Protein Struktur

- Biologen und Chemiker dürfen mit Handys spielen (leise)
- go home, go to sleep
- wake up at slide 39

Proteins - who cares?

Most important molecules in life? Ask the DNA / RNA people

- structural (keratin / hair)
- enzymes (catalysts)
- messengers (hormones)
- regulation (bind to other proteins, DNA, ..)
- industrial biosensors to washing powder
- receptors
- transporters (O₂, sugars, fats)
- anti-freeze ...

Proteins are easy

- data (protein data bank, www.rcsb.org)
 - 1.5×10^5 files
- literature on function, interactions, structure
- software
 - viewers, molecular dynamics simulators, docking, ...
- nomenclature and rules

Proteins are not friendly

- one cannot take a sequence and predict structure/function
- data formats are full of surprises
- data contains error and mistakes

Protein Rules, Physics, Folklore

Physics / Chemistry

- protein + water = set of interacting atoms
 - can be calculated (not really)

Rules (not quantified)

- proteins unfold if you heat them (exceptions?)
- many charged amino acids.. they are soluble
- if they are more than 300 residues, they have more than one domain,
- proteins fold to a unique structure (could you prove this ?)
 - lowest free energy structure

Protein chemistry

Chemists / biochemists

- sleep, go home
- one tiny surprise at the end of the lectures

Short version

- proteins are sets of building blocks (amino acids, residues, Reste)
- 20 types of residue
- chains of length few to 10³ (100 or 200 typical)
- small ones (< ≈50 residues) are peptides
- they fold up to nice stable structures why?

Longer version..

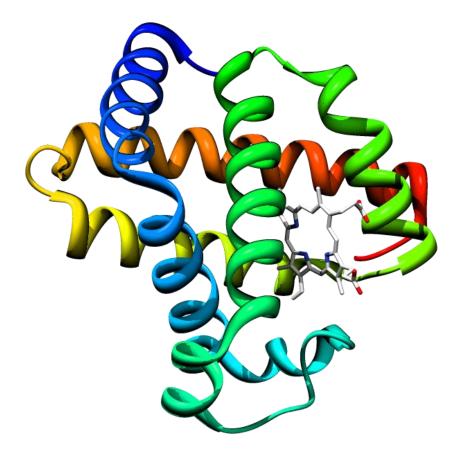
The Plan

- polymers
- different kinds of sidechain
- structure due to backbone (secondary strucure)
- properties of sidechains
- representation

Sizes

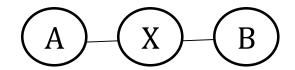
 $1 \text{ Å} = 10^{-10} \text{ m or } 0.1 \text{ nm}$

structure		size
bond	СН	1 Å
	CC	1.5 Å
protein		$10 - 10^2 \text{Å}$
radius		
α-helix		5 ½ Å
spacing		
C_i^{α} to C_{i+1}^{α}		3.8 Å

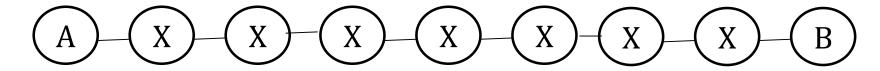


Proteins are polymers

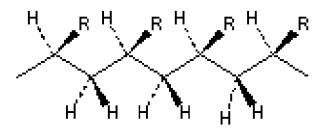
simple polymers



many times gives



example

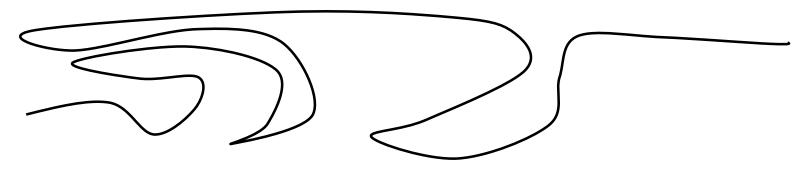


what kind of polymer would this give?

Do you know what R is?

Why are proteins interesting polymers?

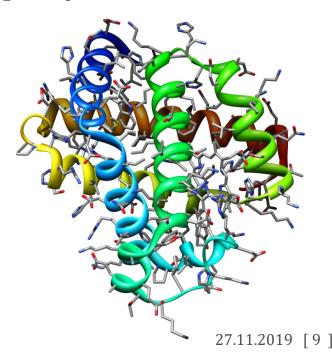
Why do boring polymers not have well-defined structures?



Each part of polymer wants to interact with all other parts equally

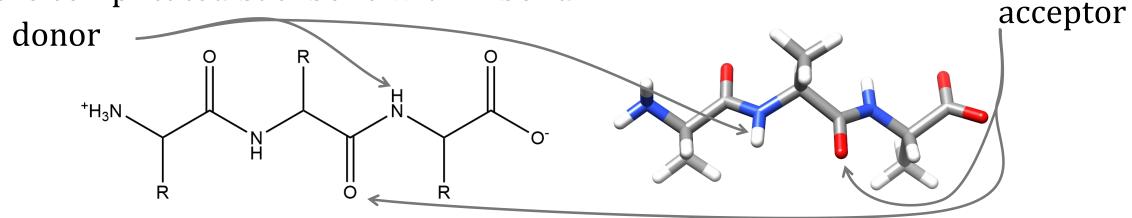
- no structural preferences
- plastic bags, Haushaltsfolie
- no regular structures

Properties that make proteins different from plastics ..



Giving proteins character 1

More complicated backbone with H-bond



- basis of standard regular structures in proteins (secondary structure)
- repeating polymer unit:

If this was all there was

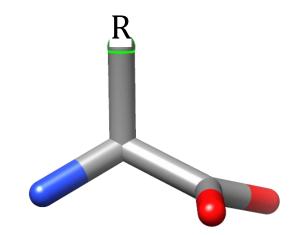
all proteins would be the same

$$^{+}H_{3}N$$
 $\stackrel{\bullet}{\underset{R}{\bigvee}}$
 $\stackrel{\bullet}{\underset{N}{\bigvee}}$
 $\stackrel{\bullet}{\underset{N}{\bigvee}}$
 $\stackrel{\bullet}{\underset{N}{\bigvee}}$
 $\stackrel{\bullet}{\underset{N}{\bigvee}}$
 $\stackrel{\bullet}{\underset{N}{\bigvee}}$

protein chemistry

amino acids (monomers) all look like:

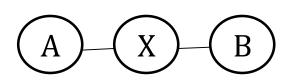
$$NH_{3}^{+} - C_{C}^{+} - C_{C}^{+}$$
 $NH_{2} - C_{R}^{+} - C_{R}^{-}$



How can we construct specific structures?

different kinds of "R" groups

Putting monomers together



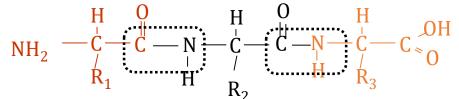
- protein synthesis story (biochemistry lectures)
- peptides and proteins
 - < 30 or 40 residues = peptide
 - > 30 or 40 residues = protein

Backbone peptide bonds

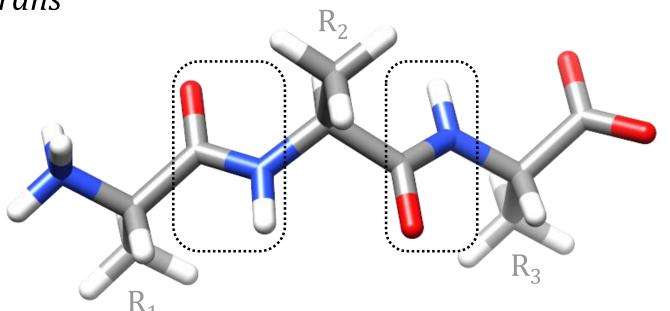
How many backbone angles?

• 3 (ϕ, ψ, ω)

Peptide bond ω is planar



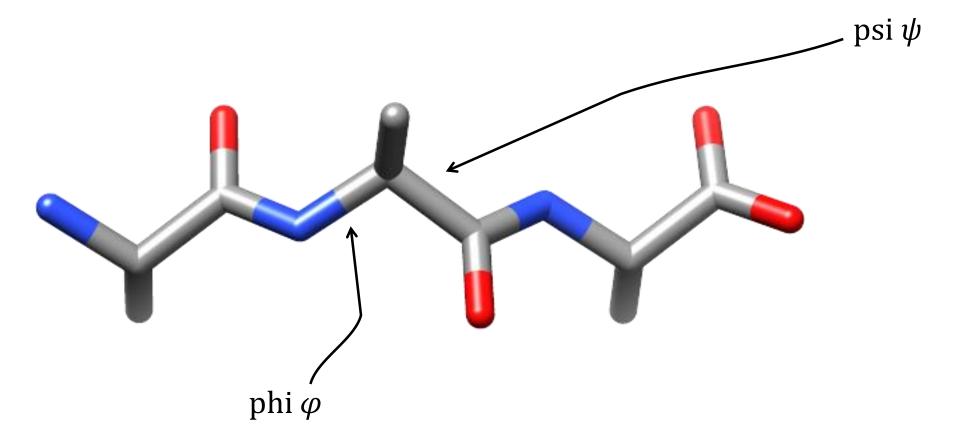
- partial double bond character (resonance forms)
- shorter than other C-N
- nearly always trans



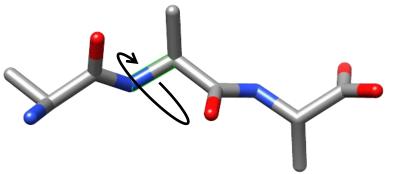
Note: usually we do not draw H atoms

Backbone rotatable angles

Two rotatable angles ϕ , ψ

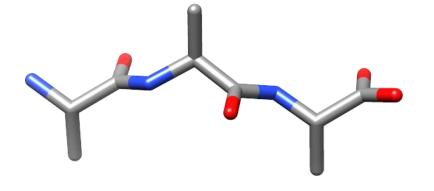






can we rotate freely?

• no... steric hindrance



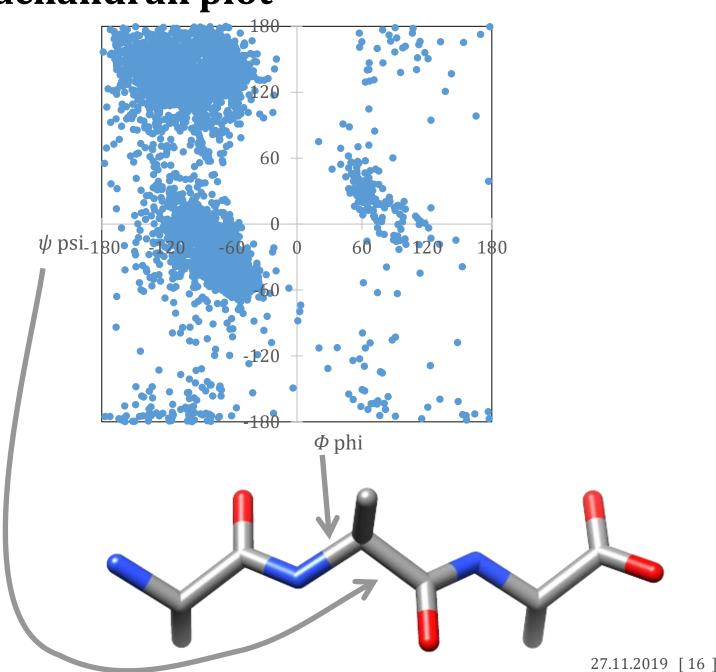
unhappy O atoms – high energy

ramachandran plot

Can we rotate freely?

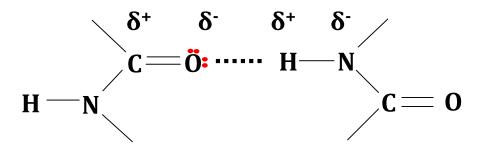
• no... steric hindrance

Ramachandran plot will reappear very often



Backbone H bonds

- oxygen is slightly negative
- NH bond is polar



H-bonds

- can be near or far in sequence
- fairly stable at room temperature

Secondary structure

Regular structures using information so far

- rotate phi (ϕ) , psi (ψ) angles so as to
 - form H-bonds where possible
 - do not force side chains to hit each other (steric clash)

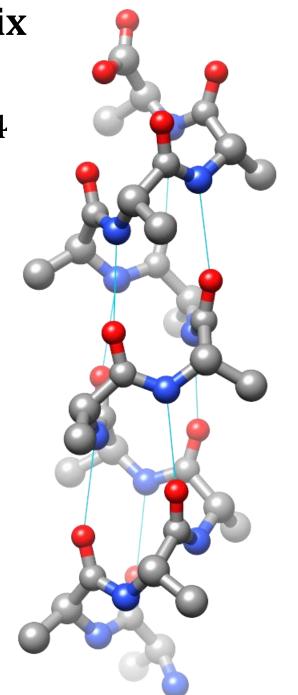
Two common structures

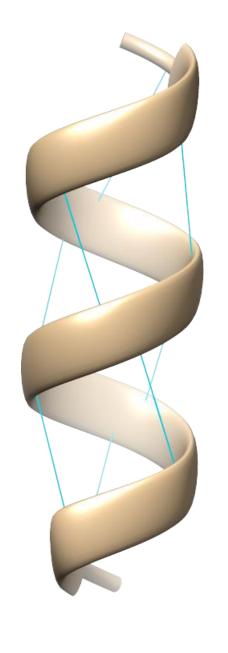
- α -helix
- β -strand / sheet

 α helix

• each CO of residue *i* H-bonded to N of *i*+4

- 3.6 residues per turn
- 2 H-bonds per residue
- side chains well separated





β-sheet

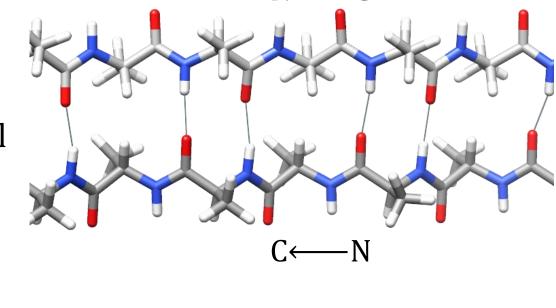
β -strand

stretch out backbone and make NH and CO groups point out

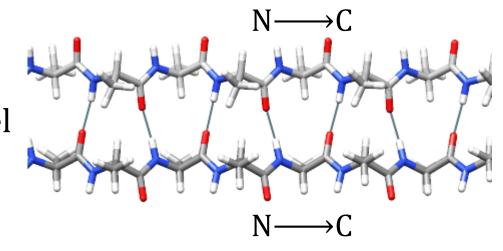
 β -sheet

• join these strands together with H-bonds (2 H-bonds/residue)

anti-parallel



or parallel



After α -helix and β -sheet

Do helices and sheets explain everything? No

• there is flexibility in the angles (look at plot)

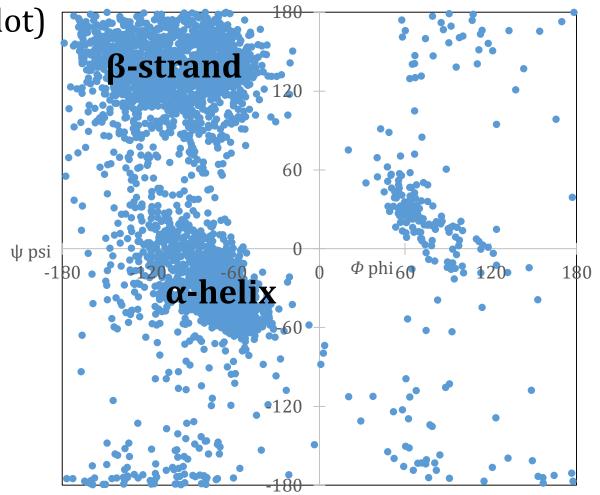
- geometry is not perfectly defined
- there are local deviations and exceptions

Other common structures

- tighter helices
- some turns

Other structure

• coil, random, not named



What determines secondary structure?

So far

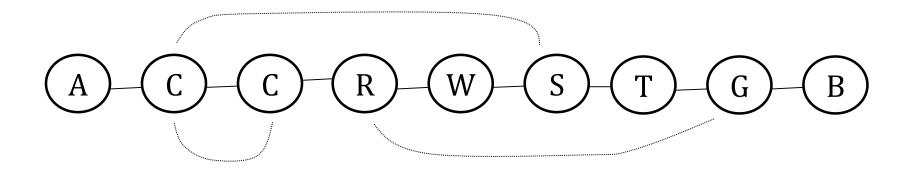
- secondary structure pattern of H-bonding
- Almost all residues have H-bond acceptor and donor
 - almost all could form α -helix or β -sheet

Difference?

- sequence of side-chains overall folding
- Why else are sidechains important
- chemistry of proteins (interactions, catalysis)
- Fundamental dogma
- the sequence of sidechains determines the protein shape

side chain possibilities

- big / small
- charged +, charged -, polar
- hydrophobic (not water soluble), polar
- interactions between sites...



• a CS interaction is different to CC is different to RG ...

Side chain properties

properties

- big / small
- neutral / polar / charged
- special (...)

example

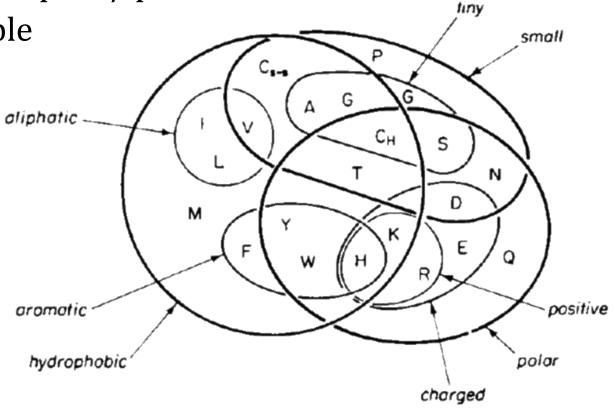
- phenylalanine side chain looks like benzene (benzin)
 - very insoluble
 - benzene would rather interact with benzene than water
 - what if you have phe-phe-phe... poly-phe?
 - does not happen in nature (can be made)
 - would be insoluble
 - not like a real peptide
 - phe is a constituent of real proteins has a role



Properties are not clear cut

You can be big / small, hydrophic / polar

• combinations are possible



Do not memorise this figure

Sidechain interactions

- ionic (if the sidechains have charge)
- hydrophobic (insoluble sidechains)
- H-bonds (some donors and acceptors)
- repulsive

Summary of amino acids (first dozen)

summary of amino acids (part 2)

Amino Acids by property

aromatic

rather hydrophobic

leucine	O N	isoleucine	O N
cysteine	s N	methionine	S
alanine	O	proline	O N
glycine	O N	valine	ON

Polar

threonine

$$\bigvee_{N}^{O}$$

serine

$$O \longrightarrow N$$

glutamine

$$N \longrightarrow 0$$

asparagine

$$0 \bigvee_{N = N} 0$$

charged

arginine

lysine

aspartate

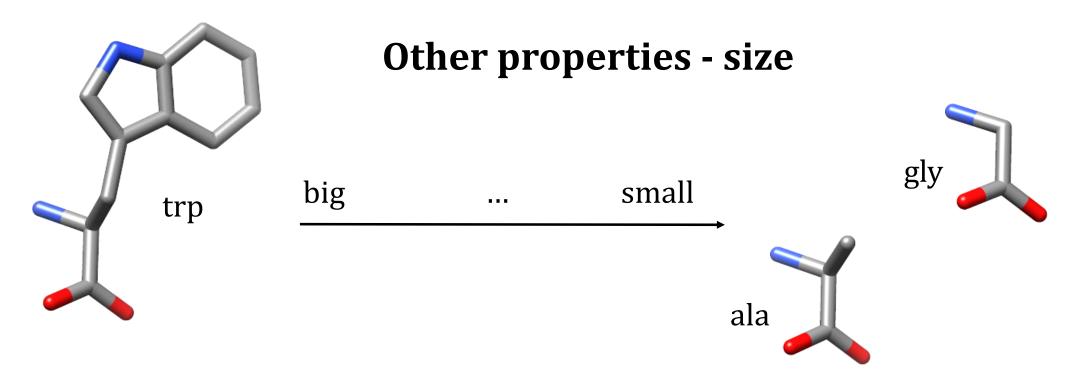
glutamate

Muss ich alle Strukturen für die Klausur wissen?

Hydrophobicity - how serious?

Very serious, but simplified

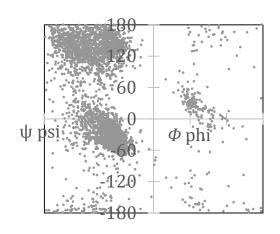
- the lists above are
 - pH dependent
 - difficult to measure experimentally (some aspects)
- Is there a single definition for hydrophobicity?



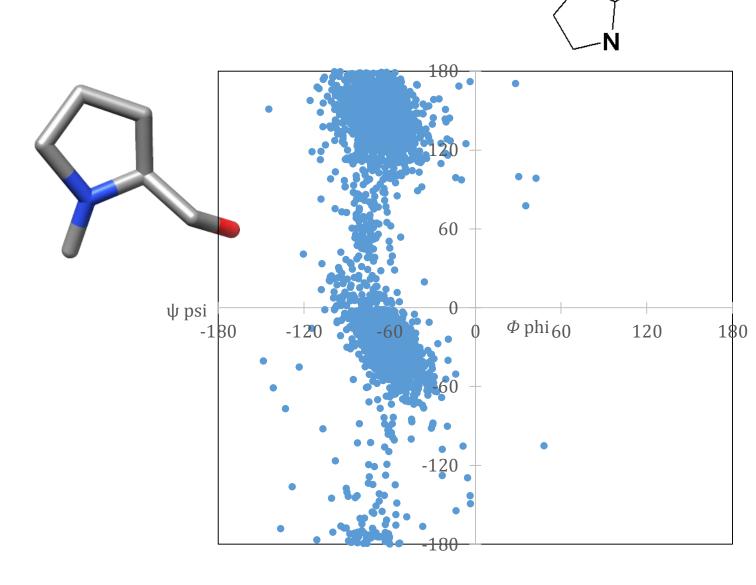
Other properties – chemistry / geometry

Proline

- only one rotatable angle!
- peptide bond sometimes cis
- pro ramachandran plot



all amino acids

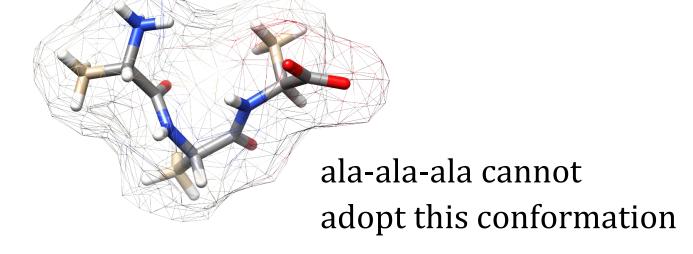


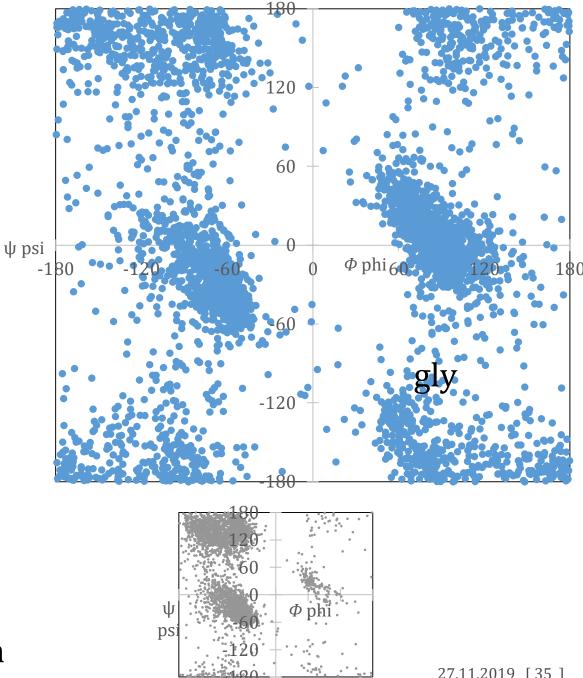
gly-gly-gly glycine no side chain

can visit forbidden parts of phi-psi map

gly

evolutionary consequences?

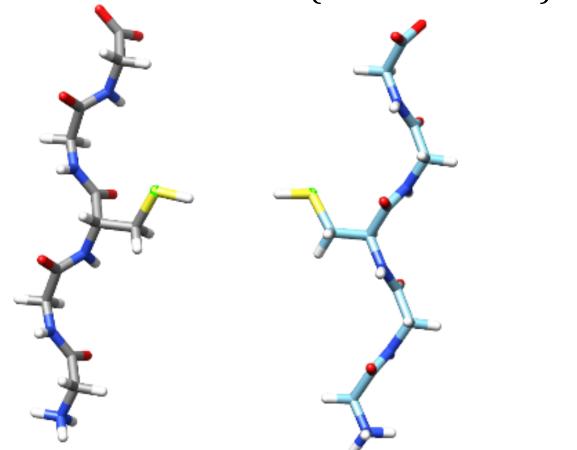


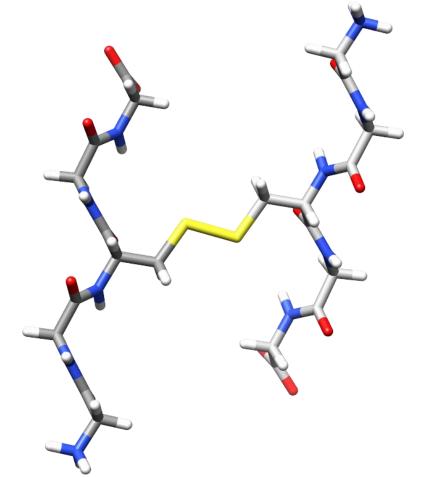


cys

cys residues can bind to each other

effect on conservation (mutation rates)?





Summary so far

- proteins are heteropolymers
- backbone forms α -helices and β -strands (and more)
 - not sequence specific
- side-chains determine the
 - pattern of secondary structure
 - overall protein shape
- special amino acids
 - cys (forms disulfide bridges)
 - gly (can visit "forbidden" regions of ramachandran plot)
 - pro (no H-bond donor)
- how many sequences can one have ? 20^n

Nomenclature

Some rules are unavoidable	Alanine	Ala	A
	Cysteine	Cys	C
	Aspartic acid	Asp	D
	Clutamia acid	C1.	Г

Glutamic acid Glu Ľ Phenylalanine Phe G Glycine Gly Histidine Η His Isoleucine Ile K Lysine Lys Leucine Leu Methionine Met M Asparagine Asn N Proline Pro Q R Glutamine Gln Arginine Arg Serine Ser Threonine Thr Valine Val Tryptophan W Trp Tyr Tyrosine

Always write from N to C terminal (convention)

Definitions, primary, secondary ...

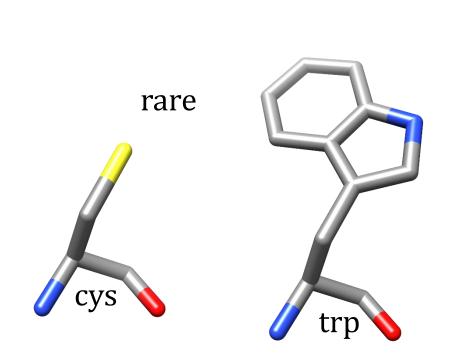
More definitions

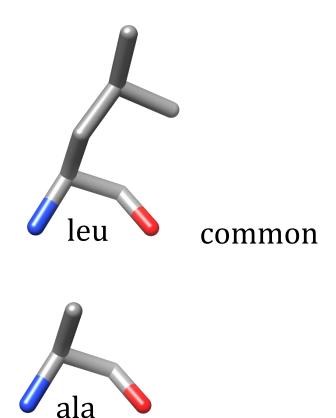
- primary structure
 - sequence of amino acids
 - ACDF (ala cys asp phe...)
- secondary structure
 - α -helix, β -sheet (+ few more)
 - structure defined by local backbone
- tertiary structure
 - how these units fold together
 - coordinates of a protein

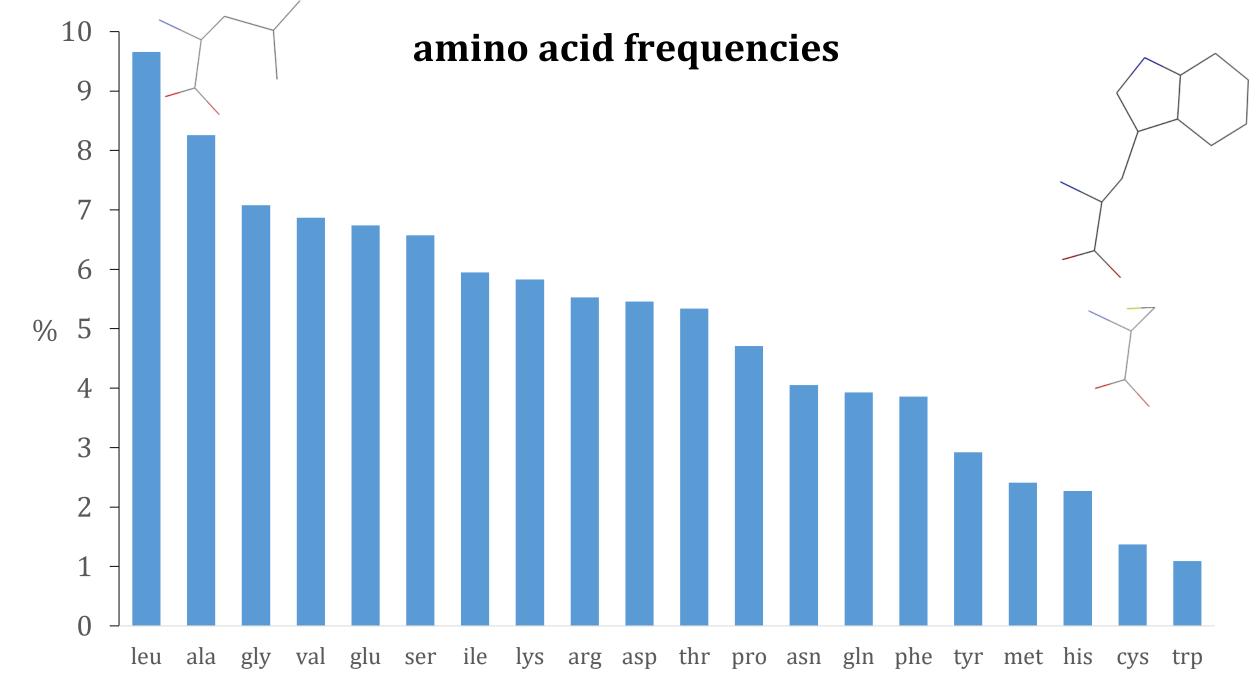
distributions of residue types

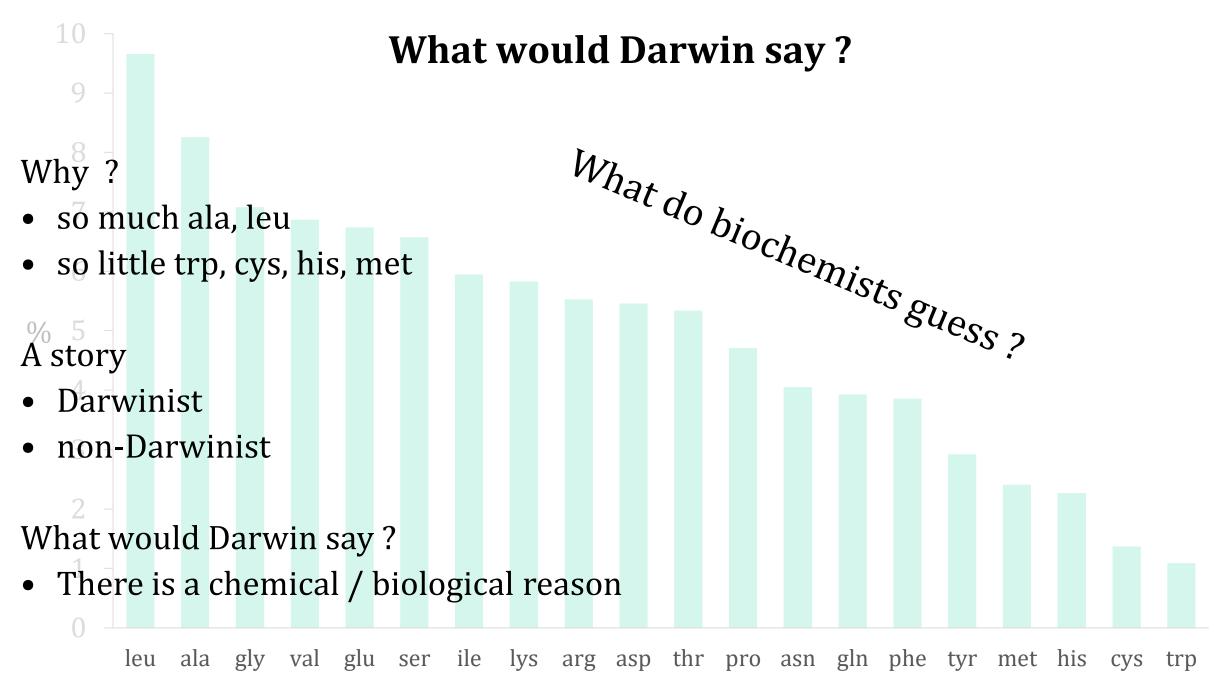
Surprise coming

- 20 amino acid types are they all equally common?
- Are you made of $\frac{1}{20}$ = 5 % of ala, leu, cys, ...?









Think Darwinist

Empirical fact

- trp, cys, met are rare in proteins
- Interpretation / explanation
- too much trp is bad for you / expensive / dangerous

Possibilities

- metabolic cost issues
 - does it cost energy to make trp? cys with its sulfur?
- protein structure lots of chemical differences between amino acid types
 - if you put lots of trp / cys / met in a protein
 - does it not fold? Does it become unstable?
- is free trp toxic?

Common amino acids

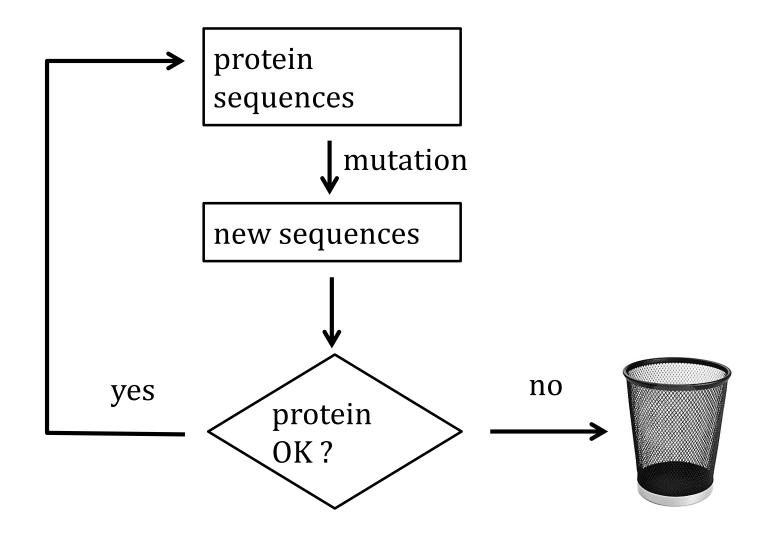
Leu and ala

- cheap to synthesise?
- do you get them as by-products from other biochemistry?
- what is their advantage in protein structure?
 - stability? rigidity? flexibility?

Forget Darwin – think neutral evolution

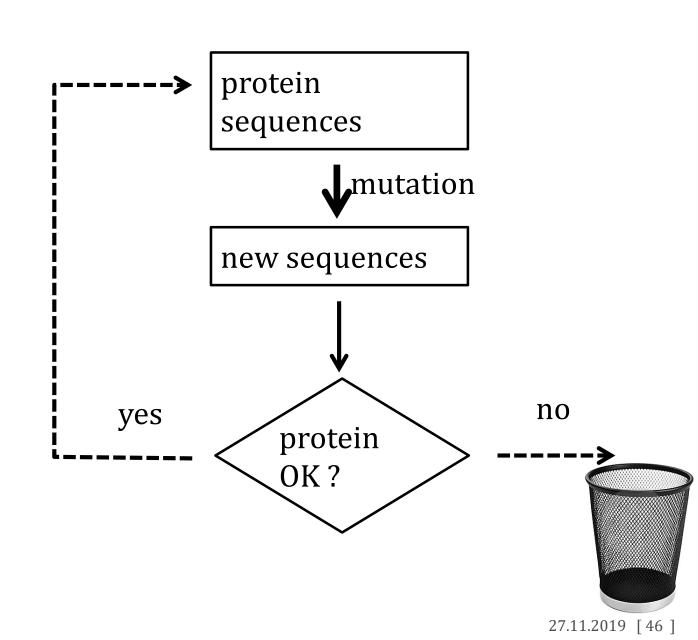
• what do we mean by Darwinism?

Very Darwinist



Think neutralist

- OK/not OK step (selection) less important
- What determines the sequences you see?
 - "mutation" step
- mutation step looks very simple
 - not really
- consider the meaning and biases



Codon bias

• look at the most rare amino acids...

ser UCU, UCA, UCC, UCG, AGU, AGC

leu CUU, CUA, CUC, CUG, UUA, UUG

...

 number of codons not quite everything his CAU, CAC

met AUG

trp UGG

some bases are more common than others

$$p(\text{his}) = p(C) \cdot p(A) \cdot p(C) + p(C) \cdot p(A) \cdot p(C)$$

= 0.22 \cdot 0.30 \cdot 0.22 + 0.22 \cdot 0.30 \cdot 0.22
\approx 0.03

U 22 %

A 30 %

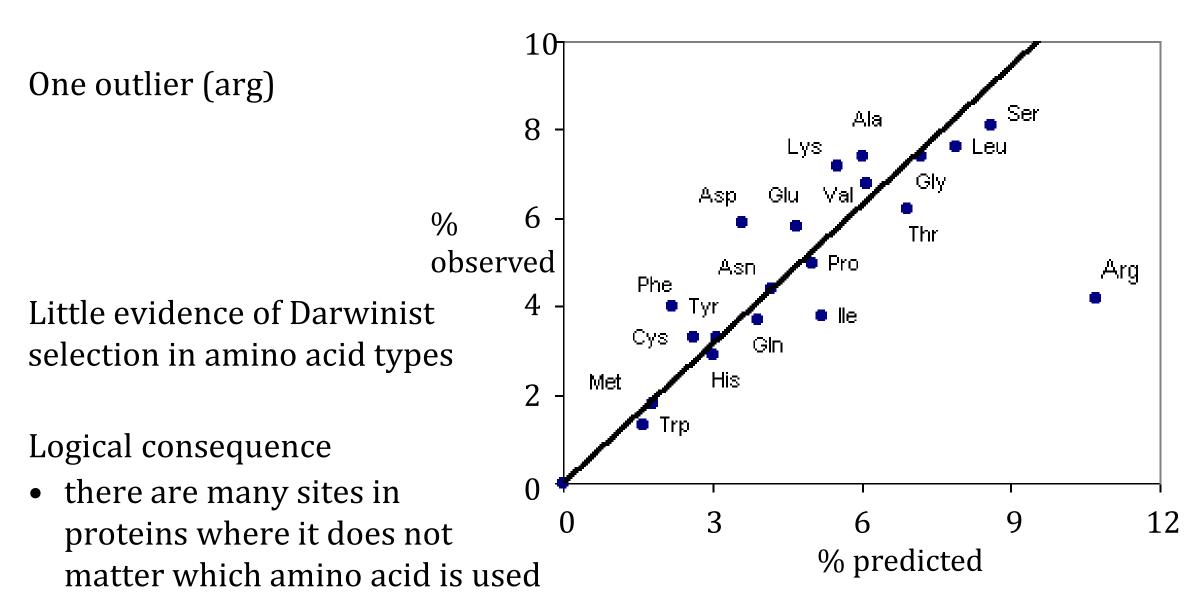
22 %

does this predict the probability of all amino acids?

26 %

• if yes, there is no selection for amino acids

How relevant is Darwinism?



Forget Darwinism and selection of amino acids?

No

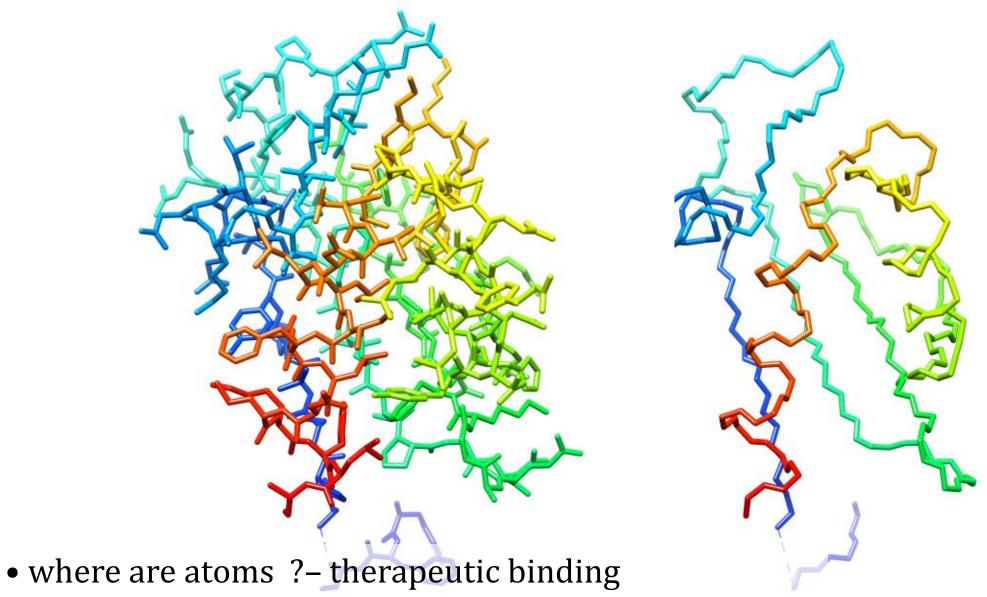
- arg example
- lots of mutation data
 - for an enzyme
 - most mutations are a bit bad, some do not matter
- Do not be a pure Darwinist
- do not interpret everything you see in terms of fitness

Representation

Ultimately, our representation of a structure...

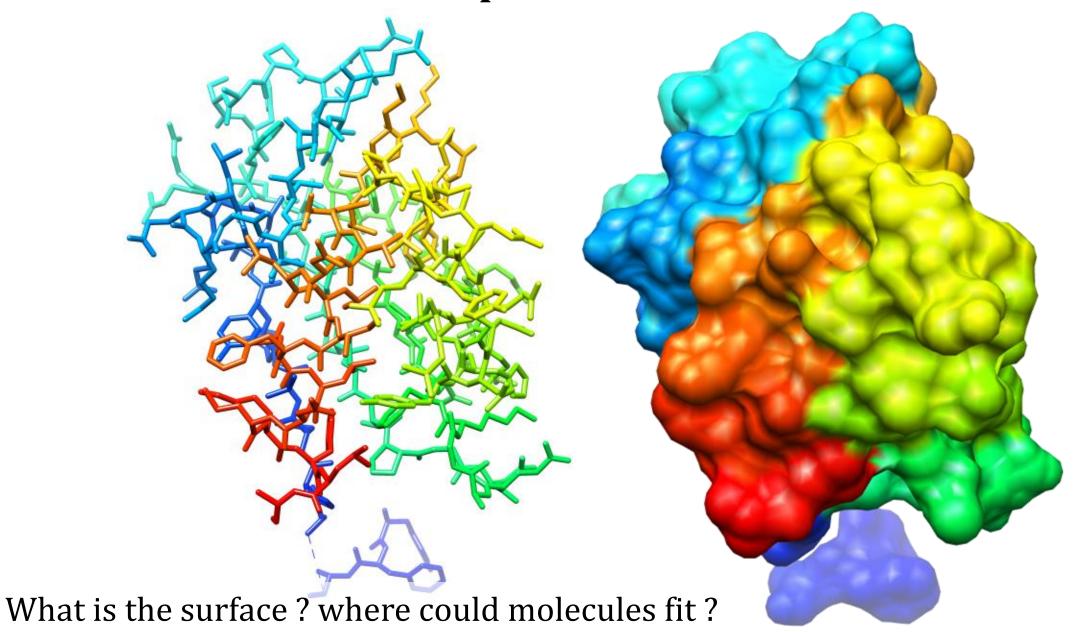
```
13.358 -13.673
                                                         1.00 18.79
ATOM
             Ν
                 ARG
                          1
                                 31.758
                                                                           1BPI 137
                                         13.292 -12.188
                                                         1.00 14.26
ATOM
             CA
                 ARG
                                 31.718
                                                                           1BPI 138
ATOM
             C
                 ARG
                                 33.154
                                         13.224 -11.664
                                                         1.00 18.25
                                                                           1BPI 139
ATOM
             0
                 ARG
                                 33.996
                                         12.441 -12.225
                                                         1.00 20.10
                                                                           1BPI 140
ATOM
             СВ
                 ARG
                                 30.886
                                         12.103 -11.724
                                                         1.00 16.74
                                                                           1BPI 141
                                                         1.00 15.96
                                                                           1BPI 142
ATOM
             CG
                 ARG
                                 29.594
                                         11.968 -12.534
                                         13.182 -12.299
                                                         1.00 15.45
ATOM
             CD
                 ARG
                                 28.700
                                                                           1BPI 143
                                         12.895 -12.546
                                                          1.00 12.82
ATOM
             NE
                 ARG
                                 27.267
                                                                           1BPI 144
                                         13.087 -13.727
                                                         1.00 17.38
                                                                           1BPI 145
ATOM
             CZ
                 ARG
                                 26.661
             NH1 ARG
                                 27.370
                                         13.558 -14.735
                                                         1.00 18.38
ATOM
         10
                                                                           1BPI 146
                                         12.797 -13.838
                                                         1.00 25.73
ATOM
         11
             NH2 ARG
                                 25.367
                                                                           1BPI 147
         12
                                 33.800
                                         13.936 -10.586
                                                         1.00 17.07
                                                                           1BPI 148
ATOM
             Ν
                 PRO
         13
                                 34.976
                                         13.367
                                                  -9.840
                                                          1.00 14.99
ATOM
             CA
                 PRO
                                                                           1BPI 149
ATOM
         14
                 PRO
                                 34.960
                                         11.922
                                                  -9.660
                                                          1.00 13.11
                                                                           1BPI 150
             C
ATOM
         15
                                 33.962
                                         11.306
                                                  -9.391
                                                         1.00 10.57
                                                                           1BPI 151
             0
                 PRO
ATOM
         16
             CB
                 PRO
                                 34.922
                                         14.145
                                                  -8.523
                                                          1.00 15.81
                                                                           1BPI 152
                                 X, 4y, 5Z 333.371
                                                         1.00 18.91
ATOM
         17
             CG
                                         15.391
                                                  -8.737
                                                                           1BPI 153
                 PRO
                                         15.273 -10.096
                                                         1.00 19.41
ATOM
         18
             CD
                                                                           1BPI 154
                 PRO
                                 coordinates
                                                  -9.707
                                                          1.00 8.73
         19
                          3
                                                                           1BPI 155
ATOM
             Ν
                 ASP
```

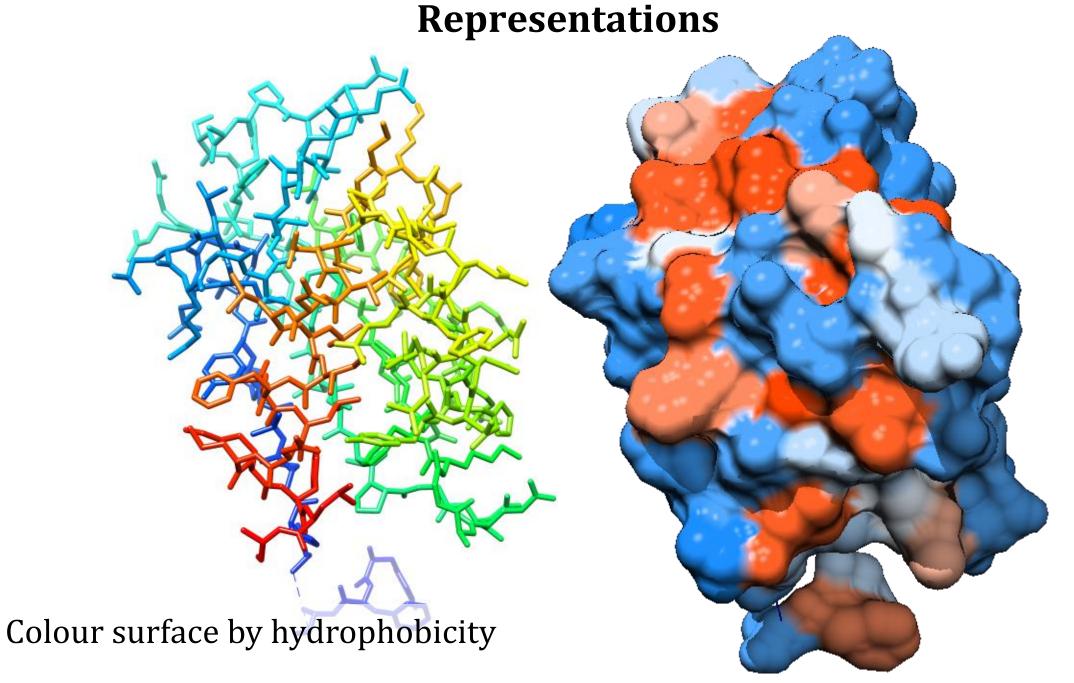
Representations



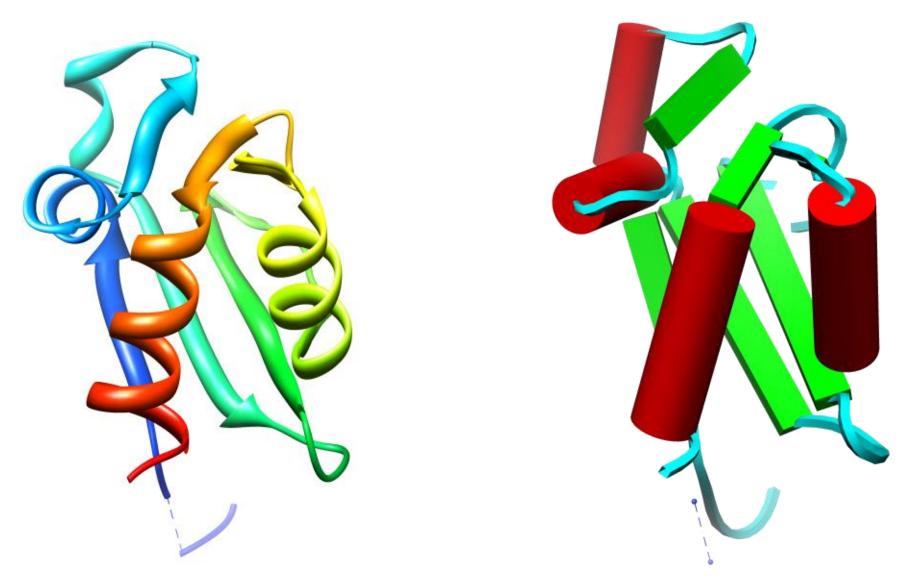
• which residues could be involved in interactions?

Representations





Representations



Highlight / emphasise regular structures

Why does structure matter?

- what residues can I change and preserve function?
- what is the reaction mechanism of an enzyme?
- what small molecules would bind and block the enzyme?
- is this protein the same shape as some other of known function?

Where do structures come from?

- X-ray crystallography
- NMR
- + a bit of small angle X-ray scattering, electron diffraction, neutron diffraction...

resolution, precision, accuracy

Coordinates 27.370 13.558 -14.735

what do they mean?

Random errors

- non-systematic / noise / uncertainty
- should be scattered around correct point

X-ray crystallography has model for data

- uncertainty (probability)
- resolution (experimental)
 - < 1 Å (unusually good)
 - > 5 Å (bad, but examples..)
 3LJ5 Full Length Bacteriophage P22 Portal Protein
 3M0C X-ray Crystal Structure of PCSK9 in Complex with the LDL receptor

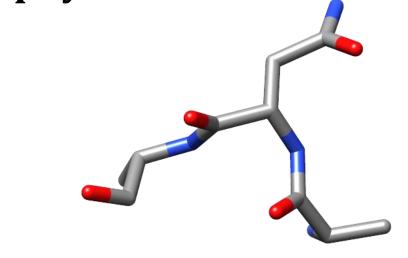
X-ray crystallography

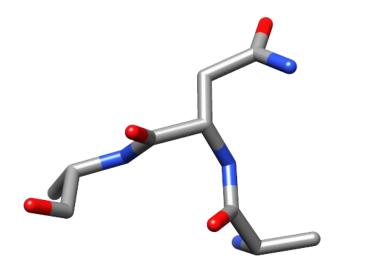
Non-systematic errors

- small problems: (O and N look the same)
- few huge problems
- newer structures are better

Proteins are not static

- overall motion
- local motion





NMR structures

Different philosophy to X-ray

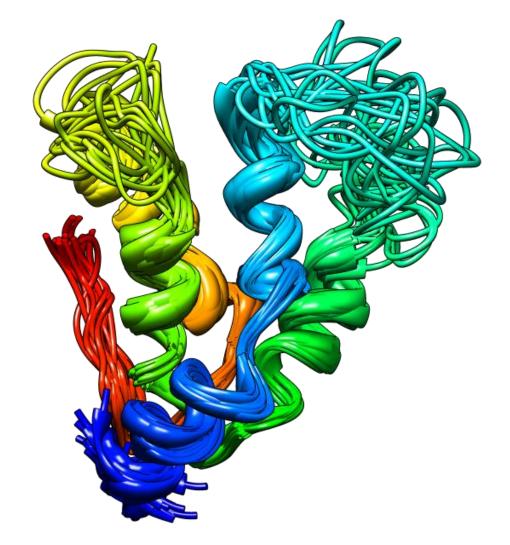
- lots of little internal distances
- do not quite define structure

Generate 50 or 10² solutions

look at scatter of solutions

As with X-ray

- some parts are well defined
- some not



structure 1sm7 27.11.2019 [59]

Summarise and stop

- roles of proteins
- heteropolymers 20 types of amino acid / residue
- geometry avoiding atomic clashes, forming H bonds
 - leads to regular secondary structure
- chemistry of amino acids very different to another
- unique structure for a sequence reflects these differences
- representations of structures
- structures in PDB are experimental have errors

some questions

- $(Asp)_{100}$
 - is it soluble? Is it acidic / basic?
 - would it form a compact regular structure?
- Why does a protein fold to a specific structure, but polyethylene not?
- Glycine can reach parts of the angle space which other residues cannot. What are the evolutionary consequences?
- How big is sequence space? How much has been tried by evolution?
- if you have a protein of poly-trp, would it form a specific structure? How would it behave in solution?
- for length n, do all / many / few of the n^{20} sequences form specific structures ?
- how would a Darwinist explain the uneven distribution of amino acid usage?
- why would you want to represent a protein by its surface?
- why might you draw it as a series of helices and strands?
- what is the biggest chain in the protein data bank? Examples
 - fatty acid synthase $> 2 \times 10^3$ residues/chain
 - dynein heavy chain motor domain > 4×10^3 residues/chain