
# Important nuclei (spin $\frac{1}{2}$ )

nucleus	sensitivity	notes
$^1\text{H}$	1	cheap and natural
$^{13}\text{C}$	$1.6 \times 10^{-2}$	expensive, but only 1% of natural abundance
$^{15}\text{N}$	$10^{-3}$	not cheap, 0.4 % natural abundance
$^{31}\text{P}$	$7 \times 10^{-2}$	DNA and other $\text{PO}_4$ chemistry, less protein

- but the natural isotopes are  $^{12}\text{C}$  and  $^{14}\text{N}$ 
  - (usually) these isotopes require labelling
- Proteins
  - $^1\text{H}$ ,  $^{13}\text{C}$ ,  $^{15}\text{N}$

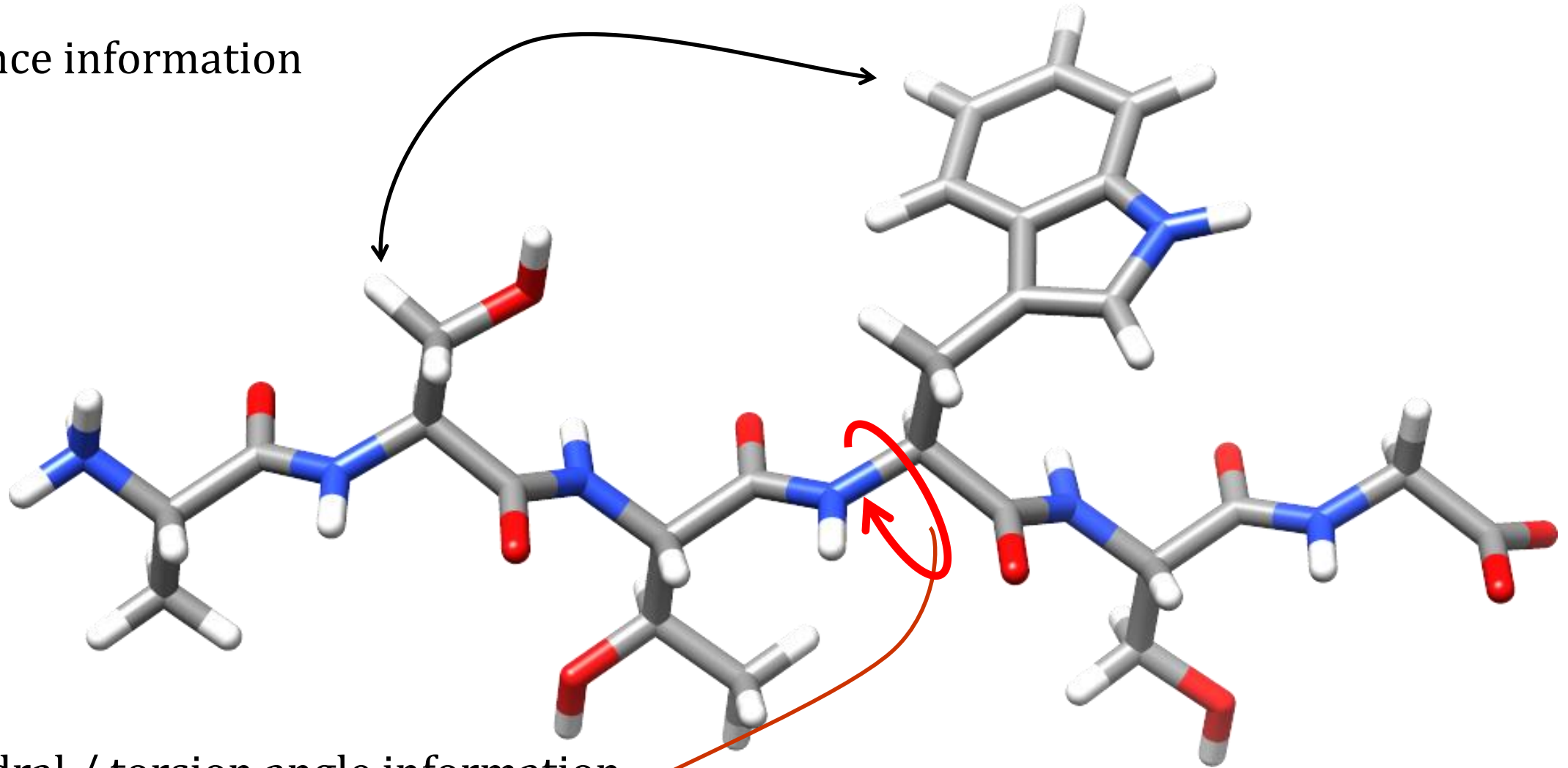
# NMR for us

- You get a spectrum (1D, 2D, ..)
- Where are the peaks ?
  - For chemists – not this course
- We care about structural information
- This nucleus affects that nucleus
  - (field splitting, relaxation, ...)
- Can be related back to structure



# To calculate structures ?

1. distance information



2. dihedral / torsion angle information

# Distance information / the NOE

Most important (NOE = nuclear overhauser effect)

- an effect which depends on how close in space nuclei are
- $\text{NOE} \propto r^{-6}$
- usually only up to about 5 - 6 Å

Story

- two spins' dipoles interact
- cross relaxation phenomenon

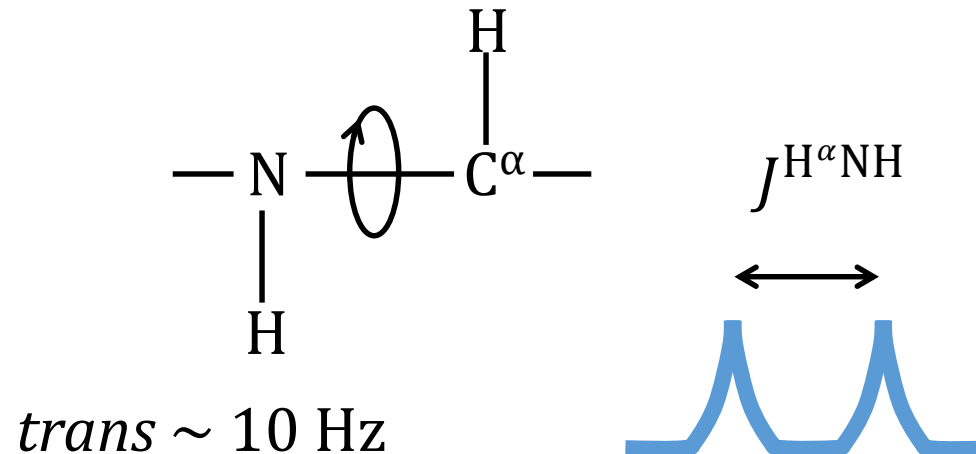
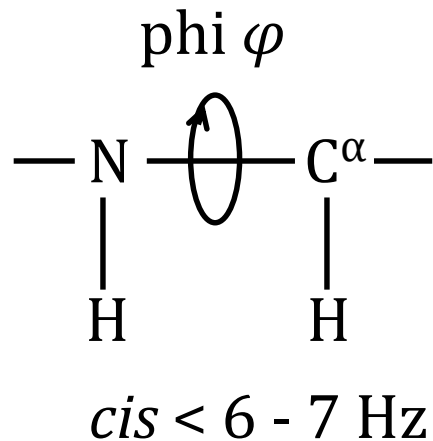


red relaxing (jumping to lower energy)  
affects black

# Other structural information

- NOE – information about short ( $< 5$  or  $6 \text{ \AA}$ ) distances
- there is more – angles
  - mainly  $J$  coupling

## Amide NH to $H^\alpha$ coupling

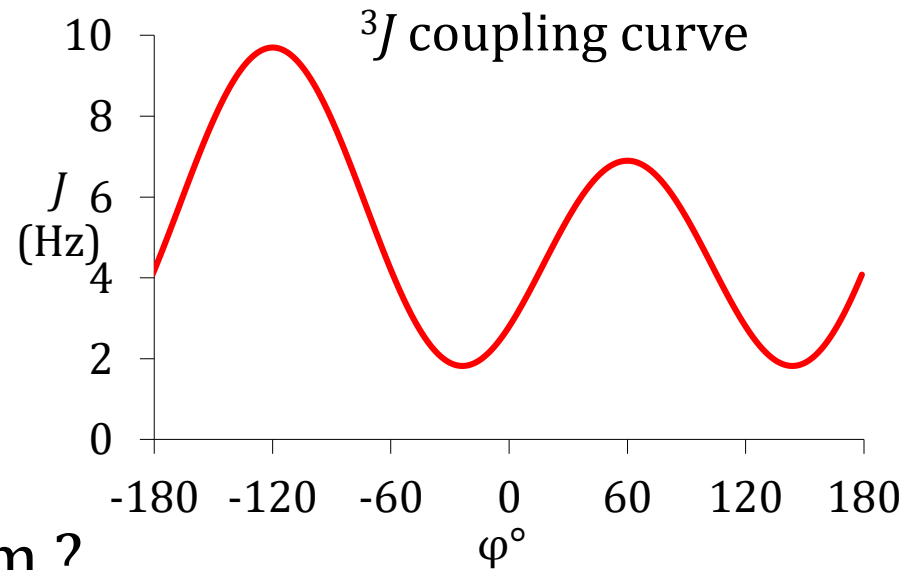


# $^3J_{\text{HN}\alpha}$ coupling

formalised as

$$^3J_{\text{H}\alpha\text{N}} = 6.4 \cos^2 \theta - 1.4 \cos \theta + 1.9$$

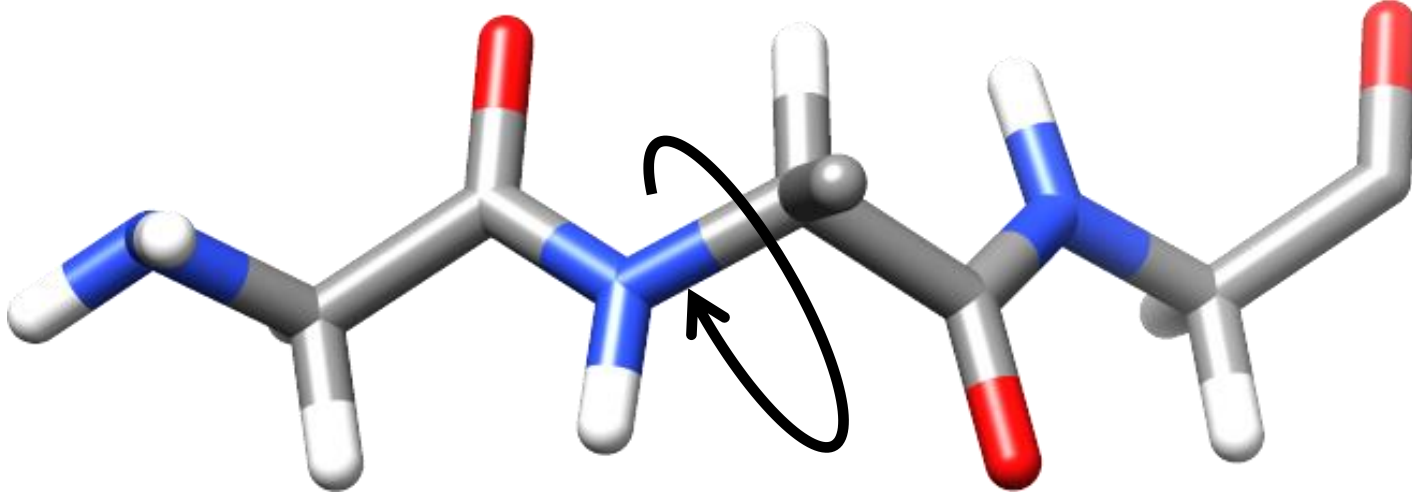
Problems...



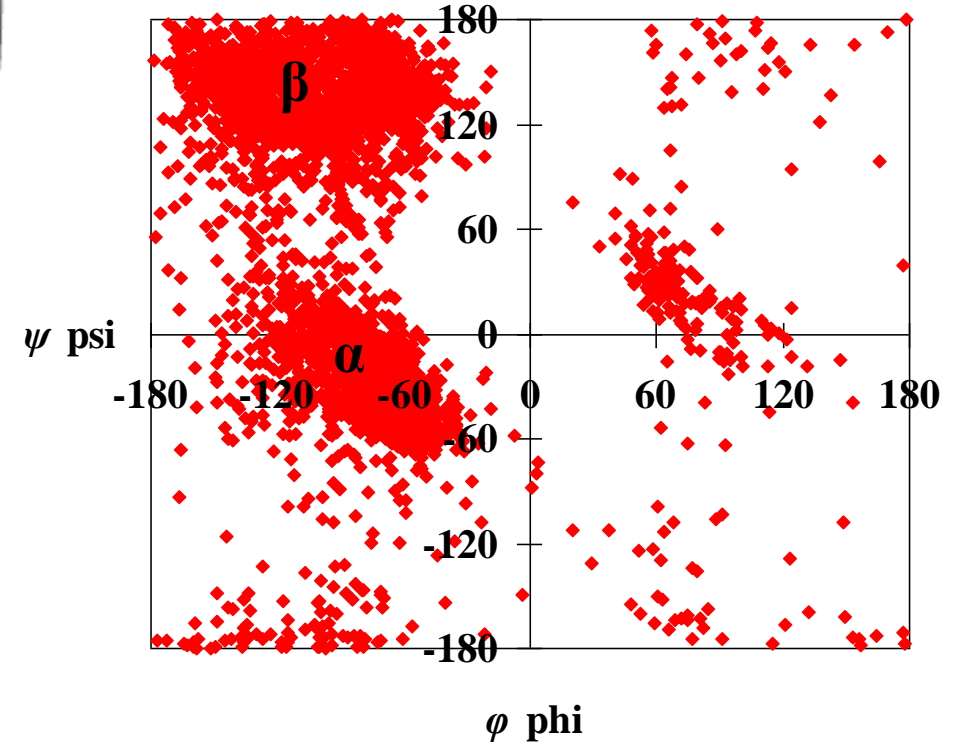
Where do 6.4, 1.4, 1.9 come from ?

Do not learn for Klausur

# Amide NH to H $^{\alpha}$ coupling



- can help distinguish  $\alpha$  from  $\beta$
- not always seen (exchange / motion)
- NH not always present
- other angles ?
  - $C^{\alpha}$  to  $C^{\beta}$





# Problems with $J$ -coupling

We have a formula

$${}^3J_{\text{N}\alpha\text{H}} = 6.4 \cos^2 \theta - 1.4 \cos \theta + 1.9$$

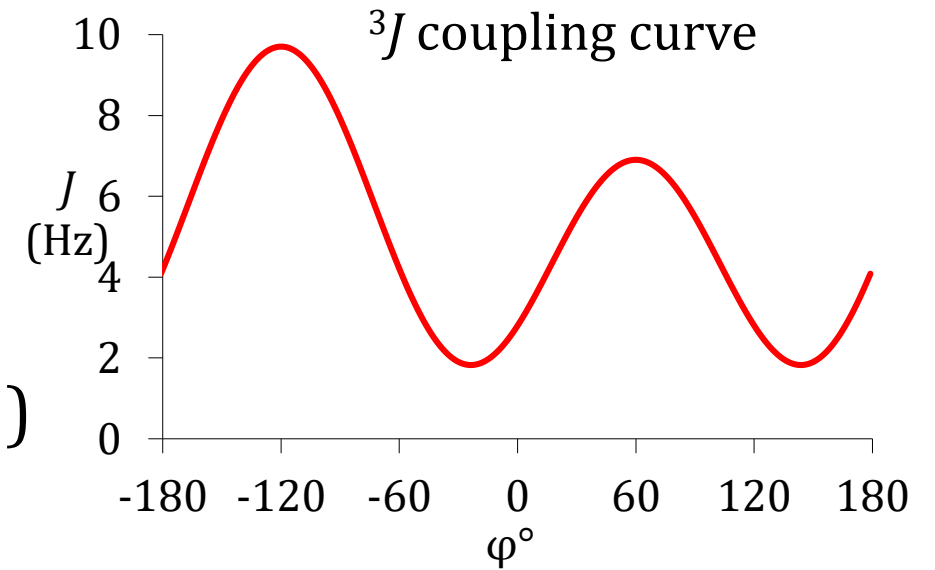
- measure  $J$ , solve for  $\theta$

Most of the time there is more than one solution ( $\theta$ )

- use only large  $J$

Dynamics and errors

- look near  $-90^\circ$



# Practical NMR

We have some basic methods

Real NMR

- more techniques
  - 2D and more
  - identifying specific kinds of atom
  - spreading peaks out

# Information summary

phenomenon	assignments	structure	
chemical shift	important	not much used	not in Folien
spin-spin ( $J$ ) coupling	important	torsion angles	
NOE	important	distances	

## More spectroscopy

- filtering according to chemistry, atom types
- $n$ -dimensional methods

## Structural information

- labels for broadening / shifting peaks
- orientation of bonds to reference ..

# Structures from NMR data

## Available information

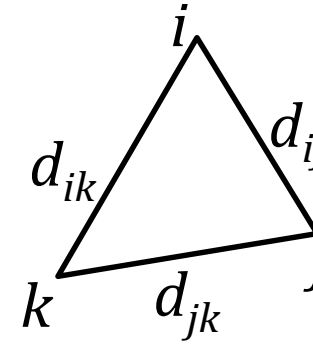
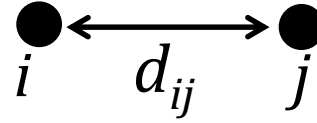
- distances
  - short (  $< 5$  to  $6 \text{ \AA}$  )
  - incomplete
- some dihedral / torsion angles
- does this define a structure ?
  - strictly no

## Coming

- distances in 2D and 3D
- Distance geometry – two versions

# Determining distances (ideal)

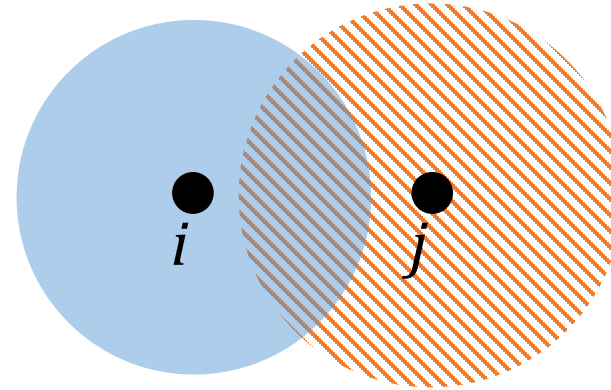
- 2 points 1 distance
- 3 points 3 distances...
  - think of  $3N_{atom}$  distances
  - remember  $N_{atom} \approx 10$  or  $20$   $N_{res}$



# Underdetermined distances

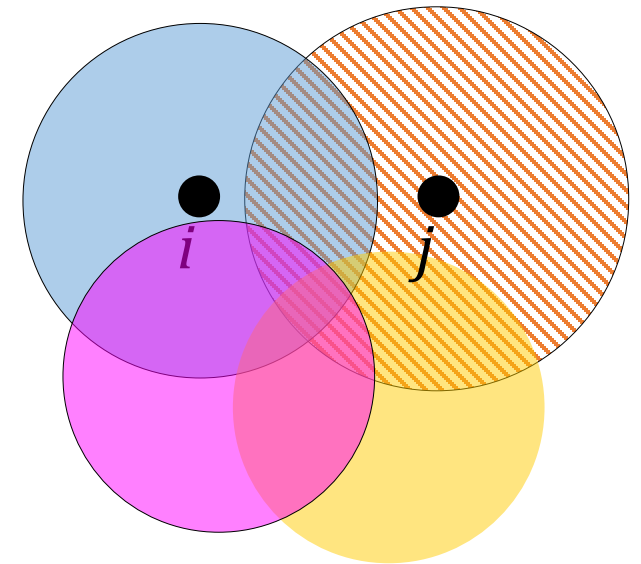
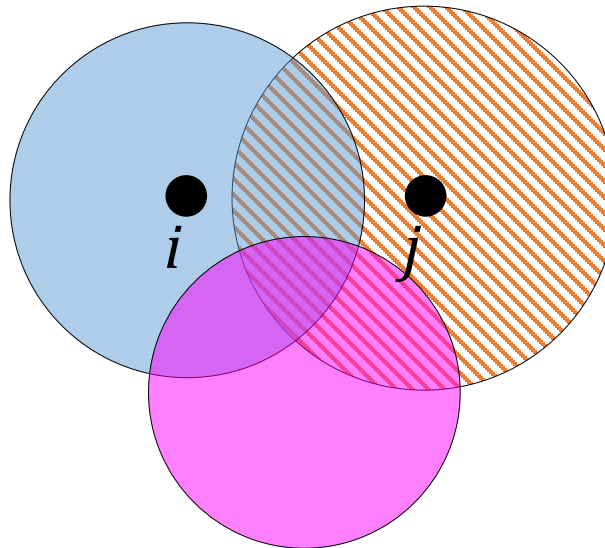
Think in terms of triangles ...

- $d_{ik} < 6 \text{ \AA}$ ,  $d_{jk} < 6 \text{ \AA}$
- where is  $k$ ?



A few more distances...

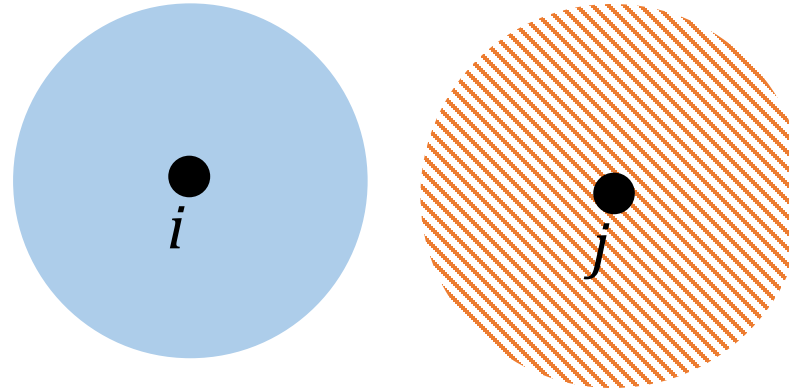
- more and more distances are useful



# Impossible distances

No overlap ?

- experimental error
- nowhere for  $k$  to go



## Real data

Protein of  $N_{res}$  residues, you might have 5 or 10  $N_{res}$  distances

- want more like  $3N_{atom}$  ( $30 - 60 N_{res}$ ) distances if perfect
  - needs much more data...
  - lots of chemical data

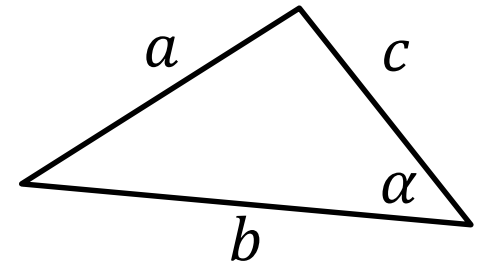
# An analytical solution ?

Is there some formula which will give you structures from distances ?

- Could I say  $a^2 = 2bc \cos \alpha$  or  $\frac{a}{\sin \alpha} = \frac{b}{\sin \beta} = \dots$  ?

There is not enough experimental data

- can be fixed partially (coming soon...)



Serious problems

- you do not know  $a, b, c, \alpha, \dots$  exactly – you cannot get other distances or angles
  - how would you deal with a range (3 – 5 Å) ?
- even if you knew many distances almost exactly
  - numerical errors accumulate (badly)



# Mission

- gather all experimental data
- mix in chemical data
- make all distance information as tight as possible
- put an upper bound on the distance between every pair of points
- put a lower bound on every distance (less important)
- somehow generate coordinates
- start with toys and triangles

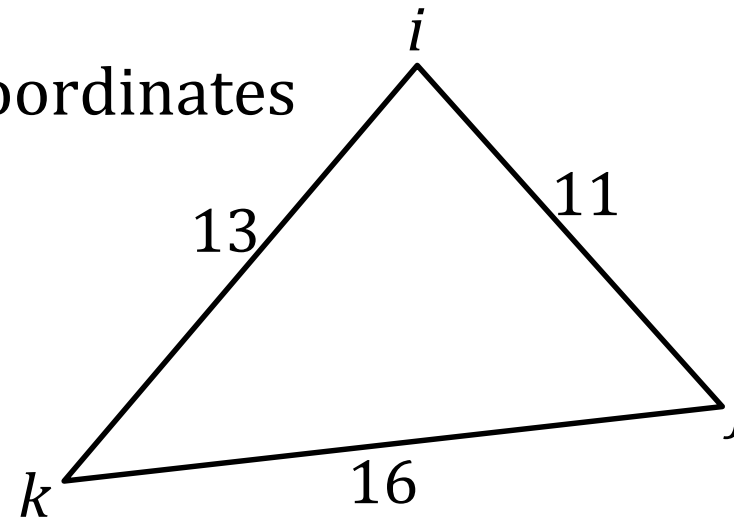
# Structures from distance information

Start in two dimensions..

Ein freundliches Dreieck

$$d_{ij} = 11 \quad d_{ik} = 13 \quad d_{jk} = 16$$

- fix  $i$ , put  $j$  on  $x$ -axis and make coordinates
- solve analytically



# Underdetermined data

$$d_{ij} = 11$$

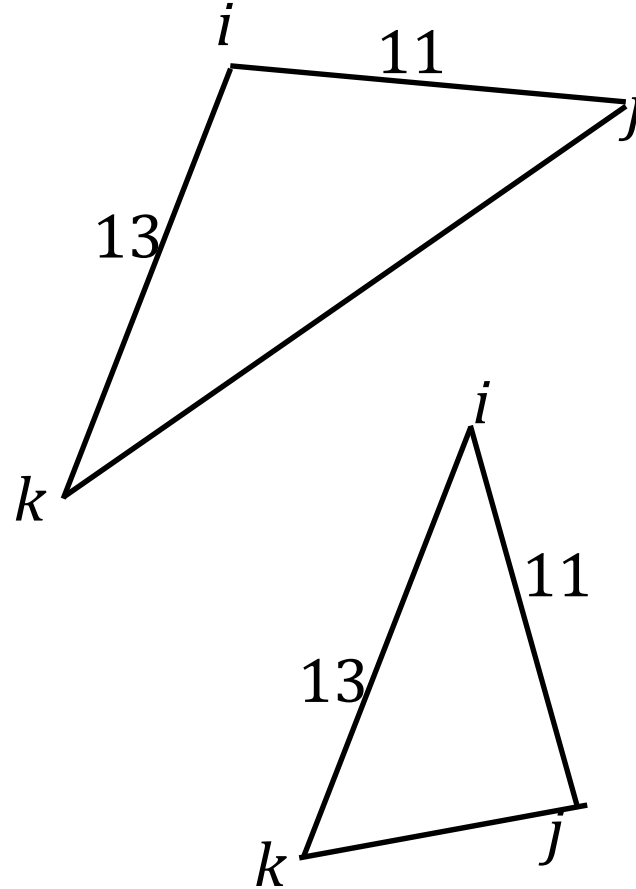
$$d_{ik} = 13$$

$$12 < d_{jk} < 20$$

More like NMR data

Unique solution ?

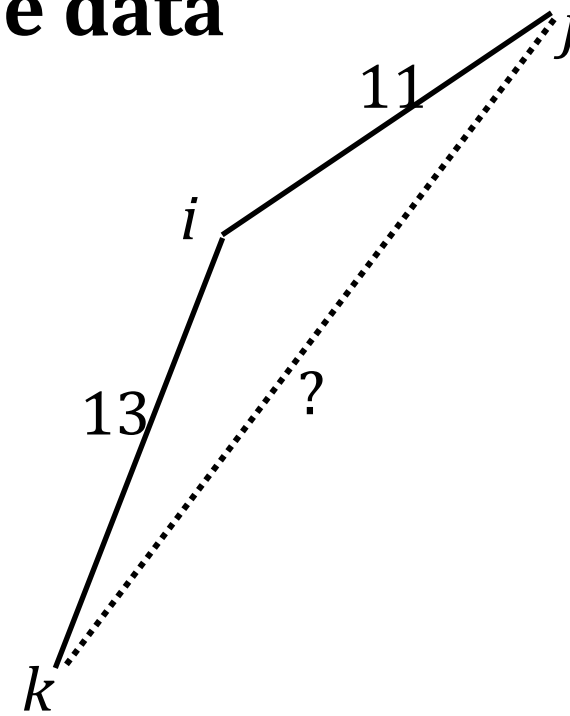
No



# Impossible data

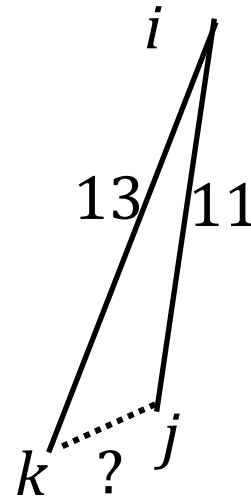
distance too big

$$d_{ij} = 11 \quad d_{ik} = 13 \quad d_{jk} = 25$$



distance too small

$$d_{ij} = 11 \quad d_{ik} = 13 \quad d_{jk} = 1$$



no 3D structure

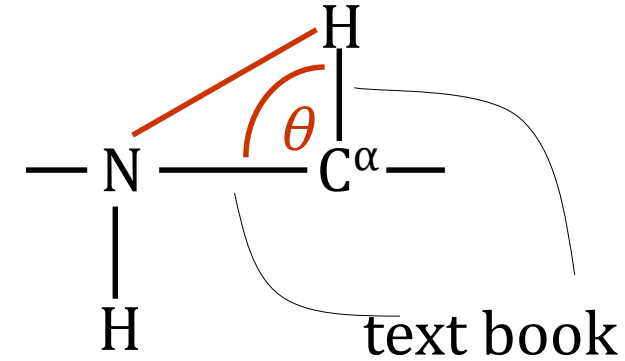
# Gathering data

- add in chemistry
- use to get more
  - mix chemistry + measurements
- what comes easily from chemistry ?

# Gather as much data as possible

Simple, geometric information

- bonds – standard
- angles – standard
- simple distances from bond angles
- dihedral / torsion angles



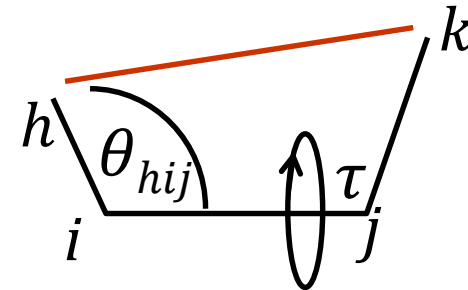
$$d_{hk}^2 = (d_{ij} - d_{hi} \cos \theta_{hij} - d_{jk} \cos \theta_{ijk})^2 + (d_{hi} \sin \theta_{hij} - d_{jk} \sin \theta_{ijk} \cos \tau_{hijk})^2 + (d_{jk} \sin \tau_{hijk})^2$$

set  $\tau = 0$

- minimum

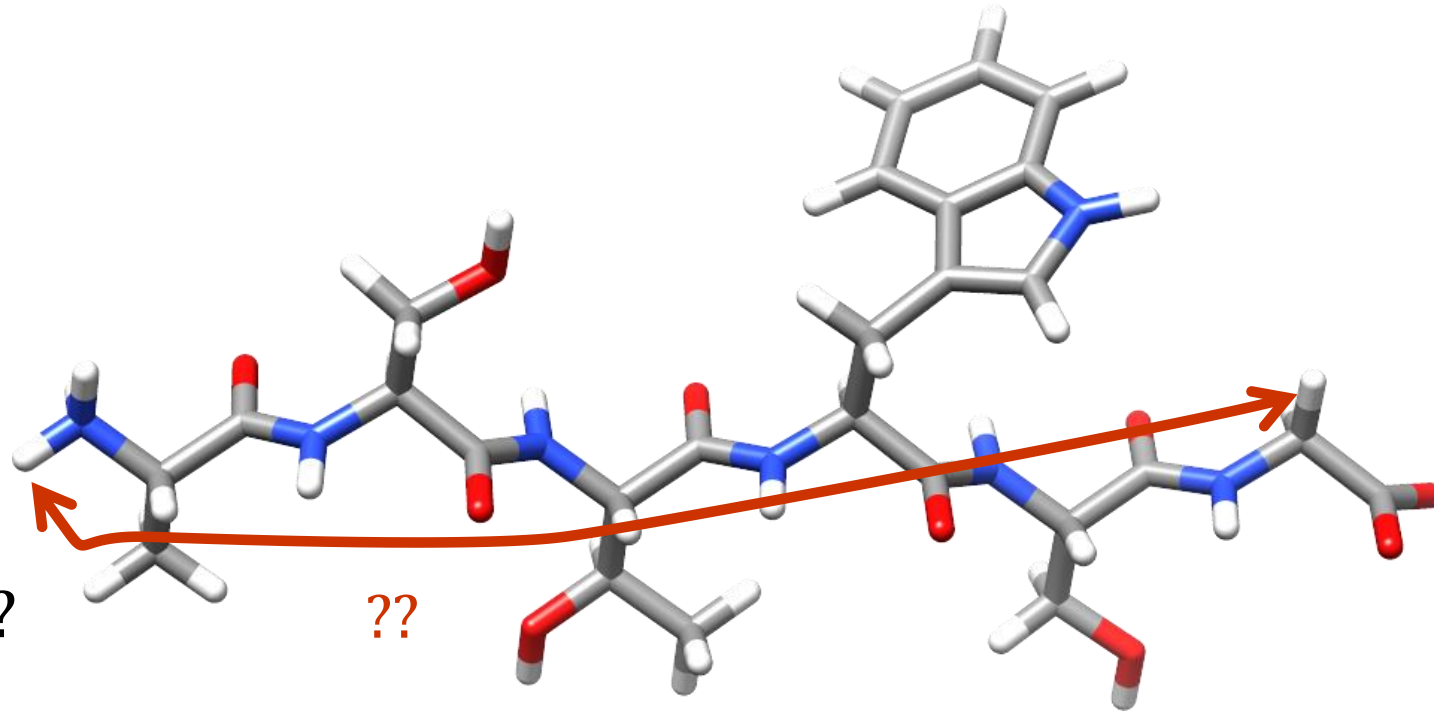
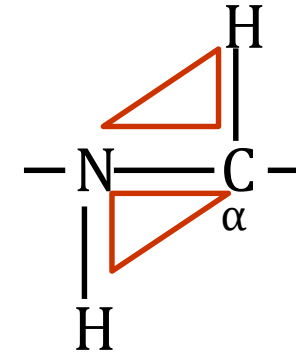
$\tau = \pi$

- maximum



# How to get more distance information

- impose some distance limits generally
- intuitively
  - stretch out a protein and there is a limit to length



Can one formalise this ?

# More general / triangle inequality

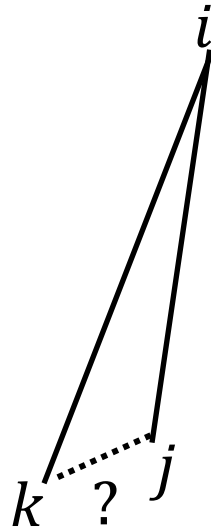
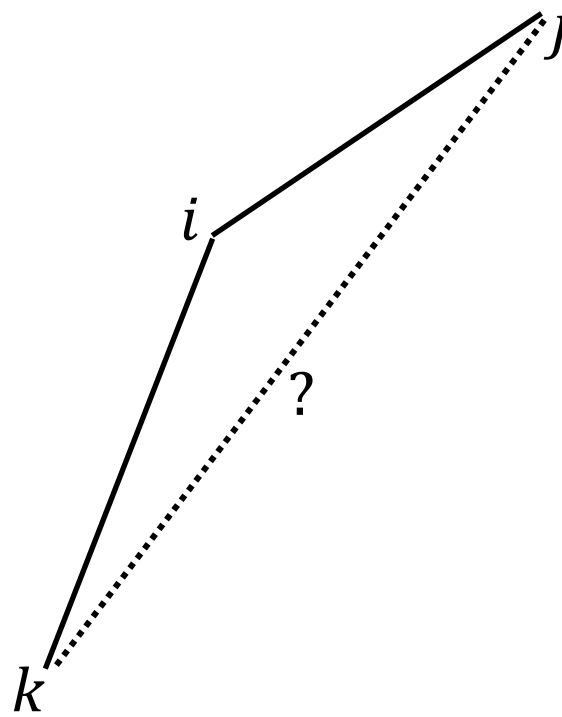
What limits can be worked out ?

upper bound

$$d_{jk} \leq d_{ij} + d_{ik}$$

lower bound

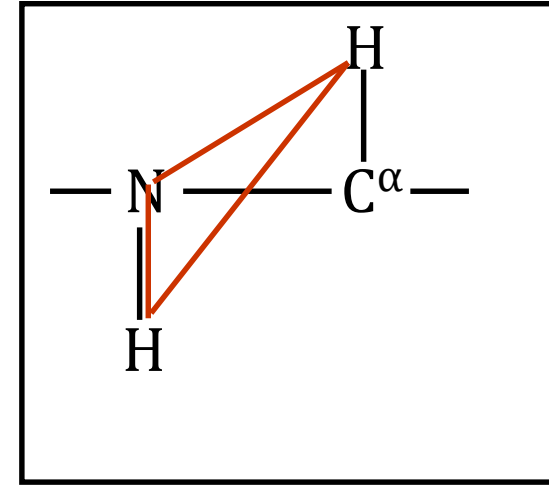
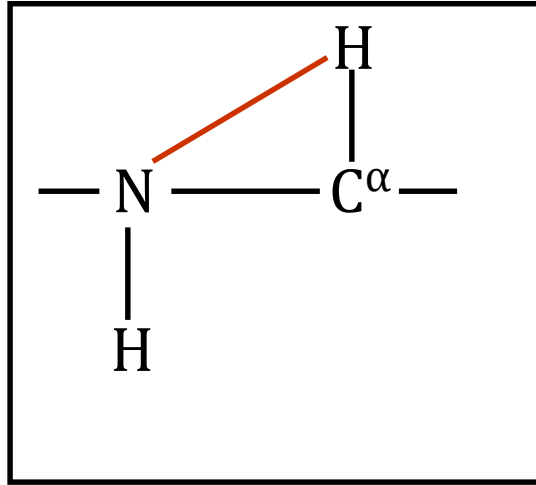
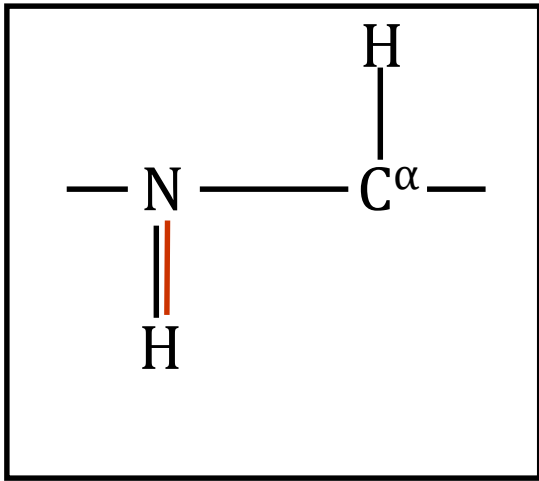
$$d_{jk} \geq |d_{ij} - d_{ik}|$$





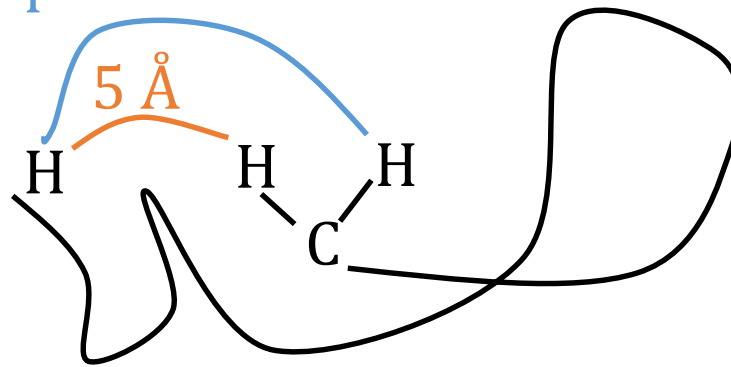
# Where to use triangle inequality

One could avoid some ugly trigonometry



more general

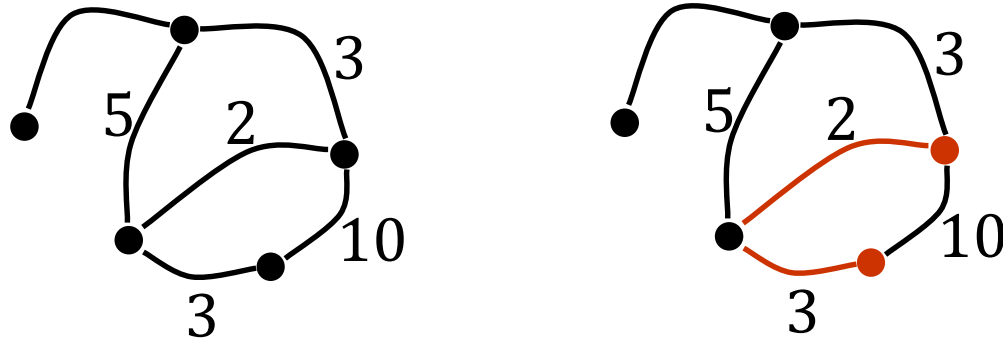
implied 6 or 7 Å



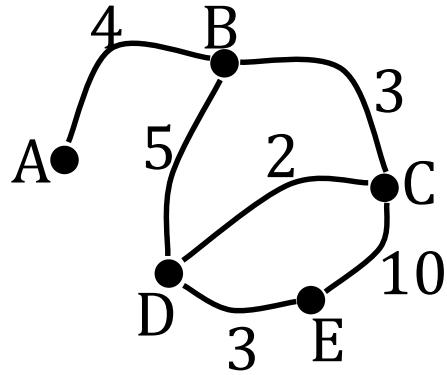
# Most general triangle bound inequality

Triangle bound should be satisfied by any three points

- chemists
  - triangle bound smoothing
- informatik
  - all points shortest path problem



# All points shortest path (Floyd)

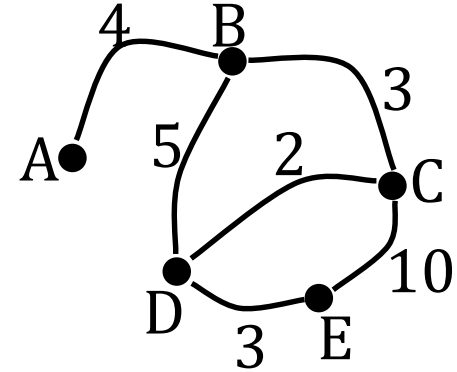


	A	B	C	D	E
A		4			
B			3	5	
C				2	10
D					3
E					

	A	B	C	D	E
A		4	max	max	max
B			3	5	max
C				2	10
D					3
E					

# Bound smoothing / Floyd

	A	B	C	D	E
A		4	max	max	max
B			3	5	max
C				2	10
D					3
E					



```

for (k = 0; k < n_last; k++)
    for (i = 0; i < n_last; i++)
        for (j = 0; j < n_last; j++)
            if ij > ik + jk
                ij := ik + jk
  
```

Running time  
 $O(n^3)$

	A	B	C	D	E
A		4	7	9	12
B			3	5	8
C				2	5
D					3
E					

# Distance matrix so far

We can build a distance matrix of upper limits

- consistent with all bonds and angles and other information

Can do the same for lower bounds

- every pair of atoms
  - invent some lower bound (atomic radii)

**Does this define a structure ?**

Almost certainly not

- still no way to get to a 3D model

# From distances to coordinates

How would you build coordinates from distances ?

- stepwise ?
  - error prone, errors add
- history
  - early 80's
  - methods which are tolerant of errors
    - metric matrix method

# Metric matrix method

- get best upper bounds
- get best lower bounds
  - guess distances between
    - trial distance matrix
  - convert to centre of mass matrix (metric matrix)
  - magic conversion to coordinates
    - if metric matrix has three positive eigenvalues
      - error free coordinates

real coordinates

- lots of errors
- initial coordinates not healthy
- refine

# Metric matrix method

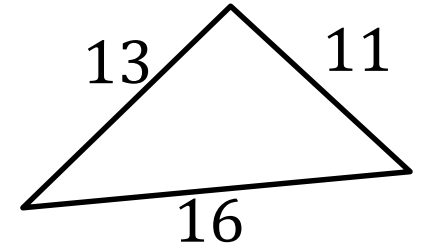
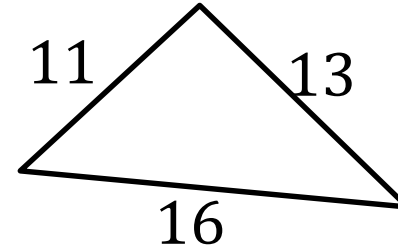
- get best lower bounds + upper bounds
  - guess distances between
    - trial distance matrix
- repeat  $n$  times
  - get  $n$  guesses
- some OK, some bad
- repeat until you have 20 or 100 OK structures
- OK = agrees with experimental data + chemically OK



# Chirality

## 2D version

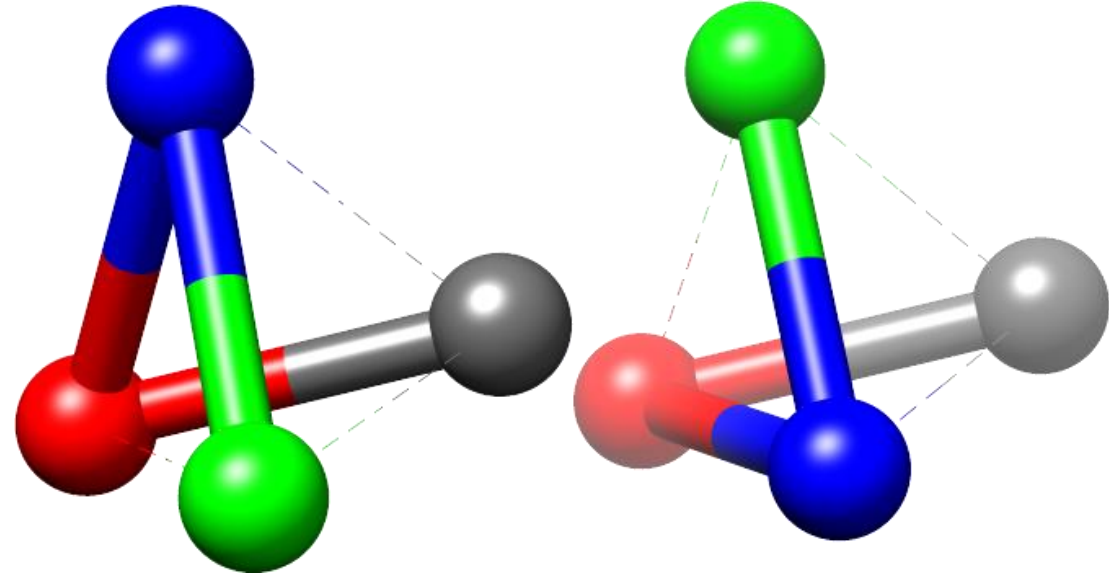
- can \*not\* be rotated on to each other
- can not be distinguished by distances



## 3D

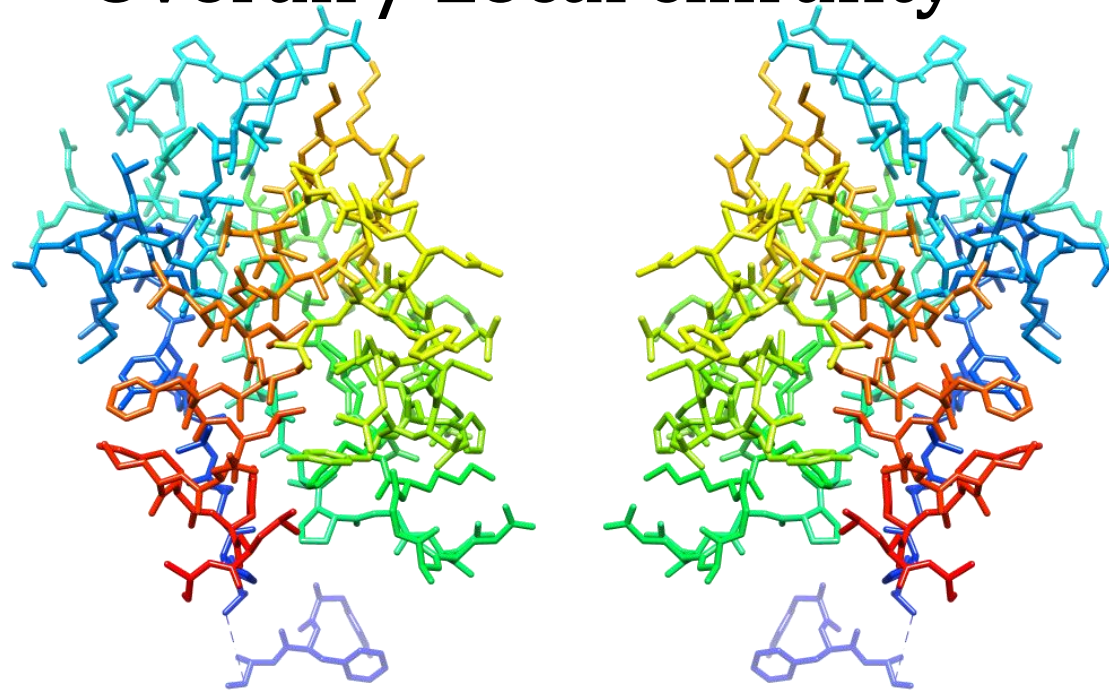
- chirality is random
- problem ? no
  - flip all coordinates and check

Local chirality ...

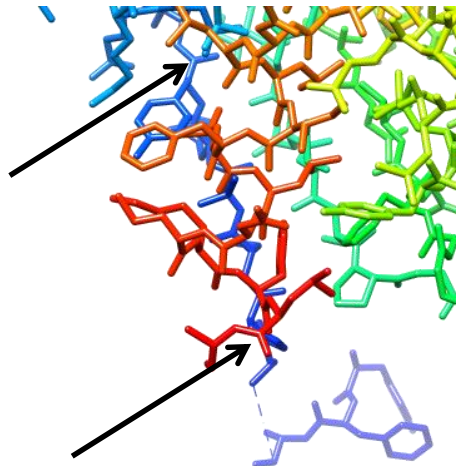


# Overall / Local chirality

overall chirality



local chirality



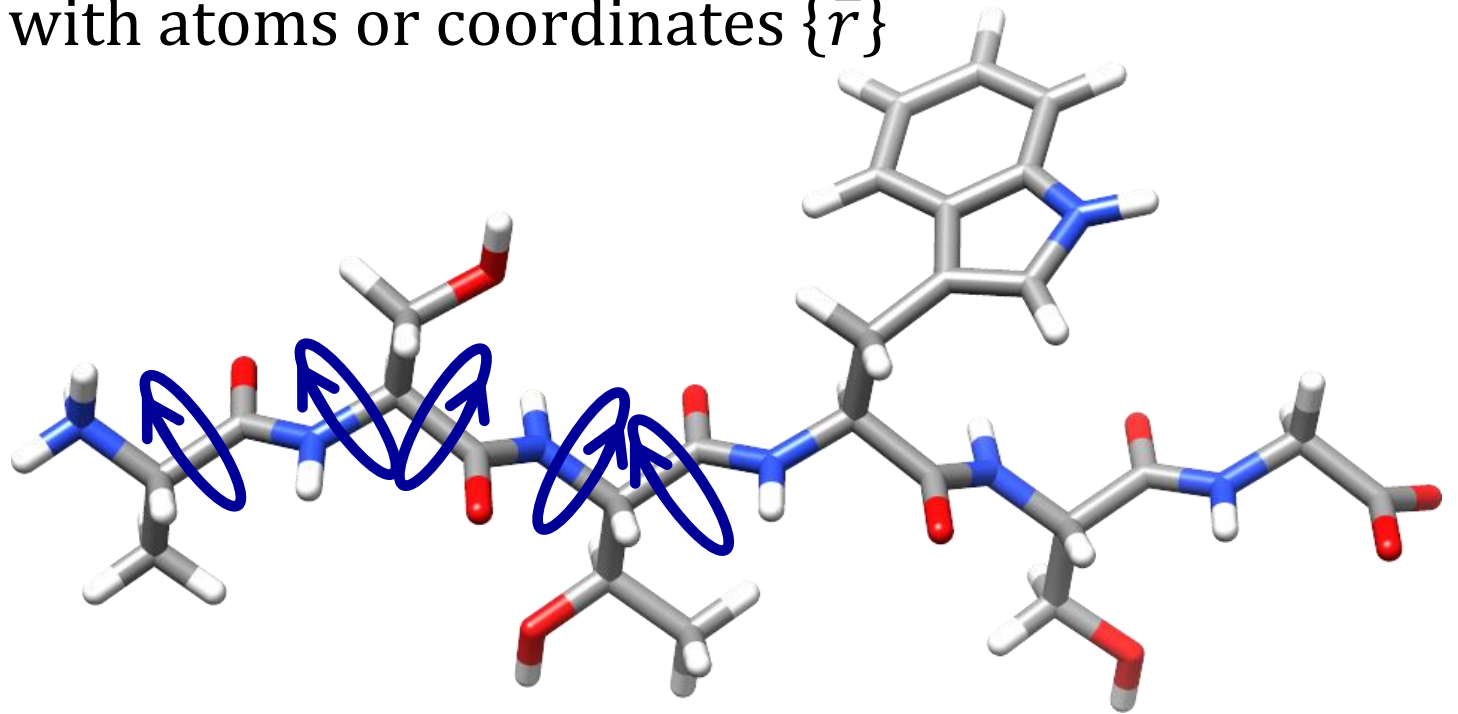
- some points correct
- some wrong
- If you invert a site, will damage other parts of structure

# The Optimisation problem

Find the coordinates that put atoms so they agree with experimental data

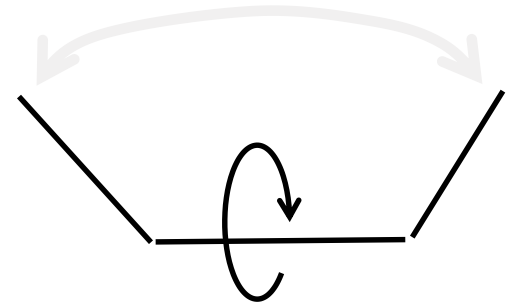
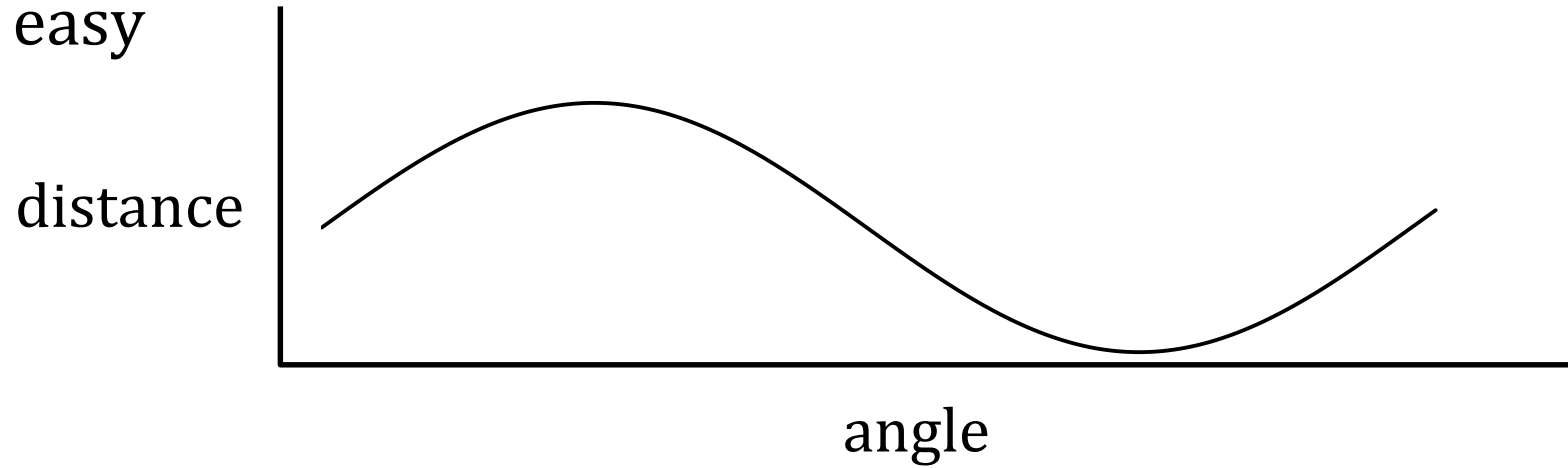
- cost  $c$  is  $\sum_i (r_i - r_i^{measured})^2$  for each measured distance  $r$

Maybe we do not work directly with atoms or coordinates  $\{\vec{r}\}$   
work with angles

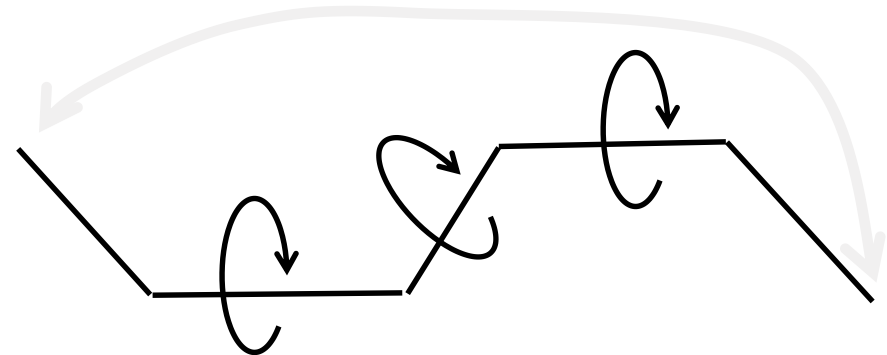


# Distances and angles

One angle is easy



longer distances depend on several angles

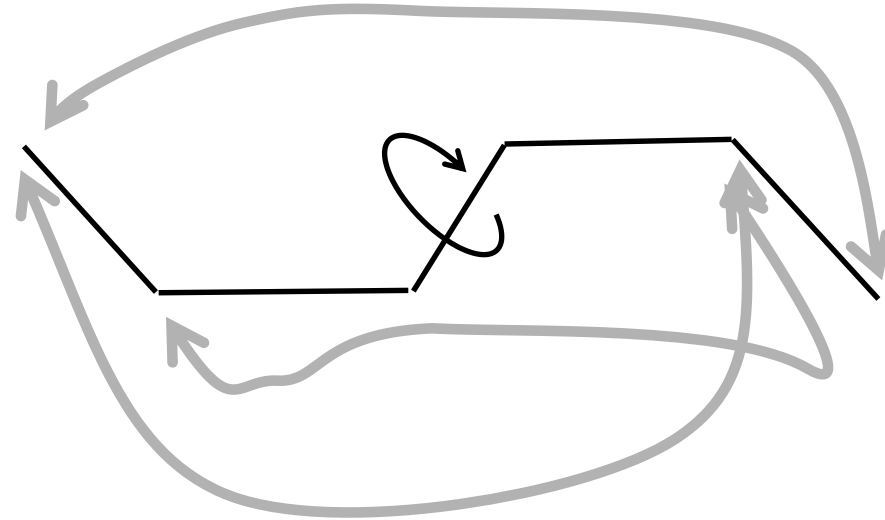


# Distances and angles

Each angle affects many distances

What does one know ?

- simple optimisation will not work



# Optimisation Strategy

Start

- concentrate on distances with few angles in between
- shorter distances become correct

Add in more distances

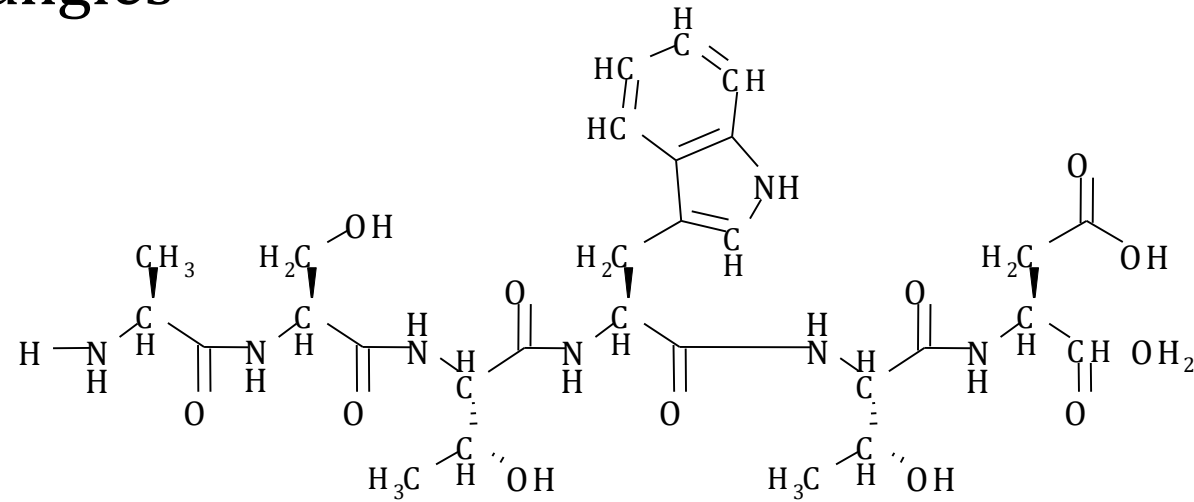
- re-optimize

Add in more distances

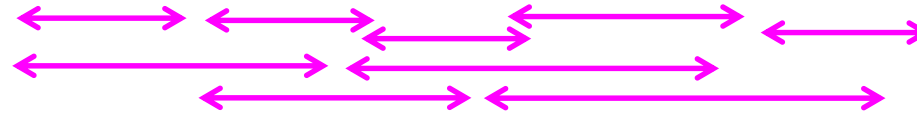
- ...

# Variable target function

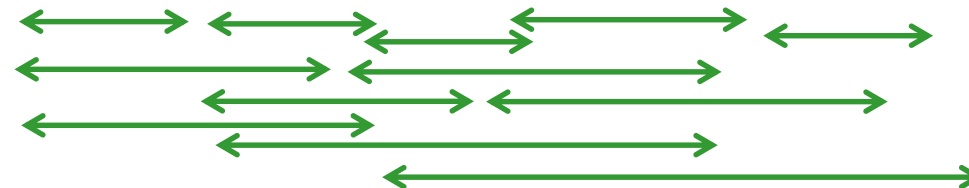
Work with torsion angles



1<sup>st</sup> step



2<sup>nd</sup> step



3<sup>rd</sup> step

ideas from Braun and Gö, 1980s

# Stepwise variable target function method

Collect experimental data

distance in sequence	residue 1	atom 1	residue 2	atom 2	distance in space (Å)
1	5	H $^{\alpha}$	6	H $^N$	4.0
0	8	H $^{\alpha}$	8	H $^{\gamma}$	4.4
80	2	H $^{\alpha}$	82	H $^N$	4.5
2	3	H $^{\alpha}$	5	H $^{\gamma}$	5.0
1	7	H $^{\beta}$	8	H $^{\gamma}$	3.8
0	3	H $^{\alpha}$	3	H $^N$	5.0

Sort according to distance in sequence



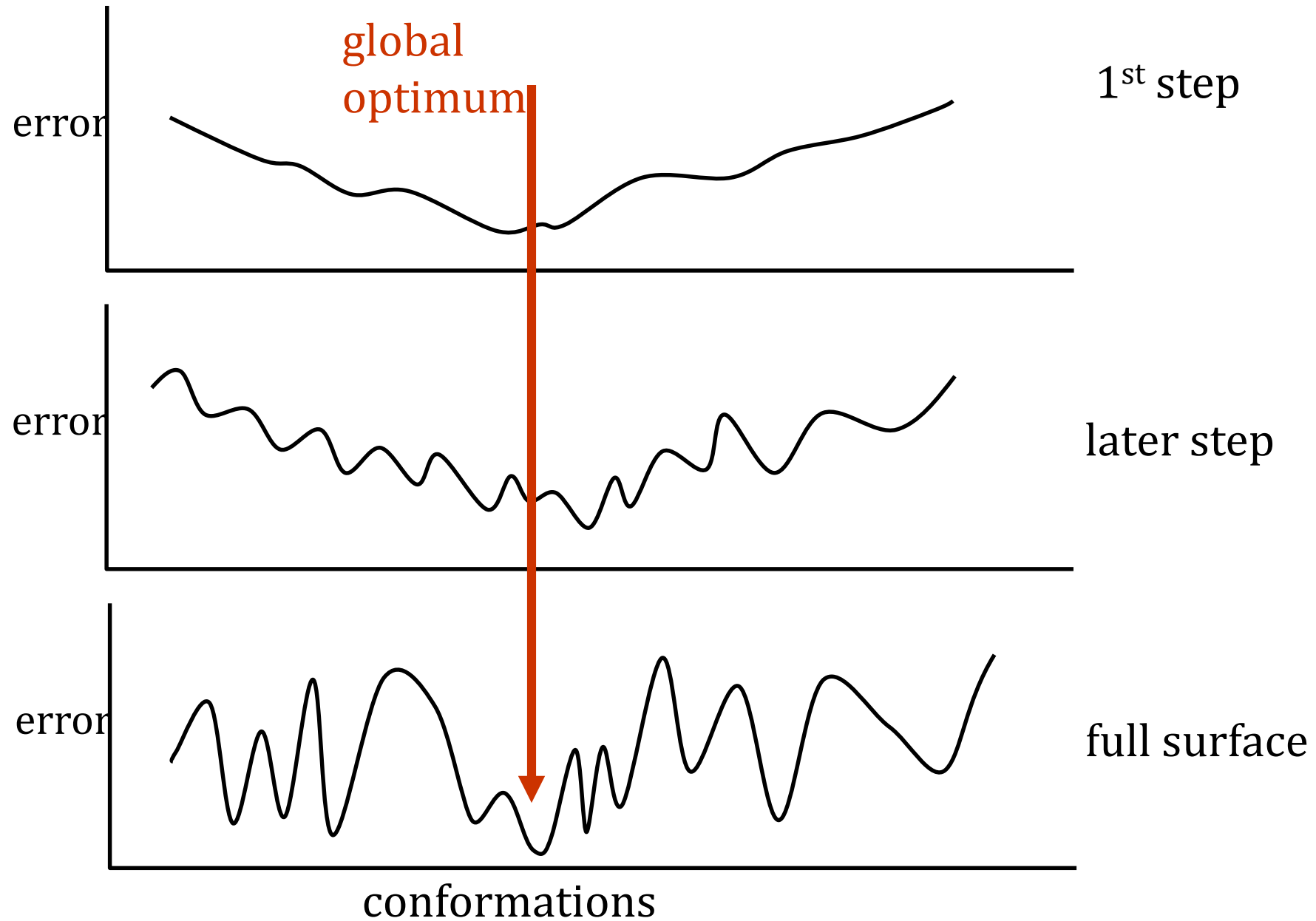
# Stepwise variable target function method

distance in sequence	residue 1	atom 1	residue 2	atom 2	distance in space (Å)
0	8	H $^{\alpha}$	8	H $^{\gamma}$	4.4
0	3	H $^{\alpha}$	3	H $^{\text{N}}$	5.0
1	5	H $^{\alpha}$	6	H $^{\text{N}}$	4.0
1	7	H $^{\beta}$	8	H $^{\gamma}$	3.8
2	3	H $^{\alpha}$	5	H $^{\gamma}$	5.0
...					
80	2	H $^{\alpha}$	82	H $^{\text{N}}$	4.5
...	...				

# Stepwise variable target function method

distance in sequence	residue 1	atom 1	residue 2	atom 2	distance in space (Å)	1 <sup>st</sup>	2 <sup>nd</sup>	3 <sup>rd</sup>	...	later
0	8	H <sup>α</sup>	8	H <sup>γ</sup>	4.4	↓	↓	↓		↓
0	3	H <sup>α</sup>	3	H <sup>N</sup>	5.0					
1	5	H <sup>α</sup>	6	H <sup>N</sup>	4.0					
1	7	H <sup>β</sup>	8	H <sup>γ</sup>	3.8		↓			
2	3	H <sup>α</sup>	5	H <sup>γ</sup>	5.0					
...										
80	2	H <sup>α</sup>	82	H <sup>N</sup>	4.5					
...	...									

# Hope..



# Variable target function vs metric matrix

Metric matrix *versus* variable target function

- proponents of both

variable target function probably more popular

- no problems with chirality

# Real implementations of distance geometry

- not small programs
- Input ?
  - list of protein sequence
  - set of distances
- most of code
  - libraries of standard amino acids
  - code to do geometry and work with standard geometries
- other information
  - angle restraints
    - convert to distances for metric matrix
    - natural for variable target function

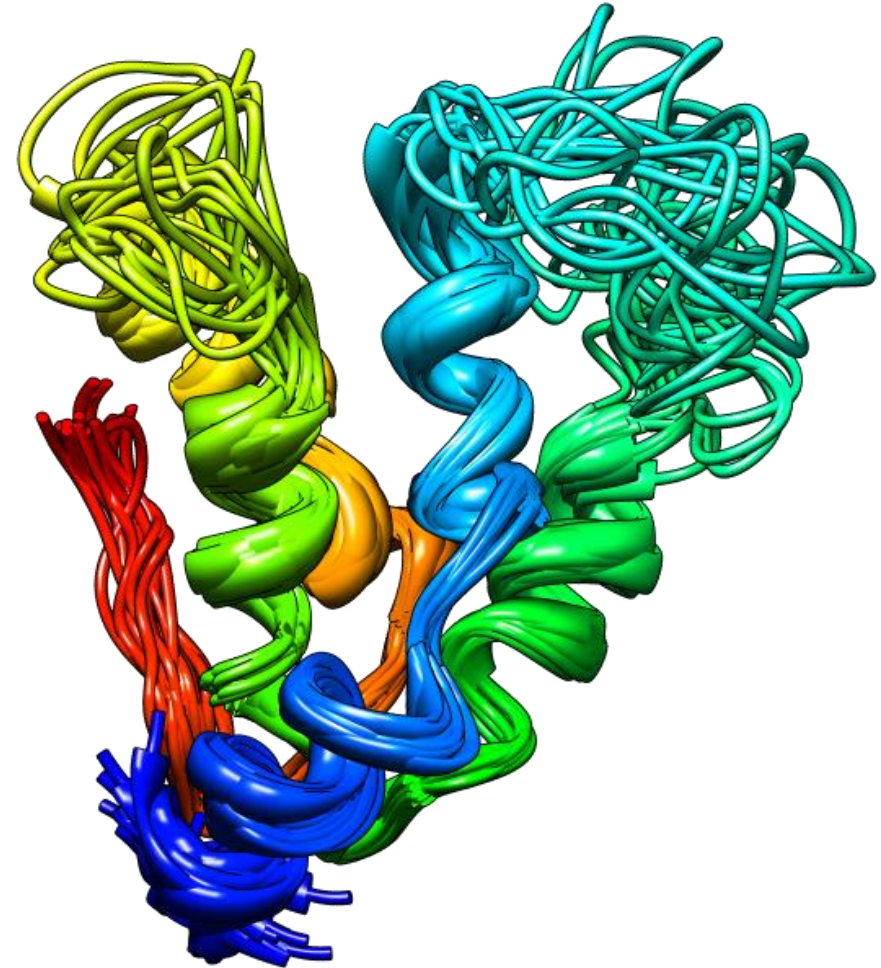
# Output from programs

Structure impossible ?

- program dies or
- best possible solution

Structure not determined ?

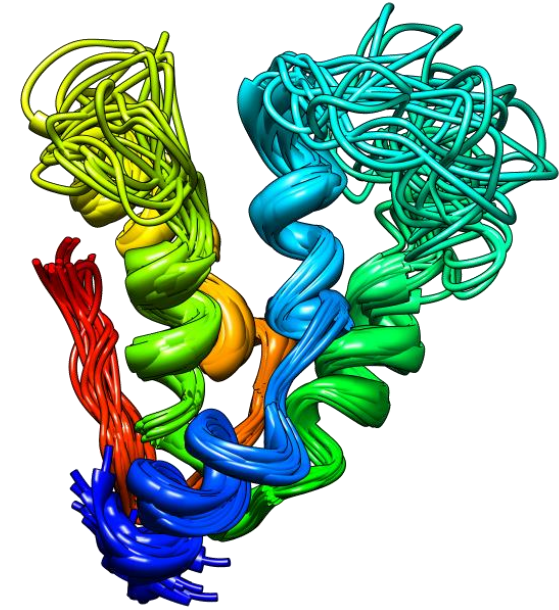
- set of possible conformations (10 to 100 )



example 1sm7

# Lots of models in a PDB file

- big difference compared to X-ray coordinates
- typical
  - ends (C- and N-termini) badly defined
  - loops poorly defined
- spectroscopists say this reflects mobility
- problems with many models
  - difficult to work with
  - arbitrary which to select for calculations
  - averaging usually not a good idea
- Is this the absolute truth ? No.
  - number of models arbitrary
  - different methods (programs /details) give different results



# Finished with making coordinates ?

- structures may not be well defined
  - can they be improved ? probably
    - restrained molecular dynamics (more next semester)
  - normal MD  $E_{phys}(\vec{r}) = \text{bonds} + \text{angles} + \text{electrostatics} \dots$
  - restrained MD  $E_{total}(\vec{r}) = E_{phys}(\vec{r}) + E_{restr}(\vec{r})$
  - and...  $E_{restr} = \sum_i k_i (r_i^{struct} - r_i^{measured})^2$
  - where  $i$  refers to the distance restraint
- Mission - to minimise  $E_{total}$
- result ?
  - structures
    - agree with restraints + low energy



# What else can one do with NMR ?

NMR sensitive to dynamics

- is this part of the protein mobile ?

Interactions

- add small molecule – which parts of spectrum change ?

Still more structural information

- residual dipolar coupling
- spin labels

# Summary

- What information does one have ?
- Is it enough ? Is it consistent ?
- Two methods to generate structures
- Differences in handling chirality